# Exploring Potential Energy Surfaces in Ground- and Excited States 

Dissertation<br>zur Erlangung des akademischen Grades<br>doctor rerum naturalium (Dr. rer. nat.)<br>vorgelegt dem Rat der Chemisch-Geowissenschaftlichen Fakultät<br>der Friedrich-Schiller-Universität Jena

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Tag der Verteidigung: 22.12.2021
"A computer is an educational device, it's in fact a direct reflection of your own imagination, your own intelligence, your own programming skills and once you're given the freedom, in which to create things and to see the immediate response on the screen, then it becomes a very enjoyable experience and you go on to involve yourself in many other things."

Computer store manager 1

## Peer-Reviewed Publications

## Included in this Thesis

- Haensch, V. G.; Neuwirth, T.; Steinmetzer, J.; Kloss, F.; Beckert, R.; Gräfe, S.; Kupfer, S.; Hertweck, C. Metal-Free Aryl Cross-Coupling Directed by Traceless Linkers. Chemistry - A European Journal 2019, 25, 16068-16073, DOI: 10.1002 / chem. 201903582
- Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: $10.1002 /$ qua. 26390


## Not Included in this Thesis

- G., U. R.; Liu, J.; Hoffmann, P.; Steinmetzer, J.; Görls, H.; Kupfer, S.; Askes, S. H. C.; Neugebauer, U.; Gräfe, S.; Schiller, A. Light-responsive paper strips as CO-releasing material with a colourimetric response. Chemical Science 2017, 8, 6555-6560, DOI: $10.1039 / \mathrm{c} 7 \mathrm{sc} 01692 \mathrm{a}$
- Liu, J.; Hoffmann, P.; Steinmetzer, J.; Askes, S. H.; Kupfer, S.; Görls, H.; Gräfe, S.; Neugebauer, U.; Gandra, U. R.; Schiller, A. Visible light-activated biocompatible photo-CORM for CO-release with colorimetric and fluorometric dual turn-on response. Polyhedron 2019, 172, 175-181, DOI: $10.1016 / \mathrm{j} . \mathrm{poly} .2019$. 04.031
- Amini, K.; Sclafani, M.; Steinle, T.; Le, A.-T.; Sanchez, A.; Müller, C.; Steinmetzer, J.; Yue, L.; Saavedra, J. R. M.; Hemmer, M.; Lewenstein, M.; Moshammer, R.; Pfeifer, T.; Pullen, M. G.; Ullrich, J.; Wolter, B.; Moszynski, R.; de Abajo, F. J. G.; Lin, C. D.; Gräfe, S.; Biegert, J. Imaging the Renner-Teller effect using laser-induced electron diffraction. Proceedings of the National Academy of Sciences 2019, 116, 8173-8177, DOI: 10.1073/pnas. 1817465116
- Liu, X.; Amini, K.; Steinle, T.; Sanchez, A.; Shaikh, M.; Belsa, B.; Steinmetzer, J.; Le, A.-T.; Moshammer, R.; Pfeifer, T.; Ullrich, J.; Moszynski, R.; Lin, C. D.; Gräfe, S.; Biegert, J. Imaging an isolated water molecule using a single electron wave packet. The Journal of Chemical Physics 2019, 151, 024306, DOI: $10.1063 / 1.5100520$
- Shillito, G. E.; Preston, D.; Traber, P.; Steinmetzer, J.; McAdam, C. J.; Crowley, J. D.; Wagner, P.; Kupfer, S.; Gordon, K. C. Excited-State Switching Frustrates the Tuning of Properties in Triphenylamine-Donor-Ligand Rhenium(I) and Platinum(II) Complexes. Inorganic Chemistry 2020, 59, 6736-6746, DOI: 10.1021 / acs. inorgchem.9b03691
- Belsa, B.; Amini, K.; Liu, X.; Sanchez, A.; Steinle, T.; Steinmetzer, J.; Le, A. T.; Moshammer, R.; Pfeifer, T.; Ullrich, J.; Moszynski, R.; Lin, C. D.; Gräfe, S.; Biegert, J. Laser-induced electron diffraction of the ultrafast umbrella motion in ammonia. Structural Dynamics 2021, 8, 014301, DOI: 10.1063/4.0000046

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## List of Acronyms

| ACN acetonitrile | DIIS direct inversion in the iterative sub- |
| :---: | :---: |
| ANEB adaptive NEB | space |
|  | DLPNO domain based local pair-natural |
| AO atomic orbital | orbital |
| ASE Atomic Simulation Environment | DM dimer method |
| BFGS Broyden-Fletcher-Goldfarb-Shanno | DNA desoxyribonucleic acid |
| BDF backward differentiation formula | DLC delocalized internal coordinates |
| BOA Born-Oppenheimer ansatz | DWI distance-weighted interpolant |
| CAM coulomb attenuated method | EulerPC Euler predictor-corrector |
| CASSCF complete active space self consistent field | ES excited state |
|  | FENEB free-end NEB |
| CBS complete basis set |  |
| D(T) singles a | FEANEB free-end adaptive NEB |
| ter with triples correction | [FF force field |
| CDD charge density difference | GDIIS geometric direct inversion in the it- |
|  | erative subspace |
| CG conjugate gradient | GEDIIS energy represented direct inversion |
| CI climbing image | in the iterative subspace |
| CIS Configuration interaction singles | GGA generalized gradient approximation |
| COS chain-of-states | GS ground state |
| CPCM conductor-like polarizable contin- | GS2 Gonzalez-Schlegel second-order algo- |
| uum model | rithm |
| CT charge transfer | GSM growing string method |
| CTS connectivity transition state | GTO Gaussian-type orbital |
| DFT density functional theory | HF Hartree-Fock |
| DFTB density functional tight-binding | HEI highest energy image |

HPC Hessian predictor-corrector
HOMO highest occupied molecular orbital
(IC internal conversion
IDPP image dependent pair potential
ILCT intra-ligand-charge-transfer
IM image method
IRC intrinsic reaction coordinate

## KS Kohn-Sham

LBFGS limited-memory BFGS
LCAO linear combination of atomic orbitals

LDA local density approximation
LEPS London-Eyring-Polanyi-Sato
LHS left-hand side
LMCT ligand-to-metal-charge-transfer
LR linear response
LUMO lowest unoccupied molecular orbital

LST linear synchronous transit

## MB-PES Müller-Brown-PES

MEP minimum energy path
MLCT metal-to-ligand-charge-transfer
MP2 Møller-Plesset perturbation theory
NEB nudged elastic band
MH model Hessian

MO molecular orbital
NA nucleic acid
NMC normal mode coordinates
NTO natural transition orbital
PCA principal component analysis
PEC potential energy curve
PES potential energy surface
PRFO partitioned rational function opti-
mization
PSB Powell's symmetric Broyden
QC quantum chemistry
QN quasi-Newton
RMSD root mean square deviation
RNA ribonucleic acid
RFO rational function optimization
RHS right-hand side
RIC redundant internal coordinates
RI resolution of identity
RMS root mean square
RS restricted step
RS-IRFO restricted step image method RFO

RS-RFO restricted step RFO
RS-PRFO restricted step partitioned RFO
SCC self-consistent charge

| SCF self-consistent field | TD-DFT time-dependent DFT |
| :--- | :---: |
| SD Slater determinant | TDSE time-dependent Schrödinger equa- |
| SM string method | tion |
| SP stationary point | TDEN transition density matrix |
| SQM semi-empirical quantum chemistry |  |
|  | TRIM trust-region image method |
| SR1 symmetric rank 1 | TS transition state |
| STO Slater-type orbital | WFO wavefunction overlap |
| SVD singular value decomposition | XTB exchange-correlation |
| STE extended tight-binding |  |

## Part I

## Introduction

## 1 Introduction

Breakthroughs in chemistry have always shaped our world, to the better or the worse. From its prescientific, alchemical origins, chemistry continuously evolved into a modern science. By providing many key technologies, some of them outlined below, chemistry enables our present standard of living.

For most of the human existence, infections with particular bacteria meant certain death, even for the most powerful people of their times 1 Today, many bacterial infections are routinely treated with a variety of sophisticated antibiotics, e.g., penicillins, discovered by Fleming, Florey and Chain. 13,14
Artificial nitrogen fixation through the Haber-Bosch-Process facilitated an unprecedented growth of the world population, by sustaining large-scale fertilizer production. 15
A main product of the chemical industry are plastics. Ranging from unwanted microplastics in the food we consume, $16-20$ to its packaging and many everyday items, plastics are omnipresent in modern society. 2123

Chemistry is also expected to play a crucial role in mitigating climate change, by providing new ways to generate and store energy, e.g., in the form of molecular hydrogen, generated via artificial photosynthesis. $24-31$

While early chemical discoveries where predominantly made in classical laboratories, where chemists meticulously carried out experiments by hand, modern chemistry is increasingly supported by computational methods. Nowadays, predictive quantum chemical and derived methods, like force fields $32 \sqrt{39}$, can be applied to more and more complex systems. [40-48]
The Haber-Bosch process is assumed to consume $1 \%$ of the world's energy production and to account for $1.4 \%$ of global $\mathrm{CO}_{2}$ emissions. (49] Computational methods are now routinely applied to investigate and evaluate greener alternatives, e.g., by electrocatalysis. 50 (53] Similarly, quantum chemical methods enable rational catalyst design for

[^0]plastic manufacturing, e.g., polyurethane materials, [54] and ethylene polymerization. [55] Computational methods are now also commonly used to study degradation pathways of antibiotics, $56-58$ or to design and evaluate entirely new drugs. $59-61$

In silico approaches are especially attractive to characterize short-lived species, such as transition states (TS) or molecular systems in excited states ( $\overline{\mathrm{ES}})$. Besides the chosen level of theory, two complementary computational approaches can be distinguished: timedependent and time-independent ones. Time-dependent approaches explicitly propagate the system under study in time, e.g., by integrating Newton's law of motion (molecular dynamics) or the time-dependent Schrödinger equation (TDSE) (quantum dynamics). 62 , 63] In the following, this thesis will outline the time-independent approach in more detail.

### 1.1 Exploring Potential Energy Surfaces

When not employing molecular or quantum dynamics, computational studies of chemical transformations are conducted by evaluating and comparing energies and molecular properties at stationary points (SP) of the potential energy surface ( $\overline{\mathrm{PES}})$, where the gradient of the potential $V(\boldsymbol{x})$ vanishes.

$$
\begin{equation*}
\frac{\mathrm{d} V(\boldsymbol{x})}{\mathrm{d} x}=\mathbf{0} \tag{1.1}
\end{equation*}
$$

SPs are distinguished by their index $l$, given by the number of significant negative eigenvalues of the potential's Hessian, 64 the matrix of second partial derivatives. Minima on the PES with $l=0$ represent possible educts, products or stable intermediates of chemical transformations and are connected through TS; with $l=1$. Involvement of higher index SP; $(l>1)$ is also discussed by some authors. 65-68] The eigenvector belonging to the single negative eigenvalue present at a TS with $l=1$ is called the transition vector. Going downhill from a TS along the transition vector in both directions yields a minimum energy path (MEP) and leads to the two SP; that are connected by this TS, A common choice for MEP calculation starting from a TS is the intrinsic reaction coordinate (IRC), the path of steepest descent in mass-weighted coordinates. 69] Or, put differently: the gradient component perpendicular to the MEP vanishes, only the parallel component along the MEP remains.

Parts of this chapter are based on:
Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: 10.1002/qua 26390

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Instead of initial TS optimization by surface-walking 70 and subsequent MEP calculation, both can be obtained simultaneously by chain-of-states (COS) optimization. A COS can be regarded as approximation of the true MEP and comprises several geometries, called images. The most widespread used COS are the nudged elastic band (NEB) and the string method (SM). (71, 72 COSs methods are especially interesting when no suitable TS guess is available, as they are often started from an interpolated path between two minima on the PES, e.g. known educts and products of a chemical transformation. 773

Both approaches, surface-walking and COS optimization, are outlined exemplarily for the two-dimensional Müller-Brown-PES (MB-PES) in Figures 1.1 to 1.2 , 74 The MB-PES is obtained as linear combination of four Gaussian potentials.

$$
\begin{align*}
V(x, y) & =\sum_{i=1}^{4} A_{i} \exp \left(a_{i}\left(x-x_{i, 0}\right)^{2}+b_{i}\left(x-x_{i, 0}\right)\left(y-y_{i, 0}\right)++c_{i}\left(z-z_{i, 0}\right)^{2}\right)  \tag{1.2}\\
\boldsymbol{A} & =(-200,-100,-170,15), \boldsymbol{x}_{\mathbf{0}}=(1,0,-0.5,-1), \boldsymbol{y}_{0}=(0,0.5,1.5,1) \\
\boldsymbol{a} & =(-1,-1,-6.5,0.7), \boldsymbol{b}=(0,0,11,0.6), \boldsymbol{c}=(-10,-10,-6.5,0.7)
\end{align*}
$$

It features three minima, two corresponding TS; of index $l=1$ and was employed numerous times for benchmarking (reaction path) optimization algorithms. 75 - 82 Even though the potential is quite simple, it exhibits several characteristics, also found in real chemical systems, e.g., normal modes of highly varying stiffness and TSk with small basins of attraction. 76
Figure 1.1 outlines the surface-walking approach. Starting from a guess geometry (I), the first SP (II, minimum) is obtained. Modification of II yields TS guess (III), which is subsequently refined to a true TS (IV) of index $l=1$. IRC integration confirms that TS IV connects II to the formerly unknown SP V. The same process can be repeated, to discover the two remaining SP; (not numbered in Figure 1.1).
PES exploration by COS optimization is shown in Figure 1.2. There, knowledge of a second SP (IV) is required. An initial COS (grey line with dots) is constructed by linear interpolation between II and IV and gradually relaxed towards the true MEP, Later COS optimization cycles are given in a darker shade. In the end, the final COS in Figure 1.2 coincides with the IRC in Figure 1.1. A COS in Cartesian coordinates and an IRC carried out in mass-weighted Cartesians only coincide when all atoms have unit mass, which is usually not true for real molecular systems. While SP geometries are independent of the chosen coordinate system, intermediate geometries along an IRC are not. 83

The surface-walking approach is usually computationally less demanding and requires only knowledge of one SP, e.g., II or V in Figure 1.1, but manual construction of a suitable TS guess may be difficult. The COS approach requires knowledge of two SPs, connected by a presumed TS, but manual construction of a TS guess is usually avoided, as the highest energy image (HEI) of a COS can be employed instead. A clear disadvantage of the second approach is its high computational cost, as many energy and gradient evaluations are required for COS optimization.

Reviews on TS searches and COS methods are given by Schlegel, Dewyer and Ásgeirsson. 82, 84, 85


Figure 1.1: Surface-walking approach for exploring the MB-PES, 74 SPs are given as bigger circles (purple $=$ minima, green $=$ TS; ; white outline), non-SP; as smaller circles. Initial (TS)-optimizations, starting from non-SPs, are given in the color of the targeted SP. Values $f(x, y) \geq 25$ are shown with the same red color. The order, in which the respective points are obtained, is indicated by roman numerals.

Starting at I, minimum II at ( $-0.044,0.465$ ) is obtained by optimization. Based on II, a TS guess (III) is constructed, leading to TS IV at ( -0.822 , 0.624). Finally, an IRC is integrated, yielding minimum V (-0.558, 1.442). Unit mass was assumed for IRC integration. By passing the inflection points (yellow circles), the curvature of the the PES changes its sign.


Figure 1.2: PES exploration by COS optimization. Please see the first paragraph of Figure 1.1 for a general comment on the plot and the employed color scheme.

Starting from I and III, minima II and IV are obtained. Subsequently, an initial, interpolated COS is gradually relaxed to the true MEP, Earlier cycles are shown in a lighter shade, COS images are shown as small dots. Evolution of the HEI (small green circle) towards the actual TS is shown by a dashed green line.

### 1.2 Photochemistry

Traversing the PES of the electronic ground state (GS) by means of surface-walking or COS optimization is usually not hampered by state crossings, as the remaining (excited) electronic states are energetically well separated. This changes in photochemical and photophysical studies, were the state of interest is an electronically ES and multiple electronic states are considered simultaneously.

Although ESTPESs are often explored using molecular or quantum dynamics, 8689 localizing SP; in ES; is still important, e.g., for Marcus theory or to rationalize ES relaxation pathways for emission (fluorescence, phosphorescence). [8, 90 92]

Typical, competing processes to be taken into account when studying ESPES; are summarized in Figure 1.3 . There, two $\mathrm{SP} s$, for instance, educts and products of a hypothetical reaction, are separated by a barrier in the GS, Depending on its height,


Figure 1.3: Schematic depiction of different photochemical processes that may take place after initial excitation from the vibrational GS at a SP. Same symmetry (spin and wavefunction) is assumed for all electronic states. Arrows indicate possible wavepackage evolution along the different electronic states.
the barrier may not be crossed in the GS, thus the reaction is prevented. By exciting a molecule into a higher lying electronic state, new pathways can become accessible, allowing crossing of otherwise insurmountable barriers.

Starting from the vibrational and electronic GS at an equilibrium geometry, a molecule is excited by light into a higher lying electronic state. According to their electronic structure, ES may be bound ( $S_{1}$ in Figure 1.3) or repulsive ( $S_{2}$ in Figure 1.3). Following the excitation, the molecule begins to evolve on the ESPES, as it is not at a SP anymore. If another ES comes energetically close, radiationless internal conversion (IC) to this ES may occur. IC takes place at conical intersections, which were found to be important regions in ES relaxation pathways. 93 99] Alternatively, relaxation to the GS can also proceed by photon emission (fluorescence). Compared to the irradiating light, the emitted light is redshifted, as some of the initial energy was already dissipated into the environment or other internal degrees of freedom. If ES; of different multiplicities have to be considered, e.g., because of strong spin-orbit coupling, additional process like intersystem crossing and phosphorescence may take place. $100-103$

A physiologically highly relevant and well studied reaction involving ES5, exhibiting several of the just discussed processes, is the photoisomerization of 11-cis-retinal to 11-trans-retinal.

11-cis-retinal acts as chromophor in the light-sensitive G protein-coupled receptor rhodopsin, which facilitates light perception in dim conditions. Together with photoreceptor proteins photopsins I-III that allow color perception, rhodopsin enables the vision process in vertebrates.

The retinals are conjugated polyenes, covalently bound in their protonated Schiff-base form to a rhodopsin lysine residue. Photoisomerization of 11-cis-retinal to 11-transretinal in rhodopsin occurs very fast in about 200 fs after initial excitation, with a high quantum yield of $0.67 .[104,105$ The protein environment greatly facilitates the isomerization reaction, as the same photoisomerization takes places much slower ( 4 ps ), with decreased quantum yield ( 0.22 ) in MeOH . It was proposed that the protein modulates the accessibility of different retinal GS conformers. 106


Figure 1.4: Schematic potential energy curves (PECs) for the photoisomerization of 11-cis-retinal to 11-trans-retinal in their protonated Schiff-base forms. PEC及 were adapted from fig 2a in 105 , licensed under CC BY 4.0.

Rhodopsin exhibits an absorption maximum at $\lambda_{\max }=498 \mathrm{~nm}$. 107 Following irradiation between 550 nm to 650 nm , a wavepackage is excited from the GS to the $S_{1}$, where it evolves towards a conical intersection (see Figure 1.4). 105,108110 Pump-probe

## 1 Introduction

experiments revealed greatly extended ES lifetimes at low temperatures, indicating the presence of a small barrier in the ES, 105

At the conical intersection, radiationless decay to the $S_{0}$ occurs, ultimately yielding 11-trans-retinal. 111 The retinal isomerization induces further conformational changes in the rhodopsin photopigment, which activate the G protein transducin. Transducin activates a phosphodiestrase, that degrades and thereby lowers the intracellular cyclic guanosine monophosphate concentration, blocking cation influx into the cell. [104 Finally, the photoreceptor cells are hyperpolarized, leading to neuron excitation and the vision process. 112

Reviews on photochemistry and calculation of ES

### 1.3 Goals and Outline of this Thesis

Computational photochemistry often requires effective EStracking, as multiple electronic states may be energetically close to a state of interest, e.g., in optimizations. When moving along the PES, root-flips may occur and state-tracking by energy ordering alone becomes unreliable.

Currently, ES-tracking is not universally available in popular quantum chemistry (QC)packages, or if present, poorly documented and only usable in a black box fashion. ORCA only recently gained EStracking capabilities in version 4.1.0. 121 In Turbomole, ESfracking is available, but undocumented. 122,123 State-tracking is enabled by default in the Gaussian program suite, but only usable in a black-box fashion, for instance, the reference step in an optimization is restricted to the previous cycle.

All QC packages just mentioned are closed-source program and their source code is not freely available, making changes to them impossible. An attractive route to add the required functionality, is to create wrappers for these packages that expose a common interface and then implement EStracking for the common interface. The feasibility of the wrapper approach was already demonstrated impressively by the Atomic Simulation Environment (ASE) project, although ASE is more tailored to solid-state applications and lacks support for internal coordinates, which are required for efficient optimizations. 124

One goal of this thesis is to implement an external, state of the art optimizer that is aware of ES; and supports efficient and effective ES tracking, for further photochemical applications, e.g., optimization of transition metal complex relaxation pathways. Secondly, this thesis provides computational insights for an elegant biaryl coupling reaction discovered by Kloss et al. 2,125

The remainder of this thesis is split in two parts. Part $\Pi$ briefly introduces several approaches to solve the Schrödinger equation and presents the theoretical foundations of ESt-tracking, surface-walking, COS methods and internal coordinates.

Results of this thesis are found in Part III. The biaryl coupling reaction is discussed in chapter 12. Implementation of the external optimizer pysisyphus and its application in the context of ES optimizations and the biaryl coupling reaction is discussed in chapter 13 . Finally, the thesis is summarized in chapter 14.

## Part II

## Theoretical Background and Methods

## 2 Schrödinger Equation

Optimizing stationary points on a PES requires, at least, the ability to calculate the energy of a given atomic configuration (system). To this end, many methods have been proposed, most of them are based on solving the TDSE

$$
\begin{equation*}
i \hbar \frac{\partial}{\partial t}|\Psi(t, \boldsymbol{r}, \boldsymbol{R})\rangle=\hat{\mathcal{H}}|\ominus(\sqcup, \nabla, \mathcal{R})\rangle \tag{2.1}
\end{equation*}
$$

The system under study is characterized by its Hamiltonian $\hat{\mathcal{H}}$ and a corresponding wave function $\Psi(t, \boldsymbol{r}, \boldsymbol{R})$, depending on time $t$, electronic coordinates $\boldsymbol{r}$ and nuclear coordinates $\boldsymbol{R}$. For stationary $\hat{\mathcal{H}}$, a time-independent form of eq. 2.1 is derived as

$$
\begin{equation*}
\hat{\mathcal{H}}|\ominus(\nabla, \mathcal{R})\rangle=\mathcal{E}_{\text {tot }}|\ominus(\nabla, \mathcal{R})\rangle \tag{2.2}
\end{equation*}
$$

with $E_{\text {tot }}$ denoting the systems total energy. The Hamiltonian $\hat{\mathcal{H}}$ comprises all interactions between the constituents of the system. For a system of $N$ electrons (subscript e) and $M$ nuclei (subscript nuc) it is

$$
\begin{equation*}
\hat{\mathcal{H}}=\hat{\mathcal{T}}_{\mathrm{e}}+\hat{\mathcal{T}}_{\mathrm{nuc}}+\underbrace{\hat{V}_{\mathrm{nuc}, \mathrm{nuc}}+\hat{V}_{\mathrm{nuc}, \mathrm{e}}+\hat{V}_{\mathrm{e}, \mathrm{e}}}_{\hat{\mathcal{V}}} \tag{2.3}
\end{equation*}
$$

with the kinetic energy operator $\hat{T}_{\mathrm{e}}\left(\hat{T}_{\text {nuc }}\right)$ of the electrons (nuclei). Coulombic interactions are taken into account by the potential energy operator $\hat{V}$. In atomic units, $\hat{\mathcal{H}}$ is given by

$$
\begin{equation*}
\hat{\mathcal{H}}=-\underbrace{\frac{1}{2} \sum_{n}^{N} \nabla_{n}^{2}}_{\hat{\mathcal{T}}_{\mathrm{e}}}-\underbrace{\frac{1}{2} \sum_{m}^{M} \frac{1}{M_{m}} \nabla_{m}^{2}}_{\hat{\mathcal{T}}_{\mathrm{nuc}}}+\underbrace{\sum_{m>m^{\prime}}^{M} \frac{q_{m} q_{m^{\prime}}}{R_{m, m^{\prime}}}}_{\hat{\mathcal{V}}_{\mathrm{nuc}, \mathrm{nuc}}}-\underbrace{\sum_{m, n}^{M, N} \frac{q_{m}}{r_{m, n}}}_{\hat{\mathcal{V}}_{\mathrm{nuc}, \mathrm{e}}}+\underbrace{\sum_{n>n^{\prime}}^{N} \frac{1}{r_{m, m^{\prime}}}}_{\hat{\mathcal{V}}_{\mathrm{e}, \mathrm{e}}} . \tag{2.4}
\end{equation*}
$$

A commonly employed approximation for $|\Psi(\boldsymbol{r}, \boldsymbol{R})\rangle$ is the Born-Oppenheimer ansatz (BOA), where the electronic wave function $\left|\Psi_{\mathrm{e}}(\boldsymbol{r}, \overline{\boldsymbol{R}})\right\rangle$ only depends parametrically on the nuclear coordinates $\boldsymbol{R}$. 126

$$
\begin{equation*}
|\Psi(\boldsymbol{r}, \boldsymbol{R})\rangle \approx\left|\Psi_{\mathrm{e}}(\boldsymbol{r}, \overline{\boldsymbol{R}})\right\rangle\left|\Psi_{\mathrm{nuc}}(\boldsymbol{R})\right\rangle \tag{2.5}
\end{equation*}
$$

This approximation is justified, as the light electrons can adapt nearly instantaneously to positional changes of much heavier nuclei. The electronic wave function $\Psi_{\mathrm{e}}(\boldsymbol{r}, \overline{\boldsymbol{R}})$ is obtained by solving the electronic Schrödinger equation

$$
\begin{equation*}
\hat{H}_{\mathrm{e}} \Psi_{\mathrm{e}}(\boldsymbol{r}, \overline{\boldsymbol{R}})=E\left|\Psi_{\mathrm{e}}(\boldsymbol{r}, \overline{\boldsymbol{R}})\right\rangle \tag{2.6}
\end{equation*}
$$

with

$$
\begin{equation*}
\hat{H}_{\mathrm{e}}=\hat{T}_{\mathrm{e}}+\hat{V} \tag{2.7}
\end{equation*}
$$

where $E$ denotes the electronic energy of the system. In the following, application of the BOA is assumed throughout and nuclear coordinates will be denoted without the overline, just as $\boldsymbol{R}$.

## 3 Solving the Electronic Schrödinger Equation

Eq. (2.6) can be solved analytically only for small systems, like $\mathrm{H}_{2}{ }^{+}$. Treating larger systems requires additional approximations. Given an approximate wave function $\Psi_{\mathrm{e}}$, a lower bound for its energy $E$ is provided by the Rayleigh-Ritz method (variational principle), 127 as the expectation value of $\hat{\mathcal{H}}$

$$
\begin{equation*}
E=\frac{\left\langle\Psi_{\mathrm{e}}\right| \hat{\mathcal{H}}\left|\ominus_{\mathrm{e}}\right\rangle}{\left\langle\Psi_{\mathrm{e}} \mid \Psi_{\mathrm{e}}\right\rangle} . \tag{3.1}
\end{equation*}
$$

Based on eq. (3.1), wave function $\Psi_{\mathrm{e}}$ can be varied, until a minimum energy $E$ is obtained. As the wave function $\Psi_{\mathrm{e}}$ describes fermionic particles, it must fulfill the Pauli exclusion principle, which requires the wave function to change sign, under the exchange of two fermions. 128 Assuming the expansion of $\Psi_{\mathrm{e}}$ into a set of $N$ orthogonal spin orbitals
$\left\{\phi_{i}\right\}$, the Pauli exclusion principle is satisfied, by employing a Slater determinant (SD) as wave function.

$$
\Psi_{\mathrm{e}}(1,2, \ldots N)=\frac{1}{\sqrt{N!}}\left|\begin{array}{cccc}
\phi_{1}\left(\boldsymbol{x}_{1}\right) & \phi_{2}\left(\boldsymbol{x}_{1}\right) & \cdots & \phi_{N}\left(\boldsymbol{x}_{1}\right)  \tag{3.2}\\
\phi_{1}\left(\boldsymbol{x}_{2}\right) & \phi_{2}\left(\boldsymbol{x}_{2}\right) & \cdots & \phi_{N}\left(\boldsymbol{x}_{2}\right) \\
\vdots & \vdots & \ddots & \vdots \\
\phi_{1}\left(\boldsymbol{x}_{N}\right) & \phi_{2}\left(\boldsymbol{x}_{N}\right) & \cdots & \phi_{N}\left(\boldsymbol{x}_{N}\right)
\end{array}\right|
$$

Here, $\phi_{a}\left(\boldsymbol{x}_{i}\right)$ denotes electron $i$ in spin orbital $a$. Spin orbitals are composed of a spatial orbital $\psi_{a}$ and a spin part, depending on spin coordinate $\omega_{i}$.

$$
\phi_{a}\left(\boldsymbol{x}_{i}\right)=\left|\psi_{a}\left(\boldsymbol{r}_{i}\right)\right\rangle \cdot\left\{\begin{array}{l}
\left|\alpha\left(\omega_{i}\right)\right\rangle  \tag{3.3}\\
\left|\beta\left(\omega_{i}\right)\right\rangle
\end{array}\right.
$$

### 3.1 Hartree-Fock

The computational bottleneck of solving eq. 2.6) is the accurate treatment of the electronelectron interaction ( $\hat{V}_{\mathrm{e}, \mathrm{e}}$, see eq. 2.4). A basic approximation to $\hat{V}_{\mathrm{e}, \mathrm{e}}$ is introduced in the Hartree-Fock (HF) method, by considering merely the interaction of electrons with the average field of the remaining electrons. 129 Minimizing the energy of a SD, while requiring the spin orbitals to stay orthonormal, yields the HF equations.

$$
\begin{equation*}
\hat{f}\left(\boldsymbol{x}_{i}\right) \phi_{a}\left(\boldsymbol{x}_{i}\right)=\varepsilon_{a} \phi_{a}\left(\boldsymbol{x}_{i}\right) \quad i=1,2 \ldots N \tag{3.4}
\end{equation*}
$$

The Fock operator $\hat{f}\left(\boldsymbol{x}_{i}\right)$ comprises a one-electron part $\hat{h}\left(\boldsymbol{x}_{i}\right)$, describing electron-nuclear attraction and electronic kinetic energy and a two-electron part $\hat{v}^{\mathrm{HF}}\left(\boldsymbol{x}_{i}\right)$, describing Coulomb interactions between electrons.

$$
\begin{equation*}
\hat{f}\left(\boldsymbol{x}_{i}\right)=\hat{h}\left(\boldsymbol{x}_{i}\right)+\hat{v}^{\mathrm{HF}}\left(\boldsymbol{x}_{i}\right) \tag{3.5}
\end{equation*}
$$

Whereas orbital-free, fully numerical solution of the HF equations (3.4) is restricted to small systems up to tri-atomics, 130133 larger systems can be treated by expanding spatial orbitals $\psi_{a}$ into atomic orbitals $\varphi_{\mu}$ (linear combination of atomic orbitals (LCAO)).

$$
\begin{equation*}
\psi_{a}=\sum_{\mu=1}^{\eta} d_{\mu a} \varphi_{\mu} \quad a=1,2 \ldots \eta \tag{3.6}
\end{equation*}
$$

Parameter $\eta$ denotes the number of basis functions used in the contraction of spatial orbital $\psi_{a}$ and $d_{\mu}$ denote contraction coefficients. Historically, Slater-type orbitals (STOs)

$$
\begin{equation*}
\varphi_{\mathrm{STO}}(r) \propto e^{-\alpha r} \tag{3.7}
\end{equation*}
$$

have been employed as basis functions, as they are analytical solutions to the hydrogen atom Schrödinger equation. Their use in quantum chemical calculations was superseded by Gaussian-type orbitals (GTOs)

$$
\begin{equation*}
\varphi_{\mathrm{GTO}}(r) \propto e^{-\alpha r^{2}} \tag{3.8}
\end{equation*}
$$

as Boys recognized that molecular integrals can be calculated much more efficiently using GTOs. 134 -138 Compared to GTOs, evaluation of molecular integrals over STO requires more elaborate algorithms, even for simple integrals like basis function overlaps. $139-144$

In contrast to STOs, GTOs show incorrect asymptotic behavior for big $r$ and their derivative vanishes at the nucleus $(r=0)$, hence multiple GTO; are summed to approximate a STO, 145] Many different GTO basis sets have been proposed until now. $146-148$ Basis set choice depends on the chemical nature of the system at hand, as well as on the employed quantum chemical method. 149,150

Expanding spatial orbitals into atomic orbitals allows transforming the differential HF equations (3.4) into a set of algebraic equations

$$
\begin{equation*}
\hat{f}\left(\boldsymbol{r}_{i}\right) \sum_{\mu} C_{\mu a} \varphi_{\mu}\left(\boldsymbol{r}_{i}\right)=\varepsilon_{a} \sum_{\mu} C_{\nu a} \varphi_{\nu}\left(\boldsymbol{r}_{i}\right) \tag{3.9}
\end{equation*}
$$

also known as the Roothaan-Hall equations

$$
\begin{equation*}
F C=S C \varepsilon \tag{3.10}
\end{equation*}
$$

with Fock matrix $\boldsymbol{F}$, molecular orbital (MO) coefficients $\boldsymbol{C}$, overlap matrix $\boldsymbol{S}$ and MO energies $\varepsilon$. 151,152 The Roothaan-Hall equations (3.10) are solved iteratively, until self-consistency of $\boldsymbol{C}$ is achieved.

If a sufficiently sized basis set is employed, HF is able to account for $99 \%$ of the total energy. The difference between the exact, non-relativistic total energy $E_{\text {exact }}$ and the HF energy $E_{\mathrm{HF}}$ is known as the correlation energy

$$
\begin{equation*}
E_{\mathrm{cor}}=E_{\mathrm{exact}}-E_{\mathrm{HF}} \tag{3.11}
\end{equation*}
$$

Two types of electron correlation are distinguished: static correlation, arising from from different, energetically (nearly) degenerate electronic configurations and dynamic correlation, arsing from the correlated motion of electrons, as they repel each other.

### 3.2 Density Functional Theory

Except for the exchange interaction, electron correlation is mostly neglected in HF A computationally efficient way to partially incorporate dynamical correlation at similar computational cost, is density functional theory (DFT). Whereas a $N$ electron wave function depends on $3 N$ spatial and $N$ spin coordinates, the electronic density

$$
\begin{equation*}
n(\boldsymbol{r})=N \int \mathrm{~d} \boldsymbol{r}_{2} \ldots \int \mathrm{~d} \boldsymbol{r}_{N} \Psi^{*}\left(\boldsymbol{r}, \boldsymbol{r}_{2}, \ldots \boldsymbol{r}_{N}\right) \Psi\left(\boldsymbol{r}, \boldsymbol{r}_{2}, \ldots \boldsymbol{r}_{N}\right) \tag{3.12}
\end{equation*}
$$

depends on only 3 spatial coordinates, independent of $N$. Hohenberg and Kohn showed that a GS wave function $\Psi_{0}\left(\boldsymbol{r}_{1}, \boldsymbol{r}_{2}, \ldots \boldsymbol{r}_{N}\right)$ is an exact functional of the GS electronic density $n_{0}(\boldsymbol{r})$ and that the GS density can be determined from the GS energy functional $E\left[n_{0}(\boldsymbol{r})\right]$ by the variational principle. 153,154

Originally, DFT was formulated orbital-free, thus making it computationally very efficient, as the dimensionality of the electronic density is only 3 , but the accurate construction of the electronic kinetic energy functional proved difficult. $155-160$ Nowadays, DFT is mostly used in the framework of Kohn and Sham (KS), where orbitals are reintroduced and used to expand the electronic density. [161, 162]

$$
\begin{equation*}
n(\boldsymbol{r})=\sum_{a}\left|\psi_{a}(\boldsymbol{r})\right|^{2} \tag{3.13}
\end{equation*}
$$

KSHFT simplifies the calculation of the electronic kinetic energy

$$
\begin{equation*}
T_{0}[n(\boldsymbol{r})]=-\frac{1}{2} \sum_{a} \int \mathrm{~d} \boldsymbol{r} \psi_{a}(\boldsymbol{r})^{*} \nabla^{2} \psi_{a}(\boldsymbol{r}) \tag{3.14}
\end{equation*}
$$

but also increases the computational costs, as the original dimensionality of the problem is restored. The energy functional comprises several terms

$$
\begin{equation*}
E[n(\boldsymbol{r})]=T_{0}[n(\boldsymbol{r})]+J[n(\boldsymbol{r})]+E_{\mathrm{XC}}[n(\boldsymbol{r})]+V_{\mathrm{ext}}[n(\boldsymbol{r})] \tag{3.15}
\end{equation*}
$$

with the electron-electron Coulomb potential $J$, the exchange-correlation (XC) functional $E_{\mathrm{XC}}$ and an arbitrary external potential $V_{\text {ext }}$, e.g., the Coulomb potential of the electrons
in the field of nuclear charges. The only unknown term in eq. (3.15) is $E_{\mathrm{XC}}$ and its choice is crucial for the accuracy of DFT calculations. (163)
The simplest XC functional is given by the local density approximation (LDA), derived from an homogeneous electron gas. It only depends on the electronic density $n(\boldsymbol{r}) .164$ A first improvement over LDA is given by the generalized gradient approximation (GGA) that also considers the electronic density gradient $\nabla n(\boldsymbol{r}) \cdot 165$ By incorporating exact HF exchange, (global) hybrid functionals are obtained. A popular hybrid functional is B3LYP, 166 consisting of the exchange-functional B88 of Becke and the correlationfunctional of Lee, Yang and Parr. [163, 167]

$$
\begin{align*}
E_{\mathrm{xc}}^{\mathrm{B} 3 \mathrm{LYP}}= & E_{\mathrm{xc}}^{\mathrm{LDA}}+a_{0}\left(E_{\mathrm{x}}^{\mathrm{HF}}-E_{\mathrm{x}}^{\mathrm{LDA}}\right)+a_{x} \Delta E_{\mathrm{x}}^{\mathrm{B} 88} \\
& +a_{c}\left(E_{\mathrm{c}}^{\mathrm{LYP}}-E_{\mathrm{c}}^{\mathrm{VWN}-\mathrm{LDA}}\right) \tag{3.16}
\end{align*}
$$

Parameters $a_{0}=0.2, a_{x}=0.72$ and $a_{c}=0.81$ were fitted, to reproduce atomization energies and ionization potentials from the G2 data set. [168]
More recent developments include meta-(hybrid)-GGAb, relying also on the second derivative of the electronic density $\nabla^{2} n(\boldsymbol{r})$ and double hybrid functionals, which include a perturbative second-order correlation part. 169 -172

Common XC functionals, like B3LYP, have several shortcomings, especially in the treatment of ESF in time-dependent DFT (TD-DFT) calculations (see section 3.3). Plain GGA and hybrid XCfunctionals often greatly underestimate excitation energies of charge transfer (CT) states and completely fail in the description of double excitations, based on the single reference nature of DFT, 171, 173, 177
Parts of these deficiencies can be traced to a wrong behavior in the long-range part of the interelectronic exchange potential. In B3LYP, it decays only as $-0.2 r_{12}^{-1}$, instead of the correct $-r_{12}^{-1}$ and the amount of exact HF exchange is independent of $r_{12}$. 178 , 179 By making the amount of exact exchange depend on the interelectronic distance $r_{12}$, greatly improved results for CT states are obtained. To this end, Yanai et al. proposed the coulomb attenuated method (CAM), where $r_{12}^{-1}$ is split into

$$
\begin{equation*}
\frac{1}{r_{12}}=\underbrace{\frac{1-\left[\alpha+\beta \operatorname{erf}\left(\mu r_{12}\right)\right]}{r_{12}}}_{\text {short range }}+\underbrace{\frac{\alpha+\beta \operatorname{erf}\left(\mu r_{12}\right)}{r_{12}}}_{\text {long range }} . \tag{3.17}
\end{equation*}
$$

Optimal values were determined as $\alpha=0.19, \beta=0.46$ and $\mu=0.33$. (179] Eq. (3.17) is used to interpolate the exact exchange ratio between $19 \%$ for short and $65 \%$ for long
interelectronic distances. In the CAM framework B3LYP uses $\alpha=0.2$ and $\beta=0.0$. XC functionals incorporating eq. (3.17) are called range-separated hybrids. $180-183$

Alternatively, local hybrid XC functionals, where the amount of exact exchange is real-space position dependent, also show promising performance for CT states. $184-186$

### 3.3 Time-Dependent Density Functional Theory

Excited states, arsing from a time-dependent external potential, e.g., an oscillating electric field with frequency $\omega$, field strength $F$ and dipole operator $\boldsymbol{\mu}$

$$
\begin{equation*}
\boldsymbol{V}_{\mathrm{ext}}(t)=\boldsymbol{\mu} F \cos (\omega t) \tag{3.18}
\end{equation*}
$$

can't be treated by classic Hohenberg-Kohn-Sham DFT, as the Hohenberg-Kohn theorems are restricted to electronic densities, describing time-independent GS5. Runge and Gross extended the time-independent DFT formalism and proved a unique one-to-one correspondence between a time-dependent external potential and the electronic density $n(\boldsymbol{r}, t) .187-189$

The time-dependent electronic density is calculated similar to the Kohn-Sham (KS) ansatz in eq. 3.13 as

$$
\begin{equation*}
n(\boldsymbol{r}, t)=\sum_{a}\left|\psi_{a}(\boldsymbol{r}, t)\right|^{2} \tag{3.19}
\end{equation*}
$$

in which the orbitals $\psi_{a}(\boldsymbol{r}, t)$ are obtained from the time-dependent KS-equations

$$
\begin{equation*}
i \frac{\partial}{\partial t} \psi_{a}(\boldsymbol{r}, t)=\left[-\frac{1}{2} \nabla^{2}+V_{s}(\boldsymbol{r}, t)\right] \psi_{a}(\boldsymbol{r}, t) \tag{3.20}
\end{equation*}
$$

with the time-dependent KS potential $V_{s}(\boldsymbol{r}, t)$.
While there exist real-space formulations of TD-DFT, 190 193 that explicitly propagate the electronic density in time, TD-DFT calculations are mostly conducted in the linear response ( $\overline{\mathrm{LR}}$ ) formulation. 194,195 Assuming a small time-dependent perturbation, the electronic density is given as

$$
\begin{equation*}
n(\boldsymbol{r}, t)=n_{0}(\boldsymbol{r})+n_{1}(\boldsymbol{r}, t) \tag{3.21}
\end{equation*}
$$

where $n_{1}(\boldsymbol{r}, t)$ is the linear response. Its Fourier transformation is given as

$$
\begin{equation*}
n_{1}(\boldsymbol{r}, \omega)=\int \mathrm{d} \boldsymbol{r}^{\prime} \chi\left(\boldsymbol{r}, \boldsymbol{r}^{\prime}, \omega\right) v_{1}(\boldsymbol{r}, \omega) \tag{3.22}
\end{equation*}
$$

with the density-density response function $\chi\left(\boldsymbol{r}, \boldsymbol{r}^{\prime}, \omega\right)$, whose poles correspond to the excitation energies of the system under study. 196, 197)

Practically, excitation energies $\omega$ and ES wave functions, expressed as linear combination of singly excited SDs , are obtained by solving the Casida equations 194,195

$$
\left[\begin{array}{cc}
A & B  \tag{3.23}\\
A^{*} & B^{*}
\end{array}\right]\binom{X}{Y}=\omega\left[\begin{array}{cc}
1 & 0 \\
0 & -1
\end{array}\right]\binom{X}{Y} .
$$

Explicit expressions for the matrices $\boldsymbol{A}$ and $\boldsymbol{B}$ are found in the literature. 195,198 The eigenvectors $\binom{\boldsymbol{X}}{\boldsymbol{Y}}$ correspond to the transition density matrices (see section 4.2).

### 3.4 Density Functional Tight-Binding

Naive HF and DFT both scale as $N^{4}$, with $N$ denoting the number of basis functions. 189 This unfavorable scaling makes them unsuitable for quick, explorative calculations and for treating large systems, with thousands of atoms. On the other hand, force fields ( FF ) enable molecular dynamic simulations of many nanoseconds for several hundred thousands of atoms. 199201 Drawbacks are their complicated setup and the absence of a wave function, for further analysis or the calculation of ESF. The realm between high scaling ab initio methods and FFs is given by semi-empirical quantum chemistry methods (SQM). They allow treating large systems, while still providing a wave function. 202, 203] A SQM closely related to DFT is density functional tight-binding (DFTB). 204 206

Starting from the DFT energy functional in eq. (3.15) and assuming $n(\boldsymbol{r})$ is the sum of an unperturbed reference density $n_{0}$ and a small fluctuation $\Delta n$, the XCenergy functional can be expanded in a Taylor series around $n_{0} .207$

$$
\begin{align*}
E_{\mathrm{XC}}\left[n_{0}+\Delta n\right] & =E_{\mathrm{XC}}\left[n_{0}\right]+\int\left[\frac{\delta E_{\mathrm{XC}}[n]}{\delta n}\right]_{n_{0}} \Delta n+\frac{1}{2} \int^{\prime} \int\left[\frac{\delta^{2} E_{\mathrm{XC}}[n]}{\delta n^{2}}\right]_{n_{0}, n_{0}^{\prime}} \Delta n \Delta n^{\prime} \\
& +\frac{1}{6} \int^{\prime \prime} \int^{\prime} \int\left[\frac{\delta^{3} E_{\mathrm{XC}}[n]}{\delta n^{3}}\right]_{n_{0}, n_{0}^{\prime}, n_{0}^{\prime \prime}} \Delta n \Delta n^{\prime} \Delta n^{\prime \prime}+\ldots \tag{3.24}
\end{align*}
$$

For simplicity, the dependence of $n$ on $\boldsymbol{r}$ has been dropped in eq. (3.24) and integration occurs over $\boldsymbol{r}, \boldsymbol{r}^{\prime}$ and $\boldsymbol{r}^{\prime \prime}$. Based on the expansion order in eq. (3.24) different DFTB approaches are distinguished, ranging from DFTB1 to DFTB , with the latter being the most accurate one. 205]

The general energy expression for self-consistent charge (SCC) DFTB comprises three terms 207

$$
\begin{equation*}
E_{\text {SCC-DFTB }}=\underbrace{\sum_{i a b} n_{i} \sum_{\mu \in a} \sum_{\nu \in b} c_{\mu i} c_{\nu i} H_{\mu \nu}^{0}}_{E^{H 0}}+\underbrace{\frac{1}{2} \sum_{a b} \Delta q_{a} \Delta q_{b} \gamma_{a b}}_{E^{\gamma}}+\underbrace{\frac{1}{2} \sum_{a b} V_{a b}^{\text {rep }}}_{E^{\text {rep }}} \tag{3.25}
\end{equation*}
$$

where $i$ runs over all KS orbitals, $a$ and $b$ over all atoms and the greek indices $\mu$ and $\nu$ over basis functions at the respective atom.
$E^{H 0}$ contains energy contributions from an atomic orbital Hamiltonian and depends entirely on precomputed reference densities, [208] making the approach, in combination with a minimal valence basis, computationally extremely efficient. Diagonalization of $H^{0}$ is the dominant step in DFTB calculations. Pair-wise repulsive terms are gathered in $E^{\text {rep }}$. Considering only $E^{H 0}$ and $E^{\text {rep }}$ leads to the non-self consistent DFTB1 method. 205 Including $E^{\gamma}$ yields the self-consistent methods (DFTB2, DFTB3). 207, 209] $\Delta q_{i}$ is the net charge of atom $i$

$$
\begin{equation*}
\Delta q_{i}=q_{i}-q_{i}^{0} \tag{3.26}
\end{equation*}
$$

and $\gamma_{a b}$ describes the electron-electron interaction. It is given as integral over two normalized Slater-type spherical charge densities. (207) To obtain meaningful geometries, a dispersion correction is often added to eq. (3.25), which also proved crucial for reliable DFT geometries. 210-213

A major drawback of the DFTB approach is its element-pair-wise parametrization, resulting in a very tedious fitting process. [214, 215] The recently proposed extended tight-binding (XTB) methods by Grimme alleviate the parametrization problem, as they avoid pair-wise parameters and are parametrized up to radon $(Z=86) .[212,216,217$

## 4 Excited State Tracking in Optimizations

In contrast to GS optimizations, where the state of interest is usually energetically well separated from other electronic states and crossings are unlikely, the picture is different for ES optimizations. Given an initial ES geometry there may be a plethora of energetically close states that may cross with the state of interest over the course of an optimization. To overcome this problem, robust algorithms for ES tracking are of vital importance. Until now, many different techniques for state-tracking have been proposed, i.e., comparison of attachment- and detachment densities, [218] quantified natural transition orbital (NTO) analysis, 219,220 overlaps between transition density matrices in the MO basis 221, 222 and the NTO basis 223,224 , overlaps between wavefunctions comprised of arbitrarily excited SD 225 and overlaps between wavefunctions constructed from singly excited SD. (226] As the latter two methods calculate overlaps between SDb, they are computationally much more expensive than the former methods. Exclusively, actual overlaps between wavefunctions - comprised of arbitrarily excited SD - can capture crossings between ground- and excited electronic states, as the other methods rely on quantities that are only defined for ESp, like transition density matrices and NTO; derived from the former. 227
ES tracking in optimizations is realized by calculating excited state overlaps between a given molecular geometry and a reference geometry with the overlaps being stored in an overlap matrix $\boldsymbol{S}$. The exact expressions for $\boldsymbol{S}$ depend on the employed tracking algorithm. In the following, three different methods for ES tracking will be discussed briefly.

## Parts of this chapter are based on:

Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: $10.1002 /$ qua 26390

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### 4.1 Wavefunction Overlaps

Wavefunction overlaps (WFO) between two sets of electronic states given by the wavefunctions $\left\{\left|\Psi_{I}\right\rangle\right\}$ and $\left\{\left|\Psi_{J}^{\prime}\right\rangle\right\}$, calculated at molecular geometries $\boldsymbol{R}_{I}$ and $\boldsymbol{R}_{J}$, can be expanded into SD:

$$
\begin{equation*}
S_{I J}=\left\langle\Psi_{I} \mid \Psi_{J}^{\prime}\right\rangle=\sum_{k=1}^{N_{\mathrm{Cl}, I}} \sum_{l=1}^{N_{\mathrm{Cl}, J}} d_{I k} d_{J l}\left\langle\Phi_{k} \mid \Phi_{l}^{\prime}\right\rangle, \tag{4.1}
\end{equation*}
$$

with $d$ being (configuration interaction) CI-coefficients, $N_{\text {CI }}$ their number and $\{\langle\Phi|\}$ denote SDb. Overlaps between Slater determinants $\left\langle\Phi_{k} \mid \Phi_{l}^{\prime}\right\rangle$ are reduced to MO overlaps.

$$
\begin{align*}
& \left\langle\Phi_{k} \mid \Phi_{l}^{\prime}\right\rangle \\
& = \\
& \left|\begin{array}{cccccc}
\left\langle\phi_{k(1)} \mid \phi_{l(1)}^{\prime}\right\rangle & \cdots & \left\langle\phi_{k(1)} \mid \phi_{l\left(n_{\alpha}\right)}^{\prime}\right\rangle & \left\langle\phi_{k(1)} \mid \bar{\phi}_{l\left(n_{\alpha}+1\right)}^{\prime}\right\rangle & \cdots & \left\langle\phi_{k(1)} \mid \bar{\phi}_{l(n)}^{\prime}\right\rangle \\
\vdots & \ddots & \vdots & & \ddots & \vdots \\
\left\langle\phi_{k\left(n_{\alpha}\right)}^{\prime} \mid \phi_{l(1)}^{\prime}\right\rangle & \cdots & \left\langle\phi_{k\left(n_{\alpha}\right)} \mid \phi_{l\left(n_{\alpha}\right)}^{\prime}\right\rangle & & & \\
\vdots & & & \left\langle\bar{\phi}_{k\left(n_{\alpha}+1\right)} \mid \bar{\phi}_{l\left(n_{\alpha}+1\right)}^{\prime}\right\rangle & \cdots & \left\langle\bar{\phi}_{k\left(n_{\alpha}+1\right)} \mid \bar{\phi}_{l(n)}^{\prime}\right\rangle \\
& \ddots & & \vdots & \ddots & \vdots \\
\left\langle\bar{\phi}_{k(n)} \mid \phi_{l(1)}^{\prime}\right\rangle & \cdots & \left\langle\bar{\phi}_{k(n)} \mid \phi_{l\left(n_{\alpha}\right)}^{\prime}\right\rangle & \left\langle\bar{\phi}_{k(n)} \mid \bar{\phi}_{l\left(n_{\alpha}+1\right)}^{\prime}\right\rangle & \cdots & \left\langle\bar{\phi}_{k(n)} \mid \bar{\phi}_{l(n)}^{\prime}\right\rangle
\end{array}\right|  \tag{4.2}\\
& = \\
& \left|\begin{array}{cccccc}
\left\langle\phi_{k(1)} \mid \phi_{l(1)}^{\prime}\right\rangle & \cdots & \left\langle\phi_{k(1)} \mid \phi_{l\left(n_{\alpha}\right)}^{\prime}\right\rangle \\
\vdots & \ddots & \vdots & & & \\
\left\langle\phi_{k\left(n_{\alpha}\right)} \mid \phi_{l(1)}^{\prime}\right\rangle & \cdots & \left\langle\phi_{k\left(n_{\alpha}\right)} \mid \phi_{l\left(n_{\alpha}\right)}^{\prime}\right\rangle & \left\langle\bar{\phi}_{k\left(n_{\alpha}+1\right)} \mid \bar{\phi}_{l\left(n_{\alpha}+1\right)}^{\prime}\right\rangle & \cdots & \left\langle\bar{\phi}_{k\left(n_{\alpha}+1\right)} \mid \bar{\phi}_{l(n)}^{\prime}\right\rangle \\
& & \vdots & \ddots & \vdots \\
& \mathbf{0} & & \left\langle\bar{\phi}_{k(n)} \mid \bar{\phi}_{l\left(n_{\alpha}+1\right)}^{\prime}\right\rangle & \cdots & \left\langle\bar{\phi}_{k(n)} \mid \bar{\phi}_{l(n)}^{\prime}\right\rangle
\end{array}\right|
\end{align*}
$$

The MO overlap matrix is block diagonal, as MOs of different spin are orthogonal and their overlap vanishes. 225] Overlaps between alpha MO; $\left(\mathcal{S}_{k l}\right)$ and beta MOs $\left(\overline{\mathcal{S}}_{k l}\right)$ can be calculated separately.

$$
\begin{align*}
& \left\langle\Phi_{k} \mid \Phi_{l}^{\prime}\right\rangle= \\
& \underbrace{\left|\begin{array}{ccc}
\left\langle\phi_{k(1)} \mid \phi_{l(1)}^{\prime}\right\rangle & \cdots & \left\langle\phi_{k(1)} \mid \phi_{l\left(n_{\alpha}\right)}^{\prime}\right\rangle \\
\vdots & \ddots & \vdots \\
\left\langle\phi_{k\left(n_{\alpha}\right)} \mid \phi_{l(1)}^{\prime}\right\rangle & \cdots & \left\langle\phi_{k\left(n_{\alpha}\right)} \mid \phi_{l\left(n_{\alpha}\right)}^{\prime}\right\rangle
\end{array}\right|}_{\mathcal{S}_{k l}} \times \underbrace{\left|\begin{array}{ccc}
\left\langle\bar{\phi}_{k\left(n_{\alpha}+1\right)} \mid \bar{\phi}_{l\left(n_{\alpha}+1\right)}^{\prime}\right\rangle & \cdots & \left\langle\bar{\phi}_{k\left(n_{\alpha}+1\right)} \mid \bar{\phi}_{l(n)}^{\prime}\right\rangle \\
\vdots & \ddots & \vdots \\
\left\langle\bar{\phi}_{k(n)} \mid \bar{\phi}_{l\left(n_{\alpha}+1\right)}^{\prime}\right\rangle & \cdots & \left\langle\bar{\phi}_{k(n)} \mid \bar{\phi}_{l(n)}^{\prime}\right\rangle
\end{array}\right|}_{\overline{\mathcal{S}}_{k l}} \tag{4.3}
\end{align*}
$$

MO overlaps are ultimately obtained in a straightforward fashion from atomic orbital (AO) overlaps

$$
\begin{equation*}
\left\langle\phi_{p} \mid \phi_{q}^{\prime}\right\rangle=\sum_{\mu=1}^{N_{\mathrm{MO}}} \sum_{\nu=1}^{N_{\mathrm{MO}}} C_{p \mu} C_{q \nu}^{\prime}\left\langle\chi_{\mu} \mid \chi_{\nu}^{\prime}\right\rangle \tag{4.4}
\end{equation*}
$$

with $\left\{\left|\phi_{p}\right\rangle\right\}$ and $\left\{\left|\phi_{q}^{\prime}\right\rangle\right\}$ being two distinct sets of MO; with coefficients $\boldsymbol{C}$ and $\boldsymbol{C}^{\prime}$. The AO sets $\left\{\left|\chi_{\mu}\right\rangle\right\}$ and $\left\{\left|\chi_{\nu}^{\prime}\right\rangle\right\}$ are centered at their respective coordinates, $\boldsymbol{R}_{I}$ and $\boldsymbol{R}_{J}$, and their overlap matrix $\boldsymbol{S}_{\mathrm{AO}}$ can be obtained from a calculation comprising all basis functions, or, if the difference between both geometries is small, recovered from the MO coefficients $\boldsymbol{C}\left(\boldsymbol{C}^{\prime}\right)$ at $\boldsymbol{R}_{I}\left(\boldsymbol{R}_{J}\right): 228,229$

$$
\begin{equation*}
\boldsymbol{S}_{\mathrm{AO}}=\left(\boldsymbol{C}^{-1}\right)^{\top} \boldsymbol{C}^{-1} \tag{4.5}
\end{equation*}
$$

Application of eq. 4.5 in geometry optimization is justified, as the geometries between two optimization cycles are very similar. The reader is referred to the original publication for a full discussion of the wavefunction overlap (WFO) algorithm. 225

### 4.2 Transition Density Overlaps

Another option for ES tracking are transition density matrix (TDEN) overlaps. Given two MO coefficient matrices $\boldsymbol{C}_{I}$ and $\boldsymbol{C}_{J}$ obtained at different molecular geometries $\boldsymbol{R}_{I}$ and $\boldsymbol{R}_{J}$, their overlap matrix $\boldsymbol{S}_{\mathrm{MO}}$ is defined according to eq. (4.4) as

$$
\begin{equation*}
\boldsymbol{S}_{\mathrm{MO}}=\boldsymbol{C}_{I} \boldsymbol{S}_{\mathrm{AO}} \boldsymbol{C}_{J}^{\top} \tag{4.6}
\end{equation*}
$$

Assuming $N_{\mathrm{o}}$ occupied and $N_{\mathrm{v}}$ virtual MOs, the overlaps between two one-electron transition density matrices $\boldsymbol{T}_{I}$ and $\boldsymbol{T}_{J}$ of dimensions $\left(N_{\mathrm{o}} \times N_{\mathrm{o}}+N_{\mathrm{v}}\right)$ in their respective

MO bases $\boldsymbol{C}_{I}$ and $\boldsymbol{C}_{J}$ are calculated as:

$$
\begin{equation*}
S_{I J}=\sum_{\sigma=1}^{N_{\mathrm{o}}} \sum_{\tau=1}^{N_{\mathrm{o}}} \sum_{\mu=1}^{N_{\mathrm{o}}+N_{\mathrm{v}}} \sum_{\nu=1}^{N_{\mathrm{o}}+N_{\mathrm{v}}} T_{I, \tau \mu} T_{J, \sigma \nu} S_{\mathrm{MO}, \sigma \tau} S_{\mathrm{MO}, \mu \nu} \tag{4.7}
\end{equation*}
$$

Equation (4.7) is efficiently implemented using matrix-matrix products.

$$
\begin{equation*}
S_{I J}=\sum_{\sigma=1}^{N_{\mathrm{o}}} \sum_{\tau=1}^{N_{\mathrm{o}}} S_{\mathrm{MO}, \tau \sigma} \boldsymbol{T}_{I} \boldsymbol{S}_{\mathrm{MO}} \boldsymbol{T}_{J}^{\top} \tag{4.8}
\end{equation*}
$$

Compared to WFOs, the evaluation of TDEN overlaps is computationally less challenging, as no determinant calculations are involved.

### 4.3 Natural Transition Orbital Overlaps

A third option for ES tracking are NTO overlaps. 223 Natural transition orbitals $\psi$ allow a more compact representation of ES $\leqslant$ through a basis transformation of the occupied and virtual MOs $\phi$ and $\phi^{\prime}$.

$$
\begin{align*}
& \left(\psi_{1}, \cdots, \psi_{N_{\mathrm{o}}}\right)=\left(\phi_{1}, \cdots, \phi_{N_{\mathrm{o}}}\right) \boldsymbol{U}  \tag{4.9}\\
& \left(\psi_{1}^{\prime}, \cdots, \psi_{N_{\mathrm{v}}}^{\prime}\right)=\left(\phi_{1}^{\prime}, \cdots, \phi_{N_{\mathrm{v}}}^{\prime}\right) \boldsymbol{V} \tag{4.10}
\end{align*}
$$

Here, the matrices $\boldsymbol{U}$ and $\boldsymbol{V}$ are conveniently calculated by a singular value decomposition (SVD) of the one-electron transition density matrix $\boldsymbol{T}$

$$
\begin{equation*}
\boldsymbol{T}=\boldsymbol{U} \boldsymbol{\Sigma} \boldsymbol{V}^{\top} \tag{4.12}
\end{equation*}
$$

with $\boldsymbol{\Sigma}$ being a diagonal matrix containing the singular values $\left\{\lambda_{i}\right\} \cdot 227$ Often, only a few singular values differ significantly from zero, thus $\boldsymbol{T}$ is well captured by a low-rank approximation and only few NTOs are needed for the ES description.

Given two one-electron transition density matrices $\boldsymbol{T}_{I}$ and $\boldsymbol{T}_{J}$, calculated at nuclear geometries $\boldsymbol{R}_{I}$ and $\boldsymbol{R}_{J}$, NTO; $\left\{\psi_{I}\right\}$ and $\left\{\psi_{J}\right\}$, their overlap is calculated as

$$
\begin{equation*}
S_{I J}=\sum_{k=1}^{N_{\mathrm{NTO}}} \lambda_{I, k}\left|\boldsymbol{C}_{I, k}^{\mathrm{NTO}} \boldsymbol{S}_{\mathrm{AO}} \boldsymbol{C}_{J, k}^{\mathrm{NTO}, \mathrm{~T}}\right| \tag{4.13}
\end{equation*}
$$

with $\boldsymbol{C}^{\mathrm{NTO}}$ containing the respective NTO coefficients for excited state $I$ or $J$ and $\lambda_{I, k}$ being the $k$-th singular value obtained from the SVD of $\boldsymbol{T}_{I} . N_{\text {NTO }}$ is determined from the number of singular values above a prescribed threshold (e.g. 0.3). A full derivation of the NTO overlap algorithm is presented in ref. [224.

### 4.4 Choosing a Reference Cycle

While tracking ESS in geometry optimizations, one of the two nuclear geometries $\boldsymbol{R}_{I}$ and $\boldsymbol{R}_{J}$ is given by the current optimization cycle, whereas the remaining one is given by a reference cycle. Different types of reference cycles can be chosen. Overlaps between the ES; at the current cycle can either be calculated with the ES; at the first cycle or with the previous optimization cycle. Alternatively, an adaptive algorithm can be employed, where the reference cycle is only updated when two conditions are met. A new reference cycle is only chosen when the highest overlap of the current excited states and the reference state is above a certain threshold (e.g. 0.5), indicating that there is a well defined correspondence between one of the current ES; and the reference ES, Furthermore, the ratio between the second highest and highest overlaps must be between 0.3 and 0.6. Low values indicate easily distinguishable states, therefore no update is needed. High values indicate very similar overlaps, consequentially an update may not be appropriate as multiple states at the current cycle are very similar to the reference state.

## 5 Obtaining Minima on Potential Energy Surfaces

### 5.1 The Quadratic Approximation

Searching for minima on PES is an optimization problem

$$
\begin{equation*}
\min _{\boldsymbol{R} \in \mathbb{R}^{n}} E(\boldsymbol{R}) \tag{5.1}
\end{equation*}
$$

with $\boldsymbol{R}$ denoting a set of (nuclear) coordinates and $E(\boldsymbol{R})$ the electronic energy. A suitable step $\boldsymbol{p}_{k}$ to minimize $E(\boldsymbol{R})$ at optimization cycle $k$ can be obtained from the Taylor expansion of $E(\boldsymbol{R})$ to second-order around $\boldsymbol{R}_{k}$, with gradient $\boldsymbol{g}_{k}$ and Hessian $\boldsymbol{H}_{k}$

$$
\begin{equation*}
E\left(\boldsymbol{R}_{k}+\boldsymbol{p}_{k}\right)=E_{k}+\boldsymbol{p}_{k}^{\top} \boldsymbol{g}_{k}+\frac{1}{2} \boldsymbol{p}_{k}^{\top} \boldsymbol{H}_{k} \boldsymbol{p}_{k} . \tag{5.2}
\end{equation*}
$$

By differentiating eq. (5.2) with respect to $\boldsymbol{p}_{k}$ and equating the result to zero, the step is given as

$$
\begin{equation*}
\boldsymbol{p}_{k}=-\left(\boldsymbol{H}_{k}\right)^{-1} \boldsymbol{g}_{k} \tag{5.3}
\end{equation*}
$$

For minimizations it has to be ensured that $\boldsymbol{H}_{k}$ has no negative eigenvalues, otherwise eq. (5.3 will result in an erroneous step. Transformed to the Hessian eigenvector basis, eq. (5.3) is rewritten as

$$
\begin{equation*}
\widetilde{p}_{k, i}=-\frac{\widetilde{g}_{k, i}}{\widetilde{H}_{k, i i}} \tag{5.4}
\end{equation*}
$$

with $i$ denoting the eigenvector index and the tilde denotes transformed quantities. For negative eigenvalues $\widetilde{H}_{k, i i}<0$, the step produced by eq. 5.4 will be uphill along the gradient, instead of downhill, against the gradient.

## Parts of this chapter are based on:

Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: 10.1002 /qua 26390

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## 5 Obtaining Minima on Potential Energy Surfaces

Downhill steps can be obtained by determining a suitable shift parameter $\lambda$, e.g., from root-finding algorithms. 230

$$
\begin{equation*}
\widetilde{p}_{k, i}=-\frac{\widetilde{g}_{k, i}}{\widetilde{H}_{k, i i}-\lambda} \tag{5.5}
\end{equation*}
$$

Alternatively, valid downhill directions are ensured by rational function optimization (RFO) where

$$
\begin{equation*}
E\left(\boldsymbol{R}_{k}+\boldsymbol{p}_{k}\right)=E_{k}+\frac{\boldsymbol{p}_{k}^{\top} \boldsymbol{g}_{k}+\frac{1}{2} \boldsymbol{p}_{k}^{\top} \boldsymbol{H}_{k} \boldsymbol{p}_{k}}{1+\boldsymbol{p}_{k}^{\top} \boldsymbol{S} \boldsymbol{p}_{k}} \tag{5.6}
\end{equation*}
$$

instead of eq. 55.2 is minimized. $70,231,232$ Matrix $S$ is usually chosen as the unit matrix. The RFO step $\boldsymbol{p}_{k}$ for minimizations is obtained by solving the eigenvalue equation

$$
\left[\begin{array}{cc}
\boldsymbol{H}_{k} & \boldsymbol{g}_{k}  \tag{5.7}\\
\boldsymbol{g}_{k}^{\top} & 0
\end{array}\right]\left[\begin{array}{c}
\boldsymbol{p}_{k} \\
1
\end{array}\right]=\nu\left[\begin{array}{cc}
\boldsymbol{S} & \mathbf{0} \\
\mathbf{0}^{\top} & 1
\end{array}\right]\left[\begin{array}{c}
\boldsymbol{p}_{k} \\
1
\end{array}\right]
$$

and scaling the eigenvector corresponding to the lowest eigenvalue $\nu$, so its last element equals to 1 .

### 5.2 Hessian Update and Initial Choice

Depending on the Hessian employed to calculate the step $\boldsymbol{p}_{k}$, e.g., by eqs. (5.3) and (5.7), several methods are distinguished. With a simple unit Hessian $\left(\boldsymbol{H}_{k}=\boldsymbol{I}\right)$ the method of steepest descent is obtained from eq. (5.3), where $\boldsymbol{p}_{k}$ is simply given by

$$
\begin{equation*}
\boldsymbol{p}_{k}=-\left(\boldsymbol{H}_{k}\right)^{-1} \boldsymbol{g}_{k}=-(\boldsymbol{I})^{-1} \boldsymbol{g}_{k}=-\boldsymbol{g}_{k} \tag{5.8}
\end{equation*}
$$

Performance of the steepest descent method is often very poor, as coordinate coupling is fully neglected. 233

Using the exact Hessian in every cycle eq. (5.3) gives rise to Newton's method. 234 If the PES is well described by a quadratic model, Newton's method shows fast local convergence. Obtaining the exact Hessian in every optimization cycle is often not feasible while searching for SPs, as its calculation is computationally very demanding. A compromise between the computationally cheap, but badly performing method of steepest descent and the performant, but computationally expensive Newton method is provided by quasi-Newton (QN) methods. In QN methods, an approximate Hessian $\boldsymbol{A}_{k}$ is used to obtain the step

$$
\begin{equation*}
\boldsymbol{p}_{k}=-\left(\boldsymbol{A}_{k}\right)^{-1} \boldsymbol{g}_{k} \tag{5.9}
\end{equation*}
$$

Based on the secant equation

$$
\begin{equation*}
\boldsymbol{A}_{k+1} \boldsymbol{s}_{k}=\boldsymbol{y}_{k} \tag{5.10}
\end{equation*}
$$

an initial Hessian is updated along the optimization, using only coordinate differences $\boldsymbol{s}_{k}=\boldsymbol{R}_{k+1}-\boldsymbol{R}_{k}$ and gradient differences $\boldsymbol{y}_{k}=\boldsymbol{g}_{k+1}-\boldsymbol{g}_{k}$, obtained at two successive cycles $k$ and $k+1$. As eq. 5.10 is under-determined, additional constraints like enforcing certain symmetry and positive definiteness of $\boldsymbol{A}_{k+1}$ are employed, to devise actual update formulas. One of the most popular formulas for minimizations is the Broyden-Fletcher-Goldfarb-Shanno ( $\overline{\text { BFGS }}$ ) update. 235 238

$$
\begin{equation*}
\boldsymbol{A}_{k+1}=\boldsymbol{A}_{k}-\frac{\boldsymbol{A}_{k} \boldsymbol{s}_{k} \boldsymbol{s}_{k}^{\top} \boldsymbol{A}_{k}}{\boldsymbol{s}_{k}^{\top} \boldsymbol{A}_{k} \boldsymbol{s}_{k}}+\frac{\boldsymbol{y}_{k} \boldsymbol{y}_{k}^{\top}}{\boldsymbol{y}_{k}^{\top} \boldsymbol{s}_{k}} \tag{5.11}
\end{equation*}
$$

The need to invert $\boldsymbol{A}_{k}$ in every optimization cycle is overcome, by directly updating the inverse of $\boldsymbol{A}_{k}$ :

$$
\begin{equation*}
\boldsymbol{A}_{k}^{-1}=\left(\boldsymbol{I}-\rho_{k} \boldsymbol{s}_{k} \boldsymbol{y}_{k}^{\top}\right) \boldsymbol{A}_{k}^{-1}\left(\boldsymbol{I}-\rho_{k} \boldsymbol{y}_{k} \boldsymbol{s}_{k}^{\top}\right)+\rho_{k} \boldsymbol{s}_{k} \boldsymbol{s}_{k}^{\top} \tag{5.12}
\end{equation*}
$$

with $\rho_{k}=1 / \boldsymbol{y}_{k}^{\top} \boldsymbol{s}_{k}$. Other popular update formulas are the symmetric rank 1 (SR1) update, Powell's symmetric Broyden ( (PSB) update, and especially for optimizing saddle points, Bofills update. [234, 239, 240] Instead of always updating $\boldsymbol{A}_{k}$ the exact Hessian could, if feasible, be recalculated periodically and used as starting point for further Hessian updates, thus mixing Newton and QN steps for improved optimization performance.

Besides the update formula, optimization outcome is greatly affected by the initial Hessian choice (or its inverse). If calculating an initial Hessian is infeasible or not desired, an approximate model Hessian can be estimated. For the purpose of geometry optimization, several model Hessians were proposed, e.g., by Lindh, 241] Fischer [242], and Swart. 243] The only requirement for calculating a model Hessian from the previously mentioned models, is the existence of a set of primitive internal coordinates (stretches, bends, dihedrals, see also Figure 6.1, defining the connectivity of the system under study. See chapter 6 for an overview on internal coordinates.

In the following, Fischers model Hessian is presented exemplarily. 242 By fitting data from reference molecules obtained at the HF/6-31G** level of theory, Fischer gave empirical formulas for the individual force constants $k$. Eqs. (5.13) to 5.15 are specified using the atomic labels $(M, N, O, P)$ shown in Figure 6.1 on page 42 . The bond length between two atoms $M$ and $N$ is denoted by $r^{M N}$. The sum of their covalent radii is denoted by $r_{\text {cov }}^{M N}$.

$$
\begin{gather*}
k_{\mathrm{stretch}}=A e^{-B\left(r^{M N}-r_{\text {cov }}^{M N}\right)}  \tag{5.13}\\
A=0.3601, B=1.944 \\
k_{\text {bend }}=A+\frac{B}{\left(r_{\text {cov }}^{O M} r_{\mathrm{cov}}^{O N}\right)^{D}} e^{-C\left(r^{O M}+r^{O N}-r_{\text {cov }}^{O M}-r_{\mathrm{cov}}^{O N}\right)}  \tag{5.14}\\
A=0.089, B=0.11, C=0.44, D=-0.42 \\
k_{\text {dihedral }}=A+\frac{B L^{D}}{\left(r^{O P} r_{\mathrm{cov}}^{O P} e^{E}\right.} e^{-C\left(r^{O P}-r_{\text {cov }}^{O P}\right)}  \tag{5.15}\\
A=0.0015, B=14.0, C=2.85, D=0.57, E=4.00
\end{gather*}
$$

Parameter $L$ in eq. (5.15) denotes the number of bond stretches connected to the central atoms $O$ and $P$ of the dihedral, without the central $O-P$ bond. Calculating a model Hessian by eqs. (5.13) to (5.15) yields a diagonal Hessian in internal coordinates. If the optimization is to be conducted in Cartesian coordinates, the estimated Hessian has to be converted to Cartesian coordinates by eq. (6.10).

### 5.3 Conjugate Gradient and Limited-Memory BFGS

For big molecules, solving the RFO eigenvalue problem eq. (5.7) may become prohibitively expensive. 244246 A popular algorithm to determine $\boldsymbol{p}_{k}$, avoiding any matrix operations and requiring minimal storage, is the nonlinear conjugate gradient (CG) method. (75] In CG methods, the search direction is given as

$$
\begin{equation*}
\boldsymbol{p}_{k+1}=-\boldsymbol{g}_{k+1}+\beta_{k+1} \boldsymbol{p}_{k} \tag{5.16}
\end{equation*}
$$

The algorithm is initialized with a steepest descent step $\boldsymbol{p}_{0}=-\alpha_{0} \boldsymbol{g}_{0}$. Many variants for $\beta$ have been proposed, a popular one was given by Polak and Ribière 247, 248

$$
\begin{equation*}
\beta_{k+1}^{\mathrm{PR}}=\frac{\boldsymbol{g}_{k+1}^{\top}\left(\boldsymbol{g}_{k+1}-\boldsymbol{g}_{k}\right)}{\boldsymbol{g}_{k}^{\top} \boldsymbol{g}_{k}} \tag{5.17}
\end{equation*}
$$

Hager and Zhang offered a more complex definition 249]

$$
\begin{equation*}
\beta_{k+1}^{\mathrm{HZ}}=\left(\boldsymbol{y}_{k}-2 \boldsymbol{p}_{k} \frac{\boldsymbol{y}_{k}^{\top} \boldsymbol{y}_{k}}{\boldsymbol{y}_{k}^{\top} \boldsymbol{p}_{k}}\right)^{\top} \frac{\boldsymbol{g}_{k+1}}{\boldsymbol{y}_{k}^{\top} \boldsymbol{p}_{k}} . \tag{5.18}
\end{equation*}
$$

As CG methods employ no Hessian information, their performance is expected to be inferior for molecular optimizations.

Limited-memory methods like limited-memory BFGS (LBFGS) offer the best of both worlds: Low storage requirements and computational efficiency similar to CG methods, while implicitly keeping an Hessian approximation. 250 By storing the latest $m$ vector pairs $\left\{\boldsymbol{s}_{i}, \boldsymbol{y}_{i}\right\}$ for $i=k-m, \ldots, k-1$, the matrix-vector product $\boldsymbol{A}_{k}^{-1} \boldsymbol{g}_{k}$ is calculated in a two-loop recursion, using only vector-vector products (see Algorithm 11). 234, 251

```
Algorithm 1 LBFGS two-loop recursion
    \(\boldsymbol{q} \leftarrow \boldsymbol{g}_{k}\)
    for \(i=k-1, \ldots, k-m\) do
        \(\alpha_{i} \leftarrow \rho_{i} \boldsymbol{s}_{i}^{\top} \boldsymbol{q} \quad \triangleright\) Store \(\alpha_{i}\)
        \(\boldsymbol{q} \leftarrow \boldsymbol{q}-\alpha_{i} \boldsymbol{y}_{i}\)
    \(\boldsymbol{r} \leftarrow\left(\boldsymbol{H}_{k}^{0}\right)^{-1} \boldsymbol{q} \quad \triangleright\left(\boldsymbol{H}_{k}^{0}\right)^{-1}=\) preconditioner or vector
    for \(i=k-m, \ldots, k-1\) do
        \(\beta_{i} \leftarrow \rho_{i} \boldsymbol{y}_{i}^{\top} \boldsymbol{r}\)
        \(\boldsymbol{r} \leftarrow \boldsymbol{r}+\boldsymbol{s}_{i}\left(\alpha_{i}-\beta_{i}\right)\)
    return \(\boldsymbol{A}_{k}^{-1} \boldsymbol{g}_{k}=\boldsymbol{r}\)
```

Two QN optimizations, using either a BFGS update or the LBFGS approach, yield the exact same results for the first $m$ cycles. The LBFGS method usually outperforms CG methods in molecular optimizations, despite slightly increased storage and computational costs. 252, 253

Care has to be taken that $\boldsymbol{s}_{k} \boldsymbol{y}_{k}>0$ is ensured, otherwise $\boldsymbol{A}_{k}$ may lose its positivedefiniteness. As the Hessian is never explicitly constructed in the LBFGS method, determination of an appropriate shift factor as in eq. (5.5) is not easily achieved. If $\boldsymbol{s}_{k} \boldsymbol{y}_{k}>0$ is violated, the Hessian update may be skipped, but this is usually not advised, as valuable curvature information is neglected. 234 By using a damped BFGS update, positive-definiteness of $\boldsymbol{A}_{k}$ can be guaranteed. 254] Recently, Goldfarb proposed a double damping procedure that modifies the $\left(s_{k}, \boldsymbol{y}_{k}\right)$ pair (Algorithm 2]. 255]

The $\boldsymbol{A}^{-1} \boldsymbol{y}$ term in Algorithm 2 is conveniently calculated using the two-loop recursion in Algorithm 1, so double damping is easily utilized in LBFGS optimizations.

### 5.4 Trust Radius

Given a search direction $\boldsymbol{p}_{k}$, a suitable step length has to be determined. To this end, several approaches exist, with the simplest being the trust radius method. If the length of a proposed step $\boldsymbol{p}_{k}$ is below or equal to a prescribed trust radius $\Delta$, it is accepted,

```
Algorithm 2 Double damping procedure, adapted from 255
    Given \(\boldsymbol{s} \leftarrow \boldsymbol{s}_{k}, \boldsymbol{y} \leftarrow \boldsymbol{y}_{k}, \boldsymbol{A}^{-1} \leftarrow \boldsymbol{A}_{k}^{-1}\) and parameters \(\mu_{1}, \mu_{2}\)
    if \(\boldsymbol{s}^{\top} \boldsymbol{y}<\mu_{1} \boldsymbol{y}^{\top} \boldsymbol{A}^{-1} \boldsymbol{y}\) then \(\theta_{1}=\frac{\left(1-\mu_{1}\right) \boldsymbol{y}^{\top} \boldsymbol{A}^{-1} \boldsymbol{y}}{\boldsymbol{y}^{\top} \boldsymbol{A}^{-1} \boldsymbol{y}-\boldsymbol{s}^{\top} \boldsymbol{y}}\)
    else \(\theta_{1}=1\)
    \(\tilde{\boldsymbol{s}}=\theta_{1} \boldsymbol{s}+\left(1-\theta_{1}\right) \boldsymbol{A}^{-1} \boldsymbol{y}\)
    if \(\tilde{\boldsymbol{s}}^{\top} \boldsymbol{y}<\mu_{2} \tilde{\boldsymbol{s}}^{\top} \tilde{\boldsymbol{s}}\) then \(\theta_{2}=\frac{\left(1-\mu_{2}\right) \tilde{\boldsymbol{s}}^{\top} \tilde{\boldsymbol{s}}}{\tilde{\boldsymbol{s}}^{\top} \tilde{\boldsymbol{s}}-\tilde{\boldsymbol{s}}^{\top} \boldsymbol{y}}\)
    else \(\theta_{2}=1\)
    \(\tilde{\boldsymbol{y}}=\theta_{2} \boldsymbol{y}+\left(1-\theta_{2}\right) \tilde{\boldsymbol{s}}\)
    return \(\tilde{s}, \tilde{y}\)
```

otherwise it can either be simply downscaled or determined from more sophisticated methods like the restricted step ( $(\overline{\mathrm{RS}})$ algorithm, as well as the level-shifted Newton method. 232,256 The trust radius $\Delta$ is updated according to the agreement between predicted energy changes $\Delta E_{\text {pred }}$, e.g., through eq. 5.6) or eq. (5.2), and actual energy changes $\Delta E_{\text {act }}$ over the course of an optimization. [234, 239


Figure 5.1: Trust radius update flowchart. Depending on the agreement between actual energy changes $\Delta E_{\text {act }}$ and predicted energy changes $\Delta E_{\text {pred }}$, the trust radius $\Delta$ is either decreased or increased.

### 5.5 Line searches

A more rigorous approach for step length determination is given by line searches that may involve multiple energy and gradient evaluations in every optimization cycle. The line search problem is formulated as

$$
\begin{equation*}
\min _{\alpha>0} E\left(\boldsymbol{R}_{k}+\alpha \boldsymbol{p}_{k}\right) \tag{5.19}
\end{equation*}
$$

with the goal to determine $\alpha$ for a given $\boldsymbol{p}_{k}$. Line searches are usually carried out iteratively, until a set of prescribed conditions is satisfied. One popular condition is to require a sufficient decrease of the function value (Armijo condition) 234

$$
\begin{equation*}
E\left(\boldsymbol{R}_{k}+\alpha \boldsymbol{p}_{k}\right) \leq E\left(\boldsymbol{R}_{k}\right)+c_{1} \alpha \boldsymbol{g}_{k}^{\top} \boldsymbol{p}_{k} . \tag{5.20}
\end{equation*}
$$

Parameter $c_{1}$ controls the degree of required energy decrease and is usually chosen quite small $\left(10^{-4}\right)$, thus allowing line search convergence even for minor energy decreases. Enforcing the Armijo condition only requires additional energy calculations. As eq. (5.20) may allow very short steps (depending on $c_{1}$ ), a second condition (curvature condition) can be enforced, guaranteeing that sufficient optimization progress is made:

$$
\begin{equation*}
\boldsymbol{g}\left(\boldsymbol{R}_{k}+\alpha_{k} \boldsymbol{p}_{k}\right)^{\top} \boldsymbol{p}_{k} \geq c_{2} \boldsymbol{g}_{k} \boldsymbol{p}_{k} . \tag{5.21}
\end{equation*}
$$

Similar to $c_{1}$ in eq. 5.20 , the parameter $c_{2} \in\left(c_{1}, 1\right)$ controls how strongly the left-hand side (LHS) of eq. (5.21) must be reduced, until the curvature condition is satisfied. Typical values for $c_{2}$ are 0.9 , if $\boldsymbol{p}_{k}$ was obtained from a (quasi)-Newton method and 0.1, if $\boldsymbol{p}_{k}$ was obtained from a conjugate gradient method. Enforcing the curvature condition requires additional gradient calculations. Applied together, the Armijo and curvature conditions are known as the Wolfe conditions. [234 An example for valid minimizers of a function, obeying the Wolfe conditions, is shown in Figure 5.2. Popular line search algorithms are those provided by Moré and Thuente [257) and by Hager and Zhang. [249]


Figure 5.2: Illustration of valid steps from $x_{0}=-7$ (red dot) towards a minimizer of $f(x)=(x-1)(x+1)(x+4)(x+7)(x-5)(x-9)$, satisfying the Wolfe conditions. $\boldsymbol{p}_{0}$ was chosen as the direction of steepest descent. The thin black line indicates points that satisfy the Armijo condition ( $c_{1}=0.0001$ ), the dashed black line indicate points that satisfy the curvature condition $\left(c_{2}=0.1\right)$. Their union, indicated by thick black dots, represent points that satisfy the Wolfe conditions.

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Conducting exact line searches satisfying the Wolfe conditions may become computationally very demanding, as additional energy and gradient evaluations are needed in every optimization cycle. Computationally cheaper, albeit less robust, are partial line searches, which are usually employed in the geometry optimizers of QC codes, as additional gradient evaluations are often costly. In practice, partial line searches are performed by fitting a polynomial between two coordinates $\boldsymbol{R}_{k}$ and $\boldsymbol{R}_{\mathrm{t}}$. Utilizing energies and projected gradients $\left(\boldsymbol{g}_{k} \cdot \alpha_{\mathrm{t}} \boldsymbol{p}_{k}\right)$ obtained at $\boldsymbol{R}_{k}$ and $\boldsymbol{R}_{\mathrm{t}}$, a constrained polynomial of fourth degree, or an unconstrained polynomial of third degree can be fitted. 75 The resulting inter- or extrapolated data (coordinates, energy and gradient) is then used to compute a new step direction $\boldsymbol{p}_{k+1}$ for the next optimization cycle, so only one energy and gradient evaluation is needed per optimization cycle. Differences between partial and full line searches are illustrated in Figure 5.3.


Figure 5.3: Flowcharts for partial (left) and full line searches (right). While partial line searches require only one energy and gradient evaluation per cycle, full line searches may require multiple energy (and gradient) evaluations per cycle, until convergence is achieved.

To the best knowledge of the author, line searches in the context of molecular optimizations are exclusively used for minimizations, but they may also be used for maximizations, e.g., maximizing the energy along the imaginary mode in partitioned rational function optimization ( $\overline{\text { PRFO }})$, presented in chapter 9 .

### 5.6 Direct Inversion in the Iterative Subspace

Instead of fitting polynomials, the geometric direct inversion in the iterative subspace (GDIIS) method can be used for inter- and extrapolation. 258 In GDIIS, a set of $k$ coordinates $\left\{\boldsymbol{R}_{i}\right\}$ is linearly combined to minimize the length of an error vector $\left\|\boldsymbol{x}^{*}\right\|_{2}$.

$$
\begin{equation*}
\boldsymbol{R}^{*}=\sum_{i}^{k} c_{i} \boldsymbol{R}_{i}, \quad \text { with } \sum_{i}^{k} c_{i}=1 \tag{5.22}
\end{equation*}
$$

The error vector $\boldsymbol{x}^{*}$ is obtained as linear combination of respective error vectors $\boldsymbol{e}_{i}$ associated with each structure $\boldsymbol{R}_{i}$.

$$
\begin{equation*}
\boldsymbol{x}^{*}=\sum_{i}^{k} c_{i} \boldsymbol{e}_{i} \tag{5.23}
\end{equation*}
$$

Common choices for $\boldsymbol{e}_{i}$ are the gradient $\boldsymbol{g}_{i}$, or the predicted quadratic step $-\boldsymbol{H}_{i}^{-1} \boldsymbol{g}_{i}$, recall eq. (5.3). 246] Coefficients $c_{i}$ are obtained by solving the least-squares problem

$$
\left(\begin{array}{cccc}
a_{1,1} & \cdots & a_{1, k} & 1  \tag{5.24}\\
\vdots & \ddots & \vdots & \vdots \\
a_{k, 1} & \cdots & a_{k, k} & 1 \\
1 & \cdots & 1 & 0
\end{array}\right)\left(\begin{array}{c}
c_{1} \\
\vdots \\
c_{k} \\
\lambda
\end{array}\right)=\left(\begin{array}{c}
0 \\
\vdots \\
0 \\
1
\end{array}\right)
$$

with $a_{i, j}=\boldsymbol{e}_{i}^{\top} \boldsymbol{e}_{j}$ and the Lagrangian multiplier $\lambda .259$ As regular GDIIS is prone to converge to the nearest SP or an inflection point, several safeguards have been proposed by Farkas and Schlegel, yielding the controlled GDIIS method. 259

Similar to $\boldsymbol{R}^{*}$, a corresponding gradient $\boldsymbol{g}^{*}$ is obtained that can be used to predict new coordinates for the next optimization cycle.

$$
\begin{equation*}
\boldsymbol{R}_{k+1}=\boldsymbol{R}^{*}-\boldsymbol{H}^{-1} \boldsymbol{g}^{*} \tag{5.25}
\end{equation*}
$$

In GDIIS, coefficients $\left|c_{i}\right|>1$ are possible, thus allowing extrapolation. Far from convergence, extrapolation may yield erroneous steps. A closely related method allowing

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only interpolation, is the energy represented direct inversion in the iterative subspace (GEDIIS). (260 Based on a first-order energy expansion

$$
\begin{equation*}
E\left(\boldsymbol{R}^{*}\right)=E_{k}+\left(\boldsymbol{R}^{*}-\boldsymbol{R}_{k}\right)^{\top} \boldsymbol{g}_{k}, \tag{5.26}
\end{equation*}
$$

the GEDIIS energy expression is obtained by multiplying with $c_{i}$ and summing over $N$ points

$$
\begin{equation*}
E\left(\boldsymbol{R}^{*}\right)=\sum_{i=1}^{N} c_{i}\left[E\left(\boldsymbol{R}_{i}\right)+\sum_{j=1}^{N} c_{j} \boldsymbol{R}_{j} \boldsymbol{g}_{i}-\boldsymbol{R}_{i} \boldsymbol{g}_{i}\right] . \tag{5.27}
\end{equation*}
$$

Coefficients $c_{i}$ are obtained by direct minimization of eq. 5.27) under the constraint $0 \leq c_{i} \leq 1$. Compared to GDIIS, GEDIIS can be enabled earlier in an optimization and has been shown to result in smooth optimizations, with less erroneous, energy increasing steps. [260]

## 6 Internal Coordinates

Coordinate system choice greatly affects the outcome of geometry optimizations. The simplest choice are Cartesian coordinates, as they are, disregarding translation and rotation, unambiguously defined and always available. A big disadvantage, preventing efficient optimizations in Cartesians, is their strong coupling, exemplified in a strongly non-diagonal Hessian. 261 By employing an optimizer that utilizes Hessian information, the coupling can be taken into account to some degree. For this, the Hessian would have to be calculated, as it is not possible to estimate an approximate Hessian using Cartesian coordinates alone (see section 5.2).
An improved coordinate choice is given by normal mode coordinates (NMC), obtained from diagonalizing the mass-weighted Hessian. [262, 263] While keeping the big advantage of being unambiguously defined, they are orthogonal to each other and don't couple. Nonetheless, a Hessian is needed for their definition.

A third choice is given by redundant internal coordinates (RIC), composed of bond stretches, bends, linear bends and dihedrals. $232,239,264,266]$ In the following, the definition and usage of RIC for molecular optimizations is summarized. Many more coordinate systems have been proposed over the years, including the Z-Matrix 75], natural internal coordinates, 267, 269, delocalized internal coordinates (DLC), 270 hybrid DLC, (271) and translation-rotation internal coordinates, 272) from which only DLC will be briefly discussed.

### 6.1 Definition of Redundant Internal Coordinates

Redundant internal coordinates for a set of atoms are assigned, based on atomic connectivity. 232] See Figure 6.1 for an illustration, how common internal coordinates are defined. A bond stretch $q_{\mathrm{s}}$ between two atoms is assigned, when their distance is equal or smaller than the scaled sum of their covalent radii. A factor of 1.2 or 1.3 is usually employed for scaling. For every pair of bond stretches sharing an atom, a bend $q_{\mathrm{b}}$ is defined. Similarly, a dihedral $q_{\mathrm{d}}$ is defined for every consecutively bonded set of four atoms. Care has to be taken that no dihedral is defined, when three of the four atoms
a)


d)



Figure 6.1: Definition of commonly used primitive internal coordinates: a) bond bondstretch $q_{\mathrm{s}}, \mathrm{b}$ ) bend $q_{\mathrm{b}} \mathrm{c}$ ) linear bend $q_{\mathrm{lb}}$ and its orthogonal complement $q_{\mathrm{lb}, \mathrm{c}}$ d) dihedral $q_{d}$.
are (nearly) collinear, as then $q_{\mathrm{d}}$ becomes undefined. For a set of three (nearly) collinear atoms, a linear bend $q_{1 \mathrm{~b}}$ and its orthogonal complement can be defined. $273-275$. In contrast to regular bends, the second derivative of a linear bend, with respect to the Cartesian coordinates making it up, is defined even for collinear atomic arrangements. If several disconnected fragments are present, it has to be ensured that coordinates connecting them are defined.

Furthermore, additional coordinates can be defined, e.g., explicit hydrogen bonds, out-of-plane angles, [276] and/or auxiliary bond stretches that use an increased scaling factor for the summed covalent radii. 232

The choice of the initial scaling factor in the bond stretch definition and the different types of internal coordinates actually used, make their overall definition highly ambiguous. It was recognized that including too many (auxiliary) bond stretches is detrimental for the optimization outcome and hampers the back-transformation of a step in internal coordinates to Cartesian coordinates (vide infra). [232, 277]

### 6.2 Wilson's B-Matrix

The central quantity when working with internal coordinates, is the Wilson $\boldsymbol{B}$ matrix. It relates changes in Cartesian coordinates $\boldsymbol{R}$, to changes in internal coordinates $\boldsymbol{q} .232$, 278

$$
\begin{equation*}
B_{i j}=\frac{\partial q_{i}}{\partial R_{j}} \tag{6.1}
\end{equation*}
$$

Matrix $\boldsymbol{B}$ is rectangular, with one row per internal coordinate $q_{i}$ and one column per Cartesian coordinate $R_{j}$.

$$
\begin{gather*}
q_{1}\left(\begin{array}{c}
R_{1} \quad R_{2} \ldots \cdots R_{n} \\
q_{2} \\
\vdots \\
\vdots \\
\vdots \\
\vdots \\
\vdots \\
q_{n} \\
\ldots \ldots \ldots \ldots \ldots \ldots \\
\text { stretches } \\
\text { bends } \\
\text { dihedrals }
\end{array}\right) \tag{6.2}
\end{gather*}
$$

Explicit expressions for eq. (6.1) are given, e.g., by Wilson and Bakken. 232, 278 With $\boldsymbol{B}$, small Cartesian displacements are transformed to displacements in internal coordinates.

$$
\begin{equation*}
\boldsymbol{B} \delta \boldsymbol{R}=\delta \boldsymbol{q} \tag{6.3}
\end{equation*}
$$

As the matrix inverse is only defined for quadratic matrices, the pseudo-inverse $\boldsymbol{B}^{+}$is used for the inverse operation.

$$
\begin{equation*}
\delta \boldsymbol{R}=\boldsymbol{B}^{+} \delta \boldsymbol{q} \tag{6.4}
\end{equation*}
$$

$\boldsymbol{B}^{+}$is readily obtained from a SVD of $\boldsymbol{G}=\boldsymbol{B} \boldsymbol{B}^{\top}$ by inverting only singular values above a certain threshold.

$$
\begin{align*}
\boldsymbol{G} & =\boldsymbol{B} \boldsymbol{B}^{\top}=\boldsymbol{U} \boldsymbol{\Sigma} \boldsymbol{V}^{\top}  \tag{6.5}\\
\boldsymbol{B}^{+} & =\boldsymbol{B}^{\top}\left(\boldsymbol{V}^{\prime} \boldsymbol{\Sigma}^{\prime-1} \boldsymbol{U}^{\prime \top}\right) \tag{6.6}
\end{align*}
$$

The prime indicates that only singular values above a threshold and corresponding singular vectors are used for the inversion. A suitable threshold for the singular values is $3.16 \times 10^{-4}$, when $\boldsymbol{G}$ was calculated in atomic units (corresponding to a threshold of $1 \times 10^{-7}$ for the eigenvalues of $\left.\boldsymbol{G}\right) \cdot 264$

Similar to the coordinates, the Cartesian gradient $\boldsymbol{g}_{R}$ and the gradient in internal coordinates $\boldsymbol{g}_{q}$ are transformed.

$$
\begin{align*}
\boldsymbol{g}_{R} & =\boldsymbol{B}^{\top} \boldsymbol{g}_{q}  \tag{6.7}\\
\boldsymbol{g}_{q} & =\left(\boldsymbol{B}^{\top}\right)^{+} \boldsymbol{g}_{R} \tag{6.8}
\end{align*}
$$

Transforming the Cartesian Hessian $\boldsymbol{H}_{R}$ and the Hessian in internal coordinates $\boldsymbol{H}_{q}$ requires $\boldsymbol{g}_{q}$ and derivatives of $\boldsymbol{B}$, with respect to Cartesian coordinates.

$$
\begin{align*}
\boldsymbol{H}_{R} & =\boldsymbol{B}^{\boldsymbol{\top}} \boldsymbol{H}_{q} \boldsymbol{B}+\boldsymbol{K}  \tag{6.9}\\
\boldsymbol{H}_{q} & =\left(\boldsymbol{B}^{\boldsymbol{\top}}\right)^{+}\left(\boldsymbol{H}_{R}-\boldsymbol{K}\right) \boldsymbol{B}^{+}  \tag{6.10}\\
K_{j k} & =\sum_{i}\left[\boldsymbol{g}_{q}\right]_{i} B_{i j k}^{\prime}=\sum_{i}\left[\boldsymbol{g}_{q}\right]_{i} \frac{\partial^{2} q_{i}}{\partial R_{j} \partial R_{k}} \tag{6.11}
\end{align*}
$$

When the chosen set of internal coordinates is redundant, matrix $\boldsymbol{B}$ is rank-deficient and has linearly dependent rows. To ensure a step is only taken in the non-redundant subspace of $\boldsymbol{B}$, the gradient $\boldsymbol{g}_{q}$ is projected by $\boldsymbol{P}$.

$$
\begin{align*}
\boldsymbol{P} & =\boldsymbol{B} \boldsymbol{B}^{+}  \tag{6.12}\\
\widetilde{\boldsymbol{g}}_{q} & =\boldsymbol{P} \boldsymbol{g}_{q} \tag{6.13}
\end{align*}
$$

Additionally, the elements of $\boldsymbol{H}_{q}$ belonging to the redundant subspace ( $\boldsymbol{I}-\boldsymbol{P}$ ) orthogonal to $\boldsymbol{P}$, are shifted to high values $(\alpha=1000)$.

$$
\begin{equation*}
\widetilde{\boldsymbol{H}}_{q}=\boldsymbol{P} \boldsymbol{H}_{q} \boldsymbol{P}+\alpha(\boldsymbol{I}-\boldsymbol{P}) \tag{6.14}
\end{equation*}
$$

Constraints are easily implemented by means of a projector $\boldsymbol{P}^{\prime}$. 265 Given a diagonal matrix $\boldsymbol{C}$, with ones on the diagonal for the constrained primitives and zeros elsewhere, $\boldsymbol{P}$ can be modified to

$$
\begin{equation*}
\boldsymbol{P}^{\prime}=\boldsymbol{P}-\boldsymbol{P C}(\boldsymbol{C P C})^{-1} \boldsymbol{C P} . \tag{6.15}
\end{equation*}
$$

This approach even allows constraining Cartesian coordinates of (single) atoms, which may appear counterintuitive at first. But Cartesian coordinates are well supported in the framework of internal coordinates, as their first and second derivatives with respect to Cartesian coordinates are easily calculated ( $\left.\partial q_{i} / \partial R_{j}=\delta_{i j}, \partial^{2} q_{i} / \partial R_{j} \partial R_{k}=0\right)$. Alternatively, constraints can be implemented by means of Lagrange multipliers or penalty functions. 84, 279,281

### 6.3 Internal-Cartesian Back-Transformation

A step in internal coordinates has to be iteratively transformed to a step in Cartesian coordinates. 232 Starting from internals $\boldsymbol{q}_{0}$ and step $\Delta \boldsymbol{q}_{0}$, the desired target coordinates $\boldsymbol{q}_{T}=\boldsymbol{q}_{0}+\Delta \boldsymbol{q}_{0}$ are obtained from multiple evaluations of

$$
\begin{equation*}
\boldsymbol{R}_{l+1}=\boldsymbol{R}_{l}+\boldsymbol{B}^{+} \Delta \boldsymbol{q}_{l} \tag{6.16}
\end{equation*}
$$

as eq. (6.4) is only valid for small displacements $\delta \boldsymbol{q}$. Figure 6.2 shows the flowchart for the iterative back-transformation.


Figure 6.2: Flowchart for the iterative back-transformation of a desired step in internal coordinates $\Delta \boldsymbol{q}_{0}$, starting from Cartesian coordinates $\boldsymbol{R}_{0}$ and corresponding internals $\boldsymbol{q}_{0}$.

It has to be noted that not all possible sets of internal coordinates $\boldsymbol{q}$ have a valid Cartesian representation. The triangle equality states for every triangle that the summed length of two sides must be greater or equal to the third side. It is easy to propose three bond stretches between three atoms that violate the triangle equality, thus this set of internals can't have a Cartesian representation. 282] Given the restriction of eq. (6.4) to small displacements and the possibility of invalid internals, the back-transformation in

Figure 6.2 occasionally fails. In such cases, optimizations are continued with $\boldsymbol{R}_{1}$, the Cartesian coordinates obtained in the first cycle of the back-transformation. A more robust approach for the back-transformation using high-order derivatives of $\boldsymbol{B}$ based on automatic differentiation was proposed by Rybkin. 283

### 6.4 Delocalized Internal Coordinates

With a computationally cheap method to evaluate energies and gradients, handling RIC can become the computational bottleneck for big molecules. Especially the number of possible dihedrals grows immensely with the molecular size, leading to big Hessians and Wilson's B-matrices. Matrix diagonalization and inversion scale cubically with matrix size $\left(\mathcal{O}\left(n^{3}\right)\right)$. Both operations are needed, e.g., to calculate a RFO step in internal coordinates and back-transform it to Cartesians. Furthermore, RIC still show some coupling, even though it is greatly reduced compared with Cartesian coordinates.

Both problems are overcome with DLC. Diagonalizing $\boldsymbol{G}=\boldsymbol{B} \boldsymbol{B}^{\top}$ yields two sets of eigenvectors. For nonlinear molecules, a set of $3 N-6$ eigenvectors $\boldsymbol{U}$ with eigenvalues $\lambda>0$ spanning the non-redundant subspace of internals and the remaining eigenvectors $\boldsymbol{V}$ with vanishing eigenvalues spanning the redundant subspace, is obtained. 270]

$$
\boldsymbol{G}(\boldsymbol{U} \boldsymbol{V})=(\boldsymbol{U} \boldsymbol{V})\left(\begin{array}{ll}
\boldsymbol{\Sigma} & 0  \tag{6.17}\\
\mathbf{0} & 0
\end{array}\right)
$$

Eigenvectors $\boldsymbol{U}$ (active set) are used to obtain the active (DLC) coordinates $\boldsymbol{q}^{\mathrm{DLC}}$ and for transforming the original $\boldsymbol{B}$ matrix.

$$
\begin{align*}
\boldsymbol{q}^{\mathrm{DLC}} & =\boldsymbol{U}^{\top} \boldsymbol{q}  \tag{6.18}\\
\boldsymbol{B}^{\mathrm{DLC}} & =\boldsymbol{U}^{\top} \boldsymbol{B} \tag{6.19}
\end{align*}
$$

When substituting $\boldsymbol{B}^{\mathrm{DLC}}$ for $\boldsymbol{B}$ in eqs. (6.8), 6.11) and (6.16), the same equations can be used to transform Cartesian gradient and Hessian as well as for the iterative back-transformation. 270 For a nonlinear molecule of $N$ atoms with $m=3 N-6$ internal degrees of freedom, the DLC Hessian has size $(m \times m)$. Thus, Hessian diagonalization will be much faster in DLC compared to RIC for big $N$. As the eigenvectors $\boldsymbol{U}$ are orthogonal to each other, DLCb show no coupling, at least for the coordinates at which they were obtained.

DLCs have two drawbacks. First, they are complicated linear combinations of RIC and not easily interpretable. Second, they are only strictly orthogonal at the geometry
they were defined at. Consequently, DLC defined at the beginning may not be suitable in later stages of an optimization, requiring repeated diagonalization of $\boldsymbol{G}$, therefore defeating the purpose of avoiding expensive operations on big matrices. Usually $\boldsymbol{G}$ is only calculated once and the same set of eigenvectors $\boldsymbol{U}$ is used throughout an optimization.
For a set of small molecules, DLC and RIC showed similar performance with respect to required optimization cycles. [270] Choosing DLC/ may become favorable for big molecules, when matrix operation (diagonalization of $\boldsymbol{H}$, inversion of $\boldsymbol{B}$ ) become costly and the overall optimization time is considered.

## 7 Preconditioning

The optimization problem

$$
\begin{equation*}
\min _{\boldsymbol{R} \in \mathbb{R}^{n}} E(\boldsymbol{R}) \tag{7.1}
\end{equation*}
$$

is well conditioned, if its solution is not greatly perturbed by small changes of its input values. [234] Molecular optimizations are often ill-conditioned, as molecular degrees of freedom (e.g. bond stretches, dihedrals, interfragment stretches) have force constants of different magnitudes. While a small change of a bond length greatly affects the energy, a small rotation around a dihedral leaves the energy nearly unaffected.
As already discussed in section 5.1, $E(\boldsymbol{R})$ can be linearized and minimized by solving $\boldsymbol{H}_{k} \boldsymbol{p}_{k}=-\boldsymbol{g}_{k}$ for $\boldsymbol{p}_{k}$. Ill-conditioning makes $\boldsymbol{p}_{k}$ susceptible to small inaccuracies in $\boldsymbol{H}_{k}$ and $\boldsymbol{g}_{k}$, e.g., arising from numerical integration in DFT, impeding efficient optimizations. 284 Furthermore, step restriction may become difficult, as steps for the different internal coordinates may be of different magnitudes, preventing the use of a simple maximum value for all step elements.
The conditioning of a problem or a matrix is measured by its condition number $\kappa$. For a normal matrix e.g., the Hessian, $\kappa$ is defined as

$$
\begin{equation*}
\kappa=\frac{\left|\lambda_{\max }\right|}{\left|\lambda_{\min }\right|}, \tag{7.2}
\end{equation*}
$$

with $\lambda_{\max }\left(\lambda_{\min }\right)$ being the greatest (smallest) matrix eigenvalue. 285
Preconditioning can be seen as transformation to a set of coordinates $\boldsymbol{R}^{\prime}=\boldsymbol{P}^{1 / 2} \boldsymbol{R}$, where the corresponding Hessian has an improved eigenvalue distribution $\left\{\lambda_{i}^{\prime}\right\}$, through a preconditioner matrix $\boldsymbol{P} .286]$ The eigenvalues of the preconditioned Hessian $\left\{\lambda_{i}^{\prime}\right\}$ are defined as

$$
\begin{equation*}
\lambda_{i}^{\prime}=\frac{\boldsymbol{u}_{i}^{\top} \boldsymbol{H}_{k} \boldsymbol{u}_{i}}{\boldsymbol{u}_{i}^{\top} \boldsymbol{P}_{k} \boldsymbol{u}_{i}} \tag{7.3}
\end{equation*}
$$

with $\boldsymbol{u}_{i}$ denoting the $i$-th eigenvector of the unconditioned Hessian $\boldsymbol{H}_{k}$. Well-conditioned optimizations are expected to converge much faster, compared to ill-conditioned optimizations. 234, 287 The effect of preconditioning for a quadratic potential is shown in Figure 7.1.

[^1]

Figure 7.1: Effect of preconditioning for an ill conditioned quadratic potential obtained from $\boldsymbol{x}^{\top} \boldsymbol{A} \boldsymbol{x}$ with $\boldsymbol{A}=\left[\begin{array}{ll}63 & 62 \\ 62 & 63\end{array}\right]$ (eigenvalues $\lambda_{1}=125, \lambda_{2}=1$ ) and $\boldsymbol{x}=\left[\begin{array}{l}x_{1} \\ x_{2}\end{array}\right]$.
Left: Original problem. Progress along the valley towards the minimizer requires large step sizes, as the potential is shallow, but small steps perpendicular to this direction lead to steep potential increases, making step scaling difficult. Right: Preconditioned problem with optimal condition number $\kappa=1$, allowing easy step scaling, as the potential is isotropic. In the present case the "optimal" preconditioner $\boldsymbol{P}=\boldsymbol{A}^{-1}$ was used, which is usually not available in real applications, e.g., searches for SP; on molecular PESs.

Optimizations are then carried out in the new coordinate system, e.g., using the method of steepest descent.

$$
\begin{align*}
\boldsymbol{R}_{k+1}^{\prime} & =\boldsymbol{R}_{k}^{\prime}-\alpha_{k} \boldsymbol{g}_{k}^{\prime}  \tag{7.4}\\
\boldsymbol{P}^{1 / 2} \boldsymbol{R}_{k+1} & =\boldsymbol{P}^{1 / 2} \boldsymbol{R}_{k}-\alpha_{k} \boldsymbol{P}^{-1 / 2} \boldsymbol{g}_{k} \tag{7.5}
\end{align*}
$$

Back-transformation of eq. 7.5 to the original coordinate system recovers the equation for a quasi-Newton step, as already given in eq. (5.9). In this sense, steepest descent in the transformed coordinate system and a QN in the original coordinate system are identical, when $\boldsymbol{P}=\boldsymbol{A}_{k}$.

$$
\begin{equation*}
\boldsymbol{R}_{k+1}=\boldsymbol{R}_{k}-\alpha_{k} \boldsymbol{P}^{-1} \boldsymbol{g}_{k} \tag{7.6}
\end{equation*}
$$

For $\boldsymbol{P}=\boldsymbol{I}$ the steepest descent method in the original coordinates $\boldsymbol{R}$ is recovered, and for $\boldsymbol{P}=\boldsymbol{H}_{k}$, eq. 7.6 yields the original Newton step. 286 Similarly, preconditioning can
be applied to a QN method, e.g., the LBFGS algorithm (see line 5 in Algorithm 1 on page 35), for improved optimization performance compared to preconditioned steepest descent. 234, 288]

A useful preconditioner $\boldsymbol{P}$ must fulfill several requirements: a) it must be easy to construct, b) cheap to invert, e.g., by exploiting sparsity, and c) must be positive definite to ensure a descent direction is obtained, see eq. (5.4). 287

Mones et al. proposed a force field based scheme to construct effective preconditioners $\boldsymbol{P}$ for molecular systems, fulfilling the above requirements. 286 Starting from a surrogate force field potential $V_{\mathrm{FF}}$

$$
\begin{equation*}
V_{\mathrm{FF}}=\sum_{i} V_{i}\left(q_{i}(\boldsymbol{R})\right) \tag{7.7}
\end{equation*}
$$

comprising individual potential energy terms $V_{i}$, e.g., quadratic potentials

$$
\begin{equation*}
V_{i, \text { Quadratic }}=\frac{1}{2} k_{i}\left(q_{i}-q_{i, 0}\right)^{2} \tag{7.8}
\end{equation*}
$$

the corresponding Hessian

$$
\begin{equation*}
\boldsymbol{H}_{\mathrm{FF}}=\frac{\partial^{2} V_{\mathrm{FF}}}{\partial \boldsymbol{R}^{2}}=\sum_{i} \frac{\partial^{2} V_{i}}{\partial \boldsymbol{R}^{2}}=\sum_{i} \boldsymbol{H}_{i} \tag{7.9}
\end{equation*}
$$

is calculated. When long-range interactions are neglected while constructing $V_{\mathrm{FF}}$, e.g., only bonding interactions are considered, $\boldsymbol{H}_{\mathrm{FF}}$ will be sparse and is easily stored and inverted. Depending on $\boldsymbol{R}$ and the actual potential energy terms $\left\{V_{i}\right\}, \boldsymbol{H}_{\mathrm{FF}}$ may not be positive definite. Mones suggested modifying the local Hessian contributions to ensure overall positivity. 286

$$
\begin{equation*}
\boldsymbol{H}_{i}=\frac{\partial^{2} V_{i}}{\partial \boldsymbol{R}^{2}}=\underbrace{\frac{\partial q_{i}}{\partial \boldsymbol{R}} \otimes \frac{\partial q_{i}}{\partial \boldsymbol{R}} \frac{\partial^{2} V_{i}}{\partial q_{i}^{2}}}_{\boldsymbol{H}_{i}^{(1)}}+\underbrace{\frac{\partial^{2} q_{i}}{\partial \boldsymbol{R}^{2}} \frac{\partial V_{i}}{\partial q_{i}}}_{\boldsymbol{H}_{i}^{(2)}} \tag{7.10}
\end{equation*}
$$

For quadratic potentials $\partial^{2} V_{i} / \partial q_{i}^{2}=k_{i}>0$, so $\boldsymbol{H}_{i}^{(1)}$ will be positive-semi-definite, whereas the definiteness of $\boldsymbol{H}_{i}^{(2)}$ is ambiguous. By dropping the second term $\boldsymbol{H}_{i}^{(2)}$ and using the absolute value of $\partial^{2} V_{i} / \partial q_{i}^{2}$, positive-definiteness of $\boldsymbol{P}$ is ensured even for non-quadratic potentials and the following, general formula for $\boldsymbol{P}$ is obtained:

$$
\begin{equation*}
\boldsymbol{P}=\sum_{i} \tilde{\boldsymbol{H}}_{i}^{(1)}=\frac{\partial q_{i}}{\partial \boldsymbol{R}} \otimes \frac{\partial q_{i}}{\partial \boldsymbol{R}}\left|\frac{\partial^{2} V_{i}}{\partial q_{i}^{2}}\right| \tag{7.11}
\end{equation*}
$$

The $\partial q_{i} / \partial \boldsymbol{R}$ terms are elements of Wilson's B-Matrix discussed in section 6.2 and are easily computed. A particular attractive, black-box way, for calculating $\partial^{2} V_{i} / \partial q_{i}^{2}$ was given by Lindh. 241

$$
\begin{equation*}
V_{\text {Lindh }}(\boldsymbol{q})=V_{\text {linear }}(\boldsymbol{q})+V_{\text {stretch }}(\boldsymbol{q})+V_{\text {bend }}(\boldsymbol{q})+V_{\text {dihedral }}(\boldsymbol{q}) \tag{7.12}
\end{equation*}
$$

The linear term assumes the availability of energy and gradient at a newly computed point $\boldsymbol{q}_{0}$.

$$
\begin{equation*}
V_{\text {linear }}(\boldsymbol{q})=V_{\text {Lindh }}\left(\boldsymbol{q}_{0}\right)+\left(\boldsymbol{q}-\boldsymbol{q}_{0}\right) \frac{\partial V_{\text {Lindh }}\left(\boldsymbol{q}_{0}\right)}{\partial \boldsymbol{q}} \tag{7.13}
\end{equation*}
$$

Quadratic expressions are used for the remaining terms.

$$
\begin{align*}
V_{\text {stretch }} & =\frac{1}{2} \sum_{j>i} k_{i j}\left(q_{i j}-q_{0, i j}\right)^{2}  \tag{7.14}\\
V_{\text {bend }} & =\frac{1}{2} \sum_{k>j>i} k_{i j k}\left(q_{i j k}-q_{0, i j k}\right)^{2}  \tag{7.15}\\
V_{\text {dihedral }} & =\frac{1}{2} \sum_{l>k>j>i} k_{i j k l}\left(q_{i j k l}-q_{0, i j k l}\right)^{2} \tag{7.16}
\end{align*}
$$

Depending on the number of subscript indices, different internal coordinates are distinguished (2: stretches, 3: bends, 4: dihedrals). Lindh suggested to use all possible terms, besides repeated and reversed index combinations. Force constants $k$ are calculated from the equations below.

$$
\begin{align*}
\rho_{i j} & =\exp \left(\alpha_{i j}\left(q_{\mathrm{ref}, i j}^{2}-q_{i j}^{2}\right)\right)  \tag{7.17}\\
k_{i j} & =k_{s} \rho_{i j}  \tag{7.18}\\
k_{i j k} & =k_{b} \rho_{i j} \rho_{j k}  \tag{7.19}\\
k_{i j k l} & =k_{d} \rho_{i j} \rho_{j k} \rho_{k l} \tag{7.20}
\end{align*}
$$

Parameters $\left\{\alpha_{i j}, q_{\mathrm{ref}, i j}, k_{s}, k_{b}, k_{d}\right\}$, obtained from calculations on water and hydroxysulphane at the HF/STO-3G level of theory, are given in Table 7.1. Figure 7.2 shows the effect of preconditioning using the Lindh model Hessian for 2-pentene. Improved preconditioners may be estimated from more sophisticated force fields like Amber, 32] CHARMM [33], or the universal force field (UFF), 289] at the cost of having to provide a suitable molecular topology to the force field engine.


Figure 7.2: Effect of preconditioning on force constants and condition number $\kappa$. a) Unconditioned Cartesian Hessian and b) preconditioned Cartesian Hessian for 2-pentene, obtained at the HF/def2-SVP level of theory. The preconditioner was obtained as outlined in [286], using the Lindh model Hessian (vide infra). 241]. Preconditioning leads to a markedly improved eigenvalue distribution and greatly reduced condition number $\kappa_{P}$. The six smallest eigenvalues belonging to the eigenvectors describing translational and rotational motions have been removed.

Table 7.1: Lindh model Hessian parameters in atomic units for the first three periods of the periodic table. The parameter for an atom pair depends entirely on their periods $i, j$ in the periodic table. Force constants $k_{s}=0.45, k_{b}=0.15$ and $k_{d}=0.005$ are employed in eqs. (7.18) to (7.20). (241]


## 8 Chain-of-States Methods

Computational studies of reaction mechanism are conducted by locating SP; on the PES of the reactants and the MEPs connecting them. Whereas educts and products of a reaction may be known beforehand and are easily optimized, the TSß connecting them are often less accessible. Unfortunately, the comprehensive description of reaction energetics requires the TS, as it is essential for the prediction of reaction barriers.

Optimizing TS/ poses several challenges: Most methods like PRFO 290 or the image method (IM) 256 require a computationally demanding Hessian calculation and the Hessian must have at least one significant negative eigenvalue with a corresponding eigenvector, suitable for the description of the reaction under study. Costly Hessian calculations may be sped up by employing techniques like resolution of identity (RI) 291, 292, however, providing a good initial guess geometry is far more difficult. Often, guesses for TSk are constructed manually, nevertheless such TS guesses may yield Hessians with wrong eigenvalue structure, or without a suitable eigenvector to describe the reaction coordinate, resulting in a laborious process of trial and error. Suitable TS guesses for further refinement can be obtained from COS methods.
In COS methods, a MEP is discretized by a set of tangents $\left\{\boldsymbol{\tau}_{i}\right\}$ defined at a set of molecular geometries (images) given by coordinates $\left\{\boldsymbol{R}_{i}\right\}$. The simplest tangent is given by the normalized distance vector between two adjacent images

$$
\begin{equation*}
\boldsymbol{\tau}_{i}=\frac{\boldsymbol{R}_{i+1}-\boldsymbol{R}_{i}}{\left|\boldsymbol{R}_{i+1}-\boldsymbol{R}_{i}\right|} \tag{8.1}
\end{equation*}
$$

but often a more sophisticated definition like an upwinding tangent is used. 72, 293, 294 A COS is usually spanned between two pre-optimized minima on the PES, while passing through a point close to the TS, (73)

## Parts of this chapter are based on:

Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: 10.1002/qua 26390

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## 8 Chain-of-States Methods

The initial path is iteratively refined until it coincides with a MEP, or represents a sufficient approximation to it. An image $i$ is considered converged to the MEP when the perpendicular component $\boldsymbol{f}_{i}^{\perp}$ of the force $\boldsymbol{f}_{i}$ acting on it vanishes or is below a prescribed threshold

$$
\begin{equation*}
\boldsymbol{f}_{i}^{\perp}=\boldsymbol{f}_{i}-\left(\boldsymbol{f}_{i} \cdot \boldsymbol{\tau}_{i}\right) \boldsymbol{\tau}_{i} \approx \mathbf{0} \tag{8.2}
\end{equation*}
$$

that is, the remaining force $\boldsymbol{f}_{i}$ acts only along the tangent $\boldsymbol{\tau}_{i}$.
Common examples for COS methods are the NEB and the growing string method (GSM). 71, 295 Both methods differ in how the image distribution along the path is achieved and in their initial path setup (see section 8.2).

### 8.1 Nudged Elastic Band

The NEB method introduces artificial spring forces, acting parallel to the path

$$
\begin{equation*}
\boldsymbol{f}_{i}^{\|}=k_{i}\left(\left|\boldsymbol{R}_{i+1}-\boldsymbol{R}_{i}\right|-\left|\boldsymbol{R}_{i-1}-\boldsymbol{R}_{i}\right|\right) \boldsymbol{\tau}_{i}, \tag{8.3}
\end{equation*}
$$

so the total force $\boldsymbol{f}_{i}^{\mathrm{NEB}}$ acting on image $i$ is given by

$$
\begin{equation*}
\boldsymbol{f}_{i}^{\mathrm{NEB}}=\boldsymbol{f}_{i}^{\perp}+\boldsymbol{f}_{i}^{\|} . \tag{8.4}
\end{equation*}
$$

If the same spring constant $k_{i}$ is used throughout all images, an equal image distribution along the band is achieved, as $\boldsymbol{f}_{i}^{\|}$vanishes for equidistant images.

### 8.2 Interpolating Initial Paths

An initial COS path is generated by interpolation between two minima. The quality of the initial path is crucial for the successful outcome of COS optimizations. 296] Given two images $\boldsymbol{R}_{0}$ and $\boldsymbol{R}_{N-1}$, intermediate images $\boldsymbol{R}_{i}$ are linearly interpolated as

$$
\begin{equation*}
\boldsymbol{R}_{i}=\boldsymbol{R}_{0}+i \frac{\boldsymbol{R}_{N-1}-\boldsymbol{R}_{0}}{N-1} \tag{8.5}
\end{equation*}
$$

with $N$ denoting the desired total number of images in the path and $i$ is running from 0 to $N-1$. Linear interpolation in Cartesian coordinates often yields poor paths, as the curvilinear nature of bends and dihedrals is neglected, leading to artificially compressed bonds and frequent atom clashes.

Improved paths are obtained from interpolating atomic distance matrices $\boldsymbol{D}_{0}$ and $\boldsymbol{D}_{N-1}$, defined at $\boldsymbol{R}_{0}$ and $\boldsymbol{R}_{N-1}$. The distance between atoms $j$ and $k$ for Cartesian coordinates $\boldsymbol{R}$ is

$$
\begin{equation*}
D_{j k}=\sqrt{\sum_{\sigma}\left(R_{j \sigma}-R_{k \sigma}\right)^{2}} \tag{8.6}
\end{equation*}
$$

with $\sigma$ running over the Cartesian axes $x, y, z$. Analogous to eq. (8.5), interpolated distance matrices are obtained as

$$
\begin{equation*}
\boldsymbol{D}_{i}=\boldsymbol{D}_{0}+i \frac{\boldsymbol{D}_{N-1}-\boldsymbol{D}_{0}}{N-1} \tag{8.7}
\end{equation*}
$$

Typically, there are many more atomic distances than internal degrees of freedom, so a desired distance matrix $\boldsymbol{D}_{i}$ can only be approximated by a set of Cartesian coordinates in a least-squares sense. Halgren and Lispcomb proposed the linear synchronous transit (LST) method, to determine a suitable set of Cartesian coordinates via minimization of an objective function $S^{\text {LST }} \cdot 297,299$

$$
\begin{equation*}
S_{i}^{\mathrm{LST}}\left(\boldsymbol{R}_{i}\right)=\underbrace{\sum_{k>j} \frac{\left(D_{j k}-D_{j k}^{(i)}\right)^{2}}{\left(D_{j k}^{(i)}\right)^{4}}}_{S^{\mathrm{IDPP}}}+\underbrace{10^{-6} \sum_{j, \sigma}\left(R_{i, j \sigma}-R_{i, j \sigma}^{(i)}\right)^{2}}_{S_{\mathrm{TR}}} \tag{8.8}
\end{equation*}
$$

Indices $j$ and $k$ denote atoms. Terms $\boldsymbol{D}^{(i)}$ and $\boldsymbol{R}_{i}^{(i)}$, determined from eq. 8.7 and eq. 8.5, remain fixed throughout the minimization of $S_{i}^{\text {LST }}$. The small term $S_{\mathrm{TR}}$ in eq. 8.8 was introduced to penalize excessive translation and rotation, with respect to a reference geometry. Full LST paths are obtained by subsequent minimizations of $S_{i}^{\text {LST }}$ for all intermediate images $(0<i<N-1)$.

The LST approach was improved by Smidstrup in the image dependent pair potential (IDPP) method. By only using the first term $S^{\mathrm{IDPP}}$ from eq. 8.8) and optimizing all images simultaneously as a NEB, overall convergence is improved. 296]

While LST and IDPP usually greatly improve upon naive linear interpolation, both methods can yield discontinuous paths, as the minimization(s) may fail to converge. The computational effort to evaluate the objective function $S$ and its gradient $\nabla S$ is usually negligible, compared to the actual electronic structure calculations in the subsequent COS optimizations.

If it is feasible to define internal coordinates at $\boldsymbol{R}_{0}$ and $\boldsymbol{R}_{N-1}$, then linear interpolation in DLC is the preferred interpolation method, as the curvilinear nature of bends and dihedrals is correctly taken into account, while lengthy minimization as in LST and


Figure 8.1: Different interpolated paths for rotation around the central dihedral in $\mathrm{CH}_{2} \mathrm{BrCHFCl}$. a) Initial and final geometries, b) linearly interpolated Cartesians and c) linearly interpolated DLC/. Linearly interpolated Cartesians yield artificially compressed bond lengths, as the curvilinear nature of dihedrals is neglected, while linearly interpolated DLC take it correctly into account.

IDPP are avoided. 300 A discussion of the different interpolation algorithms is given by Zhu. 301 Figure 8.1 illustrates the differences between linear interpolation in Cartesian and internal coordinates.

### 8.3 Growing String Method

If none of the former interpolation methods yields a reasonable initial path, a growing COS method like the GSM can be used. 300 In this case, initial construction of a full path is avoided and new images are added on-the-fly, when sufficient convergence is achieved on the innermost images of the two disconnected sub-strings. In string methods (SM;), a prescribed image distribution along the string is achieved by periodical reparametrization. Depending on the employed coordinate system of the string images, a cubic spline is used for Cartesian coordinates and linear interpolation is done for delocalized internal coordinates. [300, 302] No spring constants are needed in the string method, and the forces acting on the images in a string are given by eq. 8.2).

### 8.4 NEB and String Variants

When the initial path is poor, tight convergence is required or many images are employed, COS optimizations can become computationally very costly, as one gradient evaluation per image is needed in every optimization cycle. To this end, several COS variants have been developed.
If the goal is to obtain a suitable TS guess, there is no need to tightly converge images far from the HEI Rather, it may be beneficial, to increase the resolution around the HEI by zooming-in around it.

Originally proposed by Maragakis, this is implemented in the adaptive NEB (ANEB) variant. 303 The full NEB is optimized until a looser convergence threshold is fulfilled, then all but the energetically highest lying images are dropped and new images are interpolated between. This strategy can be applied recursively, zooming in further and further, until the ANEB is converged. Maragakis reported computational savings of up to $60 \%$. 303 An illustration of the ANEB method is given in Figure 8.2 .


Figure 8.2: Illustration of NEB variants, aiming at reducing the computational costs of NEB optimizations. The original NEB has poor resolution around the presumed TS and features a long tail with only minor energetic differences between the images. a) Black inset in the upper part: adaptive NEB. Only the highest energy images are kept and new images are interpolated in between, leading to increased resolution around the TS, b) Grey inset in the lower part: free-end NEB. Images in the long tail are cut off and only the previous images are relaxed further. The frontier image is constrained to a contour line of the PES,

Similarly, if the NEB exhibits a long tail of energetically close images, computational resources may be wasted. Zhu suggested the free-end NEB (FENEB) variant, where the path is cut off after a selected image and the image is constrained to an isocontour line of the energy. 304 Both variants, ANEB and FENEB were later combined by Zhang into to free-end adaptive NEB (FEANEB) method. 305]

### 8.5 Optimizing Chain-Of-States

As briefly mentioned in section 8.1, COS paths are optimized by minimizing $f^{\perp}$. Many different algorithms have been employed for COS optimizations, with CG and LBFGS appearing most promising (see section 5.3). 306, 307]. Sheppard and Henkelman recognized a global version of LBFGS as especially effective. 306 By optimizing the whole COS as one super-molecule with LBFGS, the inter-image coupling via the tangents $\tau$ is included to some degree.

It was shown by Melander and Herbol that removal of translational and rotational degrees of freedom, either via quaternions or the Kabsch algorithm, is crucial for the success of COS optimizations. 307, 308

### 8.6 Climbing Image

Improved TS guesses can be obtained by employing either one or two climbing images (CIs). 309, 310 After a few optimization cycles, the HEI can be converted to a CI. When two CIs are desired, the two images bracketing the HEI are converted. The force acting on CI $i$ is reversed along its tangent $\boldsymbol{\tau}_{i}$, so a CI moves uphill along the COStpath and eventually converges to the true TS,

$$
\begin{equation*}
\boldsymbol{f}_{i}^{\mathrm{CI}}=\boldsymbol{f}_{i}-2\left(\boldsymbol{f}_{i}^{\top} \boldsymbol{\tau}_{i}\right) \boldsymbol{\tau}_{i} \tag{8.9}
\end{equation*}
$$

### 8.7 Highest Energy Image

Given a converged COS, its HEI can be selected and used as guess in a subsequent TS optimization. If the chosen TSłoptimizer employs Hessian information and the Hessian has multiple negative eigenvalues, the HEI tangent can be used to select the initial eigenvector to follow uphill by an overlap criterion.

Starting TS searches in internal coordinates from a previous COS calculation also provides an additional benefit that is easily overlooked: By considering the union of internal coordinates from the first and last image in a COS an improved set of internal coordinates can be generated for the TS search, compared to automated procedures that only consider the molecular geometry at the TS guess. As broken or not yet formed bonds are common in TS resulting in an artificial restriction of the geometries degrees of freedom.

Overall, the choice of the coordinate system is extremely important for the success of (TS) optimizations, as can also be seen from Table III in ref. 270. Depending on the chosen coordinate system, a TS guess may or may not have an imaginary frequency.

## 9 Obtaining Transition States on Potential Energy Surfaces

Given a suitable guess, e.g., the HEI from a COS optimization, TS are obtained most efficiently by optimizers utilizing Hessian information. In contrast to minimizations, where the energy is minimized along all Hessian eigenvectors, the energy along one mode (transition vector) is maximized, when searching for first-order saddle points.
Suitable steps for TS optimizations are obtained from the PRFO method, where two eigenvalue equations similar to eq. (5.7) are solved. By transforming gradient $\boldsymbol{g}_{k}$ and Hessian $\boldsymbol{H}_{k}$ into the eigenvector basis of the latter, the two eigenvalue equations are

$$
\begin{align*}
{\left[\begin{array}{cc}
\widetilde{H}_{k, 11} & \widetilde{g}_{k, 1} \\
\widetilde{g}_{k, 1} & 0
\end{array}\right]\left[\begin{array}{c}
\widetilde{p}_{k, 1} \\
1
\end{array}\right] } & =\nu_{\max }\left[\begin{array}{c}
\widetilde{p}_{k, 1} \\
1
\end{array}\right]  \tag{9.1}\\
{\left[\begin{array}{cccc}
\widetilde{H}_{k, 22} & & 0 & \widetilde{g}_{k, 2} \\
& \ddots & & \vdots \\
0 & & \widetilde{H}_{k, n n} & \widetilde{g}_{k, n} \\
\widetilde{g}_{k, 2} & \cdots & \widetilde{g}_{k, n} & 0
\end{array}\right]\left[\begin{array}{c}
\widetilde{p}_{k, 2} \\
\vdots \\
\widetilde{p}_{k, n} \\
1
\end{array}\right] } & =\nu_{\min }\left[\begin{array}{c}
\widetilde{p}_{k, 2} \\
\vdots \\
\widetilde{p}_{k, n} \\
1
\end{array}\right], \tag{9.2}
\end{align*}
$$

with the tilde denoting transformed quantities and assuming the energy is to be maximized along eigenvector 1 . Whereas the eigenvector with the smallest eigenvalue is scaled in eq. (9.2), the eigenvector belonging to the largest eigenvalue is scaled in eq. (9.1). As both eigenvalue equations are solved independently, PRFO neglects the coupling between both subspaces.
This problem is solved by the image method (IM). Given the Taylor expansion $E(\boldsymbol{R})$ in eq. 55.2), an image function $\bar{E}(\boldsymbol{R})$ is constructed in a way, such that the TS of $E(\boldsymbol{R})$

## Parts of this chapter are based on:

Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: 10 . 1002 /qua 26390

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coincide with the minima of $\bar{E}(\boldsymbol{R}) \cdot 232,256$ By applying a Householder transformation

$$
\begin{equation*}
\boldsymbol{P}=\boldsymbol{I}-2 v_{i} v_{i}^{\top} \tag{9.3}
\end{equation*}
$$

with identity matrix $\boldsymbol{I}$ and transition vector $v_{i}$, to gradient and Hessian at cycle $k$

$$
\begin{align*}
\overline{\boldsymbol{g}}_{k} & =\boldsymbol{P} \boldsymbol{g}_{k}  \tag{9.4}\\
\overline{\boldsymbol{H}}_{k} & =\boldsymbol{P} \boldsymbol{H}_{k} \tag{9.5}
\end{align*}
$$

the image function $\bar{E}(\boldsymbol{R})$ is defined as

$$
\begin{equation*}
\bar{E}\left(\boldsymbol{R}_{k}+\boldsymbol{p}_{k}\right)=E_{k}+\boldsymbol{p}_{k}^{\top} \overline{\boldsymbol{g}}_{k}+\frac{1}{2} \boldsymbol{p}_{k}^{\top} \overline{\boldsymbol{H}}_{k} \boldsymbol{p}_{k} \tag{9.7}
\end{equation*}
$$

Now minima of $\bar{E}(\boldsymbol{R})$ can be obtained according to the approach outlined in section 5.1 using RFO, while correctly taking into account the coupling between both subspaces.

## 10 Dimer Method

When Hessian calculations become computationally infeasible, e.g., for large systems, or when analytical Hessian implementations are unavailable, TSF can be obtained by means of the dimer method (DM). 311] As it only utilizes first derivatives, any unfavorable $\mathcal{O}\left(N^{3}\right)$-scaling matrix operations like diagonalization or inversion are avoided.


Figure 10.1: Illustration of the DM: a) The dimer comprises two images, $\boldsymbol{R}_{1}$ and $\boldsymbol{R}_{2}$, displaced by $\Delta R$ from a common midpoint $\boldsymbol{R}_{0}$ along $\boldsymbol{N}$ (dimer orientation). The rotational force $\boldsymbol{f}^{\perp}$ acts on image $\boldsymbol{R}_{1}$. Midpoint $\boldsymbol{R}_{0}$ is displaced uphill, against the parallel force $\boldsymbol{f}^{\dagger}$, towards the TS, b) One dimer rotation step: images $\boldsymbol{R}_{1}$ and $\boldsymbol{R}_{2}$ are rotated around $\boldsymbol{R}_{0}$ towards the lowest curvature mode, indicated by a dashed line.

A dimer consists of two images $\boldsymbol{R}_{1}$ and $\boldsymbol{R}_{2}$, displaced into opposite directions by $\Delta R$ from a common midpoint $\boldsymbol{R}_{0}$, along an orientation vector $\boldsymbol{N}$. The dimer is optimized towards a TS by repeated application of two steps: rotation and translation. Dimer rotation is achieved by minimizing the rotational force

$$
\begin{equation*}
\boldsymbol{f}_{R}=\left(\boldsymbol{f}_{1}-\boldsymbol{f}_{2}\right)-\left(\left(\boldsymbol{f}_{1}-\boldsymbol{f}_{2}\right) \boldsymbol{N}\right) \boldsymbol{N} \tag{10.1}
\end{equation*}
$$

and aims at bringing its orientation $\boldsymbol{N}$ into maximum coincidence with the lowest curvature mode of the system, thus approximating the imaginary mode to follow uphill.

## 10 Dimer Method

Several approaches for minimization of $\boldsymbol{f}_{R}$ have been proposed, ranging from direct minimization through LBFGS 312, the constrained Broyden method 313], or direct inversion in the iterative subspace (DIIS) [314], to rotations based on a Fourier series expansion of the curvature $C .290$ The curvature is calculated numerically as

$$
\begin{equation*}
C=\frac{\left(f_{2}-f_{1}\right) \boldsymbol{N}}{2 \Delta \boldsymbol{R}} . \tag{10.2}
\end{equation*}
$$

Dimer translation moves it closer towards the TS, Similar to climbing images in COS optimizations, a modified force $\boldsymbol{f}^{\dagger}$ is employed: In the convex region of positive curvature ( $C>0$ ), the dimer is translated using only the reversed force component along $N$, whereas in regions of negative curvature $(C<0)$ the perpendicular component of $\boldsymbol{f}_{0}$ is minimized simultaneously

$$
\boldsymbol{f}^{\dagger}= \begin{cases}-\left(\boldsymbol{f}_{0}^{\top} \boldsymbol{N}\right) \boldsymbol{N} & C>0  \tag{10.3}\\ \boldsymbol{f}_{0}-2\left(\boldsymbol{f}_{0}^{\top} \boldsymbol{N}\right) \boldsymbol{N} & C<0\end{cases}
$$

While Hessian-based TS-optimizers require only one gradient evaluation per optimization cycle, multiple evaluations are needed for the DM, as repeated rotations may be necessary. Kästner reported an average of 2.81 gradient calculations per dimer cycle for converging the Baker-TS benchmark set. [312, 315] Depending on the system, increased costs of additional gradient evaluations may be offset by not having to calculate the Hessian. In a comparison between PRFO and DM, 290 Heyden et al. determined the DM to be computationally more efficient for systems consisting of 1.5 X atoms, when the exact Hessian is calculated every $X$ cycles in the $\overline{\text { PRFO method }}{ }^{1}$

[^2]
## 11 Intrinsic Reaction Coordinate

Given a successfully optimized TS, it may not be obvious which reactants it connects on the PES, e.g., just by examining the imaginary mode, or when the optimized TS is quite different from the initial guess structure. Actually connected reactants can be obtained by integrating a MEP, originating from the TS,

An uniquely defined MEP is the IRC, [316] the path of steepest descent in mass-weighted coordinates $\boldsymbol{x}$, along reaction coordinate $s$ :

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}(s)}{\mathrm{d} s}=-\frac{\boldsymbol{g}(\boldsymbol{x})}{|\boldsymbol{g}(\boldsymbol{x})|} \tag{11.1}
\end{equation*}
$$

Starting at the educts with $s=-\infty$, the reaction coordinate $s$ passes the TS at $s=0$ and becomes $s=+\infty$ for the products. Calculating an IRC can also be regarded as integrating Newton's equations of motion with fully damped kinetic energy, yielding an infinitely slow moving, imaginary minimum energy trajectory. 317 The resulting IRC does not exhibit any complicated rotational or vibrational motions and is easily interpreted. 69, 318 As the kinetic energy is fully neglected, dynamical effects like PESłbifurcations are not captured by an IRC, 318, 319] although multiple techniques have been developed to capture such events along an IRC, $320-322$

The TS is a stationary point with vanishing gradient, therefore eq. (11.1) cannot be used to define a direction of steepest descent at the TS. An initial step towards a geometry with non-vanishing gradient is done by displacing the TS along its imaginary mode (transition vector) by a prescribed length $\Delta x$. Alternatively, $\Delta x$ can be calculated from a quadratic potential, by requiring a certain energy lowering $\Delta E$ (e.g. $5 \times 10^{-4} E_{\mathrm{h}}$ )

$$
\begin{equation*}
\Delta E=\frac{1}{2} q \Delta x^{2} \tag{11.2}
\end{equation*}
$$

with $q$ being the force constant of the transition vector.

## Parts of this chapter are based on:

Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: 10.1002/qua. 26390

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## 11 Intrinsic Reaction Coordinate

Eq. (11.1) is a stiff differential equation (see chapter 7), making its accurate integration difficult. $323-325$ Algorithms to solve eq. (11.1) can be broadly categorized in explicit and implicit methods. Explicit methods require only evaluation of quantities (e.g. energy and gradient) at the initial point, whereas implicit methods also require quantities evaluated at intermediate points of the integration. 325 The simplest explicit scheme is the Euler method

$$
\begin{equation*}
\boldsymbol{x}_{k+1}=\boldsymbol{x}_{k}-\alpha \boldsymbol{g} . \tag{11.3}
\end{equation*}
$$

For stiff problems, the Euler method allows only small integration lengths, before the accuracy degrades. Implicit methods allow for greater integration step lengths at increased computational costs, as multiple energy and gradient evaluations are needed in every IRC cycle. 326, 327] Two examples for implicit methods will be briefly discussed in the following. Historical surveys of different integration algorithms are given by Melissas, Deng and Gonzalez, 328 [330 while a more recent overview is presented by Maeda. 318]

### 11.1 Gonzalez-Schlegel 2nd-Order Algorithm

A widely implemented IRC integrator, utilizing Hessian information and sustaining longer step lengths was proposed by Gonzalez and Schlegel. [83, 331 Starting at point $\boldsymbol{x}_{k}$, a half-step of length $\frac{1}{2} \Delta s$ is taken against the gradient towards pivot point $\boldsymbol{x}_{k+1}^{*}$

$$
\begin{equation*}
\boldsymbol{x}_{k+1}^{*}=\boldsymbol{x}_{k}-\frac{1}{2} \Delta s \frac{\boldsymbol{g}_{k}}{\left|\boldsymbol{g}_{k}\right|} . \tag{11.4}
\end{equation*}
$$

Subsequently, a constrained optimization is carried out on a hypersphere of radius $\frac{1}{2} \Delta s$, by minimizing the Lagrangian function

$$
\begin{equation*}
L(\lambda)=E_{k+1}^{\prime}+\Delta \boldsymbol{x}^{\prime \top} \boldsymbol{g}_{k+1}^{\prime}+\frac{1}{2} \Delta \boldsymbol{x}^{\prime \boldsymbol{\top}} \boldsymbol{H}_{k+1}^{\prime} \Delta \boldsymbol{x}^{\prime}-\frac{1}{2} \lambda\left[\boldsymbol{p}_{k}^{\prime \top} \boldsymbol{p}_{k}^{\prime}-\left(\frac{1}{2} \Delta s\right)^{2}\right] . \tag{11.5}
\end{equation*}
$$

No energy and gradient evaluation is needed at the pivot point, but at the successive points on the hypersphere $\left\{\boldsymbol{x}_{k+1}^{\prime}\right\}$. Primes indicate quantities evaluated on the hypersphere, $\boldsymbol{p}_{k}^{\prime}$ denotes a vector pointing from the pivot point towards $\boldsymbol{x}_{k+1}^{\prime}$. See Figure 11.1 for an illustration. The Hessian is updated using BFGS [235-238] Depending on the chosen step length, multiple additional energy and gradient evaluations are needed in every IRC cycle.


Figure 11.1: IRC step determination in the Gonzalez-Schlegel second-order algorithm (GS2) and Hessian predictor-corrector algorithms. 83, 325, 331-333 Large, filled circles indicate points, where energy and gradients are evaluated. a) Gonzalez-Schlegel 2: After an initial half step from $\boldsymbol{x}_{k}$ to a pivot point $\boldsymbol{x}_{k+1}^{*}$, a constrained optimization is carried out on a hypersphere with radius $\frac{1}{2} \Delta s$, yielding the next point on the $\boxed{\pi C} \boldsymbol{x}_{k+1}$. b) Hessian predictor-corrector integrator: After an initial predictor step to $\boldsymbol{x}_{k+1}^{*}$, an analytical surface is fitted by distance-weighted interpolants (DWI;) using energy, gradient and Hessian data. Multiple corrector integrations with decreasing step sizes are carried out on the DWIs surface and their results are combined using Richardson extrapolation, yielding the the next point on the IRC. In cycle $k$ no energy and gradient evaluation is needed at $\boldsymbol{x}_{k+1}$.

### 11.2 Predictor-Corrector Integration

Expanding $\boldsymbol{g}(\boldsymbol{x})$ to first-order around $\boldsymbol{x}_{0}$ and substituting into eq. 11.1) yields

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}(s)}{\mathrm{d} s}=-\frac{\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)+\boldsymbol{H}\left(\boldsymbol{x}_{0}\right) \Delta \boldsymbol{x}}{\left|\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)+\boldsymbol{H}\left(\boldsymbol{x}_{0}\right) \Delta \boldsymbol{x}\right|} \tag{11.6}
\end{equation*}
$$

Introducing an independent variable $t$ allows separation of eq. 11.6 into

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}(t)}{\mathrm{d} t} \frac{\mathrm{~d} t}{\mathrm{~d} s}=-\left(\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)+\boldsymbol{H}\left(\boldsymbol{x}_{0}\right) \Delta \boldsymbol{x}\right) \cdot \frac{1}{\left|\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)+\boldsymbol{H}\left(\boldsymbol{x}_{0}\right) \Delta \boldsymbol{x}\right|} \tag{11.7}
\end{equation*}
$$

yielding the equations $325,334,335$

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}}{\mathrm{~d} t}=-\left(\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)+\boldsymbol{H}\left(\boldsymbol{x}_{0}\right) \Delta \boldsymbol{x}\right) \tag{11.8}
\end{equation*}
$$

and

$$
\begin{equation*}
\frac{\mathrm{d} s}{\mathrm{~d} t}=\left|\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)+\boldsymbol{H}\left(\boldsymbol{x}_{0}\right) \Delta \boldsymbol{x}\right| \tag{11.9}
\end{equation*}
$$

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For quadratic potentials, eq. 11.8 is solved by

$$
\begin{equation*}
\boldsymbol{x}(t)=\boldsymbol{x}_{0}+\boldsymbol{A}(t) \boldsymbol{g}\left(\boldsymbol{x}_{0}\right) \tag{11.10}
\end{equation*}
$$

Matrix $\boldsymbol{A}(t)$ is defined as

$$
\begin{equation*}
\boldsymbol{A}(t)=\boldsymbol{U}_{0} \boldsymbol{\alpha}(t) \boldsymbol{U}_{0}^{\top} \tag{11.11}
\end{equation*}
$$

with $\boldsymbol{U}_{0}$ being the eigenvector matrix of $\boldsymbol{H}\left(\boldsymbol{x}_{0}\right)$ and the diagonal matrix $\boldsymbol{\alpha}(t)$ with elements

$$
\begin{equation*}
\alpha_{i i}=\frac{e^{-\lambda_{i} t}-1}{\lambda_{i}}, \tag{11.12}
\end{equation*}
$$

where $\left\{\lambda_{i}\right\}$ are the eigenvalues of $\boldsymbol{H}\left(\boldsymbol{x}_{0}\right)$. Eq. 11.9$)$ is integrated until a $t$ is obtained that produces the desired step length $\Delta s=s-s_{0}$. The integration is carried out in the basis of the Hessian eigenvectors by the Euler method.

$$
\begin{equation*}
\frac{\mathrm{d} s}{\mathrm{~d} t}=\sqrt{\sum_{i} \boldsymbol{g}_{0 i}^{\prime 2} e^{-2 \lambda_{i} t}} \tag{11.13}
\end{equation*}
$$

Please see Appendix A.1 for a full derivation of eq. 11.13). An initial step size for the Euler integration is calculated as

$$
\begin{equation*}
\delta t=\frac{1}{N_{\text {Euler }}} \frac{s-s_{0}}{\left|\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)\right|} \tag{11.14}
\end{equation*}
$$

with $N_{\text {Euler }}=5000$. Given a suitable $t$, corresponding coordinates are obtained from eq. (11.10).

Based on eqs. (11.6) to (11.14), Hratchian proposed the Hessian predictor-corrector (HPC) method. 325,336

Starting from $\boldsymbol{x}_{k}$, an initial predictor step to $\boldsymbol{x}_{k+1}^{*}$ is determined by the procedure just outlined and the gradient is evaluated. Subsequently, the predictor step is refined by modified Bulirsch-Stoer integration. $337-339$

As the corrector integration requires many energy and gradient evaluations, actual electronic structure calculations would be too costly. Instead, corrector integrations are carried out, using an analytical DWI surface, obtained from energies and derivatives at $\boldsymbol{x}_{k}$ and $\boldsymbol{x}_{k+1}^{*} \cdot 340344$

The DWI energy is defined as

$$
\begin{equation*}
E_{\mathrm{DWI}}=\sum_{i}^{N} w_{i} T_{i} \tag{11.15}
\end{equation*}
$$

with $T_{i}$ being a Taylor expansion to second-order

$$
\begin{equation*}
T_{i}\left(\Delta \boldsymbol{x}_{i}\right)=E\left(\boldsymbol{x}_{i}\right)+\Delta \boldsymbol{x}_{i}^{\top} \boldsymbol{g}\left(\boldsymbol{x}_{i}\right)+\frac{1}{2} \Delta \boldsymbol{x}_{i}^{\top} \boldsymbol{H}\left(\boldsymbol{x}_{i}\right) \Delta \boldsymbol{x}_{i}, \tag{11.16}
\end{equation*}
$$

and $\Delta \boldsymbol{x}_{i}=\boldsymbol{x}-\boldsymbol{x}_{i}$. Assuming $N=2$, e.g., interpolation between two geometries, coordinate dependent weights $w_{1}$ and $w_{2}$ are given as

$$
\begin{equation*}
w_{1}(\boldsymbol{x})=\frac{\left|\Delta \boldsymbol{x}_{2}\right|^{n}}{\left|\Delta \boldsymbol{x}_{1}\right|^{n}+\left|\Delta \boldsymbol{x}_{2}\right|^{n}}, \quad w_{2}(\boldsymbol{x})=\frac{\left|\Delta \boldsymbol{x}_{1}\right|^{n}}{\left|\Delta \boldsymbol{x}_{1}\right|^{n}+\left|\Delta \boldsymbol{x}_{2}\right|^{n}} \tag{11.17}
\end{equation*}
$$

Integer $n$ is commonly chosen as 4.345 The DWI energy expression eq. 11.15 can be differentiated with respect to Cartesian coordinates, allowing fast energy and gradient calculations in $\mathcal{O}\left(N^{2}\right)$ operations. Explicit expressions for the gradient of $E_{\text {DWI }}$ are given by Meisner. 346, 347

The Bulirsch-Stoer method comprises repeated cycles of numerical integrations with decreasing step sizes and subsequent Richardson extrapolation to zero step size, until a prescribed extrapolation error is satisfied, e.g. $1 \times 10^{-6} a_{0}$. Given a function that returns the corrector integration endpoint $\boldsymbol{x}_{k+1}$ for a given integration step size $\Delta s=\Delta s_{0} / 2^{j}$, depending on integer $j$

$$
\begin{equation*}
R(j, 0)=\boldsymbol{x}_{k+1}\left(\Delta s_{0} / 2^{j}\right) \tag{11.18}
\end{equation*}
$$

an extrapolation table can be formulated (see Table 11.1]. $348-350$ Entries in the first

Table 11.1: Richardson extrapolation table for function $R(j, k)$. Entries in the first column $R(j, k=0)$ are obtained by Euler integration on a DWI surface with step size $\Delta s=\Delta s_{0} / 2^{j}$. All remaining entries $R(j, k>0)$ can be calculated from the recursion defined in eq. 11.19.

| $R(0,0)$ |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| $R(1,0)$ | $R(1,1)$ |  |  |  |
| $R(2,0)$ | $R(2,1)$ | $R(2,2)$ |  |  |
| $R(3,0)$ | $R(3,1)$ | $R(3,2)$ | $R(3,3)$ |  |
| $\ldots$ | $\ldots$ | $\ldots$ | $\cdots$ | $\ldots$ |
| $\mathcal{O}\left(\Delta s_{0}\right)$ | $\mathcal{O}\left(\Delta s_{0}^{2}\right)$ | $\mathcal{O}\left(\Delta s_{0}^{3}\right)$ | $\mathcal{O}\left(\Delta s_{0}^{4}\right)$ |  |

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column $(k=0)$ are obtained by actual corrector integrations with decreasing step sizes. All remaining entries with $k>0$ are obtained by extrapolation, based on the recursion

$$
\begin{equation*}
R(j, k)=\frac{2^{k} R(j, k-1)-R(j-1, k-1)}{2^{k}-1} \tag{11.19}
\end{equation*}
$$

and previously calculated values. See Figure 11.1 for an illustration of the step determination in the HPC method.

While modified-midpoint integration is employed in the original Bulirsch-Stoer method, it was found to magnify the stiff character of eq. 11.1). (325 Instead, Hratchian proposed to use simple Euler integration, resulting in a modified Bulirsch-Stoer method.[325] Additionally, the backward differentiation formula ( $\overline{B D F}$ ) and Radau IIA methods were found to be suitable corrector integrators. [3, 351, 352

Determining the corrector step requires a $\mathcal{O}\left(N^{3}\right)$-scaling Hessian diagonalization, which may become a computational bottleneck for large molecules. To this end, Hratchian also suggested the simplified Euler predictor-corrector (EulerPC) integrator, 332, 333 where the predictor step is obtained as

$$
\begin{equation*}
\boldsymbol{x}_{k+1}^{*}=\boldsymbol{x}_{k}-\Delta s \frac{\boldsymbol{g}_{k}}{\left|\boldsymbol{g}_{k}\right|} . \tag{11.20}
\end{equation*}
$$

Alternatively, Hessian information can be considered in the predictor step by directly integrating

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}}{\mathrm{~d} s}=-\frac{\boldsymbol{g}_{0}+\Delta \boldsymbol{x} \boldsymbol{H}_{0}}{\left|\boldsymbol{g}_{0}+\Delta \boldsymbol{x} \boldsymbol{H}_{0}\right|}, \tag{11.21}
\end{equation*}
$$

while avoiding matrix diagonalization, which was already recognized by Page and McIver and pointed out again by Meisner. 335, 346] Compared to GS2, HPC and EulerPC only need two energy and gradient evaluations per IRC cycle.

## Part III

## Results

## 12 Biaryl Cross-Coupling

### 12.1 Introduction

Biaryl motifs play a key role in many pharmaceutical compounds[353] like antihypertensive sartans (354, non-steroidal anti-inflammatory drugs 355, 356, natural products 357 like cannabinol 358 and many alkaloids. $359-361$. Selective cross-coupling of aryl-residues is usually achieved by employing transition metal catalysts. [362-369] As transition metal catalysts are often toxic and expensive, metal-free aryl cross-coupling reactions have been proposed, as more sustainable alternatives. 370-379]
Recently, Kloss reported a novel metal-free cross-coupling reaction, where two phenyl groups tethered by a sulfonamide linker can be fused with high regio- and chemoselectivity in a single coupling product through irradiation by ultraviolet light. [2, 125]
This chapter presents computational insights and suggests an ES reaction mechanism for the biaryl cross-coupling reaction. Computational investigations were carried out for two substrates: 1a, a $p$-methyl carboxylate substituted biaryl-sulfonamide, affording good reaction yields and the unsubstituted reference biaryl-sulfonamide $\mathbf{1 b}$, for which only traces of the biphenyl photoproduct are detected (Figure 12.1).
For each substrate a conformer search was conducted, to identify important GS conformations. Subsequently, GSTS; were obtained and the corresponding IRC were integrated. ESF; of both substrates were calculated and analyzed for several geometries along the IRC.

### 12.2 Computational Details

Ground state potential energy surfaces of $\mathbf{1 a}$ and $\mathbf{1 b}$ were sampled by means of relaxed scans and conformer analysis. All ground state DFT calculations were carried out, using

[^3]

Figure 12.1: General scheme for the biaryl cross-coupling photoreaction, starting from biaryl-sulfonamides and yielding coupled biaryls, after irradiation at 254 nm $(4.88 \mathrm{eV})$. Two substrates were investigated: 1a $\left(\mathrm{R}^{1}=\mathrm{Me}, \mathrm{R}^{2}=\mathrm{COOMe}\right)$ and the unsubstituted biaryl-sulfonamide $\mathbf{1 b}\left(R^{1}=R^{2}=H\right)$, yielding the photoproducts 2a and $\mathbf{2 b}$. The species $\mathrm{NHCH}_{2}$ can hydrolyze to ammonia and formaldehyde. [125, 380
the range-separated XC functional CAM-B3LYP 179 and the all-electron def2-TZVP triple- $\zeta$ basis set. 381] Dispersion interactions were taken into account by Grimme's D3-model with Becke-Johnson damping. [382, 383]

Relaxed scans were performed with Gaussian 16, Revision B. 01384 at the DFT level (CAM-B3LYP/def2-TZVP) by varying the central (C-N-S-C)-dihedral angle in the sulfonamide-linker from $-180^{\circ}$ to $180^{\circ}$ with a step size of $10^{\circ}$, while equilibrating the remaining degrees of freedom at each step.
Possible conformers of the biaryl sulfonamide substrates were generated by a simulated annealing procedure, as implemented in Grimme's extended tight binding code GFN-XTB 5.8 using the GFN2-parametrization. (212 Effects of solvation by acetonitrile (ACN) on the conformer geometries were taken into account by the generalized Born solvent area (GBSA) continuum solvation model. 217
Improved energies 385 for the conformers (generated by XTB) were calculated with domain based local pair-natural orbital (DLPNO) singles and doubles coupled cluster with triples correction $\overline{\operatorname{CCSD}(\mathrm{T}) \text {, }}$, as implemented in ORCA 4.0.1.2. 386 388] The def2-QZVPP and corresponding auxiliary basis sets were utilized. (381, 389, 390] Solvent effects ACN $\varepsilon=36.6, n=1.344$ ) on the coupled cluster single point energies were considered by the conductor-like polarizable continuum model (CPCM). 391, 392 Tight criteria were used for convergence and truncation thresholds in the DLPNO procedure $\left(\right.$ TightPNO, $\left.T_{\text {CutPairs }}=1 \times 10^{-5}, T_{\text {CutPNO }}=1 \times 10^{-7}, T_{\text {CutMKN }}=1 \times 10^{-4}\right) .33$

To visualize the correlation between conformer bonding parameters and their coupled cluster energies, a principal component analysis (PCA) was conducted, using a set of eight primitive internal coordinates as features, describing the relative orientation of the two phenyl rings. All features were shifted to zero mean and scaled to unit variance. Conformers above a certain energy threshold $\left(30.0 \mathrm{~kJ} \mathrm{~mol}^{-1}\right.$ higher, than the minimum energy conformer, based on the DLPNO CCSD(T) energies) were excluded from the

PCA. The PCA was carried out using the python package scikit-learn. 394,395 Care has to be taken when dealing with dihedrals, as they are periodic $1_{1}^{1}$ Dihedrals were correctly taken into account for the PCA by employing their sine and cosine values. 396


Figure 12.2: Atom labels, used for defining the internal coordinate features of the PCA; for the conformers of $\mathbf{1 a}$ and $\mathbf{1 b}$.

Table 12.1: Employed internal coordinates in the PCA,

| $\#$ | Coordinate type | Indices |
| :---: | ---: | :--- |
| 1 | Bond | $\mathrm{C} 6-\mathrm{C} 7$ |
| 2 | Bond | $\mathrm{C} 5-\mathrm{C} 1$ |
| 3 | Bend | $\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3$ |
| 4 | Bend | $\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2$ |
| 5 | Bend | $\mathrm{N} 3-\mathrm{C} 2-\mathrm{C} 1$ |
| 6 | Bend | $\mathrm{C} 5-\mathrm{N} 3-\mathrm{C} 2$ |
| 7 | $\cos$ (Dihedral) | $\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2$ |
| 8 | $\sin$ (Dihedral) | $\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2$ |
| 9 | $\cos$ (Dihedral) | $\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2-\mathrm{C} 1$ |
| 10 | $\sin$ (Dihedral) | $\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2$ |

Transition state optimizations, subsequent vibrational analysis, and reaction path calculations were carried out at the DFT level. The nature of the first-order saddle points was confirmed by vibrational analysis. Minimum energy paths were obtained by IRC integration using the HPC method, 325, 336 as implemented in Gaussian, to verify that the optimized TS; connect the presumed educts sand products of the biaryl coupling reaction. Exact Hessians were recalculated every seventh IRC step. Equally spaced geometries were sampled from both sides of the IRC every sixth step, yielding reaction paths for subsequent ES calculations.

[^4]All ES calculations were carried out in Gaussian 16, using TD-DFT along the sampled IRC geometries. The same basis set and dispersion correction model as for the preliminary ground state DFT calculations was applied. This computational setup allows a balanced description of local, as well as of CT excitations among the $\pi$-systems of educt and product states. [397] Vertical excitation energies and oscillator strengths of the six lowest singlet ES; were calculated for the sampled geometries along the IRC. Solvent effects (ACN) on the vertical excitation energies and oscillator strengths were taken into account by the CPCM, Excited state characters were interpreted in terms of NTO; (see section 4.3), as calculated by Multiwfn 3.5. 227, 398] All calculations were carried out, assuming neutral charge and singlet multiplicity. If not noted otherwise, all orbital plots were created using an isovalue of 0.04 au .

### 12.3 Preferred Ground State Conformation

Molecules posses many different internal degrees of freedom. Rigid degrees, like the framework of (covalent) bonds, are not easily altered and stay close to their equilibrium values. Flexible degrees, like dihedrals, are altered more easily and may span a broad range of values, some of them far from equilibrium values, e.g., obtained at a stationary point.

This high flexibility can give rise to many different conformers, each one corresponding to a local minimum on the molecular $\overline{\mathrm{PES}}$, Naive calculation of full, $3 N-6(5)$ dimensional PESk and subsequent determination of all local minima is infeasible for all, except the smallest molecules. Efficient conformer search is a topic of ongoing research and to this date, many different methods have been proposed. 399-407

Presence of multiple, energetically accessible and interconvertible conformers complicates computational studies, as their results may depend strongly on the studied conformers. Whereas determining energetically low-lying conformers may be feasible by one of the methods mentioned above, studying their interconversion is usually out of scope, as this would require $\sum_{i}^{N-1} i$ additional TS optimizations, for $N$ conformers.

In the present study, energetically low-lying conformers of $\mathbf{1 a}$ and $\mathbf{1 b}$ have been obtained by means of simulated annealing and were interpreted in terms of corrected energies from coupled cluster calculations and PCA (see section 12.2).

## 1a

Simulated annealing conformer search yielded 38 geometries for the educt 1a, 34 remained below $30.0 \mathrm{~kJ} \mathrm{~mol}^{-1}$ above the minimum energy conformer (see Figure 12.3 and Figure 12.5). Two conformer motifs could be identified: linear conformers with increased C1 - C5 distances (e.g. Figure 12.3e) and horseshoe conformers featuring decreased C5C 1 distances and (nearly) parallel phenyl rings (e.g. Figure 12.3a) facing each other.
The minimum energy geometry for $\mathbf{1 a}$ (conformer 2) belongs to the horseshoe motif with a C5 - C1 distance of 310 pm . In typical linear conformers, e.g., conformer 19, the C5-C1 distance is about 500 pm . Small C5-C1 distances are expected to facilitate the bond formation, between these two carbon atoms.

(a) Conformer 2, $+0.0 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(b) Conformer 7, $+5.9 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(c) Conformer 33, $+9.5 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(d) Conformer 37, $+19.8 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(e) Conformer 19, $+20.3 \mathrm{~kJ} \mathrm{~mol}^{-1}$

Figure 12.3: Selected 1a conformers and energies obtained at the DLPNO-CCSD(T)/def2QZVPP/CPCM(ACN) level of theory, with respect to minimum energy at conformer 2.

PCA results for 1a are shown in Figure 12.5. The first two principal components PC1 and PC 2 capture $54.7 \%$ of the total variance. PC 1 corresponds mainly to reduced $\mathrm{C}-\mathrm{C}$ distances in both, C6-C7 ( -0.49 ) and C5-C1 ( -0.52 ) (see Table 12.2 ), allowing to distinguish between linear (negative PC1 contribution, increased distances) and horseshoe motifs (positive PC1 contribution, decreased distances). Similarly, decreases of the angle C5 - N3-C2 ( -0.52 ) contribute to PC1. Indeed, conformers 1 to 7,12 and 33 in Figure 12.5 with strong positive contributions of PC1 correspond to the horseshoe motif,
while conformers 19 and 16 with strong negative contributions of PC1 correspond to the linear motif (see also Figure 12.3). A more detailed depiction of conformers 2 and 19 including values of selected internal coordinates is found in Figure 12.4.

(a) Conformer 2, horseshoe motif.

(b) Conformer 19, linear motif.

Figure 12.4: Detailed display of prototypical horseshoe and linear motifs for 1a, including structural parameters that make up the first principal component PC1 (C5C1, C6-C7 and C5-N3-C2). The units in conformer 19 (pm) have been omitted for clarity.

Table 12.2: First two principal components PC1 and PC2 in terms of scaled and shifted internal coordinates for conformers of $\mathbf{1 a}$ and $\mathbf{1 b}$.

|  |  | 1a |  | 1b |  |
| ---: | ---: | ---: | ---: | ---: | ---: |
| $\#$ | Coordinate | PC1 | PC 2 | PC 1 | PC 2 |
| 1 | $\mathrm{C} 6-\mathrm{C} 7$ | -0.49 | -0.08 | -0.44 | -0.01 |
| 2 | $\mathrm{C} 5-\mathrm{C} 1$ | -0.52 | -0.07 | -0.49 | -0.04 |
| 3 | $\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3$ | -0.03 | 0.52 | -0.05 | 0.56 |
| 4 | $\mathrm{~S} 4-\mathrm{N} 3-\mathrm{C} 2$ | -0.06 | 0.48 | -0.07 | 0.53 |
| 5 | $\mathrm{~N} 3-\mathrm{C} 2-\mathrm{C} 1$ | 0.29 | -0.20 | 0.39 | -0.03 |
| 6 | $\mathrm{C} 5-\mathrm{N} 3-\mathrm{C} 2$ | -0.52 | -0.04 | -0.49 | -0.03 |
| 7 | $\cos (\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2)$ | 0.21 | 0.50 | 0.17 | -0.03 |
| 8 | $\sin (\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2)$ | -0.05 | -0.11 | -0.08 | -0.43 |
| 9 | $\cos (\mathrm{~S} 4-\mathrm{N} 3-\mathrm{C} 2-\mathrm{C} 1)$ | 0.26 | -0.40 | 0.34 | 0.21 |
| 10 | $\sin (\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2)$ | -0.10 | -0.15 | 0.15 | -0.42 |

## 1b

Simulated annealing conformer search yielded 17 geometries for $\mathbf{1 b}$ that can similarly be assigned to either linear or horseshoe motifs. In contrast to 1a, the minimum energy


Figure 12.5: Scatter plot of DLPNO-CCSD(T)/def2-QZVPP/CPCM(ACN) energies for the first two principal components (PC1, PC2), obtained from a PCA on a subset of primitive internals coordinates of 1a. All energies are given with respect to the minimum energy at conformer 2 . The first principal component PC1 allows distinguishing between horseshoe and linear motif. Conformers $33,12,2,1,7$, and 6 exhibit a horseshoe motif, whereas conformers 19,16 , 25 , and 29 belong to the linear motif.
geometry is linear for $\mathbf{1 b}$ (conformer 8) with a high C5-C1 distance of 471 pm . Horseshoe conformers have slightly higher energies: $+4.6 \mathrm{~kJ} \mathrm{~mol}^{-1}$ for conformer 1 and $+8.0 \mathrm{~kJ} \mathrm{~mol}^{-1}$ for conformer 4 (see Figure 12.7 and Figure 12.6 ).

By considering the first two principal components PC1 and PC2, $58.2 \%$ of the total variance is captured. As expected, this is slightly more compared with $\mathbf{1 a}$, as $\mathbf{1 b}$ has overall fewer degrees of freedom. The analysis of the $\mathbf{1 a P C A}$ also applies to the $\mathbf{1 b}$ PCA, as PC1 and PC2 are very similar in both cases (see Table 12.2). Notably, as the minimum conformer 8 does not belong to the horseshoe motif, contribution of PC1 is negligible (Figure 12.7).

(a) Conformer 8, $+0.0 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(b) Conformer 1, $+4.6 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(c) Conformer 4, $+8.0 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(d) Conformer 9, $+12.9 \mathrm{~kJ} \mathrm{~mol}^{-1}$

(e) Conformer 12, $+22.7 \mathrm{~kJ} \mathrm{~mol}^{-1}$

Figure 12.6: Selected 1b conformers and energies obtained at the DLPNO-CCSD(T)/def2QZVPP/CPCM(ACN) level of theory with respect to minimum energy conformer 8.


Figure 12.7: Scatter plot of DLPNO-CCSD(T)/def2-QZVPP/CPCM(ACN) energies for the first two principal components ( $\mathrm{PC} 1, \mathrm{PC} 2$ ) obtained from a PCA on a subset of primitive internals coordinates of $\mathbf{1 b}$. All energies are given with respect to the minimum energy at conformer 8 . The first principal component PC1 allows distinguishing between horseshoe and linear motif. Conformers 1-4 exhibit a horseshoe motif, whereas conformers $9,12,13$, and 17 belong to the linear motif.

In summary, the conformer search results for $\mathbf{1 a}$ and $\mathbf{1 b}$ give a first hint on the different photoreaction yields. The minimum energy conformer 2 for 1a belongs to the horseshoe motif, exhibiting a small C5-C1 distance ( 310 pm ), which facilitates efficient $\mathrm{C}-\mathrm{C}$ bond formation. However, the minimum energy conformer 8 for $\mathbf{1 b}$ belongs to the linear motif exhibiting an increased C5-C1 distance ( 471 pm ), hindering efficient bond formation in the photoreaction, while its lowest horseshoe conformer was predicted at slightly higher energy ( $4.4 \mathrm{~kJ} \mathrm{~mol}^{-1}$ ).

## Relaxed Scan Around the Central Dihedral Angle

Additionally, the GSIPES of $\mathbf{1 a}$ and $\mathbf{1 b}$ were sampled by relaxed scans around the central dihedral angle $\phi(\mathrm{C} 5-\mathrm{S} 4-\mathrm{N} 3-\mathrm{C} 2)$ at the DFT (CAM-B3LYP/def2-TZVP) level of theory.

For 1a, results from the relaxed scan are in line with the conformer search. Both approaches predict a horseshoe conformer as minimum energy geometry (see Figure 12.8). The horseshoe conformer at $\phi=-80^{\circ}$ exhibits a small $\mathrm{C} 5-\mathrm{C} 1$ distance of only 326 pm and is $10 \mathrm{~kJ} \mathrm{~mol}^{-1}$ more stable, compared to the linear conformer at $\phi=70^{\circ}$, with an increased C5 - C1 distance of 480 pm .


Figure 12.8: Energy differences and C5-C1 distances for the relaxed scan around the central dihedral (C5-S4-N3-C2) in 1a at the CAM-B3LYP/def2-TZVP level of theory. A horseshoe conformer is predicted as minimum energy geometry at a) $\phi=-80^{\circ}$ with b) a C5-C1 distance of 326 pm . The second marked minimum corresponds to a linear conformer with increased C5-C1 distance.

In contrast to the conformer search, the relaxed scan predicts a horseshoe conformer as minimum for $\mathbf{1 b}$, but it is only $5.2 \mathrm{~kJ} \mathrm{~mol}^{-1}$ more stable, compared to the linear conformer at $\phi=70^{\circ}$ (see Figure A.1).
Overall, sampling the GS PES of 1a and $\mathbf{1 b}$ revealed two distinct, structural motifs: horseshoe conformations with shorter C5-C1 distances, pivotal for efficient bond formation and linear conformations with increased C5-C1 distances, rendering efficient bond formation unlikely.

### 12.4 Ground State Reaction Coordinate and Electronic Structure

To evaluate the potential course of the biaryl coupling, formation of photoproducts $\mathbf{2 a}$ and 2b (fig. 12.1) was assessed along a GS IRC, obtained at the CAM-B3LYP/def2TZVP/CPCM(ACN) level of theory. The IRC connects educt and product states via a cyclic (C1 - C2 - N3-S4 - C5)-containing TS (see Figure 12.91).


Figure 12.9: Selected distances along the IRC, describing the formation of 2a from 1a by photosplicing, obtained at the CAM-B3LYP/def2-TZVP/CPCM(ACN) level of theory. a) Atomic distances with significant changes when going from the educt to the TS and b) distances that change significantly in the second IRC half, when going from TS to the photoproducts.

Towards the five-membered TS of the reaction from 1a to $\mathbf{2 a}$, the $\mathrm{C} 5-\mathrm{C} 1$ distance decreases from 348 pm to 164 pm , which is associated with $\mathrm{C} 5-\mathrm{C} 1$ bond formation.

Simultaneously, partial cleavage of the C5-S4 bond is indicated by a distance increase from 176 pm to 239 pm (see Figure 12.9). In stark contrast, changes within the sulfonamide linker are less pronounced until the TS is reached. Only on relaxation towards the product state and thus formation of photoproduct 2a increased S4-N3 (169 pm to 257 pm ) as well as $\mathrm{C} 1-\mathrm{C} 2$ distances ( 165 pm to 355 pm ) are observed, indicating extrusion of $\mathrm{SO}_{2}$ and $\mathrm{NHCH}_{2}$. The latter species can hydrolyze to ammonia and formaldehyde. 125,380
Similar results are obtained for the formation of $\mathbf{2 b}$ from $\mathbf{1 b}$, which are omitted here (see Figure A. 2 on page 138). Selected atomic distances for educts, TS. and products of both photoreactions are given in Table 12.3 .

Table 12.3: Selected distances in educts, transition states and products in the biaryl coupling reactions of $\mathbf{1 a}$ and $\mathbf{1 b}$ in pm. While the C5-C1 bond is formed along the IRC, the remaining bonds are broken.

|  |  | 1a |  |  | 1b |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
|  |  | Educt | TS | Products | Educt | TS | Products |
| Bond | Change | Distance $/ \mathrm{pm}$ |  | Distance $/ \mathrm{pm}$ |  |  |  |
| C5 - C1 | formed | 348 | 164 | 148 | 340 | 170 | 148 |
| C5- S4 | broken | 176 | 239 | 352 | 177 | 241 | 354 |
| S4- N3 | broken | 165 | 169 | 257 | 164 | 170 | 256 |
| C1-C2 | broken | 150 | 165 | 355 | 151 | 163 | 347 |

According to the calculations, TS-formation requires an activation energy of 4.08 eV in the GS for $\mathbf{1 a}$, and 4.07 eV for $\mathbf{1 b}$. Irradiation of the biaryl sulfonamides is carried out at $254 \mathrm{~nm}(4.88 \mathrm{eV})$, energetically well above the obtained GS barrier heights of about 4.0 eV . Even though the barriers are presumably not crossed in the electronic GS, but in an ES, the obtained pathways are consistent with the experimental conditions and energetically feasible, as the GSk barrier heights are below the irradiation energy.
The electronic structure of the biaryl sulfonamides is briefly discussed, at the example of a 1a horseshoe conformer. Its frontier MO, (HOMO-3 to LUMO +3 ) have mainly $\pi$ and $\pi^{*}$ character (see Figure 12.12) and are expected to be major contributors, to the low-lying ESS of 1a. Whereas, the orbital lobes at C5 and C1 in the HOMO of 1a have opposite signs and thus, are antibonding, the C5-C1 interaction is bonding in the LUMO, A simplified scheme is shown in Figure 12.11. All investigated biaryl sulfonamides have high HOMO LUMO gaps of about 8.0 eV . In contrast to $\mathbf{1 b}$, the conjugated $\pi$-system in 1a is extended by a methyl carboxylate residue, nearly coplanar to the connected phenyl
ring (rotated by $2.3^{\circ}$ ). Extending $\pi$-systems is known to modulate excitation energies and oscillator strengths, usually resulting in redshifted and brighter excitations. 408412


Figure 12.10: Isocontour plots of the frontier MO of 1a obtained at the CAM-B3LYP/def2-TZVP/CPCM(ACN) level of theory.

(a) MO 84, HOMO

(b) MO 85, LUMO

Figure 12.11: Simplified illustration of the orbital lobes centered at C 5 and C 1 for HOMO and LUMO in 1a. Whereas the interaction in the HOMO is antibonding, it is bonding in the LUMO. The nodal plane between C5 and C1 in the HOMO is indicated by a dashed line.

### 12.5 Excited State Reaction Mechanism

To elucidate a potential reaction mechanism, low-lying ESk in the Franck-Condon region of $\mathbf{1 a}$ and $\mathbf{1 b}$ were calculated. In the following, $\pi$-orbitals located on the S-linked phenyl residue are denoted by a S-subscript $\left(\pi_{\mathrm{S}}\right)$, and $\pi$-orbitals located on the $\mathrm{CH}_{2}-\mathrm{NH}$-linked phenyl residue are denoted with a N -subscript $\left(\pi_{\mathrm{N}}\right)$.
The simulated absorption spectrum of $\mathbf{1 a}$ (see Figure 12.13a and Table 12.4) shows several bright ES; in the vicinity of the irradiating light source centered at 4.88 eV


Figure 12.12: Energies and isocontour plots of the HOMO and LUMO for horseshoe and linear conformations of $\mathbf{1 a}$ and $\mathbf{1 b}$, obtained at the CAM-B3LYP/def2TZVP/CPCM(ACN) level of theory.
$(254 \mathrm{~nm})$ that is $S_{2}$ at $5.30 \mathrm{eV}(234 \mathrm{~nm}$, oscillator strength $f=0.0664)$ and $S_{3}$ at 5.40 eV ( $230 \mathrm{~nm}, f=0.3611$ ). In terms of NTO, the $S_{3}$ is well described by one $\pi_{\mathrm{N}} / \pi_{\mathrm{N}}^{*}$ NTO pair with a contribution of $70 \%$. Considering the original MO basis, the biggest contribution $(41 \%)$ to the $S_{3}$ arises from a HOMOLUMO transition (MO) 84 and 85 in Figure 12.10), corresponding to depopulation of a $\mathrm{C} 5-\mathrm{C} 1$ antibonding orbital and subsequent population of a C5-C1 bonding orbital.

Similar ESb are obtained for $\mathbf{1 b}$, albeit hypsochromically blue-shifted and with lower oscillator strengths. Compared to the $S_{3}$ in $\mathbf{1 a}$ at 5.30 eV , the similar $S_{4}$ in $\mathbf{1 b}$ is found at 6.02 eV with a decreased oscillator strength $f=0.0550$ (see Figure 12.13). It is well described by one NTO pair $(74 \%)$ and corresponds to a transition from a $\pi_{\mathrm{N}}$ into a strongly C5-C1 bonding $\pi$ orbital. The brighter $S_{3}$ of $\mathbf{1 b}$ at 5.81 eV with $f=0.1169$ is also well captured by one $\pi_{\mathrm{S}} / \pi_{\mathrm{S}}^{*}$ NTO pair ( $86 \%$ contribution). It is not expected to participate effectively in the photoreaction, as the transition density is constrained to the S-linked phenyl residue. Due to a mismatch between the energetic location of $S_{3}$ and $S_{4}$ with the available irradiation energy (centered at 4.88 eV ), population of these ESp is unlikely. Therefore, disadvantageous ES properties at the Franck-Condon geometry of 1b prevent an efficient photoreaction at 254 nm irradiation, which is in agreement with the experiment, as only traces of the photoproduct biphenyl are detected for $\mathbf{1 b}$.

Table 12.4: Calculated low-lying excited singlet states, excitation wavelengths $(\lambda / \mathrm{nm})$, excitation energies ( $\Delta E / \mathrm{eV}$ ), oscillator strengths $f$ and leading NTO pairs of 1a for a horseshoe conformation at the educt side of the IRC obtained at the CAM-B3LYP/def2-TZVP/CPCM(ACN) level of theory.

| State | $\lambda / \mathrm{nm}$ | $\Delta E / \mathrm{eV}$ | $f$ | NTO | Weight $/ \%$ |
| :---: | :---: | :---: | :---: | :--- | :---: |
| $S_{1}$ | 242 | 5.12 | 0.0157 | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{N}}^{*}$ | 74 |
|  |  |  |  | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{N}}^{*}$ | 25 |
| $S_{2}$ | 234 | 5.30 | 0.0664 | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{S}}^{*}, \sigma_{\mathrm{CC}}^{*}$ | 57 |
|  |  |  |  | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 28 |
| $S_{3}$ | 230 | 5.40 | 0.3611 | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{N}}^{*}$ | 70 |
|  |  |  |  | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 16 |
| $S_{4}$ | 228 | 5.44 | 0.0158 | $\mathrm{n}_{\mathrm{O}} \rightarrow \pi_{\mathrm{N}}^{*}$ | 98 |
| $S_{5}$ | 217 | 5.73 | 0.2116 | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 82 |
| $S_{6}$ | 196 | 6.33 | 0.0444 | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{N}}^{*}, \sigma_{\mathrm{CC}}^{*}$ | 58 |
|  |  |  |  | $\pi_{\mathrm{S}}, \pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 24 |

Analyzing the ESF of 1a along the GS IRC (Figure 12.13) corroborates the results obtained at the Franck-Condon geometry. Product formation in the GS is prevented by a high barrier $(4 \mathrm{eV})$ that is reduced to only 0.69 eV in the $S_{1}$ at 5.12 eV . Assuming a most likely initial population of the energetically higher lying, bright $S_{3}(5.40 \mathrm{eV})$ and subsequent adiabatic evolution, the barrier is further reduced to $0.41 \mathrm{eV}^{2}$ allowing efficient product formation. The calculated excitation energy of the $S_{3}$ also coincides with the experimentally determined maximum substrate turnover at $5.32 \mathrm{eV}(233 \mathrm{~nm}) . \sqrt{125}$

Noteworthy, an even further reduced activation energy or even a barrier-free reaction is expected along a suitable excited state reaction pathway; however, determining such coordinate is far from trivial.
In contrast to 1a, the ES barrier height in the $S_{1}$ of $\mathbf{1 b}$ is slightly higher $(0.77 \mathrm{eV})$. However, efficient photoreaction is already prevented by the hypsochromically shifted ESF and low oscillator strengths, as reflected by the photochemical experiment yielding merely traces of photoproduct $\mathbf{2 b}$ at the applied conditions.

### 12.6 Summary

The computational results are in full accordance with the experimental observations and thus allowed elucidating the mechanism of photosplicing. Quantum chemical simulations show that electron-withdrawing and electron-donating groups are a prerequisite for the

[^5]Table 12.5: Calculated low-lying excited singlet states, excitation wavelengths ( $\lambda / \mathrm{nm}$ ), excitation energies ( $\Delta E / \mathrm{eV}$ ), oscillator strengths $f$ and leading NTO pairs of $\mathbf{1 b}$ for a horseshoe conformation at the educt side of the IRC obtained at the CAM-B3LYP/def2-TZVP/CPCM(ACN) level of theory.

| State | $\lambda / \mathrm{nm}$ | $\Delta E / \mathrm{eV}$ | $f$ | NTO | Weight $/ \%$ |
| :---: | :---: | :---: | :---: | :--- | :---: |
| $S_{1}$ | 231 | 5.38 | 0.0125 | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 67 |
|  |  |  |  | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 33 |
| $S_{2}$ | 227 | 5.47 | 0.0007 | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{N}}^{*}$ | 52 |
|  |  |  |  | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{N}}^{*}$ | 48 |
| $S_{3}$ | 213 | 5.81 | 0.1169 | $\pi_{\mathrm{S}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 86 |
| $S_{4}$ | 206 | 6.02 | 0.0550 | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{N}}^{*}, \pi_{\mathrm{S}}^{*}, \sigma_{\mathrm{CC}}^{*}$ | 74 |
|  |  |  |  | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{N}}^{*}$ | 19 |
| $S_{5}$ | 200 | 6.19 | 0.0028 | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 87 |
| $S_{6}$ | 195 | 6.37 | 0.0172 | $\pi_{\mathrm{N}} \rightarrow \pi_{\mathrm{S}}^{*}$ | 94 |

photoreaction, because these substituents control the energy of the $\mathrm{ES} ;$ and determine the overlap with the applied light source. The course of the photoreaction can be illustrated by means of the frontier orbitals of 1a (Figure 12.11) contributing to $S_{3}$. Specifically, the HOMO exhibits antibonding character between C5 and C1, whereas the LUMO shows bonding character between these two carbon atoms. The photoinduced population of the LUMO lowers the activation energy substantially and enables the photochemical formation of 2a. The small energy gap ( 1.0 eV ) between GS and ES $S_{1}$ in the vicinity of the TS facilitates the relaxation into the product state.


## 13 Pysisyphus

### 13.1 Introduction

This chapter discusses the implementation of pysisyphus, an external optimizer written in python, 413,414 that also accounts for ESF. Pysisyphus allows localizing SPs in groundand excited states, using surface-walking, COS methods and IRC calculations. Several algorithms for effective state-tracking are available. Although already several external optimizers with impressive functionalities like ASE or DL-FIND exist, [124, 415] none of them is tailored to the optimization of ES5. Only recently, García et al. presented their SDNTO[223] approach for localizing minima in ES $\beta$, but the program is restricted to a steepest descent optimizer and a rather limited selection of QC packages for ES gradient calculations, e.g., the programs ORCA and Gaussian. 121, 384
The chapter is organized as follows: section 13.2 verifies the correct implementation of pysisyphus and analyzes its performance by optimizing several test sets. Pysisyphus' ES tracking capabilities are demonstrated exemplarily for ES optimizations of cytosin and three transition metal complexes in section 13.3. The process of obtaining full reaction paths starting from only educts and products is shown in section 13.4 exemplarily for a GS reaction coordinate. Section 13.5 gives the list of QC programs, interfaced by pysisyphus and contains a short comment on its general structure. Implementation of RIC is discussed briefly.

### 13.2 Benchmarks and Verification

Effectiveness and efficiency of a geometry optimizer depend strongly on the careful selection of employed algorithms and the tuning of many parameters. Considering only initial Hessian choice and subsequent Hessian update in pysisyphus (see Table 13.1),

Parts of this chapter are based on:
Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: 10.1002/qua. 26390

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## 13 Pysisyphus

already gives rise to $7 \cdot 8=56$ possible algorithmic combinations. Also including convergence accelerators, which are either enabled or disabled, yields a total of $2^{3} \cdot 56=448$ possible combinations. If further important decisions, like coordinate system or initial, minimum and maximum trust radius are considered, scanning the full space of possible algorithm and parameter combinations quickly becomes infeasible. Therefore, only some appropriate combinations are usually tested. 245

To this date, several molecular benchmark sets have been put forward, allowing unambiguous evaluation of optimizer performance and its specific implementation. The most frequently employed test set for ground state optimizations was proposed by Baker and comprises 30 small molecules, containing 3 to 29 atoms (Figure 13.1). 268 The intended level of theory is HF/STO-3G, although often different basis sets are employed. (159, 232, 241, 246, 416, 417

Subsequently, Baker also released a test set for evaluating TS optimizers at the $\mathrm{HF} / 3$ 21G level of theory. [315] It includes 25 geometries from 3 to 16 atoms, close to a TS geometry. Notably, two cases (10 and 11) start in the convex region of the PES and exhibit no imaginary frequencies at the given geometry.

Accompanying the test sets, Baker proposed the following convergence criteria: no gradient component greater than $3.0 \times 10^{-4} E_{\mathrm{h}} \cdot a_{0}^{-1}\left(\mathrm{rad}^{-1}\right)$ and either an energy difference, compared to the previous cycle, below $1.0 \times 10^{-6} E_{\mathrm{h}}$, or no step component greater than $3.0 \times 10^{-4} a_{0}(\mathrm{rad})$. ${ }^{1}$

In the following, correct implementation of pysisyphus and its performance for molecular and reaction path optimization is verified for several benchmark sets and compared, if possible, to established optimizers and literature data.

[^6]Table 13.1: Possible keywords, controlling the behavior of the default optimizer for minimizations and TS searches in pysisyphus.

| Keyword | Comment | Reference |
| :---: | :---: | :---: |
| Initial Hessian calc | Calculate exact Hessian. Expensive, not available always. | - |
| unit | Unit matrix. | - |
| simple | Fixed force constants: 0.5 for bonds, 0.2 for bends and 0.1 for dihedrals. | 232 |
| fischer | MH. parametrized against HF/6-31G**, default (see eqs. $\sqrt{5.13}$ ) to (5.15)). | 242 |
| lindh | MH, parametrized against HF/STO-3G. | 241 |
| swart | MH, similar to lindh. | 243 |
| xtb/xtb1 | Calculation at GFN2/GFN1-xtb level of theory. | 418 |
| Hessian update |  |  |
| bfgs | Default, see eq. 5.11. | 235238 |
| damped_bfgs | Modifies gradient difference $\boldsymbol{y}$, when curvature condition is not satisfied $(\boldsymbol{s y}<0)$. | (234] |
| sr1 | Suitable for TSfoptimization. | 419 |
| psb | Suitable for TS-optimization. | 420 |
| bofill | Suitable for TS-optimization, combines SR1 and PSB. | 240 |
| flowchart | Switches between SR1, BFGS and PSB. | 239 |
| (hessian_recalc) | See calc, recalculation of exact Hessian every $n$-th cycle. | - |
| Convergence acceleration/stabilization |  |  |
| gediis | Disabled by default. | 260 |
| gdiis | Enabled by default. At most 5 gradients are used as error vectors. | $\begin{array}{\|l\|l\|} \hline 258,259] \end{array}$ |
| line_search | Fitting of 1d quartic/cubic polynomials, enabled by default. Fallback if GDIIS,GEDIIS failed or disabled. | 75 |

## 13 Pysisyphus

### 13.2.1 Baker Test Set

Aiming at determining most suitable default options for the pysisyphus optimizer, different choices for initial Hessian, its update, and convergence acceleration (Table 13.1) have been tested. A similar survey was already conducted by Bakken. 232

Energies and gradients were obtained with ORCA 4.2.1 at the HF/STO-3G level of theory, enforcing tight self-consistent field (SCF) convergence criteria (keyword tightscf), to reduce numerical noise in the gradients. 388] Neutral charge and singlet multiplicity were assumed throughout. Optimizations were conducted by RFO in RIC. If the proposed RFO step exceeded a prescribed trust radius (initial value $0.5 a_{0}$ (rad), maximum value $\left.1.0 a_{0}(\mathrm{rad})\right)$, it was scaled to the trust radius.

Table 13.2: Required cycles to converge Baker's test set using pysisyphus, for different keyword choices (see Table 13.1). If not noted otherwise, the initial Hessian is calculated according to Fischer, updated by the BFGS formula and convergence is accelerated by partial line searches and GDIIS,

| Initial Hessian | Cycles |
| :--- | :---: |
| fischer | 207 |
| lindh | 212 |
| simple | 254 |
| swart | 228 |
|  |  |
| Hessian update | Cycles |
| bfgs | 207 |
| damped_bfgs | 207 |
| flowchart | 225 |
| bofill | 232 |

Convergence acceleration

| line_search | gdiis | Cycles |
| :--- | :---: | :---: |
| yes | yes | 207 |
| no | yes | 208 |
| no | no | 208 |

Results of the initial survey are shown in Table 13.2. The smallest number of required cycles (207), to converge Baker's test set at the HF/STO-3G level of theory, is achieved by combining Fischers model Hessian, 242 BFGS updating, partial line searches, and GDIIS, $75,225,259$ In the following, this combination is referred to as the standard method.

Initial Hessian choice has the biggest influence on the number of cycles. The highest number of cycles (254) is obtained with a simple model Hessian, using fixed values of 0.5/0.2/0.1 for stretches, bends and dihedrals. Only 228 cycles are required with Swarts model Hessian. (243] Using Lindh's model Hessian yields results (212 cycles) close to the optimal choice (Fischer, 207 cycles). 241]

Whether the BFGS Hessian update is damped or not, has no influence for the Baker test set. Damping is only important when the curvature condition is violated, which is never the case for these well behaved optimizations. Using the flowchart update, which selects between SR1, BFGS and PSB updates, requires 225 cycles. 239] Bofills update, originally proposed for TS optimizations, needs even more cycles (232).
Disabling convergence accelerators like partial line searches and GDIIS, leaves the number of required cycles nearly unaffected, which is consistent with the findings of Bakken.[232] As the test set geometries are already close to a stationary point, the local quadratic expansion seems to approximate the true $\overline{P E S}$ well and produces reasonable, hardly improvable steps. GDIIS is only activated, after the root mean square (RMS) of the step falls below $2.5 \times 10^{-3} a_{0}(\mathrm{rad})$, which usually happens quite late in the course of an optimization, so GDIIS is employed only a few times. [259] Enabling GEDIIS led to inferior results in many cases, even though it appears to be correctly implemented. It is disabled by default in pysisyphus. 260

The number of 207 cycles in pysisyphus compares favorably to the original 240 cycles reported by Baker, 268 but is slightly higher than the 185 cycles reported by Fischer. 232 Bakken's result was obtained using a set of extra-redundant coordinates, which may lead to problems in the internal-Cartesian back-transformation (see section 6.3) of the optimization step, even though no such problems were reported by Bakken. Similar results are obtained when the bigger $6-31 \mathrm{G}^{*}$ basis is employed. Bakken reported 198 cycles, whereas 215 cycles are needed in pysisyphus. Distributed over the whole test set, most optimizations differ only in one or two cycles. Recently, Raggi reported 262 cycles for Baker's test set using RFO, without any step size restriction at the HF/631G level of theory in OpenMolcas. [417, 421, 422 Using the same convergence criteria, pysisyphus needs only 209 cycles, showing a marked improvement. Especially, three cases converged slowly in Raggis work: histidine (case 27, 29 cycles), 2,3-dimethylpentane (case 28,27 cycles) and menthone (case 30,32 cycles). This highlights the fact that optimization performance is probably highly sensitive to numerous (implementation) details and can differ greatly from software to software, even for simple cases. The present implementation is also competitive with results reported by Meyer ( 232 cycles, Baker's convergence criteria), who used a novel Gaussian process regression approach and internal coordinates. 159

Table 13.3: Optimization cycles, required to converge the Baker test set (Figure 13.1), according to Baker's criteria (see section 13.2) for different basis sets.[268] Convergence to the correct geometry was ensured by comparing the final energy to the published values (only for STO-3G basis). Bakken's results correspond to their most efficient method, using extra-redundant internal coordinates, as presented in Table X in 232 . Code to reproduce the results of this work is given in Listing A. 1 on page 139 .

| \# | Molecule | HF/STO-3G |  |  | HF/6-31G* |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | This work | Baker $268$ | Bakken $232$ | This work | Bakken $232$ |
| 1 | Water | 4 | 6 | 4 | 4 | 5 |
| 2 | Ammonia | 7 | 6 | 5 | 7 | 5 |
| 3 | Ethane | 4 | 5 | 3 | 4 | 4 |
| 4 | Acetylene | 4 | 6 | 4 | 4 | 4 |
| 5 | Allene | 4 | 5 | 4 | 5 | 5 |
| 6 | Hydroxysulphane | 7 | 8 | 7 | 8 | 6 |
| 7 | Benzene | 3 | 4 | 3 | 3 | 3 |
| 8 | Methylamine | 5 | 6 | 4 | 5 | 5 |
| 9 | Ethanol | 6 | 6 | 4 | 6 | 5 |
| 10 | Acetone | 6 | 6 | 4 | 6 | 4 |
| 11 | Disilyl ether | 10 | 8 | 8 | 14 | 11 |
| 12 | 1,3,5-Trisilacyclohexane | 8 | 8 | 9 | 7 | 8 |
| 13 | Benzaldehyde | 6 | 6 | 4 | 7 | 5 |
| 14 | 1,3-Difluorobenzene | 6 | 5 | 4 | 7 | 5 |
| 15 | 1,3,5-Trifluorobenzene | 5 | 5 | 4 | 4 | 5 |
| 16 | Neopentane | 4 | 5 | 4 | 4 | 4 |
| 17 | Furan | 7 | 8 | 5 | 7 | 5 |
| 18 | Naphthalene | 6 | 5 | 5 | 6 | 5 |
| 19 | 1,5-Difluoronaphthalene | 7 | 6 | 5 | 7 | 5 |
| 20 | 2-Hydroxybicyclopentane | 10 | 15 | 9 | 11 | 9 |
| 21 | ACHTAR10 | 8 | 12 | 8 | 9 | 9 |
| 22 | ACANIL01 | 8 | 8 | 7 | 7 | 6 |
| 23 | Benzidine | 8 | 9 | 9 | 9 | 9 |
| 24 | Pterin | 9 | 10 | 8 | 8 | 9 |
| 25 | Difuropyrazine | 8 | 9 | 6 | 8 | 7 |
| 26 | Mesityl oxide | 6 | 7 | 5 | 6 | 5 |
| 27 | Histidine | 14 | 19 | 16 | 16 | 16 |
| 28 | Dimethylpentane | 10 | 12 | 9 | 8 | 9 |
| 29 | Caffeine | 7 | 12 | 6 | 7 | 7 |
| 30 | Menthone | 10 | 13 | 12 | 11 | 13 |
|  | Sum | 207 | 240 | 185 | 215 | 198 |



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Figure 13.1: Initial geometries, contained in Baker's test set, comprising 30 small molecules. [268] A color code is given in the last row.

## 13 Pysisyphus

### 13.2.2 Hobzas S22 Test Set - Noncovalent Interactions

Compared to Baker's test set, the more challenging S22 set (Figure 13.2) was proposed by Hobza and coworkers. 423 Originally, it was compiled to provide reference energies at the Møller-Plesset perturbation theory (MP2) and $\operatorname{CCSD(T)}$ levels of theory in the complete basis set ( (CBS) limit, for benchmarking lower level computational methods. 424


Figure 13.2: Initial geometries, contained in the S 22 test set, comprising 22 dimers. 423 A color code is given in the last row.

S22 comprises 22 small molecule dimers, dominated by hydrogen bonds and dispersion interactions, see Table 13.4. Such interactions give rise to very flat PESs that are often insufficiently described by simple quadratic approximations (eq. 5.2) and eq. (5.6) on page 31. As molecular optimizers are usually based on quadratic models, searching SP; on flat PES; is especially challenging. For this reason, optimizing the S22 set should be a better test of optimizer robustness, compared to Baker's test set. 268

Energies and gradients were obtained from ORCA 4.2.1 at the RI-MP2/6-31G** level of theory. $388,425-429$ As no correlation fitting basis, explicitly optimized for $6-31 \mathrm{G}^{* *}$ is available, the def2-SVP/C correlation fitting basis set was used instead. 430 Tight SCF convergence criteria were enforced (keyword tightscf), to reduce numerical noise in the gradients. Neutral charge and singlet multiplicity were assumed throughout. Optimizations were carried out with pysisyphus, using the standard method outlined in section 13.2.1. Initial geometries of the S 22 set were taken from the original publication. 423]

Convergence was indicated when all absolute values of the gradient were equal or less than $4.5 \times 10^{-4} E_{\mathrm{h}} \cdot a_{0}^{-1}\left(\mathrm{rad}^{-1}\right)$, the RMS of the gradient was equal or less than $3.0 \times 10^{-4} E_{\mathrm{h}} \cdot a_{0}^{-1}\left(\mathrm{rad}^{-1}\right)$, all absolute values of the proposed step were equal or less than $1.8 \times 10^{-3} a_{0}(\mathrm{rad})$ and the RMS of the step was equal or less than $1.2 \times 10^{-3} a_{0}$ (rad). These thresholds conform to the defaults in the Gaussian program suite and will be denoted as Gaussian convergence criteria from now on. 384

Results for the S 22 set are given in Table 13.4 and compared to reference results published by Lindh. 417 Optimization success was judged according to the RMSD between the final geometries of this work and Lindh's work. Some geometries in the latter work exhibit a different atom ordering with respect to the original geometries of Hobza employed in this work, therefore, complicating the RMSD calculation. Before the RMSD calculations, the atoms were automatically reordered using the Hungarian method, trying to bring both atom orderings into maximum coincidence. 431, 432 Pysisyphus implements the Hungarian method according to Allen and Wagner. 433, 434 Cases 2, 8, $9,11,17$ and 18 exhibit high RMSD values ( $>0.1 a_{0}$, corresponding to $>5.3 \mathrm{pm}$ ), even after resorting. They were examined and aligned manually in the visualization system UCSF Chimera (Figure A.3). 435

Pysisyphus obtains SP; for all cases of the S22 set. All geometries, beside case 17 (benzene • water dimer) are virtually indistinguishable from the reference geometries, as evident from small RMSDs $\left(\ll 0.1 a_{0}\right)$ or perfect superposition in Figure A.3. The benzene - water dimer converged to a higher lying $\left(0.2 \mathrm{~kJ} \mathrm{~mol}^{-1}\right) \mathrm{SP} 417$ Pysisyphus required only 310 energy and gradient evaluations, whereas Lindh reported 358 evaluations. Without

## 13 Pysisyphus

case 17, pysisyphus required 260 evaluations, a marked improvement over Lindh's 323 evaluations. Overall, pysisyphus shows a promising performance for the S22 test set, even though one case converged to an energetically, slightly higher lying SP .

Table 13.4: Number of optimization cycles required to converge the S 22 test set (Figure 13.2 ) at the RI-MP2/6-31G** level of theory. 423 Lindh's results are taken from the restricted step RFO (RS-RFO) column of Table 3 in 417. Convergence to the correct geometry was ensured by calculating the RMSD with geometries published by Lindh. 417 , 436 Code to reproduce the results of this work is given in Listing A. 2 on page 140

| \# | Molecule | This work | Lindh $268$ | RMSD / $a_{0}$ |
| :---: | :---: | :---: | :---: | :---: |
| Hydrogen bonded complexes |  |  |  |  |
| 1 | ammonia dimer | 5 | 5 | 0.0034 |
| 2 | water dimer | 13 | 6 | $1.0579^{\text {a }}$ |
| 3 | formic acid dimer | 9 | 7 | 0.0045 |
| 4 | formamide dimer | 5 | 7 | 0.0090 |
| 5 | uracil dimer | 5 | 7 | 0.0105 |
| 6 | 2 -pyridoxine $\cdot 2$-aminopyridine | 16 | 15 | 0.0097 |
| 7 | adenine • thymine WC | 6 | 18 | 0.0981 |
| Dispersion dominated complexes |  |  |  |  |
| 8 | methane dimer | 7 | 20 | $1.4933{ }^{\text {a }}$ |
| 9 | ethene dimer | 3 | 5 | $2.8812^{\text {a }}$ |
| 10 | benzene $\cdot$ methane | 4 | 4 | 0.0054 |
| 11 | benzene dimer $\mathrm{C}_{2 \mathrm{~h}}$ | 9 | 7 | $0.1071^{\text {a }}$ |
| 12 | pyrazine dimer | 10 | 8 | 0.0144 |
| 13 | uracil dimer stack | 18 | 24 | 0.0196 |
| 14 | indole • benzene | 44 | 61 | 0.0247 |
| 15 | adenine • thymine stack | 33 | 29 | 0.0177 |
| Mixed complexes |  |  |  |  |
| 16 | ethene - ethine | 4 | 6 | 0.0077 |
| 17 | benzene . water | 50 | 35 | $0.1383^{\text {a,b }}$ |
| 18 | benzene • ammonia | 17 | 33 | $0.7585^{\text {a }}$ |
| 19 | benzene • hydrogen cyanide | 14 | 31 | 0.0245 |
| 20 | benzene dimer $\mathrm{C}_{2 \mathrm{v}}$ | 7 | 5 | 0.0218 |
| 21 | indole • benzene, T-shape | 12 | 4 | 0.0073 |
| 22 | phenole dimer | 19 | 21 | 0.0690 |
|  | Sum | 310 | 358 |  |

${ }^{\text {a }}$ Overlays of both geometries are given in Figure A. 3 at page 138
${ }^{\mathrm{b}}$ Converged to a higher lying SP

### 13.2.3 External Validation

The RFO implementation of pysisyphus was also employed by Ahuja et al., as one of multiple references, in an effort to benchmark a new reinforcement-learning-based optimization approach. 437] Starting from a set of perturbed organic molecules, they found their approach to be superior, requiring on average 94 cycles until convergence, whereas pysisyphus required 105. The next best optimizer (LBFGS) already required 129 cycles on average. In a second test set, pysisyphus seemed competitive with their new method, as it required the fewest cycles for 2 out of 12 cases, and the same cycle number for one case (table 1 in 437$) \cdot{ }^{2}$

### 13.2.4 Baker Transition State Test Set

Compared to the optimizations just described, where the energy was minimized along all Hessian eigenvectors, TS searches are more challenging, as they require energy maximization along one eigenvector. Foremost, this necessitates selecting an appropriate eigenvector and keeping track of it, along the optimization.

Besides a test set for GS minimizations, Baker also proposed a set, concerning TS searches. 315 This section investigates the performance of pysisyphus for the Baker TS test set, using two different approaches: Hessian based TS optimization and TS optimizations using only first-derivatives, by means of the DM (see chapter 10).

## Hessian-based Optimizer

All calculations were carried out using ORCA 4.2.1 at the HF/3-21G level of theory. 388 Singlet multiplicity was assumed throughout, except for cases 4,5 and 8 , which were calculated as doublets. Total charges were chosen according to Table 13.6. The initial (maximum) trust radius was decreased to $0.1 a_{0}$ (rad) ( $0.3 a_{0}$ (rad)). Analytical Hessian were calculated in the first optimization cycle and the imaginary mode belonging to the most negative eigenvalue, if present, was followed uphill. The Hessian was subsequently updated with Bofills formula. 240

Converging cases 10 and 11 (Figure 13.3), which start in the convex region of the PES, requires additional measures. Case 10 is supposed to yield a TS for the dissociation of $1,2,4,5$-tetrazine into two HCN molecules and $\mathrm{N}_{2}$, whereas case 11 is supposed to yield the TS for the isomerization of trans-butadiene to cis-butadiene. Even though various authors reported successful optimizations for one or both cases, they never discussed how the initial root was obtained. 232, 256, 315

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Figure 13.3: Initial geometries of the Baker TS test set, comprising 25 small molecules and ions. 315 A color code is given in the last row.

Pysisyphus offers an elegant way (keyword rx_mode), for initial root selection, requiring minimal user input. By specifying a list of primitive internals and an additional number, an approximate eigenvector is constructed. For tetrazine (Figure 13.4) rx_mode: [ [ [0, 2], 1], [ [1, 3], 1]] yields a vector, suitable for describing in-phase change of the two CN-bonds, stretched at the TS,
a)


Figure 13.4: Starting geometries for cases 10 and 11 from Baker's TS test set. a) 1,2,4,5tetrazine and b) trans-butadiene. Bonds, stretched at the TS, are crossed by a dashed line. Atom labels start at 0 , to be consistent with the actual pysisyphus input.

In this example, the exact signs of the additional numbers do not matter, as long as both are the same. Swapping one sign to a negative number would result in a vector, describing stretching of one and compression of the other bond. After normalization, overlaps between the approximate mode and Hessian eigenvectors are calculated and the eigenvector corresponding to the highest overlap is select as initial root (Listing 13.1). Care has to be taken that only overlaps within the non-redundant subspace are considered. In the present implementation, pysisyphus defines 28 primitive internals for tetrazine.

Listing 13.1: Python code for the rxmode root selection algorithm.

```
def root_select(eigvals, eigvecs, rx_mode, thresh=1e-8):
    # Construct approximate mode/eigenvector
    mode = np.zeros_like(eigvals) # Initialize zero-vector
    for prim_coord, number in rx_mode: # Loop over rx_mode input
        # Determine location of primitive internal in mode
        index = get_index_of_prim_coord(prim_coord)
        mode[index] = number # Set number at appropriate location
    mode /= np.linalg.norm(mode) # Normalize to unit length
    # Take absolute value of overlaps, because eigenvector sign is ambiguous
    ovlps = np.abs(np.einsum("ij,i}>>\textrm{j}", eigvecs, mode)
    # Only consider overlaps in non-redundant subspace by zeroing overlaps
    # in the redundant subspace.
    small_inds = np.abs(eigvals) < thresh
    ovlps[small_inds] = 0.0
    root = ovlps.argmax() # Select root with highest overlap
    return root
```

Using the aforementioned rx_mode input, the algorithm determines the 16th Hessian eigenvector for maximization. If a TSS for a bond breaking reaction is sought one cannot just blindly maximize along the lowest (first) Hessian eigenvector, as the lowest modes usually describe dihedrals motions. For the cis-trans isomerization of butadiene (case 11), specification of one dihedral is enough (rx_mode: [[[2, 0, 1, 3], 1] ], see

## 13 Pysisyphus

Figure 13.4). In this case, maximization is actually carried out along the first Hessian eigenvector, with a non-vanishing eigenvalue.

Overall, the presented approach allows easy, initial root selection. By controlling the sign of the additional number, even complicated phase relationships between different internal coordinates can be expressed, if required.

Pysisyphus implements three TS optimizers that utilize Hessian information: restricted step partitioned RFO (RS-PRFO), restricted step image method RFO (RS-IRFO), and the trust-region image method (TRIM) (see eq. (9.7) on page 62 and eq. (5.7) on page 32 ). TRIM combines the IM with step determination through eq. (5.5). [256, 438 All three optimizers were applied to Baker's TS test set. Additionally, the use of line searches, that is maximization in one subspace and minimization in the other, was investigated for RS-PRFO.

Contrary to the published value of $-242.25529 E_{\mathrm{h}}$, a lower final energy of $-242.25696 E_{\mathrm{h}}$ is used as reference energy for case 22 . The former value corresponds to a planar, symmetry constrained solution, which relaxes to the latter value without constraints.

Results for the different optimizers are shown in Table 13.5. RS-PRFO, with enabled line search is the only optimizer that converged all 25 investigated test cases, requiring 325 cycles. Disabling line search yields a failure for case 2 and a total of 326 cycles are required to converge the remaining 24 test cases. RS-IRFO fails for case 15 and requires a total of 297 cycles. TRIM fails for five cases (10, 11, 14, 16, 22), while the remaining cases require 254 cycles.

Considering only the 18 cases that converged for all optimizers (excludes $2,10,11,15$, $14,16,22$ ), then RS-IRFO needs the fewest cycles (221), closely followed by RS-PRFO with enabled line search (224). Disabling line search increases the required cycles to 238 for RS-PRFO, TRIM requires a total of 227 cycles.

Table 13.5: Performance of different TS optimizers in pysisyphus for converging Baker's TS test set. The numbers in parentheses give the required cycles, to converge the 18 cases that did not fail with any optimizer (excludes cases $2,10,11,14$, 15, 16 and 22), recall Figure 13.3 .

| Optimizer | Converged | Failed cases | Cycles |
| :--- | :---: | :---: | :---: |
| RS-PRFO with line search | 25 | - | $325(224)$ |
| RS-PRFO without line search | 24 | 2 | $326(238)$ |
| RS-IRFO | 24 | 15 | $297(221)$ |
| TRIM | 20 | $10,11,14,16,22$ | $254(227)$ |

Table 13.6: Number of optimization cycles required to converge Baker's TS test set (Figure 13.3), according to Baker's criteria (see section 13.2 . 315 Convergence to the correct geometry was ensured by comparing the final energy to the published values. Bakken's results correspond to their most efficient method, as presented in Table XV in [232]. Besalu's results correspond to the RS-PRFO column in Table 1 in 256], whereas Baker's original results are taken from Table IV in 315]. Code to reproduce the results of this work is given in Listing A. 3 on page 141 .

| \# | Reaction | This work | Bakken $232$ | Besalu $256$ | Baker $268$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $\mathrm{HCN} \rightleftarrows \mathrm{HNC}$ | 12 | 9 | - | 9 |
| 2 | $\mathrm{HCCH} \rightleftarrows \mathrm{CCH}_{2}$ | 13 | 7 | - | 8 |
| 3 | $\mathrm{H}_{2} \mathrm{CO} \rightleftarrows \mathrm{H}_{2}+\mathrm{CO}$ | 13 | 13 | - | 13 |
| 4 | $\mathrm{CH}_{3} \mathrm{O} \rightleftarrows \mathrm{CH}_{2} \mathrm{OH}$ | 9 | - | - | 11 |
| 5 | cyclopropyl ring opening | 20 | - | - | 11 |
| 6 | bicyclo[1.1.0]butane | 8 | - | - | 7 |
|  | ring opening |  |  |  |  |
| 7 | bicyclo[1.1.0]butane | 11 | - | - | 9 |
|  | ring opening |  |  |  |  |
| 8 | 1,2-migration (formyloxy)ethyl | 10 | - | - | 26 |
| 9 | butadiene + ethylene $\rightleftarrows$ cyclo- | 18 | 15 | x | 13 |
|  | hexene |  |  |  |  |
| 10 | s-tetrazine $\rightleftarrows 2 \mathrm{HCN}+\mathrm{N}_{2}$ | 11 | 9 | 32 | 14 |
| 11 | trans-butadiene $\rightleftarrows$ cis-butadiene | 10 | - | - | 8 |
| 12 | $\mathrm{CH}_{3} \mathrm{CH}_{3} \rightleftarrows \mathrm{CH}_{2} \mathrm{CH}_{2}+\mathrm{H}_{2}$ | 11 | 8 | 15 | 12 |
| 13 | $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{~F} \rightleftarrows \mathrm{CH}_{2} \mathrm{CH}_{2}+\mathrm{HF}$ | 15 | 15 | - | 11 |
| 14 | vinyl alcohol $\rightleftarrows$ acetaldehyde | 23 | 19 | - | 13 |
| 15 | $\mathrm{HCOCl} \rightleftarrows \mathrm{HCl}+\mathrm{CO}$ | 15 | - | - | 12 |
| 16 | $\mathrm{H}_{2} \mathrm{O}+\mathrm{PO}_{3}{ }^{-} \rightleftarrows \mathrm{H}_{2} \mathrm{PO}_{4}{ }^{-}$ | 15 | - | - | 23 |
| 17 | Claisen rearrangement | 13 | - | 36 | 8 |
| 18 | $\mathrm{SiH}_{2}+\mathrm{CH}_{3} \mathrm{CH}_{3} \rightleftarrows \mathrm{SiH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{3}$ | 12 | 12 | - | 7 |
| 19 | $\mathrm{HNCCS} \rightleftarrows \mathrm{HNC}+\mathrm{CS}$ | 13 | 13 | - | 10 |
| 20 | $\mathrm{HCONH}_{3}^{+} \rightleftarrows \mathrm{NH}_{4}^{+}+\mathrm{CO}$ | 15 | - | 14 | 10 |
| 21 | rotational TS in acrolein | 5 | 5 | - | 4 |
| 22 | $\mathrm{HCONHOH} \rightleftarrows \mathrm{HCOHNHO}$ | 14 | 5 | 10 | 5 |
| 23 | $\mathrm{HNC}+\mathrm{H}_{2} \rightleftarrows \mathrm{H}_{2} \mathrm{CNH}$ | 7 | 8 | 22 | 8 |
| 24 | $\mathrm{H}_{2} \mathrm{CNH} \rightleftarrows \mathrm{HCNH}_{2}$ | 23 | 18 | 20 | 14 |
| 25 | $\mathrm{HCNH}_{2} \rightleftarrows \mathrm{HCN}+\mathrm{H}_{2}$ | 9 | 8 | 56 | 9 |
|  | Sum | 325 | 162 | 205 | 275 |

A more detailed listing of the RS-PRFO results with enabled line search is given in Table 13.6. Compared with Baker's original work, the most efficient method in this work requires more cycles ( 325 cycles versus 275 cycles). 315] However, the present implementation can be used mostly in a black box way, whereas Baker's results were obtained with manually constructed Z-matrices. Comparisons with other published results is difficult, as often only results for a subset of the 25 cases are reported. In contrast to the present work, Bakken found TRIM to provide the best performance for 15 tested cases. Similarly, Besalu found the IM based optimizers more stable (see Table 13.5 . 256 But it must be kept in mind that both studies employed only a subset of Baker's TS test set. Considering only cases that converged for all optimizers in this work, then RS-IRFO also compares favorably to RS-PRFO (221 versus 224 cycles).
In summary, RS-PRFO with enabled line searches shows robust performance on Baker's TS test set. In case, a TS search is started from a convex region on the PES, pysisyphus offers a simple way to select an appropriate root for uphill following, by means of the rx.mode keyword.

## Dimer method

While utilizing Hessian information enables efficient and effective TS optimization, computational demand of obtaining and handling Hessian matrices (diagonalization/inversion, see chapter (10) grows quickly with system size. (439 Recently, Hermes showed that QNmethods easily loose track of the leftmost Hessian eigenvector and accumulate significant error in the Hessian, when standard update methods (SR1, BFGS) are employed. [440] Only one Hessian calculation at the beginning may be sufficient to converge well-behaved TS optimizations, as in Baker's TS set. But more challenging TS searches may require additional, intermediate Hessian calculations, to circumvent aforementioned error accumulation, which may become prohibitive for big systems.

Pysisyphus implements the DM, which requires only gradients, thus avoiding the problems outlined above. In every optimization cycle, the dimer is converged onto the leftmost Hessian eigenvector, up to a prescribed tolerance, by means of dimer rotations. Translation steps move the whole dimer closer towards the TS,

Energies and gradients were calculated at the same level of theory as previously outlined. The DM (chapter 10) was employed for converging dimer orientation $\boldsymbol{N}$ to the leftmost Hessian eigenvector, corresponding to the smallest eigenvalue, in every optimization cycle. A reasonable choice for the initial dimer orientation $\boldsymbol{N}$ is crucial for a successful outcome of the TS search. Starting from optimized TSk, obtained with a Hessian based optimizer, IRC were integrated towards the educts and products, connected by the TSk. Initial
guesses for $\boldsymbol{N}$ were calculated as normalized difference between the Cartesian coordinates of aligned educt and product geometries. If no initial $\boldsymbol{N}$ is provided to pysisyphus, a random orientation is chosen. In general, picking a random orientation should be avoided, as the dimer rotations may fail to converge for bad initial guesses. An example for this is found in Table 1 of 312 , where even simple cases like the HCN isomerization failed to converge for a random initial orientation.

A dimer separation of $\Delta R=0.0189 a_{0}$ was chosen (Figure 10.1 on page 63). Rotations were carried out as described by Kästner, with a tolerance of $\phi_{\text {tol }}=5^{\circ}$ and steps obtained from a LBFGS optimizer. 312 Performance of LBFGS has proven superior to the CG and steepest descent approaches outlined elsewhere. 290, 311 For cases 10 and 11, starting in the convex PES region, dimer rotations were disabled until the convex region was left and the curvature became negative. In contrast to the original proposal by Jonsson, 311 it was critical for the convergence of case 10, to use parallel and perpendicular force components, even in the convex PES region, recall eq. 10.3). These choices correspond to the default settings for the DM in pysisyphus.

Dimer translation was carried out by a (preconditioned) LBFGS optimizer. Preconditioners $\boldsymbol{P}$ were constructed as outlined in chapter 7, using Lindh's model Hessian. 241, 286 All stretches, bends and dihedrals, automatically defined by the internal coordinate setup algorithm in pysisyphus, are used for constructing $\boldsymbol{P}$. The present implementation utilizes the sparse linear algebra capabilities of the SciPy package. 441 As some of the molecules are quite small, using all primitive internals resulted in dense preconditioners. For large molecules, care has to be taken for $\boldsymbol{P}$ to be actually sparse, otherwise the computational benefits of preconditioning may not be realized.

Gas-phase Hessians are at least six fold degenerate, so $\boldsymbol{P}$ was regularized $(\boldsymbol{P} \rightarrow \boldsymbol{P}+c \boldsymbol{I})$ with $c=0.103 E_{\mathrm{h}} a_{0}^{-2}$. 286 If the absolute value of any element in the proposed step exceeded $0.25 a_{0}$, the whole step was scaled accordingly, so the absolute values are equal to or less than the threshold. Cartesian coordinates and Baker's convergence criteria were used throughout.

Results obtained with and without preconditioner are given in Table 13.7. The DM in combination with (preconditioned) LBFGS, as implemented in pysisyphus, managed to converge 24 out of 25 cases from Baker's TS set. Only case 11 failed, regardless of whether a preconditioner was used. Employing preconditioning is clearly superior, as only 358 optimization cycles (dimer translations) are needed, while 577 cycles are required without preconditioner. Disregarding the additional energy and gradient evaluations required for dimer rotations, the present implementation nearly rivals the performance of optimizations that utilized Hessian information (325 cycles, Table 13.6). For a total of

1092 evaluations, an average of 46 energy and gradient evaluations are needed to converge 24 cases with preconditioner. Without preconditioner, the total (average) is significantly higher 1706 (71).

Table 13.7: Number of optimization cycles required to converge Baker's TS test set (Figure 13.3) according to Baker's criteria (see section 13.2), using the DM and (preconditioned) LBFGS, 234, 251, 286, 315 Required energy and gradient evaluations are given in parentheses. Convergence to the correct geometry was ensured by comparing the final energy to the published values. Code to reproduce these results is given in Listing A. 4 on page 142 .

| \# | Reaction | Preconditioning |  |
| :---: | :---: | :---: | :---: |
|  |  | yes | no |
| 1 | $\mathrm{HCN} \rightleftarrows \mathrm{HNC}$ | 12 (42) | 13 (44) |
| 2 | $\mathrm{HCCH} \rightleftarrows \mathrm{CCH}_{2}$ | 13 (45) | 11 (36) |
| 3 | $\mathrm{H}_{2} \mathrm{CO} \rightleftarrows \mathrm{H}_{2}+\mathrm{CO}$ | 10 (32) | 12 (38) |
| 4 | $\mathrm{CH}_{3} \mathrm{O} \rightleftarrows \mathrm{CH}_{2} \mathrm{OH}$ | 10 (25) | 10 (24) |
| 5 | cyclopropyl ring opening | 14 (46) | 32 (95) |
| 6 | bicyclo[1.1.0]butane | 12 (40) | 25 (79) |
| 7 | ring opening bicyclo[1.1.0]butane | 14 (49) | 42 (131) |
|  | ring opening |  |  |
| 8 | 1,2-migration (formyloxy) ethyl | 18 (47) | 34 (73) |
| 9 | butadiene + ethylene $\rightleftarrows$ cyclo- | 17 (57) | 31 (105) |
| 10 | hexene <br> s-tetrazine $\rightleftarrows 2 \mathrm{HCN}+\mathrm{N}_{2}$ | 12 (43) | 21 (63) |
| 11 | trans-butadiene $\rightleftarrows$ cis-butadiene | ( | (63) |
| 12 | $\mathrm{CH}_{3} \mathrm{CH}_{3} \rightleftarrows \mathrm{CH}_{2} \mathrm{CH}_{2}+\mathrm{H}_{2}$ | 12 (29) | 16 (37) |
| 13 | $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{~F} \rightleftarrows \mathrm{CH}_{2} \mathrm{CH}_{2}+\mathrm{HF}$ | 14 (35) | 19 (45) |
| 14 | vinyl alcohol $\rightleftarrows$ acetaldehyde | 17 (51) | 29 (80) |
| 15 | $\mathrm{HCOCl} \rightleftarrows \mathrm{HCl}+\mathrm{CO}$ | 12 (50) | 29 (114) |
| 16 | $\mathrm{H}_{2} \mathrm{O}+\mathrm{PO}_{3}{ }^{-} \rightleftarrows \mathrm{H}_{2} \mathrm{PO}_{4}{ }^{-}$ | 15 (50) | 31 (104) |
| 17 | Claisen rearrangement | 19 (58) | 38 (113) |
| 18 | $\mathrm{SiH}_{2}+\mathrm{CH}_{3} \mathrm{CH}_{3} \rightleftarrows \mathrm{SiH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{3}$ | 23 (53) | 43 (109) |
| 19 | $\mathrm{HNCCS} \rightleftarrows \mathrm{HNC}+\mathrm{CS}$ | 22 (67) | 29 (90) |
| 20 | $\mathrm{HCONH}_{3}{ }^{+} \rightleftarrows \mathrm{NH}_{4}^{+}+\mathrm{CO}$ | 16 (61) | 21 (74) |
| 21 | rotational TS in acrolein | 11 (36) | 22 (69) |
| 22 | $\mathrm{HCONHOH} \rightleftarrows \mathrm{HCOHNHO}$ | $27(75)^{\text {a }}$ | $18(56)^{\text {b }}$ |
| 23 | $\mathrm{HNC}+\mathrm{H}_{2} \underset{\mathrm{H}}{\leftrightarrows} \mathrm{H}_{2} \mathrm{CNH}$ | 12 (30) | 16 (42) |
| 24 | $\mathrm{H}_{2} \mathrm{CNH} \rightleftharpoons \mathrm{HCNH}_{2}$ | 18 (52) | 22 (56) |
| 25 | $\mathrm{HCNH}_{2} \rightleftarrows \mathrm{HCN}+\mathrm{H}_{2}$ | 8 (19) | 13 (29) |
| Sum |  | 358 (1092) | 577 (1706) |

${ }^{\text {a }}$ Final energy $-242.25696 E_{\mathrm{h}}$, relaxed TS geometry.
${ }^{\text {b }}$ Final energy $-242.25530 E_{\mathrm{h}}$, published value. 315

Comparison to other implementations is difficult. Heyden also used the DM for Baker's TS set, but the starting geometries were obtained as interpolated HEI from a previous GSM run. [290 Kästner directly started from the test set geometries, but used a random initial dimer orientation $\boldsymbol{N}$, resulting in failures for several cases. He reported between 2211 to 2689 energy and gradient evaluations to converge on average 22 of the 25 test cases. 312 Shang et al. employed a different level of theory ${ }^{3}$, direct dimer rotation using constrained Broyden minimization, a more complex force definition, and different convergence criteria. They achieved an average of 35 energy and gradient evaluations, while converging all 25 cases, which appears superior to the present implementation. 313 But is has to be noted that looser convergence criteria were employed in this study, as convergence was already indicated, when the maximum absolute value of the gradient entries fell below $6.9 \times 10^{-3} E_{\mathrm{h}} a_{0}{ }^{-1}$.

### 13.2.5 Chain-Of-States Test Set

Pysisyphus offers several ways for obtaining TS guesses, e.g., by means of COS methods like NEB or the GSM Here, the performance of the GSM implementation in pysisyphus is investigated for a diverse test set, comprising 20 reactions (Figure 13.5).
The set is mostly based on a set presented by Birkholz and Schlegel. [265, 315, 442, 444 It includes educt, TS and product geometries obtained at the PM6 445 and B3LYP/6$31 \mathrm{G}(\mathrm{d}, \mathrm{p})$ level of theory. $446-449$ For the present benchmark, the given TSk were reoptimized at the GFN2-XTB level of theory, using pysisyphus. Subsequently, IRC; were integrated towards educts and products of the reactions, to obtain a consistent set of structures. The IRC endpoints were then employed as initial geometries for the GSM, Two cases were excluded from the original set: the $\mathrm{S}_{\mathrm{N}} 2$ reaction of fluoride and methyl chloride (case 19) does not have a TS at the GFN2-XTB level of theory, which was also noted by Birkholz for the B3LYP/6-31G(d,p) PES. Similarly, explorative calculations for the addition of difluorocarbene to ethylene (case 9) revealed the same absence of a TS at the tight-binding level of theory. These cases were replaced by two hydride shift reactions (cases 14 and 30) from the MOBH35 test set (metal-organic barrier heights with 35 members). 450

The ability of the GSM to yield useful TS guesses for the MOBH35 set was recently probed by Grimme, using the original GSM implementation of Zimmerman. [300, 451] Unfortunately, no details on the GSM calculations (number of images, optimization algorithm, threshold for growing images, among others) were reported by Grimme.

[^8]$1 \mathrm{C}_{2} \mathrm{H}_{4}+\mathrm{N}_{2} \mathrm{O} \rightarrow\left[+[-\mathrm{O}-\mathrm{N}]_{-}^{\ddagger} \rightarrow\left[\begin{array}{c}\mathrm{O} \\ \mathrm{NN}^{\prime N}\end{array}\right.\right.$


13

$3 \mathrm{HCN} \rightarrow\left[\underset{\mathrm{C}^{\prime}=\mathrm{S}}{\stackrel{\mathrm{H}}{=}}\right]^{\ddagger} \rightarrow \mathrm{HNC}$



5

16

$6 \Downarrow / / \rightarrow\left[\left[_{[-i}^{-i}\right]\right]^{\ddagger} \rightarrow \square$

7

18

$8 \leftrightarrows+\mathrm{C}_{2} \mathrm{H}_{4} \rightarrow \square$

19




Figure 13.5: Investigated reactions in the COS test set.

Some geometries in the present set differ significantly from the originals. Notably, the educts for the two Cope-rearrangements (cases 4 and 16 in Figure 13.5) exhibit a different conformation (see Figure 13.6). When a GSM is initiated from the original conformers, subsequent optimization of the HEI yields a wrong TS. If the GSM is initiated from the IRC endpoints, the correct TS is obtained. Starting from reasonable conformers is crucial for the successful outcome of COS optimizations, as already noted by Birkholz and others. 442, 451
a)

b)

c)

d)


Figure 13.6: Different educt ( $\mathrm{a}, \mathrm{c}$ ) and TS geometries ( $\mathrm{b}, \mathrm{d}$ ) for the Cope rearrangement of 1,4 -hexadiene (case 4 in the COS test set). a) Conformer, as published in [442], b) (wrong) TS, obtained from GSM started from a) and subsequent TS optimization. c) Conformer used in this benchmark and d) correct TS, obtained when starting GSM from the conformer c).

Likewise, the high symmetry of HCN and CNH (case 3) was broken slightly, to allow successful interpolation. A similar approach was recently discussed by Palenik. 282 Interestingly, correct TSß are obtained by the connectivity transition state (CTS) approach, also available in pysisyphus, even when starting from the original conformers (see Listing A.6). (442
Energies and gradients were obtained at the GFN2-XTB level of theory using XTB 6.3.3. 212] Singlet multiplicity and neutral charge was assumed throughout, except for case 15 (charge -1). Initial educt and product geometries were preoptimized for a maximum of 5 cycles, using the standard method in pysisyphus (see section 13.2.1). Starting from the preoptimized geometries, a string in DLC was grown to a total of 12 images ( 10 inner images). 300 New images were added, when the RMS of the perpendicular force acting on the frontier images fell below $0.05 E_{\mathrm{h}} \cdot a_{0}^{-1}\left(\mathrm{rad}^{-1}\right)$. String reparametrization occurred every other cycle during the growth phase, and every third cycle after the string was fully grown. Originally, Behn and Zimmerman proposed a reparametrization approach, where every image is shifted many times along the string using very small steps. 299, 300 Pysisyphus uses a different approach: The required step is estimated and taken directly for every image that needs reparametrization. This is repeated iteratively, until the desired string parametrization is achieved. Usually, only one or two steps are needed per image and thus, excessive internal-Cartesian back-transformations are avoided. While still in the growth phase, the DLC were reset after every reparametrization, and a new active set $\boldsymbol{U}$ was calculated for every image (see section 6.4). Resetting of DLC; was stopped after the string was fully grown, to not interfere with the LBFGS optimizer, which relies on a history of previous steps and gradient differences.
When the RMS of the perpendicular force dropped below $0.0075 E_{\mathrm{h}} \cdot a_{0}{ }^{-1}\left(\mathrm{rad}^{-1}\right)$ and the string was fully grown, the HEI was converted to a CI aimed at yielding an improved TS guess. During the growth phase, string optimization was achieved via
steepest descent steps after reparametrizations and CG (Fletcher-Reeves) otherwise. [234, 300] After the string was fully grown, optimization continued with LBFGS, A history of 10 previous cycles was retained and a doubly damped BFGS update was used (Algorithm 2 on page 36). [255] To the best knowledge of the author, a doubly damped BFGS update was not yet applied in the context of COS optimization. The whole step vector was scaled accordingly, if the absolute value of any element in it exceeded $0.2 a_{0}$ (rad). All optimizers were employed in their global variant, that is the whole COS was optimized as super-molecule, instead of optimizing every image on its own. While this makes no difference for steepest descent, as image-coupling is neglected, Sheppard found global LBFGS superior over per-image LBFGS 306
GSM convergence was signaled, when the RMS of the perpendicular force fell below or was equal to $0.005 E_{\mathrm{h}} \cdot a_{0}^{-1}\left(\mathrm{rad}^{-1}\right)$. In the present study, GSM is employed to obtain a reasonable TS guess, so there is no need for overly tight convergence. Finally, after convergence an interpolated HEI was obtained by means of cubic splining, which is superior over just picking the HEI without interpolation. 309 It was ensured that all primitive bond stretches that are either present in the educt or the product geometry, are also defined for the HEI derived TS guess. In the following and in the context of COS optimizations, this approach will be referred as the standard method.

Two of the twenty cases required slight adjustments, to ensure convergence: The GSM for case 3 (HCN isomerization) was optimized in Cartesian coordinates instead of DLC and for case 20, the convergence criterion was tightened to $0.003 E_{\mathrm{h}} \cdot a_{0}^{-1}\left(\mathrm{rad}^{-1}\right)$.

Subsequent TS optimizations were started from the splined HEI in RIC, using RS-IRFO, An exact Hessian was calculated before the first optimization cycle and the eigenvector for energy maximization was selected according to its overlap with the HEI tangent. The trust radius was restricted to $0.5 a_{0}(\mathrm{rad})$, to avoid overly long steps, which can result in a loss of the correct eigenvector. 440 Convergence of the TS optimization was indicated when the standard Gaussian criteria were fulfilled (see section 13.2.2). A Hessian was calculated at the final geometry to verify the nature of the SP as first-order TS Success of the TS optimizations was verified by calculating the RMSD with a reference TS geometry.

Results for the COS test set are summarized in Table 13.8. The correct TS is obtained for all twenty cases in the test set by the standard method with only two minor adjustments. Similarly, all previous GSM optimizations converged. String optimizations of cases 11 (formaldehyde decomposition) and 14 ( $\mathrm{H}_{2}$ addition to formaldehyde) required the most cycles (26 and 22) until convergence, even though they belong to the smallest test cases. Overall, a total of 204 GSM cycles are required, each cycle comprising multiple energy and gradient evaluations. Subsequent TS optimizations required a total of 228 cycles
(minimum 2, median 10, maximum 26 cycles). In nearly all cases, the overlap of the HEI tangent with the eigenvector, belonging to the most negative eigenvalue of the HEI Hessian, was close to 1.

Disabling LBFGS and relying only on steepest descent and CG steps for string optimization, leads to a failure in the subsequent TS optimizations of cases 11 and 19. For case 11, already the GSM optimization failed, as it did not converge in 50 cycles. For case 19 it required only 7 cycles, but converged to a wrong TS, as the eigenvector selection fails and the vector belonging to the second, most negative eigenvalue is chosen for uphill following. Compared to the standard method disabling LBFGS results in 54 additional COS optimization cycles ( 225 vs. 171) , if the two failing cases are neglected.

Turning off the doubly damped BFGS update leads to 45 additional COS optimization cycles (249), compared to the standard method. COS optimization of case 11 terminates prematurely, as the coordinates of two images became too similar. The number of required TS optimization cycles increases to 278, corresponding to 50 additional cycles, compared to the standard method. Nonetheless, all 20 TS optimizations converged to the correct TSF.

Only the combination of LBFGS and doubly damped update appears to sustain long optimization step lengths (maximum absolute element $0.2 a_{0}(\mathrm{rad})$ ). The influence of whether LBFGS with double damping is enabled or not, could probably be decreased by using smaller step lengths, at the cost of additional energy and gradient evaluations.
The present setup also allows to investigate the influence of the chosen primitive internals at the TS guess, on the outcome of the TS optimization. When bends and dihedrals, defined at educt and product geometries, are also included in the coordinate set at the TS guess, 2 additional cycles (230) are required compared to the standard method that only considers the bonding skeleton. Fully neglecting, which primitives are defined at educt and product, leads to deteriorated TS optimization performance, as overall 256 cycles are required. Care must be taken, not to include superfluous linear bends at the TS guess, which can lead to spurious eigenvectors with strongly negative eigenvalues, as observed for case 1 in Figure 13.7. Even though the Cartesian Hessian at the TS guess has one significant imaginary mode, retaining the linear bend results in two imaginary modes in RIC (Table 13.9). If the linear bend is omitted at the TS guess, only one imaginary mode is present. At least for this case, the spurious eigenvector is easily detected, as its overlap with the HEI tangent is basically zero (Table 13.9).

In summary, the present GSM implementation offers a robust way to obtain TS guesses in a black-box fashion, while requiring only educt and product geometries as input. By


|  |  |  | 877 | Ø07 | UnS |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $78^{\circ} 0$ | $\varepsilon$ | 9980＊0 | ¢\％ | 6 |  | 07 |
| L9．0 | $\varepsilon$ | 9910．0 | 07 | $L$ |  | 6 I |
| L6．0 | I | \＆LZ0＊0 | 8 | 9 | әиә！̣pełnq of uo！̣！ppe ${ }^{\text {}} \mathrm{OS}$ | 8I |
| \＆6．0 | $\zeta$ | L800＊0 | ZI |  |  | LI |
| 76.0 | $\checkmark$ | 五0000 | 97 | II |  | 9I |
| 96.0 | I | $0000^{\circ} 0$ |  | 9 |  | GI |
| 96.0 | $\checkmark$ | 2T00＊0 | OI | \％7 | әрКчәргешлол от ио！т！рре ${ }^{7} \mathrm{H}$ | 玵 |
|  | $\checkmark$ | も¢L0．0 | GI | 0I |  | \＆I |
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| 96.0 | I | $0900{ }^{\circ}$ | \＆I | G |  | 0I |
| $06^{\circ}$ | E | 9800＊0 | 㕵 | 6 |  | 6 |
| 86.0 | I | $9100{ }^{\circ}$ | 历 | 9 |  | 8 |
| 86.0 | $\checkmark$ | $6800^{\circ} 0$ | OL | 9 |  | $L$ |
| 96.0 | I | LL00．0 | 9 | 0I |  | 9 |
| $66^{\circ} 0$ | I | 9000＊0 | 9 | 8 |  | 9 |
| L6．0 | $\checkmark$ | 2100＊0 | 6 I | 6 |  | 万 |
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| ¢ејләл O | $N$ | $0 p \longdiv { \text { USTVY } }$ | SL | WSO | иоџəวеәу | \＃ |
|  |  |  |  |  |  |  |

[^9]Table 13.9: Negative eigenvalues and overlaps of the corresponding eigenvectors with the HEI tangent, depending on whether a linear bend is defined at the HEI or not (see Figure 13.7).


Figure 13.7: Case 1 from the COS test set. Whereas a linear bend is a suitable coordinate for the educt a) $\left(\varangle_{\mathrm{ONN}}=179.6^{\circ}\right)$, it isn't suitable anymore at TS geometry b) $\left(\varangle_{\text {ONN }}=143.7^{\circ}\right)$. Including the linear bend at b$)$ leads to a spurious imaginary mode.
taking into account the union of bonds defined at educt and product geometries for the TS guess, improved optimization performance is achieved. LBFGS combined with double damping appears to be a promising strategy for COS optimization, whereas using plain LBFGS seems less suitable.

### 13.3 Excited State Optimization

The performance of pysisyphus for optimizing and tracking ES; is verified against two previously published systems: the optimizations of the first two excited singlet states of cytosin $(\mathbf{C y})$ and the $S_{9}$ of the ruthenium nitrosyl complex cis- $(\mathrm{Cl}, \mathrm{Cl})\left[\mathrm{RuCl}_{2}\right](\mathrm{NO})($ tpy $\left.)\right]^{+}$ (RuNO). 223 Additionally, two exemplary optimizations of excited triplet states for two platinum complexes $\left(\left[\mathrm{Pt}(\mathrm{C} \equiv \mathrm{C}-\mathrm{Tol})_{2}(\right.\right.$ phen-TPA-R $\left.\left.)\right] ; \mathrm{R}=\mathrm{OMe}, \mathrm{CN}\right)$ investigated by Shillito et al., are presented. [8] Structural formulas for all compounds are given in Figure 13.8 .

All optimizations were carried out in RIC using the standard method outlined in section 13.2.1, with enabled GDIIS and quartic line search. Initial Hessians were estimated according to Fischer and updated with the BFGS formula. 234, 242, ES; were tracked by the WFO approach. The reference cycle for the overlap calculation was updated using
a)

c)

b)

d)


Figure 13.8: Compounds investigated in this chapter. a) 1H-amino-keto tautomer of cytosin ( $\mathbf{C y}$ ), b) reaction of $N$-benzylbenzenesulfonamide ( $\mathbf{1 b}$ ) to biphenyl $(\mathbf{2 b})$, c) ruthenium nitrosyl complex cis $-(\mathrm{Cl}, \mathrm{Cl})\left[\mathrm{RuCl}_{2}\right](\mathrm{NO})($ tpy $\left.)\right]^{+}$ $($ RuNO $)$ and d) two platinum complexes $\left[\mathrm{Pt}(\mathrm{C} \equiv \mathrm{C}-\mathrm{Tol})_{2}(\right.$ phen $\left.-\mathrm{TPA}-\mathrm{R})\right]$ (Pt-TPA-OMe with $\mathrm{R}=\mathrm{OMe}, \mathbf{P t - T P A}-\mathrm{CN}$ with $\mathrm{R}=\mathrm{CN})$.
the adaptive formalism described in section 4. Convergence was indicated when the Gaussian criteria were fulfilled (see section 13.2.2).

### 13.3.1 Cytosin

As one of the five canonical nucleobases making up nucleic acids (NA;), the photochemistry of cytosin received considerable scientific interest.[452-458] Absorption of ultraviolet light through nucleic acids may lead to their degradation and can induce harmful processes like cell apoptosis. [458, 459] Extensive photo-degradation of nucleic acids is avoided by very fast nonradiative decay, as the excitation energy is dissipated in the surroundings. 460 462 Cytosin shows fast IC] from a bright $\pi \rightarrow \pi^{*}$ to a dark $\mathrm{n} \rightarrow \pi^{*}$ state. [456] Its small size makes it suitable for benchmarking ES tracking algorithms, as shown by García et al. 223 Here, similarly the performance of pysisyphus for optimizing the first two singlet ES of the 1H-amino-keto-tautomer of cytosin (Cy, Figure 13.8a) was investigated.

All calculations at the DFT and TD-DFT level of theory were carried out via Turbomole V7.3. 122 Excited state properties (such as vertical excitation energies and oscillator strengths) for the first two excited singlet states were obtained from TD-DFT calculations at the PBE0/def2-SVP 381,463 level of theory, using the starting geometry provided by García. 223 The RIJ approximation with corresponding auxiliary basis set was used to speed up the calculations. 389,464

Optimization results are shown in Figure 13.9, Pysisyphus successfully optimizes both $E S$ to SP; while correctly tracking their diabatic character. In both cases,
the state crossings occur shortly after the start of optimization. The charge density differences (CDDs) along the optimization paths stay nearly constant. García reported 160 steps for the $S_{1}$ optimization and ca. 324 steps for the $S_{2}$ optimization, until a SP was reached. In pysisyphus only 9 (10) gradient calculations are needed for the $S_{1}\left(S_{2}\right)$ optimization, showing a remarkable, although unsurprising, improvement. Enhanced performance of internal coordinates in combination with an optimizer, utilizing Hessian information, was already recognized many years ago by Schlegel. 75


Figure 13.9: Electronic energies of the ground and the first two excited singlet states of cytosin ( $\mathbf{C y}$ ) along the optimization of the a) bright $S_{1}$ and b) dark $S_{2}$, with respect to the ground state energy at the first cycle. The tracked excited state is indicated by a dashed black line. Every marker corresponds to an optimization cycle and missing markers indicate root flips. CDDs illustrate the excited state character at the start and the end of the optimization; electron density is excited from blue to red.

### 13.3.2 Ruthenium Nitrosyl Complex

Besides carbon monoxide and hydrogen sulfide, nitric oxide (NO) is an important physiologically relevant gasotransmitter. 465 In the human body, NO acts as strong
vasodilator and can activate macrophages.|466-470] Even long before its important physiological role was understood, NO-precursor drugs like nitroglycerin and amyl nitrate were in wide use. 471,472 Direct and targeted application of gaseous NO in humans is difficult, resulting in the development of transition metal complexes, where the NO release can be triggered by light. 473 - 478 Transition metal nitrosyls exhibit a complex photochemistry, as NO may undergo either photo-release or photo-isomerization. $479-481$ By computing the relaxation pathways of ESk, the photochemistry of transition metal nitrosyls can be rationalized. $482-484$

As a challenging example, García recently presented the optimization of the bright $S_{9}$ of RuNO (see Figure 13.8:) using NTO overlaps. 223 Similar to the Cy example, the performance of pysisyphus in the scope of ES optimizations in the singlet manifold is investigated.

All calculations at the DFT and TD-DFT level of theory were carried out by Turbomole V7.3. 122 Excited state properties for the first ten excited singlet states were obtained from TD-DFT calculations using the BHandLYP exchange-correlation functional and the starting geometry provided by García. $[163,223,447,485$. The double- $\zeta$ basis $6-31 \mathrm{G}(\mathrm{d}) 486$ was employed for hydrogen, carbon, oxygen and nitrogen, the augmented $6-31+\mathrm{G}(\mathrm{d}) 487,488$ basis set for chlorine and the LANL2DZ basis with corresponding effective core potential for ruthenium. 489, 490 The RIJ approximation with corresponding auxiliary basis set was used to speed up the calculations. 389,464 Initial and maximum trust radius $\Delta$ were set to $0.3 a_{0}(\mathrm{rad})$.

The optimization was started in the bright $S_{9}$ state $(\Delta E=4.08 \mathrm{eV}, f=0.1251)$, corresponding to a ligand-to-metal-charge-transfer (LMCT) excitation from $\pi_{\text {tpy }}$ and $p_{\mathrm{Cl}, \mathrm{ax}}$ orbitals into an antibonding $\pi_{\mathrm{Ru}, \mathrm{NO}}^{*}$ orbital; results are shown in Figure 13.10. The NO group tilts down, and the Ru-N-O angle decreases from $176.4^{\circ}$ to $145.5^{\circ}$ along the optimization. Several root flips occur and the final ES, e.g., the respective LMCT at the $\mathrm{SP}^{\mathrm{SP}}$ is the $S_{6}$. García reported the need for 619 optimization cycles, whereas pysisyphus only needs 28 cycles to reach a SP while overachieving on the RMS (maximum) of the Cartesian gradient with $3 \times 10^{-6} E_{\mathrm{h}} a_{0}^{-1}\left(1.9 \times 10^{-5} E_{\mathrm{h}} a_{0}^{-1}\right)$, again showing a marked improvement. [223]

Just by visually inspecting, the CDDs of the initial $S_{9}$ and the final $S_{6}$ in Figure 13.10, seem quite different and it may be assumed that the ES tracking failed. But analyzing the overlaps reveals that they never drop below 0.85 and are usually well above 0.90 , confirming a successful state-tracking. This example clearly shows that small differences in ES character between successive optimization cycles may amount to big differences
between the first and the final cycle, highlighting again the need for a robust automated state tracking procedure.

If desired, pysisyphus offers easy CDD visualization along the optimization, as it is also interfaced to Multiwfn 491] and Jmol 492, allowing automated calculation and rendering of CDD; for Gaussian, ORCA and Turbomole. 121, 122, 384


Figure 13.10: Electronic energies of the ground and the first ten excited singlet states of a ruthenium nitrosyl complex (RuNO) along the optimization of the bright $S_{9}$, with respect to the ground state energy at the first cycle. The tracked excited state is indicated by a dashed black line. Excited states that do not cross with the state of interest are shown with reduced opacity. Every marker corresponds to an optimization cycle, and missing markers indicate root flips. CDDs illustrate the excited state character at the start and the end of the optimization; electron density is excited from blue to red.

### 13.3.3 Platinum Complexes

Modifying ligands in transition metal complexes, e.g., by substitution with electrondonating or electron-withdrawing groups, influences their photophysical properties considerably. By adding such electron donating or withdrawing groups to the triphenylamine (TPA) moiety in a rhenium complex, Larsen reported the effective emission energy
tuning from a short-lived 1 ILCT state (intra-ligand-charge-transfer). 493] In an effort to adjust the interaction of ILCT and metal-to-ligand-charge-transfer (MLCT) states in $f a c-\left[\operatorname{Re}(\mathrm{L})(\mathrm{CO})_{3}(\alpha \text {-diimine })\right]^{n+}$, Shillito investigated the role of the ancillary ligand L. 494 The $\alpha$-diimine ligand corresponds to 1,10 -phenanthroline, with TPA appended in 5 position (phen-TPA). Recently, the effect of further modifying TPA in the phenTPA ligand by electron-donating methoxy or electron-withdrawing cyano-groups on the photophysics of two rhenium and platinum complexes was analyzed by a combined experimental and computational study. [8]
Herein, optimizations of two excited triplet states for the two platinum complexes Pt-TPA-OMe and Pt-TPA-CN (see Figure 13.8d) are presented, thus, expanding the application of pysisyphus to excited state optimizations of triplet nature.

All quantum chemical calculations were performed using Gaussian 16. 384 Ground state equilibrium structures were obtained at the DFT level of theory utilizing the B3LYP35 functional,495 498 comprising $35 \%$ of exact-exchange, $58.5 \%$ of non-local B88 163 exchange and the LYP correlation. 447] The effective core potential MWB60499 and its corresponding valence basis set were used for platinum, the $6-31 \mathrm{G}(\mathrm{d})$ basis set was employed for all other atoms. 486 Excited state properties for the first 10 triplet states of Pt-TPA-OMe and the first 25 triplet states of Pt-TPA-CN were obtained by TD-DFT from a singlet reference state. Solvent effects $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} \epsilon=8.93, n=1.4070\right)$ were considered by means of the integral equation formalism of the polarizable continuum model. 500 Dispersion interactions were taken into account by the D3-model with BeckeJohnson damping. 382, 383 The initial (maximum) trust radius $\Delta$ was set to $0.05 a_{0}$ (rad) ( $0.3 a_{0}(\mathrm{rad})$ ).
Results of both optimizations are shown in Figure 13.11. Optimizing the $T_{9}$ of $\mathbf{P t}-$ TPA-CN to a SP took 43 optimization cycles, as the initial ES becomes the $T_{8}$ at the SP. Compared to RuNO with 34 atoms and 384 basis functions, where only the first 10 excited states were considered, Pt-TPA-CN is a much larger system with 89 atoms, 872 basis functions and 15 additional excited states, showing that pysisyphus in conjunction with WFO; can also be employed to optimize sizeable molecular systems. Optimizing the $T_{3}$ of Pt-TPA-OMe to a SP point took 35 cycles, and no state switching occurred. Successful state tracking is confirmed by the CDD shown in Figure 13.11, as they are very similar between the start and the end of the optimizations in both cases.


Figure 13.11: Electronic energies of the singlet ground state and a) the first ten triplet states of Pt-TPA-CN along the optimization of the $T_{9}$ and b ) the first five triplet states of Pt-TPA-OMe along the optimization of the $T_{3}$, with respect to the ground state energy at the first cycle. The tracked excited state is indicated by a dashed black line. Excited states that do not cross with the state of interest are shown with reduced opacity. Every marker corresponds to an optimization cycle, and missing markers indicate root flips. CDD illustrate the excited state character at the start and the end of the optimization; electron density is excited from blue to red.

### 13.4 Ground State Calculations - Biaryl Cross-Coupling

Pysisyphus' capabilities to obtain reaction coordinates was already outlined in section 13.2.5. Here, the calculation of a GS reaction coordinate for the photoreaction discussed in chapter 12 is presented, as it is an example for a complex and computationally challenging reaction, where three bonds are broken and one bond is formed. Even though the true reaction coordinate involves ESk, the GS coordinate is a good approximation to it. The unsubstituted $N$-benzylbenzenesulfonamide $\mathbf{1 b}$ (compound $\mathbf{1 p}$ in [2]) was chosen, as this allows all calculations to be performed in a timely manner, even on common desktop hardware.

All DFT calculations were performed in ORCA 4.2.0 121 with the range-separated exchange correlation functional CAM-B3LYP and the def2-SVP double- $\zeta$ basis set. 381, 501 The RIJCOSX approximation was employed to speed up the calculations. 291 Dispersion interactions were taken into account by the D3-model with Becke-Johnson damping. 382, 383 To reduce noise in the calculated gradients arsing from the numerical integration in DFT, tight convergence criteria for the self consistent field iterations and finer integration grids were used (keywords tightscf, grid4, finalgrid5, gridx6).

The GSM in DLC, as implemented in pysisyphus, was used to obtain a suitable TS guess for the biaryl cross-coupling reaction (see Figure 13.8b). Initial molecular geometries for the GSM were taken from the endpoints of an IRC, calculated in an earlier study 2 of the system at the CAM-B3LYP/def2-TZVP level of theory. Both molecular geometries were pre-optimized for 5 cycles in RIC using the standard method, outlined in section 13.2.1. A conjugate gradient optimizer 300 was employed to relax the string to the MEP The string was grown to a total length of 13 images (11 inner images). First and last image of the string remained fixed over the course of the optimization. After the string was fully grown, optimization was terminated and the HEI was used as initial guess for a TS optimization.

Subsequent TS optimization in RIC was carried out using RS-IRFO. Analytical Hessians were recalculated at every 5th optimization cycle. The nature of the obtained TS as first-order saddle point was verified by a Hessian calculation.

IRC; in mass-weighted Cartesian coordinates were traced from the TS to confirm that the TS connects presumed educts and products of the biaryl cross-coupling reaction. A predictor-corrector integrator utilizing Hessian information was used to integrate an IRC (eq. (11.1)) with a step length of $0.1 a_{0} .332$, 346] Corrector integration on the DWI surface was carried out using the Radau integrator, as provided by SciPy. [352, 441]

Analytical Hessians were recalculated every 10th IRC cycle. After the IRC converged to a Cartesian gradient RMS of $1.0 \times 10^{-3} E_{\mathrm{h}} a_{0}{ }^{-1}$, its endpoints were optimized to SPs. Along the IRC towards the products of the reaction, the TS fragments into three separate molecules. If requested by the user, pysisyphus can automatically detect the fragments and optimize them separately after convergence of the IRC is achieved.


Figure 13.12: Image energies of the growing string, describing the biaryl cross-coupling reaction over the course of the string optimization, with respect to the energy of the first image in the first cycle. Earlier cycles are shown in a lighter shade, later cycles in a darker shade. Images are enumerated by blue numbers. Inner images are introduced (grown) in later cycles. Image 1 and 13 remained fixed, while the other images were allowed to relax. The HEI was employed as guess for a subsequent TS optimization and is highlighted in orange.

Results of the calculations are shown in Figures 13.12 to 13.14 . The two starting images were pre-optimized for 5 cycles, lowering their RMS Cartesian gradient from $6.4 \times 10^{-3}$ $E_{\mathrm{h}}{a_{0}}^{-1}$ to $2.0 \times 10^{-4} E_{\mathrm{h}} a_{0}^{-1}$ for image 1 and from $6.8 \times 10^{-3} E_{\mathrm{h}} a_{0}^{-1}$ to $3.9 \times 10^{-4}$ $E_{\mathrm{h}} a_{0}{ }^{-1}$ for image 13. As educts and products will be optimized later on, starting from the IRC endpoints, it is not necessary to optimize the starting images tightly. If one of the starting images consists of multiple weakly bound fragments, the optimization may even fail, as such optimizations are notoriously difficult to converge. [272] Figure 13.12 displays the energy profile of the string images over the course of the optimization. The RMS of the DLC gradient of the string images quickly drops from $7.7 \times 10^{-3} E_{\mathrm{h}} a_{0}{ }^{-1}$ in the first cycle and never exceeds $6.0 \times 10^{-3} E_{\mathrm{h}} a_{0}{ }^{-1}$ after cycle 6 , showing that the
gradual growing of new images avoids any high energy geometries. Image 7 is the last one to be grown and image 8 is finally selected as HEI and used for the TS optimization.

Calculation of a Hessian in RIC at the HET revealed three significant negative eigenvalues. The overlap between the HEI tangent $\boldsymbol{\tau}$ and the eigenvector belonging to the most negative eigenvalue was 0.93 , indicating that the string was converged well enough for $\boldsymbol{\tau}$ to be a good approximation to the reaction coordinate. In this case, the DM would also be expected to allow a successful TS optimization, as $\boldsymbol{\tau}$ would be a very good guess for the initial dimer orientation $\boldsymbol{N}$. 290 An overlay of the HEI geometry and the optimized TS is given in Figure 13.13 , confirming that the HEI is already a good approximation to the true TS, The TS optimization converged in 16 cycles; besides the initial Hessian at the HEI three additional Hessians were calculated. An optimization starting from the same HEI without additional Hessian calculations took 50 cycles to converge the correct TS, but still needed roughly $10 \%$ less computational time compared to the optimization with Hessian recalculations ( 73 min vs. 65 min ). If analytical Hessians are available for the chosen level of theory, it is usually beneficial to recalculate them periodically for effective optimizations.

Starting from the TS, an IRC calculation (see Figure 13.14) confirmed that the optimized TS indeed connects the sulfonamide 1b educt and the biaryl coupling product. Without taking into account any thermochemical corrections, the reaction barrier in the GS, as calculated in the IRC is $335 \mathrm{~kJ} \mathrm{~mol}^{-1}$ and thus the cross-coupling reaction cannot proceed in the GS, If the involvement of ES; is considered, the barrier height is substantially lowered in the $S_{1}$ and the coupling reaction can take place. A full discussion of the reaction mechanism is found in chapter 12 and reference (2).

The example presented here exemplifies that full reaction paths can be obtained in pysisyphus with minimal user intervention, by using robust and efficient algorithms. Starting from the pre-optimization and ending with the optimization of the IRC endpoints, all calculations took 9.6 h on a desktop computer equipped with 8 physical CPU cores (Ryzen 1700X CPU).
a)



Figure 13.13: TS optimization of the biaryl cross-coupling reaction started from HEI guess using RS-IRFO: a) Energy change between HEI and optimized TS and b) RMS of the Cartesian gradient over the course of the optimization. The molecular geometries of HEI guess (in orange) and the optimized TS is shown in the upper panel.
a)

b)


Figure 13.14: Forward and backward IRC/ of the biaryl cross-coupling reaction started from the optimized TS, using a predictor-corrector integrator: a) Energy profile with respect to the educt energy at the start of the $\boxed{I R C}$ and b) RMS of the Cartesian gradient along the [RC.

## 13 Pysisyphus

### 13.5 Implementation

Pysisyphus is licensed under GNU GPL 3 and implemented in python 3. 413, 414, 502] Python is a popular, interpreted, high-level programming language that found widespread adoption in the chemistry community. [124, 272, 503 507] It features a rich ecosystem of mature, well tested and versatile libraries for scientific computing. 441,508

Inputs to pysisyphus can be given via simple to create YAML files or it can be used as a python library, to implement custom workflows, tailored to the users needs.

Required electronic energies and their derivatives are not calculated by pysisyphus itself, but by calling external QC packages. ES gradients and ES-tracking are currently available for Gaussian 16[384], ORCA 4[121], TURBOMOLE 7[122] PySCF[504 and the tight-binding code DFTB+. [206] GS calculations are possible with the already mentioned packages, Psi 4[503] and the fast semiempirical packages MOPAC2016[509] and xtb. 418 An even broader variety of QC packages is available through the QCEngine project, however with varying degree of supported features. [510] QCEngine provides basic wrappers for many additional codes, for instance NWChem, 511 GAMESS 512 and Q-Chem[513]. Advanced features like reusing wavefunction information from previous calculations and ES tracking is currently only possible with native pysisyphus calculators, nonetheless the applicability of pysisyphus is greatly increased through QCEngine. Parallel execution of the many required gradient calculations for COS methods is possible through the dask library, thus available resources like HPC clusters can be utilized efficiently. 514

Quick testing and rapid prototyping of new methods in pysisyphus is possible through analytical implementations of many potentials like the Lennard-Jones potential or the London-Eyring-Polanyi-Sato (LEPS) method, as their evaluation time is negligible compared to ab initio calculations and even semiempirical methods. [71] New 2D-potentials for testing can be easily specified just by their energy expression. First and second derivatives are then automatically determined by the sympy computer algebra system. 515

Using internal coordinates requires the calculation of the so called Wilson- $\boldsymbol{B}$ matrix that relates changes of Cartesian coordinates to changes in internal coordinates. 232 , 278 Transforming a Cartesian Hessian to internal coordinates also requires derivatives of the $\boldsymbol{B}$ matrix-elements with respect to the Cartesian coordinates. Pysisyphus uses code generation to avoid error-prone hand coding of these lengthy expressions. Algebraic expressions for the first and second derivatives of stretches, (linear) bends and dihedrals with respect to Cartesian coordinates were generated with sympy, simplified using common subexpression elimination and translated in a python module. Figure 13.15
shows valid python code that generates all expressions for the first and second derivatives of a dihedral coordinate with respect to Cartesian coordinates of the atoms making up the dihedral. Linear bends are implemented as given by Hoy. 273 275]

```
import sympy as sym
from sympy.vector import CoordSys3D
m0, m1, m2, n0, n1, n2, o0, o1, o2, p0, p1, p2 = sym.symbols("m:3 n:3 o:3 p:3")
Sys = CoordSys3D("Sys")
M = Sys.origin.locate_new("M", m0*Sys.i + m1*Sys.j + m2*Sys.k)
N = Sys.origin.locate_new("N", n0*Sys.i + n1*Sys.j + n2*Sys.k)
0 = Sys.origin.locate_new("0", 00*Sys.i + 01*Sys.j + 02*Sys.k)
P = Sys.origin.locate_new("P", p0*Sys.i + p1*Sys.j + p2*Sys.k)
U = M.position_wrt(0)
V = N.position_wrt(P)
W = P.position_wrt(0)
U_ = U.normalize()
W = W.normalize()
\mp@subsup{V}{-}{-}}=\textrm{V}.normalize(
phi_u = sym.acos(U_.dot(W_))
phi_v = sym.acos(-\mp@subsup{\overline{W}}{-}{\prime}\cdot\operatorname{dot}(\mp@subsup{\overline{v}}{-}{\prime}))
q_ d = sym.acos(U_}\cdot\operatorname{cross}(\mp@subsup{W}{-}{\prime})\cdot\operatorname{dot}(\mp@subsup{V}{-}{\prime}\cdot\operatorname{cross}(\mp@subsup{W}{-}{\prime}))/(\operatorname{sym}.\operatorname{sin}(phi_u)*\operatorname{sym}.\operatorname{sin}(phi_v)))
dx_d = (m0, m1, m2, 00, o1, o2, p0, p1, p2, n0, n1, n2)
deriv1 = sym.derive_by_array(q_d, dx_d) # deriv1.shape == (12,)
deriv2 = sym.derive_by_array(deriv1, dx_d) # deriv2.shape == (12, 12)
```

Figure 13.15: Python code for the sympy computer algebra system to generate first and second derivatives for a primitive dihedral coordinate with respect to the Cartesian coordinates of the atoms making up the dihedral.

The code is developed mainly in an object-oriented way with superclasses for (TS) optimizations, IRC calculations, COS methods, line searches and interpolation, providing commonly needed functionality. Actual algorithms like optimizers and IRC integrators are implemented as subclasses, specifying only the functionality specific to the respective method. This way, pysisyphus is extended easily with new methods, as the existing infrastructure from the superclasses can be reused.

Interoperability with the well-established ASE package is possible by an automated conversion of the central Geometry class from pysisyphus to the central Atoms class in ASE 124 Hence, functionality unique to pysisyphus, e.g., EStracking can be combined with ASE optimizers, increasing the applicability of pysisyphus.

While interfacing to ASE requires the user to write additional python code, pysisyphus also offers a socket-interface, supporting a superset of the i-PI communication protocol. [516, 517] The original protocol is restricted to simultaneous sending of energy and forces. Pysisyphus extends it to also support sending Hessians or energies only, as required for TSfoptimizations and line searches. According to the ASE documentation, a

## 13 Pysisyphus

variety of quantum chemical codes implement the (original) i-PI protocol, e.g., Quantum Espresso, [518, 519] Siesta, [520] DFTB+ [206] and NWChem. [511, 521] Communication via sockets was successfully tested with DFTB+, for which pysisyphus can act as external optimizer.

The present ES-tracking functionality is easily accessible, as interfaces to new programs only have to provide the MO coefficients and the one-electron transition density matrix. Care was taken to allow a quick and easy visualization of running and completed calculations via the pysisplot command. A subset of possible visualizations is shown in Figure 13.16 .

Fully automated interpolation of molecular geometries with the methods discussed in section 8.2, including interpolation in DLC, is available through a simple to use command line interface. This way interpolated paths, e.g. for Marcus theory, are easily obtained. 90

Pysisyphus is developed using continuous integration and features an extensive test suite, to ensure that further code development does not break existing functionality and introduces new bugs. 522 The code is fully available at https://github.com/eljost/pysisyphus and is easily installed from the python Package index with pip install pysisyphus. Comprehensive documentation and examples are found online $4^{4}$ Contributions and bug reports are welcome. To this date, pysisyphus already received a code contribution from an external researcher. 523

[^10]

Figure 13.16: Selection of possible visualizations available via the pysisplot command. Besides removing superfluous whitespace, figures were used, as produced by pysisyphus. While not intended for direct inclusion into publications, these visualizations allow a quick control of (running) calculations.

## 13 Pysisyphus

### 13.6 Summary

This chapter described the capabilities and implementation of pysisyphus - an external optimizer for stationary points and reaction coordinates in ground and ESk. By providing ES tracking capabilities and optimization in internal coordinates, pysisyphus allows highly efficient ES optimizations. The performance of pysisyphus was verified extensively against several test sets and previously published results for ES optimizations of cytosin and a ruthenium nitrosyl complex. Furthermore, it was applied to the optimization of excited triplet states of two sizeable platinum complexes, (long-lived) key states in their ES relaxation cascades. [8] The ability of pysisyphus to obtain full reaction paths by means of COS methods and IRC calculations was demonstrated for a real world example in the electronic ground state, but all presented methods are also applicable in ES calculations. Besides the already mentioned platinum complexes, pysisyphus was already successfully applied for elucidating the reaction mechanisms of (metal-free) biaryl cross-couplings and iron(0) mediated $(3+2)$ cycloaddition of thiochalcones. [2, 524

Possible extensions of pysisyphus would be the generalization of ES tracking to arbitrary multiplicities and support for wavefunctions beyond CIS, as they occur in complete active space self consistent field (CASSCF) calculations. By combining the interpolation and ES optimization capabilities of pysisyphus, the automated calculation of rate constants, e.g. in the scope of electron transfer processes using semi-classical Marcus theory, could be realized. A pure python implementation of Plassers wfoverlap 225 program is already underway. 525 By incorporating changes of the WFO algorithm proposed by Sapunar, overlap calculations for configuration interaction singles (CIS) type wavefunction could be made even more efficient. 226

## 14 Summary

Chemical reactivity of atoms, molecules and ions is governed by their underlying potential energy surface (PES). $N$ atoms in the gas-phase give rise to a PES with $3 N-6$ degrees of freedom ( $3 N-5$ for linear arrangements). Calculating the whole PES within reasonable bounds, is impossible for all but the smallest $N$. Usually, only parts of the full PES can be studied, namely stationary points (SP) and the minimum energy paths (MEPs) connecting them. By comparing energies of SP , and their separating barriers, conclusions regarding possible reactions mechanism, or their infeasibility, can be drawn. Taking excited states (ESk) into account leads to further complications, as now multiple PESs have to be considered and root flips between different ESS may occur, requiring effective ESt-tracking.

Part $\Pi$ describes the required methods to locate SP; and MEP; on PES; by using surface-walking, chain-of-states (COS) optimization and intrinsic reaction coordinate (IRC) integration.
Special focus was set on ES-tracking in chapter 4, where three approaches are outlined and an adaptive formalism to update the reference cycle is presented. State-tracking by wavefunction overlaps is the most general approach, but computationally also the most demanding, as many determinant calculations may be required.
Different coordinate systems, including their advantages and disadvantages are discussed in chapter 6. The most promising coordinates for molecular optimizations are redundant internal coordinates (RIC), as they usually exhibit small coupling and allow easy estimation of initial model Hessians. Drawbacks of internal coordinates in general, are that they are not unambiguously defined and their need for an iterative internal-Cartesian back-transformation that may fail to converge.
Chapter 7 briefly reviews preconditioning. Preconditioning shows a different route to the omnipresent Newton step (eq. (5.3) on page 31) in molecular optimizations, apart from the usual derivation via Taylor expansion. Convergence of optimizations carried out in Cartesian coordinates is often greatly improved by preconditioning. Optimizing Baker's transition state (TS) set using the dimer method (DM) and preconditioned

## 14 Summary

limited-memory BFGS (LBFGS) (section 13.2.4, Table 13.7) showed that preconditioning can rival the performance of optimizations carried out in internal coordinates.

A COS comprises several images, discretizing a MEP, onto which it can be gradually relaxed. The highest energy image (HEI) of a relaxed COS is often a good guess for subsequent TS optimizations. If converged tightly, a COS may even be a good approximation to the true MEP, Commonly employed COS methods (nudged elastic band (NEB) and string method (SM)), including their variants are discussed in chapter 8. Obtaining TS; by utilizing Hessian information is outlined in chapter 9, TS optimization using only first derivatives of the energy by means of the DM is reviewed in chapter 10. Chapter 11 describes how to integrate an IRC, the path of steepest descent in mass-weighted Cartesian coordinates. IRC/ are usually easily interpreted, as they don't exhibit complicated rotational and vibrational motions. Similar to normal optimizations, IRC integrators (Gonzalez-Schlegel second-order algorithm (GS2) and Euler predictor-corrector (EulerPC) utilizing Hessian information can sustain longer step lengths, compared to integrators relying only on the energy gradient.

Results of this thesis are presented in Part III. This thesis makes two contributions to the field of photochemistry: chapter 12 provides a possible ES reaction mechanism for the biaryl cross-coupling reaction shown in Figure 14.1 and offers a plausible explanation for its high regioselectivity.


Figure 14.1: Biaryl cross-coupling photoreaction.

The coupling reaction was investigated for two substrates: 1a, affording significant photoproduct yields and the unsubstituted $\mathbf{1 b}$, providing only photoproduct traces. Their ground state (GS) PES; were sampled by simulated annealing conformer searches and relaxed scans, leading to the identification of two major structural motifs: Linear conformers, characterized by increased distances between the C5-C1 carbons supposed to couple and horseshoe conformers exhibiting decreased C5-C1 distances. According to the conformer search, the horseshoe motif is preferred for 1a, while a linear minimum energy conformer is predicted for 1b. A five-membered TS in the GS was obtained for both substrates. IRC integration revealed high barriers, preventing a GS reaction. ES calculations along selected IRC geometries confirmed the existence of several ES;
that facilitate the photoreaction with greatly reduced barriers. Compared to 1a, the relevant ES in $\mathbf{1 b}$ are shifted hypsochromically and exhibit lower oscillator strengths, thus impeding the photoreaction which is in full accordance with the experimental results.

The second contribution is the development pysisyphus (chapter 13), an external optimizer implemented in python, aware of ES; and thus the core of this thesis. Besides the recently proposed SDNTO program by García et al., 223 which is restricted to a steepest descent optimizer and does not support RIC, the author of this thesis is not aware of other external optimizers tailored to deal with ES5. All ES-tracking approaches outlined in chapter 4 (wavefunction overlaps, transition density matrix (TDEN) overlaps and natural transition orbital (NTO) overlaps) are implemented in pysisyphus.

Pysisyphus features comprehensive documentation and is developed using contemporary practices like continuous integration and unit testing, to assert the correctness of the software.

Chapter 13 presents extensive benchmarks, confirming that pysisyphus is ahead of, or at least competitive with similar implementations. Optimization of minima was benchmarked for Baker's test set and the more challenging S22 set of Hobza. 268,423 TS optimization performance was investigated for Baker's TS test set, 315 using two different approaches: utilizing-Hessian information and utilizing only first derivatives, by means of the DM. Finally, pysisyphus' ability to obtain TS; from preceding COS optimizations was verified against a diverse test set, containing educts and products of 20 reactions.

Results for several ES optimizations are presented in section 13.3. Compared to SDNTO, pysisyphus shows greatly improved performance for ES optimizations. Optimizing the bright $S_{9}$ of RuNO required only 28 cycles, compared to 619 cycles reported for SDNTO. 479 Applicability of the ESfoptimizer to triplet states of sizable systems was demonstrated for two Pt -complexes in section 13.3.3.

Care was taken to integrate pysisyphus into the bigger ecosystem of computational chemistry software, by providing an Atomic Simulation Environment (ASE) interface and supporting an extended version of the i-PI socket protocol, allowing users to leverage the many thousand development hours already spent on these projects.
Internal coordinates were implemented using code generated by the sympy computer algebra system, to avoid error prone manual coding of the lengthy but necessary internal coordinate derivatives.

Besides the calculation of energies and their derivatives, which are carried out using external quantum chemistry (QC) software, pysisyphus is self-contained. It does not rely
on optimization algorithms implemented in the QC packages, thus offering full control, if required by the user. Pysisyphus implements a multitude of relevant algorithms for surface-walking, COS optimizations and IRC integration. All algorithms discussed in chapters 5 to 11, and many more, are available in pysisyphus. Further development of pysisyphus is especially attractive, as it would expand the applicability of all interfaced QC packages.

## 15 Zusammenfassung

Die chemische Reaktivität von Atomen, Molekülen und Ionen wird durch ihre unterliegende Potentialhyperfäche (PES) bestimmt. $N$ Atome in der Gasphase werden durch eine PES mit $3 N-6$ Freiheitsgraden beschrieben ( $3 N-5$ für lineare Anordnungen). Berechnung der vollständigen PES, in angemessenen Grenzen, ist nur für kleinste Systeme möglich. Überlicherweise können nur Ausschnitte der PES untersucht werden: Stationäre Punkte (SPs) und die sie verbindenen Minimal-Energie-Pfade (MEPs). Durch Vergleich der Energien von SPs und der sie trennenden Barrieren können Aussagen über die Möglichkeit, oder Unmöglichkeit, von Reaktiosmechanismen getroffen werden. Werden zum elektronischen Grundzustand noch zusätzlich angeregte Zustände (ESs) berücksichtigt, verkompliziert sich die Interpretation der PESs, da nun auch Wechsel zwischen verschiedenen elektronischen Zuständen auftreten können.

Die vorliegende Arbeit beschäftigt sich mit der Implementierung von Methoden zur PES-Aufklärung in Grund- und angeregten Zuständen, sowie ihrer Anwendung auf photochemische Fragestellungen.
Der Theorieteil diskutiert die benötigten theoretischen Grundlagen, um SPs und MEPs auf PESs zu finden. Dabei können drei Algorithmenklassen unterschieden werden:

- Schreiten entlang der PES (surface walking): Eine vermutete Startgeometrie wird zum nächstgelegenen SP (Minimum oder Sattelpunkt) optimiert.
- Optimierung einer Kette von Zuständen (chain-of-states, COS): Eine Kette von mehreren Geometrien approximiert einen MEP und wird gemeinsam optimiert, bis alle Geometrien auf dem MEP liegen.
- Integration einer intrinsischen Reaktionskoordinate (IRC): Pfad des steilsten Abstieges in massengewichteten, kartesischen Koordinaten, üblicherweise ausgehend von einem Sattelpunkt, hin zu Edukt(en) und Produkt(en).

Ein besonderer Fokus der Arbeit liegt auf Methoden, um einen initial gewählten ES entlang einer Optimierung zu verfolgen (state-tracking), auch über Kreuzungen mit anderen Zuständen hinweg. Der Ergebnisteil diskutiert die Resultate der vorliegenden Dissertation in zwei Kapiteln.

## Biaryl-Kopplungs-Photoreaktion

Kapitel 12 beschäftigt sich mit der Aufklärung des Reaktionsmechanismus einer Biaryl-Kopplungs-Photoreaktion. Mittels Konformerensuche und Optimierungen unter Nebenbe-

dingungen (relaxed scans) konnten zwei wichtige Strukturmotive identifiziert werden, welche die Sulfonamid-Edukte einnehmen: lineare Geometrien, mit großem Abstand zwischen den zu koppelnden Kohlenstoffatomen und Hufeisen-Geometrien, mit stark verringertem Abstand. Für den elektronischen Grundzustand wurde ein fünf-gliedriger Übergangszustand gefunden, welcher die Reaktion vermittelt. Eine IRC-Integration zeigte jedoch die Existenz einer hohen Barriere, welche die Reaktion im Grundzustand unmöglich macht. Durch Berechnungen von ESs entlang des IRCs wurden verschiedene Zustände erhalten, in denen die Barrierenhöhe stark reduziert ist, so dass die Reaktion effizient stattfinden kann. Die Ergebnisse der Berechnungen zeigen eine gute Überstimmung mit experimentellen Messungen.

## Pysisyphus

Kapitel 13 stellt die Software pysisyphus vor, einen externen Optimierer mit spezieller Unterstützung für die Verfolgung von ESs. Pysisyphus erlaubt state-tracking mittels Überlapps zwischen Wellenfunktionen, Übergangsdichte-Matrizen oder natural transition orbitals. Für verschiedene ES-Optimierungen zweier Referenzsysteme (Cytosin und ein Ruthenium-Nitrosyl-Komplex) konnte der benötigte Rechenaufwand im Vergleich zur Literatur auf $3.1 \%$ bis $5.6 \%$ reduziert werden. Die Anwendbarkeit der vorliegenden Implementierung für Optimierungen von Triplett-Zuständen wurde für zwei PlatinKomplexe gezeigt.

Weiterhin implementiert pysisyphus eine Vielzahl von Algorithmen für surface-walking, COS-Optimierung und IRC-Integration. Die Korrektheit der vorliegenden Implementierungen wurde ausgiebig gegen eine Vielzahl von Referenzsystemen verifiziert. Neben den untersuchten Referenzsystemen wurde die Anwendbarkeit von pysisyphus für die Sattelpunktsuche auch für die o.g. Biaryl-Kopplungs-Reaktion gezeigt.
Pysisyphus wurde mittels moderner Praktiken wie unit-tests, continuous integration und Code-Generierung entwickelt.

## A Appendix

## A. 1 Numerical integration of the Local Quadratic Approximation

The IRC

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}(s)}{\mathrm{d} s}=-\frac{\boldsymbol{g}(\boldsymbol{x})}{|\boldsymbol{g}(\boldsymbol{x})|} \tag{A.1}
\end{equation*}
$$

is parametrized by the arc length $s$. Assuming $3 N$ mass-weighted Cartesians the arc length is calculated as 335

$$
\begin{equation*}
\mathrm{d} s^{2}=\sum_{i=1}^{3 N} \mathrm{~d} x_{i}^{2} . \tag{A.2}
\end{equation*}
$$

Pechukas noted that integrating

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}(t)}{\mathrm{d} t}=-\boldsymbol{g}(\boldsymbol{x}) \tag{A.3}
\end{equation*}
$$

yields the same paths, as integration of eq. A.1). (334 Eq. A.3) can be related to eq. A.1) via

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}}{\mathrm{~d} t}=\frac{\mathrm{d} \boldsymbol{x}}{\mathrm{~d} s} \frac{\mathrm{~d} s}{\mathrm{~d} t} \tag{A.4}
\end{equation*}
$$

Expanding $\boldsymbol{g}(\boldsymbol{x})$ to first-order around $\boldsymbol{x}_{0}$ and substituting into eq. A.3) yields

$$
\begin{equation*}
\frac{\mathrm{d} \boldsymbol{x}(t)}{\mathrm{d} t}=-\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)-\boldsymbol{H}\left(\boldsymbol{x}_{0}\right)\left(\boldsymbol{x}-\boldsymbol{x}_{0}\right) . \tag{A.5}
\end{equation*}
$$

For quadratic potentials eq. A.5) is solved by

$$
\begin{equation*}
\boldsymbol{x}(t)=\boldsymbol{x}_{0}+\boldsymbol{A}(t) \boldsymbol{g}\left(\boldsymbol{x}_{0}\right) \tag{A.6}
\end{equation*}
$$

Matrix $\boldsymbol{A}(t)$ is defined as

$$
\begin{equation*}
\boldsymbol{A}(t)=\boldsymbol{U}_{0} \boldsymbol{\alpha}(t) \boldsymbol{U}_{0}^{\top} \tag{A.7}
\end{equation*}
$$

## A Appendix

with $\boldsymbol{U}_{0}$ being the eigenvector matrix of $\boldsymbol{H}\left(\boldsymbol{x}_{0}\right)$ and $\boldsymbol{\alpha}(t)$ is a diagonal matrix with elements

$$
\begin{equation*}
\alpha_{i i}=\frac{e^{-\lambda_{i} t}-1}{\lambda_{i}} \tag{A.8}
\end{equation*}
$$

where $\left\{\lambda_{i}\right\}$ are the eigenvalues of $\boldsymbol{H}\left(\boldsymbol{x}_{0}\right)$. From now on, $\boldsymbol{g}\left(\boldsymbol{x}_{0}\right)$ and $\boldsymbol{H}\left(\boldsymbol{x}_{0}\right)$ will be denoted as $\boldsymbol{g}_{0}$ and $\boldsymbol{H}_{0}$. Substituting Eq. A.6) in the right-hand side (RHS) of eq. A.5) and multiplying from the left with $\boldsymbol{U}_{0}^{\top}$ yields:

$$
\begin{align*}
\boldsymbol{U}_{0}^{\top} \frac{\mathrm{d} \boldsymbol{x}}{\mathrm{~d} t} & =-\boldsymbol{U}_{0}^{\top} \boldsymbol{g}_{0}-\boldsymbol{U}_{0}^{\top} \boldsymbol{H}_{0}\left(\boldsymbol{x}_{0}+\boldsymbol{A}(t) \boldsymbol{g}_{0}-\boldsymbol{x}_{0}\right)  \tag{A.9}\\
& =-\boldsymbol{U}_{0}^{\top} \boldsymbol{g}_{0}-\boldsymbol{U}_{0}^{\top} \boldsymbol{H}_{0} \boldsymbol{A}(t) \boldsymbol{g}_{0}  \tag{A.10}\\
& =-\boldsymbol{U}_{0}^{\top} \boldsymbol{g}_{0}-\boldsymbol{U}_{0}^{\top} \boldsymbol{H}_{0} \boldsymbol{U}_{0} \boldsymbol{\alpha}(t) \boldsymbol{U}_{0}^{\top} \boldsymbol{g}_{0}  \tag{A.11}\\
& =-\boldsymbol{U}_{0}^{\top} \boldsymbol{g}_{0}-\boldsymbol{\lambda}_{0} \boldsymbol{\alpha}(t) \boldsymbol{U}_{0}^{\top} \boldsymbol{g}_{0} \tag{A.12}
\end{align*}
$$

$\boldsymbol{\lambda}_{\mathbf{0}}$ denotes a diagonal matrix containing the eigenvalues of $\boldsymbol{H}_{0}$. Eq. A.12) is further simplified to

$$
\begin{align*}
\boldsymbol{U}_{0}^{\top} \frac{\mathrm{d} \boldsymbol{x}}{\mathrm{~d} t} & =-\left(\boldsymbol{\lambda}_{0} \boldsymbol{\alpha}(t)+1\right) \boldsymbol{U}_{0}^{\top} \boldsymbol{g}_{0}  \tag{A.13}\\
\frac{\mathrm{~d} \boldsymbol{x}^{\prime}}{\mathrm{d} t} & =-\left(\boldsymbol{\lambda}_{0} \boldsymbol{\alpha}(t)+1\right) \boldsymbol{g}_{0}^{\prime} \tag{A.14}
\end{align*}
$$

where the prime denotes quantities transformed to the basis of the Hessian eigenvectors $\boldsymbol{U}_{0}$. Matrix $\boldsymbol{\lambda}_{\mathbf{0}} \boldsymbol{\alpha}(t)$ is diagonal with elements

$$
\begin{align*}
\left(\boldsymbol{\lambda}_{0} \boldsymbol{\alpha}(t)\right)_{i i} & =\lambda_{i} \frac{e^{-\lambda_{i} t}-1}{\lambda_{i}}  \tag{A.15}\\
& =e^{-\lambda_{i} t}-1 \tag{A.16}
\end{align*}
$$

Substituting eq. A.16 into eq. A.14 yields

$$
\begin{align*}
\left(\frac{\mathrm{d} \boldsymbol{x}^{\prime}}{\mathrm{d} t}\right)_{i} & =-\left(e^{-\lambda_{i} t}-1+1\right) g_{0 i}^{\prime}  \tag{A.17}\\
& =-e^{-\lambda_{i} t} g_{0 i}^{\prime} \tag{A.18}
\end{align*}
$$

Arc length $s$ satisfies the equation

$$
\begin{equation*}
\frac{\mathrm{d} s}{\mathrm{~d} t}=\sqrt{\frac{\mathrm{d} \boldsymbol{x}}{\mathrm{~d} t} \frac{\mathrm{~d} \boldsymbol{x}}{\mathrm{~d} t}} \tag{A.19}
\end{equation*}
$$

Together with eq. A.18) and the fact that the (unitary) transformation $\boldsymbol{x}^{\prime}=\boldsymbol{U}_{0}^{\top} \boldsymbol{x}$ is norm conserving $\left(\boldsymbol{x}^{\top} \boldsymbol{x} \boldsymbol{x}^{\prime}=\boldsymbol{x}^{\boldsymbol{\top}} \boldsymbol{x}\right)$, the final equation for integrating $\frac{\mathrm{d} s}{\mathrm{~d} t}$ in the basis of the Hessian eigenvectors is obtained.

$$
\begin{equation*}
\frac{\mathrm{d} s}{\mathrm{~d} t}=\sqrt{\sum_{i} g_{0 i}^{\prime} e^{-2 \lambda_{i} t}} \tag{A.20}
\end{equation*}
$$

## A. 2 Biaryl Cross-Coupling Results for 1b



Figure A.1: Energy differences and $\mathrm{C} 5-\mathrm{C} 1$ distances for the relaxed scan around the central dihedral (C5-S4-N3-C2) in 1b at the CAM-B3LYP/def2-TZVP level of theory. A horseshoe conformer is predicted as minimum energy geometry at $\phi=-90^{\circ}$ with a C5-C1 distance of 337 pm .


Figure A.2: Selected distances along the IRC describing the formation of $\mathbf{2 b}$ from $\mathbf{1 b}$ by photosplicing, obtained at the CAM-B3LYP/def2-TZVP/CPCM(ACN) level of theory. a) Atomic distances with significant changes when going from the educt to the TS and b) distances that change significantly in the second IRC half, when going from TS to the photoproducts.

## A. 3 S22 Geometries with High root mean square deviations (RMSDs)

,

2




11
8

17
9

18

Figure A.3: Overlay of selected, optimized geometries from the S22 test set, as discussed in section 13.2.2, 417, 423 The numbering is consistent with Table 13.4 . Reference geometries as provided by Lindh are given in blue, whereas the geometries obtained from pysisyphus (this work) are colored by element. All geometries, beside case 17 (benzene • water dimer) are virtually indistinguishable, from the reference geometries, even though they show high RMSD k , in spite of resorting the atoms with the Hungarian algorithm. 433, 434

## A. 4 Code Listings

This section contains several python code listings, allowing to reproduce (part of) the results presented in this thesis. Listings are provided in the form of unit tests, that can be executed by the pytest python package. 526-529 Exact git commits are given as docstring below the function header.

Listing A.1: test_baker.py, Optimization of Bakers test set in 207 cycles at the HF/STO3 G level of theory, as discussed in section 13.2.1.

```
import pytest
from pysisyphus.benchmarks import Benchmark
from pysisyphus.calculators.ORCA import ORCA
from pysisyphus.optimizers.RFOptimizer import RFOptimizer
def calc_getter(charge, mult):
    return ORCA(keywords="hf_sto - 3g", pal=6, charge=charge, mult=mult)
BakerBm = Benchmark("baker", coord_type="redund", calc_getter=calc_getter)
@pytest.mark.parametrize("fn, „geom, rref_energy", BakerBm)
def test_baker(fn, geom, ref_energy):
    """pysisyphus @ 7cc0a296493b797ae934dfe891109414f9f8afc6"""
    opt_kwargs = {
        "thresh": "baker",
        "hessian_init": "fischer",
        "hessian_update": "bfgs",
        "line_search": True,
        "gdiis": True,
    }
    opt = RFOptimizer(geom, **opt_kwargs)
    opt.run()
    assert opt.is_converged
    assert geom.energy = pytest.approx(ref_energy)
```


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Listing A.2: test_s22_set.py, Optimization of S22 TS test set in 310 cycles at the RI-MP2/6-31G** level of theory, as discussed in section 13.2 .2 .

```
import pytest
from pysisyphus.benchmarks import Benchmark
from pysisyphus.calculators import ORCA
from pysisyphus.optimizers.RFOptimizer import RFOptimizer
from pysisyphus.testing import using
S22Bm = Benchmark("s22", coord_type="redund")
@using("orca")
@pytest.mark.parametrize(
    "fn, „geom, „charge, mult, rref_energy",
    S22Bm.geom_iter
)
def test_s22_set(fn, geom, charge, mult, ref_energy):
    """pysisyphus @ e56df2973cdcf3140216637b4ae4b5dbf2f30542"""
    calc = ORCA(
        keywords="RI-MP2_6-31G**_def2-SVP/C\_tightscf",
        pal=4,
        mem=1500,
        charge=charge,
        mult=mult ,
    )
    geom.set_calculator(calc)
    opt_kwargs = {
        "thresh": "gau",
        "dump": True,
    }
    opt = RFOptimizer(geom, **opt_kwargs)
    opt.run()
    assert opt.is_converged
```

Listing A.3: test_baker_ts.py, Optimization of Bakers TS test set in 304 cycles at the HF $/ 3-21 \mathrm{G}$ level of theory, as discussed in section 13.2.4, using restricted step partitioned RFO (RS-PRFO) with enabled line search.

```
import pytest
from pysisyphus.benchmarks import Benchmark
from pysisyphus.calculators.ORCA import ORCA
from pysisyphus.tsoptimizers import RSPRFOptimizer
def calc_getter(charge, mult):
    return ORCA(keywords="HF\_3-21g", pal=6, mem=500, charge=charge ,
                        mult=mult )
BakerTSBm = Benchmark(
    "baker_ts",
    coord_type="redund",
    calc_getter=calc_getter,
)
@pytest.mark.parametrize("fn, sgeom, ьref_energy", BakerTSBm)
def test_baker_ts(fn, geom, ref_energy):
    """pysisyphus @ 1ce48d3eb95715111b4758c3e6a1f0da4e7319d4"""
    if fn= "22_hconhoh.xyz":
            ref_energy = - 242.25695787 # Unconstrained solution
    opt_kwargs={
        "thresh": "baker",
        "trust_radius": 0.1,
            "trust_max": 0.3,
            "min_line_search": True,
            "max_line_search": True,
        }
        # Cases 10&11 start in the convex region of the PES, so we must
    # designate an initial mode/root/direction.
    if fn= "10_tetrazine.xyz":
        opt_kwargs["rx_mode"] = (((0, 2), 1), ((1, 3), 1))
        elif fn == "11_trans_butadiene.xyz":
            opt_kwargs["rx_mode"] = (((2, 0, 1, 3), 1), )
    opt = RSPRFOptimizer(geom, **opt_kwargs)
    opt.run()
    assert opt.is_converged
    assert geom.energy = pytest.approx(ref_energy)
```


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Listing A.4: test_baker_ts_dimer.py, Optimization of Bakers TS test set in 358 cycles (1092 gradient evaluations in total) at the HF/3-21G level of theory, using the DM and preconditioned LBFGS, as discussed in section 13.2.4 and summarized in Table 13.7 .

```
import pickle
import pytest
from pysisyphus.benchmarks import Benchmark
from pysisyphus.calculators import Dimer, ORCA
from pysisyphus.optimizers.PreconLBFGS import PreconLBFGS
BTS = Benchmark(
    "baker_ts",
    coord_type="cart",
    exclude=(10,),
)
@pytest.mark.parametrize(
    "fn, geom, „charge, „mult, ヶref_energy",
    BTS.geom_iter
)
def test_baker_ts_dimer(fn, geom, charge, mult, ref_energy):
    """pysisyphus@6d9c004d"""
    with open("Ns", "rb") as handle:
        N_INITS = pickle.load(handle) # Load initial orientations
    calc = ORCA("hf_3-21g\_tightscf", charge=charge, mult=mult, pal=6)
    dimer_kwargs = {
        "rotation_method": "fourier",
        "calculator": calc,
        "N_raw": N_INITS[fn [:2]],
        "length": 0.0189,
        "rotation_tol": 5,
        "rotation_disable_pos_curv": True,
        "trans_force_f_perp": True,
    }
    dimer = Dimer(**dimer_kwargs)
    geom.set_calculator(dimer)
    opt_kwargs = {
            "thresh": "baker",
            "precon": True,
            "max_step_element": 0.25,
            "max_cycles": 50,
            "c_stab": 0.103,
            "dump": True,
    }
    opt = PreconLBFGS(geom, **opt_kwargs)
    opt.run()
    assert opt.is_converged
    assert geom.energy == pytest.approx(ref_energy)
```

Listing A.5: test_xtb_rx.py,TS optimizations for the COS test set at the GFN2-XTB level of theory, using the GSM and restricted step image method RFO (RS-IRFO), as discussed in section 13.2 .5 .

```
from pathlib import Path
import tempfile
import pytest
from pysisyphus.benchmarks import Benchmark
from pysisyphus.xyzloader import write_geoms_to_trj
from pysisyphus.run import run_from_dict
@pytest.mark.parametrize(
    "fn,„geoms,_charge,_mult, „ref_energy", Benchmark("xtb_rx").geom_iter
)
def test_xtb_rx(fn, geoms, charge, mult, ref_energy):
    """pysisyphus@ @779c3485bb5048dc6b58fa6f035e9cb493c1ec6"""
    start, ts_ref_org, end = geoms
    id_ = fn [:2]
    with tempfile.TemporaryDirectory() as tmp_dir:
        tmp_path = Path(tmp_dir)
        inp_trj = str(tmp_path / "gs_inputs.trj")
        write_geoms_to_trj((start, end), inp_trj)
        run_dict = {
            "geom": {
                            "type": "dlc" if (id_ != "02") else "cart",
                            "fn": inp_trj,
            },
            "calc": {
                            "type": "xtb", # GFN2-XTB 6.3.3
                            "pal": 1,
                            "mem": 750,
                            "charge": charge,
                            "mult": mult,
            },
            "preopt": {"max_cycles": 5},
            "cos": {
                    "type": "gs",
                            "climb": True,
                            "climb_rms": 0.0075,
            },
            "opt": {
                    "type": "string",
                            "max_step": 0.2,
                            "rms_force": 0.005 if (id_ != "19") else 0.003,
                            "rms_force_only": True,
            },
            "tsopt": {
                        "type": "rsirfo",
                            "thresh": "gau",
                            "trust_max": 0.5,
                            "do_hess": True,
            },
        }
        results = run_from_dict(run_dict)
        assert results.ts_opt.is_converged
```


## A Appendix

Listing A.6: test_birkholz_interpolation.py, TS optimizations for the HCN isomerization and the Cope rearrangement of 1,4 -hexadiene using the connectivity transition state (CTS) method, as mentioned in section 13.2.5.

```
import pytest
from pysisyphus.calculators.PySCF import PySCF
from pysisyphus.helpers import (
        geom_loader,
        do_final_hessian ,
        highlight_text
)
from pysisyphus.tsoptimizers import RSPRFOptimizer, birkholz_interpolation
from pysisyphus.testing import using
@using("pyscf")
@pytest.mark.parametrize(
    "name, _fn, _ref_energy",
    [
        ("hcn", "lib:birkholz_rx/02_hcn_original.trj", - 91.56485102),
        ("cope", "lib:birkholz_rx/03_cope.trj", -230.06026151),
    ] ,
)
def test_birkholz_interpolation(name, fn, ref_energy):
    """pysisyphus@ f67a70a2a565c42c7d422190288aac96d4fa6fd2"""
    geoms = geom_loader(fn)
    def calc_getter():
        return PySCF(basis="sto 3g", verbose=0)
    ts_guess = birkholz_interpolation(geoms, calc_getter)
    print(highlight_text("TS-Optimization"))
    tsopt_kwargs = {
        "dump": True,
        "thresh": "gau",
        "trust_max": 0.3,
        "prefix": name,
    }
    tsopt = RSPRFOptimizer(ts_guess, **tsopt_kwargs)
    tsopt.run()
    do_final_hessian(ts_guess, write_imag_modes=True)
    assert ts_guess.energy= pytest.approx(ref_energy)
```


## B Acknowledgements

My sincerest gratitude goes to everyone that supported me and allowed this thesis to become reality. First and foremost I'd like to thank Prof. Dr. Stefanie Gräfe for supervising my thesis and giving me the opportunity to be part of her group, while still providing the freedom to pursue my scientific interests.

This thesis would not have been possible without the guidance of Dr. Stephan Kupfer. Thank you for the many collaborations and insights. I appreciate the many stimulating discussions with Philipp Traber and Phillip Seeber. They really broadened my horizon. The good work atmosphere greatly facilitated working on this thesis. I also wish to thank Philipp Traber and fellow colleagues Clara Zens and Alexander Schwab for continued use and testing of pysisyphus and providing me with invaluable feedback.

I'd like to thank Prof. Dr. Christian Hertweck for the ability to collaborate on the biaryl cross-coupling project and Veit Hänsch for the associated experimental measurements and the continued scientific exchange on this topic.

I would like express my deepest gratitude to my wife Agnes, for keeping up with our two daughters, in times of the COVID-19 pandemic and closed kindergartens. Alma and Josephine, you kept us always on our toes, which made this endeavor not easier but your presence makes everything much more worthwhile.

A heartfelt thank you goes to my parents and my grandmother for continued support and spending countless hours with Alma.

Finally I'd like to thank my friends for the rare and enjoyable encounters, that were possible in these times. Better times are on their way.

## C Selbständigkeitserklärung

Ich erkläre, dass ich die vorliegende Arbeit selbständig und unter Verwendung der angegebenen Hilfsmittel, persönlichen Mitteilungen und Quellen angefertigt habe.

Jena, 16.02.2022

## Bibliography

(1) Hoffman, D. 1979 Computer Store Manager Predicts Future Youtube, https: //www.youtube.com/watch?v=eNT1L3jGjbA,
(2) Haensch, V. G.; Neuwirth, T.; Steinmetzer, J.; Kloss, F.; Beckert, R.; Gräfe, S.; Kupfer, S.; Hertweck, C. Metal-Free Aryl Cross-Coupling Directed by Traceless Linkers. Chemistry - A European Journal 2019, 25, 16068-16073, DOI: 10. 1002/chem. 201903582 .
(3) Steinmetzer, J.; Kupfer, S.; Gräfe, S. pysisyphus: Exploring potential energy surfaces in ground and excited states. International Journal of Quantum Chemistry 2020, 121, DOI: 10.1002 /qua. 26390 .
(4) G., U. R.; Liu, J.; Hoffmann, P.; Steinmetzer, J.; Görls, H.; Kupfer, S.; Askes, S. H. C.; Neugebauer, U.; Gräfe, S.; Schiller, A. Light-responsive paper strips as CO-releasing material with a colourimetric response. Chemical Science 2017, 8, 6555-6560, DOI: $10.1039 / \mathrm{c} 7 \mathrm{sc} 01692 \mathrm{a}$.
(5) Liu, J.; Hoffmann, P.; Steinmetzer, J.; Askes, S. H.; Kupfer, S.; Görls, H.; Gräfe, S.; Neugebauer, U.; Gandra, U. R.; Schiller, A. Visible light-activated biocompatible photo-CORM for CO-release with colorimetric and fluorometric dual turn-on response. Polyhedron 2019, 172, 175-181, DOI: $10.1016 / \mathrm{j}$. poly. 2019.04.031.
(6) Amini, K.; Sclafani, M.; Steinle, T.; Le, A.-T.; Sanchez, A.; Müller, C.; Steinmetzer, J.; Yue, L.; Saavedra, J. R. M.; Hemmer, M.; Lewenstein, M.; Moshammer, R.; Pfeifer, T.; Pullen, M. G.; Ullrich, J.; Wolter, B.; Moszynski, R.; de Abajo, F. J. G.; Lin, C. D.; Gräfe, S.; Biegert, J. Imaging the Renner-Teller effect using laser-induced electron diffraction. Proceedings of the National Academy of Sciences 2019, 116, 8173-8177, DOI: $10.1073 /$ pnas. 1817465116 .
(7) Liu, X.; Amini, K.; Steinle, T.; Sanchez, A.; Shaikh, M.; Belsa, B.; Steinmetzer, J.; Le, A.-T.; Moshammer, R.; Pfeifer, T.; Ullrich, J.; Moszynski, R.; Lin, C. D.; Gräfe, S.; Biegert, J. Imaging an isolated water molecule using a single electron
wave packet. The Journal of Chemical Physics 2019, 151, 024306, DOI: 10 . 1063/1.5100520.
(8) Shillito, G. E.; Preston, D.; Traber, P.; Steinmetzer, J.; McAdam, C. J.; Crowley, J. D.; Wagner, P.; Kupfer, S.; Gordon, K. C. Excited-State Switching Frustrates the Tuning of Properties in Triphenylamine-Donor-Ligand Rhenium(I) and Platinum(II) Complexes. Inorganic Chemistry 2020, 59, 6736-6746, DOI: 10.1021/acs.inorgchem.9b03691.
(9) Belsa, B.; Amini, K.; Liu, X.; Sanchez, A.; Steinle, T.; Steinmetzer, J.; Le, A. T.; Moshammer, R.; Pfeifer, T.; Ullrich, J.; Moszynski, R.; Lin, C. D.; Gräfe, S.; Biegert, J. Laser-induced electron diffraction of the ultrafast umbrella motion in ammonia. Structural Dynamics 2021, 8, 014301, DOI: $10.1063 / 4.0000046$.
(10) Chalmers, C.; Chaloner, E. 500 years later: Henry VIII, leg ulcers and the course of history. J R Soc Med 2009, 102, 514-517, DOI: $10.1258 /$ jrsm. 2009.090286 .
(11) Bonkat, G. Syphilis und Heinrich VIII. Urologe 2018, 57, 1433-1435, DOI: 10.1007/s00120-018-0798-9.
(12) Weissfeld, A. S. Infectious Diseases and Famous People Who Succumbed to Them. Clinical Microbiology Newsletter 2009, 31, 169-172, DOI: $10.1016 /$ j. Clinmicnews.2009.10.002
(13) Fleming, A. On the Antibacterial Action of Cultures of a Penicillium, with Special Reference to their Use in the Isolation of B. influenzæ. British journal of experimental pathology 1929, 10, PMC2048009, 226-236.
(14) Bennett, J. W.; Chung, K.-T. In Advances in Applied Microbiology; Elsevier: 2001, pp 163-184, DOI: $10.1016 / \mathrm{s} 0065-2164$ (01) 49013-7.
(15) Leigh, G. J. In Catalysts for Nitrogen Fixation; Springer Netherlands: 2004, pp 33-54, DOI: 10.1007/978-1-4020-3611-8_2.
(16) Andrady, A. L. Microplastics in the marine environment. Marine Pollution Bulletin 2011, 62, 1596-1605, DOI: $10.1016 / \mathrm{j}$. marpolbul.2011.05.030.
(17) Law, K. L.; Thompson, R. C. Microplastics in the seas. Science 2014, 345, $144-$ 145, DOI: $10.1126 /$ science. 1254065 .
(18) Andrady, A. L. The plastic in microplastics: A review. Marine Pollution Bulletin 2017, 119, 12-22, DOI: $10.1016 / \mathrm{j}$. marpolbul.2017.01.082.
(19) Cox, K. D.; Covernton, G. A.; Davies, H. L.; Dower, J. F.; Juanes, F.; Dudas, S. E. Human Consumption of Microplastics. Environ. Sci. Technol. Technology 2019, 53, 7068-7074, DOI: $10.1021 /$ acs.est.9b01517.
(20) Hale, R. C.; Seeley, M. E.; Guardia, M. J. L.; Mai, L.; Zeng, E. Y. A Global Perspective on Microplastics. J. Geophys. Res. Oceans 2020, 125, DOI: 10 . 1029/2018jc014719.
(21) Andrady, A. L.; Neal, M. A. Applications and societal benefits of plastics. Phil. Trans. R. Soc. B 2009, 364, 1977-1984, DOI: $10.1098 /$ rstb. 2008.0304 ,
(22) Thompson, R. C.; Swan, S. H.; Moore, C. J.; vom Saal, F. S. Our plastic age. Phil. Trans. R. Soc. B 2009, 364, 1973-1976, DOI: $10.1098 /$ rstb. 2009.0054 .
(23) Haider, T. P.; Völker, C.; Kramm, J.; Landfester, K.; Wurm, F. R. Plastics of the Future? The Impact of Biodegradable Polymers on the Environment and on Society. Angew. Chem. Int. Ed. 2018, 58, 50-62, DOI: 10.1002 /anie. 201805766.
(24) Dogutan, D. K.; Nocera, D. G. Artificial Photosynthesis at Efficiencies Greatly Exceeding That of Natural Photosynthesis. Acc. Chem. Res. 2019, 52, 3143-3148, DOI: 10.1021 /acs.accounts.9b00380.
(25) House, R. L.; Iha, N. Y. M.; Coppo, R. L.; Alibabaei, L.; Sherman, B. D.; Kang, P.; Brennaman, M. K.; Hoertz, P. G.; Meyer, T. J. Artificial photosynthesis: Where are we now? Where can we go? Journal of Photochemistry and Photobiology C: Photochemistry Reviews 2015, 25, 32-45, DOI: $10.1016 / j . j p h o t o c h e m r e v, ~$ 2015.08.002.
(26) Zhang, B.; Sun, L. Artificial photosynthesis: opportunities and challenges of molecular catalysts. Chem. Soc. Rev. 2019, 48, 2216-2264, DOI: $10.1039 /$ c8cs00897c.
(27) Su, J.; Vayssieres, L. A Place in the Sun for Artificial Photosynthesis? ACS Energy Lett. Energy Letters 2016, 1, 121-135, DOI: 10.1021 /acsenergylett. 6 b 00059 .
(28) Butburee, T.; Chakthranont, P.; Phawa, C.; Faungnawakij, K. Beyond Artificial Photosynthesis: Prospects on Photobiorefinery. ChemCatChem 2020, 12, 18731890, DOI: $10.1002 /$ cctc. 201901856 .
(29) Kim, D.; Sakimoto, K. K.; Hong, D.; Yang, P. Artificial Photosynthesis for Sustainable Fuel and Chemical Production. Angew. Chem. Int. Ed. 2015, 54, 3259-3266, DOI: 10.1002 /anie. 201409116 .
(30) El-Khouly, M. E.; El-Mohsnawy, E.; Fukuzumi, S. Solar energy conversion: From natural to artificial photosynthesis. Journal of Photochemistry and Photobiology C: Photochemistry Reviews 2017, 31, 36-83, DOI: $10.1016 / \mathrm{j}$. jphotochemrev, 2017.02.001.
(31) Ye, S.; Ding, C.; Liu, M.; Wang, A.; Huang, Q.; Li, C. Water Oxidation Catalysts for Artificial Photosynthesis. Adv. Mater. 2019, 31, 1902069, DOI: $10.1002 /$ adma.201902069.
(32) Wang, J.; Wolf, R. M.; Caldwell, J. W.; Kollman, P. A.; Case, D. A. Development and testing of a general amber force field. Journal of Computational Chemistry 2004, 25, 1157-1174, DOI: $10.1002 /$ jcc. 20035
(33) Vanommeslaeghe, K.; Hatcher, E.; Acharya, C.; Kundu, S.; Zhong, S.; Shim, J.; Darian, E.; Guvench, O.; Lopes, P.; Vorobyov, I.; Mackerell, A. D. CHARMM general force field: A force field for drug-like molecules compatible with the CHARMM all-atom additive biological force fields. Journal of Computational Chemistry 2009, NA-NA, DOI: $10.1002 /$ jcc. 21367 .
(34) Van Duin, A. C. T.; Dasgupta, S.; Lorant, F.; Goddard, W. A. ReaxFF: A Reactive Force Field for Hydrocarbons. J. Phys. Chem. A 2001, 105, 9396-9409, DOI: $10.1021 / \mathrm{jp004368u}$.
(35) Clark, M.; Cramer, R. D.; Opdenbosch, N. V. Validation of the general purpose tripos 5.2 force field. J. Comput. Chem. 1989, 10, 982-1012, DOI: $10.1002 /$ jcc. 540100804 .
(36) Cornell, W. D.; Cieplak, P.; Bayly, C. I.; Gould, I. R.; Merz, K. M.; Ferguson, D. M.; Spellmeyer, D. C.; Fox, T.; Caldwell, J. W.; Kollman, P. A. A Second Generation Force Field for the Simulation of Proteins, Nucleic Acids, and Organic Molecules. J. Am. Chem. Soc. 1995, 117, 5179-5197, DOI: 10.1021/ ja00124a002.
(37) Harder, E.; Damm, W.; Maple, J.; Wu, C.; Reboul, M.; Xiang, J. Y.; Wang, L.; Lupyan, D.; Dahlgren, M. K.; Knight, J. L.; Kaus, J. W.; Cerutti, D. S.; Krilov, G.; Jorgensen, W. L.; Abel, R.; Friesner, R. A. OPLS3: A Force Field Providing Broad Coverage of Drug-like Small Molecules and Proteins. J. Chem. Theory Comput. 2015, 12, 281-296, DOI: $10.1021 /$ acs.jctc.5b00864.
(38) Nerenberg, P. S.; Head-Gordon, T. New developments in force fields for biomolecular simulations. Current Opinion in Structural Biology 2018, 49, 129-138, DOI: 10.1016/j.sbi.2018.02.002.
(39) Doherty, B.; Zhong, X.; Gathiaka, S.; Li, B.; Acevedo, O. Revisiting OPLS Force Field Parameters for Ionic Liquid Simulations. J. Chem. Theory Comput. 2017, 13, 6131-6145, DOI: $10.1021 /$ acs.jctc. 7b0 0520 .
(40) Senn, H. M.; Thiel, W. QM/MM Methods for Biomolecular Systems. Angewandte Chemie International Edition 2009, 48, 1198-1229, DOI: $10.1002 /$ anie. 200802019.
(41) Perilla, J. R.; Goh, B. C.; Cassidy, C. K.; Liu, B.; Bernardi, R. C.; Rudack, T.; Yu, H.; Wu, Z.; Schulten, K. Molecular dynamics simulations of large macromolecular complexes. Current Opinion in Structural Biology 2015, 31, 64-74, DOI: 10. 1016/j.sbi.2015.03.007.
(42) Marrink, S. J.; Corradi, V.; Souza, P. C.; Ingólfsson, H. I.; Tieleman, D. P.; Sansom, M. S. Computational Modeling of Realistic Cell Membranes. Chem. Rev. 2019, 119, 6184-6226, DOI: $10.1021 /$ acs.chemrev. 8b00460.
(43) Ingólfsson, H. I.; Arnarez, C.; Periole, X.; Marrink, S. J. Computational 'microscopy' of cellular membranes. J Cell Sci 2016, 129, 257-268, DOI: $10.1242 /$ jes. 176040 .
(44) Feig, M.; Yu, I.; Wang, P.-h.; Nawrocki, G.; Sugita, Y. Crowding in Cellular Environments at an Atomistic Level from Computer Simulations. J. Phys. Chem. B 2017, 121, 8009-8025, DOI: $10.1021 /$ acs.jpcb. 7b03570.
(45) Lagardère, L.; Jolly, L.-H.; Lipparini, F.; Aviat, F.; Stamm, B.; Jing, Z. F.; Harger, M.; Torabifard, H.; Cisneros, G. A.; Schnieders, M. J.; Gresh, N.; Maday, Y.; Ren, P. Y.; Ponder, J. W.; Piquemal, J.-P. Tinker-HP: a massively parallel molecular dynamics package for multiscale simulations of large complex systems with advanced point dipole polarizable force fields. Chem. Sci. 2018, 9, 956-972, DOI: $10.1039 / \mathrm{c} 7 \mathrm{sc} 04531 \mathrm{j}$.
(46) Kmiecik, S.; Gront, D.; Kolinski, M.; Wieteska, L.; Dawid, A. E.; Kolinski, A. Coarse-Grained Protein Models and Their Applications. Chem. Rev. 2016, 116, 7898-7936, DOI: $10.1021 /$ acs.chemrev. 6b00163.
(47) Li, C.; Strachan, A. Molecular scale simulations on thermoset polymers: A review. J. Polym. Sci. Part B: Polym. Phys. 2014, 53, 103-122, DOI: $10.1002 /$ polb. 23489.
(48) Suter, J. L.; Anderson, R. L.; Greenwell, H. C.; Coveney, P. V. Recent advances in large-scale atomistic and coarse-grained molecular dynamics simulation of clay minerals. J. Mater. Chem. 2009, 19, 2482, DOI: $10.1039 / \mathrm{b} 820445 \mathrm{~d}$.
(49) Kyriakou, V.; Garagounis, I.; Vourros, A.; Vasileiou, E.; Stoukides, M. An Electrochemical Haber-Bosch Process. Joule 2020, 4, 142-158, DOI: $10.1016 /$ j. joule.2019.10.006.
(50) Abghoui, Y.; Garden, A. L.; Hlynsson, V. F.; Björgvinsdóttir, S.; Ólafsdóttir, H.; Skúlason, E. Enabling electrochemical reduction of nitrogen to ammonia at ambient conditions through rational catalyst design. Physical Chemistry Chemical Physics 2015, 17, 4909-4918, DOI: $10.1039 / \mathrm{c} 4 \mathrm{cp} 04838 \mathrm{e}$.
(51) Zhao, J.; Chen, Z. Single Mo Atom Supported on Defective Boron Nitride Monolayer as an Efficient Electrocatalyst for Nitrogen Fixation: A Computational Study. Journal of the American Chemical Society 2017, 139, 12480-12487, DOI: 10.1021/jacs.7b05213.
(52) Li, Q.; He, L.; Sun, C.; Zhang, X. Computational Study of MoN2 Monolayer as Electrochemical Catalysts for Nitrogen Reduction. J. Phys. Chem. C 2017, 121, 27563-27568, DOI: $10.1021 /$ acs.jpcc. 7b10522.
(53) Guo, X.; Huang, S. Tuning nitrogen reduction reaction activity via controllable Fe magnetic moment: A computational study of single Fe atom supported on defective graphene. Electrochimica Acta 2018, 284, 392-399, DOI: $10.1016 / \mathrm{j}$. electacta.2018.07.168,
(54) Muuronen, M.; Deglmann, P.; Tomovic, Z. Design Principles for Rational Polyurethane Catalyst Development. The Journal of Organic Chemistry 2019, 84, 8202-8209, DOI: $10.1021 /$ acs.joc. $9 \mathrm{b01319}$.
(55) Chen, M.; Chen, C. Rational Design of High-Performance Phosphine Sulfonate Nickel Catalysts for Ethylene Polymerization and Copolymerization with Polar Monomers. ACS Catal. Catalysis 2017, 7, 1308-1312, DOI: $10.1021 /$ acscatal.6b03394.
(56) Dou, M.; Wang, J.; Gao, B.; Xu, C.; Yang, F. Photocatalytic difference of amoxicillin and cefotaxime under visible light by mesoporous g-C3N4: Mechanism, degradation pathway and DFT calculation. Chemical Engineering Journal 2020, 383, 123134, DOI: $10.1016 / \mathrm{j} . \mathrm{cej} .2019 .123134$.
(57) Yin, R.; Guo, W.; Wang, H.; Du, J.; Zhou, X.; Wu, Q.; Zheng, H.; Chang, J.; Ren, N. Selective degradation of sulfonamide antibiotics by peroxymonosulfate alone: Direct oxidation and nonradical mechanisms. Chemical Engineering Journal 2018, 334, 2539-2546, DOI: 10.1016/j.cej.2017.11.174.
(58) Pelalak, R.; Alizadeh, R.; Ghareshabani, E.; Heidari, Z. Degradation of sulfonamide antibiotics using ozone-based advanced oxidation process: Experimental, modeling, transformation mechanism and DFT study. Science of The Total Environment 2020, 734, 139446, DOI: $10.1016 /$ j.scitotenv.2020.139446.
(59) Abbas, A. M.; Fisal, S. R.; Orabi, A. S. Novel $\beta$-lactam antibiotic derivative and its complexes: DFT, frontier energy levels, DNA interaction, docking, physicochemical and antimicrobial properties. Journal of Molecular Structure 2020, 1218, 128487, DOI: $10.1016 / j . m o l s t r u c .2020 .128487$.
(60) Abdel-Rahman, L. H.; Abdelhamid, A. A.; Abu-Dief, A. M.; Shehata, M. R.; Bakheet, M. A. Facile synthesis, X-Ray structure of new multi-substituted aryl imidazole ligand, biological screening and DNA binding of its $\mathrm{Cr}(\mathrm{III}), \mathrm{Fe}$ (III) and $\mathrm{Cu}(\mathrm{II})$ coordination compounds as potential antibiotic and anticancer drugs. Journal of Molecular Structure 2020, 1200, 127034, DOI: $10.1016 / j . m o l s t r u c$. 2019.127034.
(61) Rajamanikandan, S.; Soundarya, S.; Paramasivam, A.; Prabhu, D.; Jeyakanthan, J.; Ramasamy, V. Computational identification of potential lead molecules targeting rho receptor of Neisseria gonorrhoeae. Journal of Biomolecular Structure and Dynamics 2021, 1-11, DOI: $10.1080 / 07391102.2021 .1885491$.
(62) Adcock, S. A.; McCammon, J. A. Molecular Dynamics: Survey of Methods for Simulating the Activity of Proteins. Chem. Rev. 2006, 106, 1589-1615, DOI: $10.1021 /$ cr040426m,
(63) Miller, W. H. Quantum dynamics of complex molecular systems. Proceedings of the National Academy of Sciences 2005, 102, 6660-6664, DOI: $10.1073 / \mathrm{pnas}$. 0408043102 .
(64) Khait, Y. G.; Panin, A. I.; Averyanov, A. S. Search for stationary points of arbitrary index by augmented Hessian method. International Journal of Quantum Chemistry 1995, 54, 329-336, DOI: 10.1002 /qua. 560540602 .
(65) Pradhan, R.; Lourderaj, U. Can reactions follow non-traditional second-order saddle pathways avoiding transition states? Physical Chemistry Chemical Physics 2019, 21, 12837-12842, DOI: $10.1039 / \mathrm{c} 9 \mathrm{cp} 02431 \mathrm{j}$.
(66) Heidrich, D.; Quapp, W. Saddle points of index 2 on potential energy surfaces and their role in theoretical reactivity investigations. Theoretica Chimica Acta 1986, 70, 89-98, DOI: $10.1007 / \mathrm{bf} 00532206$.
(67) Yin, J.; Zhang, L.; Zhang, P. High-Index Optimization-Based Shrinking Dimer Method for Finding High-Index Saddle Points. SIAM Journal on Scientific Computing 2019, 41, A3576-A3595, DOI: $10.1137 / 19 \mathrm{~m} 1253356$.
(68) Yang, C.-H.; Bhattacharyya, S.; Liu, L.; Fang, W.-h.; Liu, K. Real-time tracking of the entangled pathways in the multichannel photodissociation of acetaldehyde. Chemical Science 2020, DOI: $10.1039 / \mathrm{d0sc} 00063$ a.
(69) Fukui, K. The path of chemical reactions - the IRC approach. Accounts of Chemical Research 1981, 14, 363-368, DOI: 10.1021/ar00072a001.
(70) Simons, J.; Joergensen, P.; Taylor, H.; Ozment, J. Walking on potential energy surfaces. The Journal of Physical Chemistry 1983, 87, 2745-2753, DOI: 10 . 1021/j100238a013.
(71) JÓNSSON, H.; MILLS, G.; JACOBSEN, K. W. In Classical and Quantum Dynamics in Condensed Phase Simulations, WORLD SCIENTIFIC: 1998, DOI: 10.1142/9789812839664_0016.
(72) E, W.; Ren, W.; Vanden-Eijnden, E. String method for the study of rare events. Physical Review B 2002, 66, DOI: $10.1103 /$ physrevb. 66.052301.
(73) Quapp, W.; Bofill, J. M. A comment to the nudged elastic band method. Journal of Computational Chemistry 2010, NA-NA, DOI: $10.1002 /$ jcc. 21540 .
(74) Müller, K.; Brown, L. D. Location of saddle points and minimum energy paths by a constrained simplex optimization procedure. Theoret. Chim. Acta 1979, 53, 75-93, DOI: $10.1007 / \mathrm{bf} 00547608$.
(75) Schlegel, H. B. Optimization of equilibrium geometries and transition structures. Journal of Computational Chemistry 1982, 3, 214-218, DOI: $10.1002 / \mathrm{jcc}$. 540030212 .
(76) Birkholz, A. B.; Schlegel, H. B. Path optimization by a variational reaction coordinate method. I. Development of formalism and algorithms. The Journal of Chemical Physics 2015, 143, 244101, DOI: $10.1063 / 1.4937764$.
(77) Birkholz, A. B.; Schlegel, H. B. Path optimization by a variational reaction coordinate method. II. Improved computational efficiency through internal coordinates and surface interpolation. The Journal of Chemical Physics 2016, 144, 184101, DOI: $10.1063 / 1.4948439$,
(78) Quapp, W.; Bofill, J. M. Mechanochemistry on the Müller-Brown surface by Newton trajectories. Int J Quantum Chem 2017, 118, e25522, DOI: $10.1002 /$ qua. 25522 .
(79) Koistinen, O.-P.; Dagbjartsdóttir, F. B.; Ásgeirsson, V.; Vehtari, A.; Jónsson, H. Nudged elastic band calculations accelerated with Gaussian process regression. The Journal of Chemical Physics 2017, 147, 152720, DOI: 10.1063/1.4986787.
(80) Unke, O. T.; Brickel, S.; Meuwly, M. Sampling reactive regions in phase space by following the minimum dynamic path. J. Chem. Phys. 2019, 150, 074107, DOI: 10.1063/1.5082885.
(81) Denzel, A.; Haasdonk, B.; Kästner, J. Gaussian Process Regression for Minimum Energy Path Optimization and Transition State Search. J. Phys. Chem. A 2019, 123, 9600-9611, DOI: $10.1021 / \mathrm{acs} . j \mathrm{pca} .9 \mathrm{~b} 08239$.
(82) Dewyer, A. L.; Argüelles, A. J.; Zimmerman, P. M. Methods for exploring reaction space in molecular systems. WIREs Comput Mol Sci 2017, 8, e1354, DOI: $10.1002 / \mathrm{wcms} .1354$.
(83) Gonzalez, C.; Schlegel, H. B. Reaction path following in mass-weighted internal coordinates. The Journal of Physical Chemistry 1990, 94, 5523-5527, DOI: 10.1021/j100377a021.
(84) Schlegel, H. B. Geometry optimization. WIREs Computational Molecular Science 2011, 1, 790-809, DOI: $10.1002 / \mathrm{wcms} .34$.
(85) Ásgeirsson, V.; Jónsson, H. In Handbook of Materials Modeling; Springer International Publishing: 2020, pp 689-714, DOI: $10.1007 / 978-3-319-44677$ 6_28.
(86) Richter, M.; Marquetand, P.; Gonzalez-Vazquez, J.; Sola, I.; Gonzalez, L. SHARC: ab Initio Molecular Dynamics with Surface Hopping in the Adiabatic Representation Including Arbitrary Couplings. J. Chem. Theory Comput. 2011, 7, 1253-1258, DOI: $10.1021 /$ ct1007394.
(87) Zhang, L.; Roither, S.; Xie, X.; Kartashov, D.; Schöffler, M.; Xu, H.; Iwasaki, A.; Gräfe, S.; Okino, T.; Yamanouchi, K.; Baltuska, A.; Kitzler, M. Path-selective investigation of intense laser-pulse-induced fragmentation dynamics in triply charged 1,3-butadiene. J. Phys. B: At. Mol. Opt. Phys. 2012, 45, 085603, DOI: 10.1088/0953-4075/45/8/085603.
(88) Mai, S.; Richter, M.; Marquetand, P.; Gonzalez, L. The DNA nucleobase thymine in motion - Intersystem crossing simulated with surface hopping. Chemical Physics 2017, 482, 9-15, DOI: $10.1016 / \mathrm{j}$. chemphys.2016.10.003.
(89) Gonzalez, L.; Marquetand, P.; Richter, M.; Gonzalez-Vazquez, J.; Sola, I. In Springer Series in Chemical Physics; Springer International Publishing: 2014, pp 145-170, DOI: 10.1007/978-3-319-02051-8_7.
(90) Koch, A.; Kinzel, D.; Dröge, F.; Gräfe, S.; Kupfer, S. Photochemistry and Electron Transfer Kinetics in a Photocatalyst Model Assessed by Marcus Theory and Quantum Dynamics. The Journal of Physical Chemistry C 2017, 121, 1606616078, DOI: $10.1021 /$ acs. jpcc. 7b02812.
(91) Staniszewska, M.; Kupfer, S.; Guthmuller, J. Theoretical Investigation of the Electron-Transfer Dynamics and Photodegradation Pathways in a HydrogenEvolving Ruthenium-Palladium Photocatalyst. Chem. Eur. J. 2018, 24, 1116611176, DOI: $10.1002 /$ chem. 201801698.
(92) Staniszewska, M.; Kupfer, S.; Guthmuller, J. Effect of the Catalytic Center on the Electron Transfer Dynamics in Hydrogen-Evolving Ruthenium-Based Photocatalysts Investigated by Theoretical Calculations. J. Phys. Chem. C 2019, 123, 16003-16013, DOI: $10.1021 /$ acs. jpcc.9b03621.
(93) Toniolo, A.; Olsen, S.; Manohar, L.; Martinez, T. J. Conical intersection dynamics in solution: The chromophore of Green Fluorescent Protein. Faraday Discuss. 2004, 127, 149-163, DOI: $10.1039 / \mathrm{b} 401167 \mathrm{~h}$.
(94) Worth, G. A.; Cederbaum, L. S. BEYOND BORN-OPPENHEIMER: Molecular Dynamics Through a Conical Intersection. Annu. Rev. Phys. Chem. 2004, 55, 127-158, DOI: $10.1146 /$ annurev.physchem.55.091602.094335.
(95) Galvan, I. F.; Delcey, M. G.; Pedersen, T. B.; Aquilante, F.; Lindh, R. Analytical State-Average Complete-Active-Space Self-Consistent Field Nonadiabatic Coupling Vectors: Implementation with Density-Fitted Two-Electron Integrals and Application to Conical Intersections. J. Chem. Theory Comput. 2016, 12, 3636-3653, DOI: $10.1021 /$ acs.jctc. 6b00384.
(96) Tuna, D.; Sobolewski, A. L.; Domcke, W. Conical-Intersection Topographies Suggest That Ribose Exhibits Enhanced UV Photostability. J. Phys. Chem. B 2016, 120, 10729-10735, DOI: $10.1021 / \mathrm{acs} . j p \mathrm{cb} .6 \mathrm{~b} 09048$.
(97) Polli, D.; Altoe, P.; Weingart, O.; Spillane, K. M.; Manzoni, C.; Brida, D.; Tomasello, G.; Orlandi, G.; Kukura, P.; Mathies, R. A.; Garavelli, M.; Cerullo, G. Conical intersection dynamics of the primary photoisomerization event in vision. Nature 2010, 467, 440-443, DOI: $10.1038 /$ nature 09346 .
(98) Soto, J.; Otero, J. C.; Avila, F. J.; Pelaez, D. Conical intersections and intersystem crossings explain product formation in photochemical reactions of aryl azides. Phys. Chem. Chem. Phys. 2019, 21, 2389-2396, DOI: $10.1039 / \mathrm{c} 8 \mathrm{cp} 06974 \mathrm{c}$.
(99) Barbatti, M.; Ruckenbauer, M.; Szymczak, J. J.; Aquino, A. J. A.; Lischka, H. Nonadiabatic excited-state dynamics of polar $\pi$-systems and related model compounds of biological relevance. Phys. Chem. Chem. Phys. 2008, 10, 482-494, DOI: $10.1039 / \mathrm{b} 709315 \mathrm{~m}$.
(100) Marian, C. M. Spin-orbit coupling and intersystem crossing in molecules. WIREs Comput Mol Sci 2011, 2, 187-203, DOI: $10.1002 / \mathrm{wcms} .83$.
(101) Penfold, T. J.; Gindensperger, E.; Daniel, C.; Marian, C. M. Spin-Vibronic Mechanism for Intersystem Crossing. Chem. Rev. 2018, 118, 6975-7025, DOI: $10.1021 / \mathrm{acs}$. chemrev. 7b00617.
(102) Noda, H.; Nakanotani, H.; Adachi, C. Excited state engineering for efficient reverse intersystem crossing. Sci. Adv. 2018, 4, eaao6910, DOI: $10.1126 /$ sciadv. aao6910.
(103) Wada, Y.; Nakagawa, H.; Matsumoto, S.; Wakisaka, Y.; Kaji, H. Organic light emitters exhibiting very fast reverse intersystem crossing. Nat. Photonics 2020, 14, 643-649, DOI: $10.1038 / \mathrm{s} 41566-020-0667-0$.
(104) Kaila, V. R. I.; Send, R.; Sundholm, D. The Effect of Protein Environment on Photoexcitation Properties of Retinal. J. Phys. Chem. B 2012, 116, 2249-2258, DOI: $10.1021 / \mathrm{jp} 205918 \mathrm{~m}$.
(105) Kiefer, H. V.; Gruber, E.; Langeland, J.; Kusochek, P. A.; Bochenkova, A. V.; Andersen, L. H. Intrinsic photoisomerization dynamics of protonated Schiff-base retinal. Nat Commun 2019, 10, DOI: 10.1038/s41467-019-09225-7.
(106) Zgrablić, G.; Novello, A. M.; Parmigiani, F. Population Branching in the Conical Intersection of the Retinal Chromophore Revealed by Multipulse Ultrafast Optical Spectroscopy. J. Am. Chem. Soc. 2011, 134, 955-961, DOI: $10.1021 /$ ja205763x.
(107) Nathans, J. Determinants of visual pigment absorbance: role of charged amino acids in the putative transmembrane segments. Biochemistry 1990, 29, 937-942, DOI: $10.1021 / \mathrm{bi00456a013}$.
(108) Andersen, L. H.; Nielsen, I. B.; Kristensen, M. B.; Ghazaly, M. O. A. E.; Haacke, S.; Nielsen, M. B.; Petersen, M. A. Absorption of Schiff-Base Retinal Chromophores in Vacuo. J. Am. Chem. Soc. 2005, 127, 12347-12350, DOI: $10.1021 /$ ja051638j.
(109) Coughlan, N. J. A.; Adamson, B. D.; Gamon, L.; Catani, K.; Bieske, E. J. Retinal shows its true colours: photoisomerization action spectra of mobility-selected isomers of the retinal protonated Schiff base. Phys. Chem. Chem. Phys. 2015, 17, 22623-22631, DOI: $10.1039 / \mathrm{c} 5 \mathrm{cp} 03611 \mathrm{a}$.
(110) Knudsen, J. L.; Kluge, A.; Bochenkova, A. V.; Kiefer, H. V.; Andersen, L. H. The UV-visible action-absorption spectrum of all-trans and 11-cis protonated Schiff base retinal in the gas phase. Physical Chemistry Chemical Physics 2018, 20, 7190-7194, DOI: 10.1039/c7cp07512j.
(111) Polli, D.; Altoè, P.; Weingart, O.; Spillane, K. M.; Manzoni, C.; Brida, D.; Tomasello, G.; Orlandi, G.; Kukura, P.; Mathies, R. A.; Garavelli, M.; Cerullo, G. Conical intersection dynamics of the primary photoisomerization event in vision. Nature 2010, 467, 440-443, DOI: $10.1038 /$ nature 09346 .
(112) Arshavsky, V. Y.; Lamb, T. D.; Pugh, E. N. G Proteins and Phototransduction. Annual Review of Physiology 2002, 64, 153-187, DOI: $10.1146 /$ annurev. physiol.64.082701.102229.
(113) Garavelli, M. Computational Organic Photochemistry: Strategy, Achievements and Perspectives. Theor Chem Acc 2006, 116, 87-105, DOI: $10.1007 / \mathrm{s} 00214-$ 005-0030-z.
(114) Persico, M.; Granucci, G., Photochemistry; Springer International Publishing: 2018, DOI: $10.1007 / 978-3-319-89972-5$.
(115) Hill, N. S.; Coote, M. L. In Annual Reports in Computational Chemistry; Elsevier: 2019, pp 203-285, DOI: $10.1016 / \mathrm{bs}$. arcc. 2019.08 .008 .
(116) Francés-Monerris, A.; Gros, P. C.; Assfeld, X.; Monari, A.; Pastore, M. Toward Luminescent Iron Complexes: Unravelling the Photophysics by Computing Potential Energy Surfaces. ChemPhotoChem 2019, 3, 666-683, DOI: $10.1002 / \mathrm{cptc}$. 201900100
(117) Loos, P.-F.; Jacquemin, D. Evaluating 0-0 Energies with Theoretical Tools: A Short Review. ChemPhotoChem 2019, 3, 684-696, DOI: $10.1002 / \mathrm{cptc}$. 201900070
(118) Plasser, F. Visualisation of Electronic Excited-State Correlation in Real Space. ChemPhotoChem 2019, 3, 702-706, DOI: $10.1002 /$ cptc. 201900014.
(119) Mai, S.; Gonzalez, L. Molecular Photochemistry: Recent Developments in Theory. Angew. Chem. Int. Ed. 2020, 59, 16832-16846, DOI: 10.1002 /anie. 201916381.
(120) Westermayr, J.; Marquetand, P. Machine Learning for Electronically Excited States of Molecules. Chem. Rev. 2020, DOI: 10.1021 /acs.chemrev. 0 c00749.
(121) Neese, F. Software update: the ORCA program system version 4.0. WIREs Computational Molecular Science 2017, 8, DOI: $10.1002 / \mathrm{wcms}$. 1327 .
(122) Furche, F.; Ahlrichs, R.; Hättig, C.; Klopper, W.; Sierka, M.; Weigend, F. Turbomole. Wiley Interdisciplinary Reviews: Computational Molecular Science 2013, 4, 91-100, DOI: $10.1002 / \mathrm{wcms} .1162$.
(123) Balasubramani, S. G. et al. TURBOMOLE: Modular program suite for ab initio quantum-chemical and condensed-matter simulations. J. Chem. Phys. 2020, 152, 184107, DOI: $10.1063 / 5.0004635$.
(124) Larsen, A. H. et al. The atomic simulation environment-a Python library for working with atoms. Journal of Physics: Condensed Matter 2017, 29, 273002, DOI: $10.1088 / 1361-648 \mathrm{x} / \mathrm{aa} 680 \mathrm{e}$
(125) Kloss, F.; Neuwirth, T.; Haensch, V. G.; Hertweck, C. Metal-Free Synthesis of Pharmaceutically Important Biaryls by Photosplicing. Angewandte Chemie 2018, 130, 14684-14689, DOI: 10.1002 /ange. 201805961 .
(126) Mustroph, H. Potential-Energy Surfaces, the Born-Oppenheimer Approximations, and the Franck-Condon Principle: Back to the Roots. ChemPhysChem 2016, 17, 2616-2629, DOI: $10.1002 /$ cphc. 201600243 .
(127) Klahn, B.; Bingel, W. A. The convergence of the Rayleigh-Ritz Method in quantum chemistry. Theoretica Chimica Acta 1977, 44, 9-26, DOI: $10.1007 /$ bf00548026.
(128) Pauli, W. The Connection Between Spin and Statistics. Physical Review 1940, 58, 716-722, DOI: $10.1103 /$ physrev.58.716.
(129) Hartree, D. R. The Wave Mechanics of an Atom with a Non-Coulomb Central Field. Part I. Theory and Methods. Mathematical Proceedings of the Cambridge Philosophical Society 1928, 24, 89-110, DOI: $10.1017 /$ s0305004100011919.
(130) Shiozaki, T.; Hirata, S. Grid-based numerical Hartree-Fock solutions of polyatomic molecules. Physical Review A 2007, 76, DOI: $10.1103 /$ physreva. 76 . 040503.
(131) Morrison, J. C.; Kobus, J. In Novel Electronic Structure Theory: General Innovations and Strongly Correlated Systems; Elsevier: 2018, pp 103-116, DOI: $10.1016 / \mathrm{bs.aiq} .2017 .06 .001$.
(132) Lehtola, S. Fully numerical Hartree-Fock and density functional calculations. II. Diatomic molecules. International Journal of Quantum Chemistry 2019, 119, DOI: 10.1002 /qua. 25944 .
(133) Jiao, L. G.; Ma, J.; Ho, Y. K. Development of the numerical Hartree-Fock method for screened atoms with Debye-Hückel potential. Journal of Physics: Conference Series 2020, 1412, 132018, DOI: $10.1088 / 1742-6596 / 1412 / 13 / 132018$.
(134) Electronic wave functions - I. A general method of calculation for the stationary states of any molecular system. Proceedings of the Royal Society of London. Series A. Mathematical and Physical Sciences 1950, 200, 542-554, DOI: $10.1098 /$ rspa.1950.0036.
(135) Electronic wave functions II. A calculation for the ground state of the beryllium atom. Proceedings of the Royal Society of London. Series A. Mathematical and Physical Sciences 1950, 201, 125-137, DOI: $10.1098 /$ rspa.1950.0047.
(136) Huzinaga, S. Gaussian-Type Functions for Polyatomic Systems. I. The Journal of Chemical Physics 1965, 42, 1293-1302, DOI: $10.1063 / 1.1696113$.
(137) Gill, P. M. In Advances in Quantum Chemistry; Elsevier: 1994, pp 141-205, DOI: $10.1016 / \mathrm{s} 0065-3276(08) 60019-2$.
(138) Shavitt, I. The History and Evolution of Gaussian Basis Sets. Israel Journal of Chemistry 1993, 33, 357-367, DOI: $10.1002 /$ ijch. 199300044 .
(139) Silverstone, H. J. On the Evaluation of Two-Center Overlap and Coulomb Integrals with Noninteger-n Slater-Type Orbitals. The Journal of Chemical Physics 1966, 45, 4337-4341, DOI: $10.1063 / 1.1727493$.
(140) Jones, H. W. Computer-generated formulas for overlap integrals of slater-type orbitals. International Journal of Quantum Chemistry 1980, 18, 709-713, DOI: $10.1002 /$ qua. 560180306 .
(141) Guseinov, I. I. Evaluation of two-center overlap and nuclear-attraction integrals for Slater-type orbitals. Physical Review A 1985, 32, 1864-1866, DOI: $10.1103 /$ physreva.32.1864.
(142) Jones, H. W. Comprehensive strategy for the calculation of overlap integrals with Slater-type orbitals. International Journal of Quantum Chemistry 1997, 61, 881-889, DOI: $10.1002 /($ sici) 1097-461x(1997) 61:6<881: : aid-qual>3.0.co;2-s.
(143) Guseinov, I.; Mamedov, B. Computation of molecular integrals over Slater type orbitals I. Calculations of overlap integrals using recurrence relations. Journal of Molecular Structure: THEOCHEM 1999, 465, 1-6, DOI: $10.1016 / \mathrm{s} 0166$ -1280(98)00129-8.
(144) Guseinov, I.; Mamedov, B. Evaluation of overlap integrals with integer and noninteger n Slater-type orbitals using auxiliary functions. Journal of Molecular Modeling 2002, 8, 272-276, DOI: $10.1007 / \mathrm{s} 00894-002-0098-5$.
(145) Stewart, R. F. Small Gaussian Expansions of Slater-Type Orbitals. The Journal of Chemical Physics 1970, 52, 431-438, DOI: $10.1063 / 1.1672702$.
(146) Hariharan, P. C.; Pople, J. A. The influence of polarization functions on molecular orbital hydrogenation energies. Theoretica Chimica Acta 1973, 28, 213-222, DOI: $10.1007 / \mathrm{bf0} 0533485$.
(147) McLean, A. D.; Chandler, G. S. Contracted Gaussian basis sets for molecular calculations. I. Second row atoms, $\mathrm{Z}=11-18$. The Journal of Chemical Physics 1980, 72, 5639-5648, DOI: $10.1063 / 1.438980$.
(148) Weigend, F.; Furche, F.; Ahlrichs, R. Gaussian basis sets of quadruple zeta valence quality for atoms H-Kr. The Journal of Chemical Physics 2003, 119, 1275312762, DOI: $10.1063 / 1.1627293$.
(149) Martin, J. M.; El-Yazal, J.; François, J.-P. Basis set convergence and performance of density functional theory including exact exchange contributions for geometries and harmonic frequencies. Molecular Physics 1995, 86, 1437-1450, DOI: 10. 1080/00268979500102841.
(150) Helgaker, T.; Klopper, W.; Tew, D. P. Quantitative quantum chemistry. Molecular Physics 2008, 106, 2107-2143, DOI: 10.1080/00268970802258591.
(151) Hall, G. G. The molecular orbital theory of chemical valency VIII. A method of calculating ionization potentials. Proceedings of the Royal Society of London. Series A. Mathematical and Physical Sciences 1951, 205, 541-552, DOI: 10. 1098/rspa.1951.0048.
(152) Roothaan, C. C. J. New Developments in Molecular Orbital Theory. Reviews of Modern Physics 1951, 23, 69-89, DOI: $10.1103 /$ revmodphys.23.69.
(153) Hohenberg, P.; Kohn, W. Inhomogeneous Electron Gas. Physical Review 1964, 136, B864-B871, DOI: $10.1103 /$ physrev.136.b864.
(154) Sahni, V. In Quantal Density Functional Theory; Springer Berlin Heidelberg: 2004, pp 99-123, DOI: 10.1007/978-3-662-09624-6_4.
(155) Wang, Y. A.; Carter, E. A. In Theoretical Methods in Condensed Phase Chemistry; Kluwer Academic Publishers, pp 117-184, DOI: 10.1007/0-306-46949-9_5.
(156) Lignères, V. L.; Carter, E. A. In Handbook of Materials Modeling; Springer Netherlands: 2005, pp 137-148, DOI: 10.1007/978-1-4020-3286-8_9.
(157) Xia, J.; Huang, C.; Shin, I.; Carter, E. A. Can orbital-free density functional theory simulate molecules? The Journal of Chemical Physics 2012, 136, 084102, DOI: $10.1063 / 1.3685604$.
(158) Karasiev, V. V.; Trickey, S. B. In Advances in Quantum Chemistry; Elsevier: 2015, pp 221-245, DOI: $10.1016 /$ bs.aiq.2015.02.004.
(159) Meyer, R.; Weichselbaum, M.; Hauser, A. W. Machine Learning Approaches toward Orbital-free Density Functional Theory: Simultaneous Training on the Kinetic Energy Density Functional and Its Functional Derivative. Journal of Chemical Theory and Computation 2020, 16, 5685-5694, DOI: $10.1021 /$ acs. jctc.0c00580.
(160) Golub, P.; Manzhos, S. CONUNDrum: A program for orbital-free density functional theory calculations. Computer Physics Communications 2020, 256, 107365, DOI: $10.1016 / \mathrm{j} . \mathrm{cpc} .2020 .107365$,
(161) Kohn, W.; Sham, L. J. Self-Consistent Equations Including Exchange and Correlation Effects. Physical Review 1965, 140, A1133-A1138, DOI: $10.1103 /$ physrev.140.a1133.
(162) Becke, A. D. Perspective: Fifty years of density-functional theory in chemical physics. The Journal of Chemical Physics 2014, 140, 18A301, DOI: 10.1063/ 1.4869598 .
(163) Becke, A. D. Density-functional exchange-energy approximation with correct asymptotic behavior. Physical Review A 1988, 38, 3098-3100, DOI: $10.1103 /$ physreva.38.3098.
(164) Dirac, P. A. M. Note on Exchange Phenomena in the Thomas Atom. Mathematical Proceedings of the Cambridge Philosophical Society 1930, 26, 376-385, DOI: 10.1017/s0305004100016108.
(165) Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient Approximation Made Simple. Physical Review Letters 1996, 77, 3865-3868, DOI: $10.1103 /$ physrevlett.77.3865.
(166) Becke, A. D. Density-functional thermochemistry. III. The role of exact exchange. The Journal of Chemical Physics 1993, 98, 5648-5652, DOI: 10.1063/1. 464913.
(167) Lee, C.; Yang, W.; Parr, R. G. Development of the Colle-Salvetti correlationenergy formula into a functional of the electron density. Physical Review $B 1988$, 37, 785-789, DOI: $10.1103 /$ physrevb. 37.785 .
(168) Curtiss, L. A.; Raghavachari, K.; Trucks, G. W.; Pople, J. A. Gaussian-2 theory for molecular energies of first- and second-row compounds. The Journal of Chemical Physics 1991, 94, 7221-7230, DOI: $10.1063 / 1.460205$.
(169) Perdew, J. P.; Kurth, S.; Zupan, A.; Blaha, P. Accurate Density Functional with Correct Formal Properties: A Step Beyond the Generalized Gradient Approximation. Physical Review Letters 1999, 82, 2544-2547, DOI: $10.1103 /$ physrevlett.82.2544.
(170) Adamo, C.; Ernzerhof, M.; Scuseria, G. E. The meta-GGA functional: Thermochemistry with a kinetic energy density dependent exchange-correlation functional. The Journal of Chemical Physics 2000, 112, 2643-2649, DOI: 10.1063/1. 480838 .
(171) Grimme, S.; Neese, F. Double-hybrid density functional theory for excited electronic states of molecules. The Journal of Chemical Physics 2007, 127, 154116, DOI: $10.1063 / 1.2772854$.
(172) Goerigk, L.; Grimme, S. Efficient and Accurate Double-Hybrid-Meta-GGA Density Functionals-Evaluation with the Extended GMTKN30 Database for General Main Group Thermochemistry, Kinetics, and Noncovalent Interactions. Journal of Chemical Theory and Computation 2010, 7, 291-309, DOI: 10.1021/ ct100466k.
(173) Neese, F. A critical evaluation of DFT, including time-dependent DFT, applied to bioinorganic chemistry. JBIC Journal of Biological Inorganic Chemistry 2006, 11, 702-711, DOI: $10.1007 / \mathrm{s} 00775-006-0138-1$.
(174) Maitra, N. T.; Zhang, F.; Cave, R. J.; Burke, K. Double excitations within timedependent density functional theory linear response. The Journal of Chemical Physics 2004, 120, 5932-5937, DOI: $10.1063 / 1.1651060$.
(175) Elliott, P.; Goldson, S.; Canahui, C.; Maitra, N. T. Perspectives on doubleexcitations in TDDFT. Chemical Physics 2011, 391, 110-119, DOI: $10.1016 /$ j.chemphys.2011.03.020.
(176) Sagredo, F.; Burke, K. Accurate double excitations from ensemble density functional calculations. The Journal of Chemical Physics 2018, 149, 134103, DOI: 10.1063/1.5043411.
(177) Loos, P.-F.; Boggio-Pasqua, M.; Scemama, A.; Caffarel, M.; Jacquemin, D. Reference Energies for Double Excitations. Journal of Chemical Theory and Computation 2019, 15, 1939-1956, DOI: $10.1021 /$ acs. jctc. 8 b 01205 .
(178) Tawada, Y.; Tsuneda, T.; Yanagisawa, S.; Yanai, T.; Hirao, K. A long-rangecorrected time-dependent density functional theory. The Journal of Chemical Physics 2004, 120, 8425-8433, DOI: $10.1063 / 1.1688752$.
(179) Yanai, T.; Tew, D. P.; Handy, N. C. A new hybrid exchange-correlation functional using the Coulomb-attenuating method (CAM-B3LYP). Chemical Physics Letters 2004, 393, 51-57, DOI: $10.1016 / \mathrm{j}$. cplett. 2004.06 .011 .
(180) Gerber, I. C.; Ángyán, J. G. Hybrid functional with separated range. Chemical Physics Letters 2005, 415, 100-105, DOI: $10.1016 / \mathrm{j}$. cplett. 2005.08.060.
(181) Vydrov, O. A.; Scuseria, G. E. Assessment of a long-range corrected hybrid functional. The Journal of Chemical Physics 2006, 125, 234109, DOI: $10.1063 /$ 1.2409292 ,
(182) Gerber, I. C.; Angyan, J. G.; Marsman, M.; Kresse, G. Range separated hybrid density functional with long-range Hartree-Fock exchange applied to solids. The Journal of Chemical Physics 2007, 127, 054101, DOI: $10.1063 / 1.2759209$.
(183) Mardirossian, N.; Head-Gordon, M. $\omega$ B97X-V: A 10-parameter, range-separated hybrid, generalized gradient approximation density functional with nonlocal correlation, designed by a survival-of-the-fittest strategy. Physical Chemistry Chemical Physics 2014, 16, 9904, DOI: $10.1039 / \mathrm{c} 3 \mathrm{cp} 54374 \mathrm{a}$.
(184) Jaramillo, J.; Scuseria, G. E.; Ernzerhof, M. Local hybrid functionals. The Journal of Chemical Physics 2003, 118, 1068-1073, DOI: $10.1063 / 1.1528936$.
(185) Maier, T. M.; Arbuznikov, A. V.; Kaupp, M. Local hybrid functionals: Theory, implementation, and performance of an emerging new tool in quantum chemistry and beyond. Wiley Interdisciplinary Reviews: Computational Molecular Science 2018, 9, e1378, DOI: $10.1002 /$ wcms.1378.
(186) Haasler, M.; Maier, T. M.; Grotjahn, R.; Gückel, S.; Arbuznikov, A. V.; Kaupp, M. A Local Hybrid Functional with Wide Applicability and Good Balance between (De)Localization and Left-Right Correlation. Journal of Chemical Theory and Computation 2020, 16, 5645-5657, DOI: $10.1021 /$ acs.jctc.0c00498.
(187) Runge, E.; Gross, E. K. U. Density-Functional Theory for Time-Dependent Systems. Physical Review Letters 1984, 52, 997-1000, DOI:10.1103/physrevlett. 52.997.
(188) Gross, E. K. U.; Dobson, J. F.; Petersilka, M. In Topics in Current Chemistry; Springer-Verlag, pp 81-172, DOI: $10.1007 / \mathrm{bfb0} 016643$.
(189) Jensen, F., Introduction to Computational Chemistry; Wiley: 2017.
(190) Lopata, K.; Govind, N. Modeling Fast Electron Dynamics with Real-Time TimeDependent Density Functional Theory: Application to Small Molecules and Chromophores. Journal of Chemical Theory and Computation 2011, 7, 13441355 , DOI: $10.1021 /$ ct 200137 z .
(191) Fischer, S. A.; Cramer, C. J.; Govind, N. Excited State Absorption from RealTime Time-Dependent Density Functional Theory. Journal of Chemical Theory and Computation 2015, 11, 4294-4303, DOI: 10.1021/acs.jctc.5.b00473.
(192) Goings, J. J.; Lestrange, P. J.; Li, X. Real-time time-dependent electronic structure theory. WIREs Computational Molecular Science 2017, 8, DOI: $10.1002 / \mathrm{wcms}$. 1341.
(193) Müller, C.; Sharma, M.; Sierka, M. Real-time time-dependent density functional theory using density fitting and the continuous fast multipole method. Journal of Computational Chemistry 2020, 41, 2573-2582, DOI: $10.1002 /$ jcc. 26412.
(194) CASIDA, M. E. In Recent Advances in Density Functional Methods; WORLD SCIENTIFIC: 1995, pp 155-192, DOI: 10.1142/9789812830586_0005.
(195) Casida, M.; Huix-Rotllant, M. Progress in Time-Dependent Density-Functional Theory. Annual Review of Physical Chemistry 2012, 63, 287-323, DOI: 10 . 1146/annurev-physchem-032511-143803.
(196) Giuliani, G.; Vignale, G., Quantum Theory of the Electron Liquid; Cambridge University Press: 2005, DOI: $10.1017 / \mathrm{cbo9780511619915}$.
(197) Ullrich, C. A., Time-Dependent Density-Functional Theory; Oxford University Press: 2011, DOI: 10.1093/acprof:oso/9780199563029.001.0001.
(198) Furche, F.; Ahlrichs, R. Adiabatic time-dependent density functional methods for excited state properties. The Journal of Chemical Physics 2002, 117, 7433-7447, DOI: $10.1063 / 1.1508368$.
(199) Abraham, M. J.; Murtola, T.; Schulz, R.; Pall, S.; Smith, J. C.; Hess, B.; Lindahl, E. GROMACS: High performance molecular simulations through multi-level parallelism from laptops to supercomputers. SoftwareX 2015, 1-2, 19-25, DOI: 10.1016/j.softx.2015.06.001.
(200) Eastman, P.; Swails, J.; Chodera, J. D.; McGibbon, R. T.; Zhao, Y.; Beauchamp, K. A.; Wang, L.-P.; Simmonett, A. C.; Harrigan, M. P.; Stern, C. D.; Wiewiora, R. P.; Brooks, B. R.; Pande, V. S. OpenMM 7: Rapid development of high performance algorithms for molecular dynamics. PLOS Computational Biology 2017, 13, ed. by Gentleman, R., e1005659, DOI: $10.1371 /$ journal.pcbi. 1005659 .
(201) Kutzner, C.; Pall, S.; Fechner, M.; Esztermann, A.; Groot, B. L.; Grubmüller, H. More bang for your buck: Improved use of GPU nodes for GROMACS 2018. Journal of Computational Chemistry 2019, 40, 2418-2431, DOI: $10.1002 / \mathrm{jcc}$. 26011.
(202) Thiel, W. Semiempirical quantum-chemical methods. Wiley Interdisciplinary Reviews: Computational Molecular Science 2013, 4, 145-157, DOI: $10.1002 /$ wcms.1161.
(203) Akimov, A. V.; Prezhdo, O. V. Large-Scale Computations in Chemistry: A Bird's Eye View of a Vibrant Field. Chemical Reviews 2015, 115, 5797-5890, DOI: 10.1021/cr500524c.
(204) Koskinen, P.; Mäkinen, V. Density-functional tight-binding for beginners. Computational Materials Science 2009, 47, 237-253, DOI: $10.1016 /$ j.commatsci. 2009.07.013.
(205) Elstner, M.; Seifert, G. Density functional tight binding. Philosophical Transactions of the Royal Society A: Mathematical, Physical and Engineering Sciences 2014, 372, 20120483, DOI: $10.1098 /$ rsta.2012.0483.
(206) Hourahine, B. et al. DFTB+, a software package for efficient approximate density functional theory based atomistic simulations. The Journal of Chemical Physics 2020, 152, 124101, DOI: $10.1063 / 1.5143190$.
(207) Gaus, M.; Cui, Q.; Elstner, M. DFTB3: Extension of the Self-Consistent-Charge Density-Functional Tight-Binding Method (SCC-DFTB). Journal of Chemical Theory and Computation 2011, 7, 931-948, DOI: $10.1021 / \mathrm{ct} 100684 \mathrm{~s}$.
(208) Porezag, D.; Frauenheim, T.; Köhler, T.; Seifert, G.; Kaschner, R. Construction of tight-binding-like potentials on the basis of density-functional theory: Application to carbon. Physical Review B 1995, 51, 12947-12957, DOI: $10.1103 /$ physrevb. 51.12947.
(209) Elstner, M.; Porezag, D.; Jungnickel, G.; Elsner, J.; Haugk, M.; Frauenheim, T.; Suhai, S.; Seifert, G. Self-consistent-charge density-functional tight-binding method for simulations of complex materials properties. Physical Review $B 1998$, 58, 7260-7268, DOI: $10.1103 /$ physrevb 58.7260 .
(210) Brandenburg, J. G.; Grimme, S. Accurate Modeling of Organic Molecular Crystals by Dispersion-Corrected Density Functional Tight Binding (DFTB). The Journal of Physical Chemistry Letters 2014, 5, 1785-1789, DOI: $10.1021 /$ jz500755u.
(211) Caldeweyher, E.; Bannwarth, C.; Grimme, S. Extension of the D3 dispersion coefficient model. The Journal of Chemical Physics 2017, 147, 034112, DOI: 10.1063/1.4993215.
(212) Bannwarth, C.; Ehlert, S.; Grimme, S. GFN2-xTB-An Accurate and Broadly Parametrized Self-Consistent Tight-Binding Quantum Chemical Method with Multipole Electrostatics and Density-Dependent Dispersion Contributions. Journal of Chemical Theory and Computation 2019, 15, 1652-1671, DOI: $10.1021 /$ acs.jctc.8b01176.
(213) Caldeweyher, E.; Ehlert, S.; Hansen, A.; Neugebauer, H.; Spicher, S.; Bannwarth, C.; Grimme, S. A generally applicable atomic-charge dependent London dispersion correction. The Journal of Chemical Physics 2019, 150, 154122, DOI: $10.1063 /$ 1.5090222 ,
(214) Gaus, M.; Goez, A.; Elstner, M. Parametrization and Benchmark of DFTB3 for Organic Molecules. Journal of Chemical Theory and Computation 2012, 9, 338-354, DOI: $10.1021 /$ ct 300849 w .
(215) Jenness, G. R.; Bresnahan, C. G.; Shukla, M. K. Adventures in DFTB: Toward an Automatic Parameterization Scheme. Journal of Chemical Theory and Computation 2020, 16, 6894-6903, DOI: $10.1021 /$ acs.jctc.0c00842.
(216) Grimme, S.; Bannwarth, C.; Shushkov, P. A Robust and Accurate Tight-Binding Quantum Chemical Method for Structures, Vibrational Frequencies, and Noncovalent Interactions of Large Molecular Systems Parametrized for All spd-Block Elements (Z = 1-86). Journal of Chemical Theory and Computation 2017, 13, 1989-2009, DOI: $10.1021 /$ acs. jctc. 7b00118.
(217) Bannwarth, C.; Caldeweyher, E.; Ehlert, S.; Hansen, A.; Pracht, P.; Seibert, J.; Spicher, S.; Grimme, S. Extended tight-binding quantum chemistry methods. WIREs Computational Molecular Science 2020, DOI: $10.1002 / \mathrm{wcms} .1493$.
(218) Closser, K. D.; Gessner, O.; Head-Gordon, M. Simulations of the dissociation of small helium clusters with ab initio molecular dynamics in electronically excited states. The Journal of Chemical Physics 2014, 140, 134306, DOI: $10.1063 / 1$. 4869193.
(219) Li, J.-H.; Chai, J.-D.; Guo, G.-Y.; Hayashi, M. The quantified NTO analysis for the electronic excitations of molecular many-body systems. Chemical Physics Letters 2011, 514, 362-367, DOI: $10.1016 /$ j.cplett.2011.08.066.
(220) Li, J.-H.; Chai, J.-D.; Guo, G.-Y.; Hayashi, M. Significant role of the DNA backbone in mediating the transition origin of electronic excitations of B-DNA implication from long range corrected TDDFT and quantified NTO analysis. Physical Chemistry Chemical Physics 2012, 14, 9092, DOI: $10.1039 / \mathrm{c} 2 \mathrm{cp} 23676 \mathrm{a}$
(221) Plasser, F. TheoDORE: A toolbox for a detailed and automated analysis of electronic excited state computations. The Journal of Chemical Physics 2020, 152, 084108, DOI: $10.1063 / 1.5143076$.
(222) Song, H.; Fischer, S. A.; Zhang, Y.; Cramer, C. J.; Mukamel, S.; Govind, N.; Tretiak, S. First Principles Nonadiabatic Excited-State Molecular Dynamics in NWChem. Journal of Chemical Theory and Computation 2020, 16, 6418-6427, DOI: 10.1021/acs.jctc.0c00295.
(223) García, J. S.; Boggio-Pasqua, M.; Ciofini, I.; Campetella, M. Excited state tracking during the relaxation of coordination compounds. Journal of Computational Chemistry 2019, 40, 1420-1428, DOI: $10.1002 /$ jcc. 25800 .
(224) Campetella, M.; García, J. S. Following the evolution of excited states along photochemical reaction pathways. Journal of Computational Chemistry 2020, DOI: $10.1002 /$ jcc. 26162 ,
(225) Plasser, F.; Ruckenbauer, M.; Mai, S.; Oppel, M.; Marquetand, P.; González, L. Efficient and Flexible Computation of Many-Electron Wave Function Overlaps. Journal of Chemical Theory and Computation 2016, 12, 1207-1219, DOI: 10. 1021/acs.jctc.5b01148.
(226) Sapunar, M.; Piteša, T.; Davidović, D.; Došlić, N. Highly Efficient Algorithms for CIS Type Excited State Wave Function Overlaps. Journal of Chemical Theory and Computation 2019, 15, 3461-3469, DOI: 10.1021/acs.jctc.9b00235,
(227) Martin, R. L. Natural transition orbitals. The Journal of Chemical Physics 2003, 118, 4775-4777, DOI: $10.1063 / 1.1558471$.
(228) Plasser, F.; Lischka, H. Analysis of Excitonic and Charge Transfer Interactions from Quantum Chemical Calculations. Journal of Chemical Theory and Computation 2012, 8, 2777-2789, DOI: $10.1021 /$ ct 300307 c .
(229) Shepard, R. In Modern Electronic Structure Theory; World Scientific Publishing Company: 1995, pp 345-458, DOI: $10.1142 / 9789812832108 \_0007$.
(230) Sorensen, D. C. Newton's Method with a Model Trust Region Modification. SIAM Journal on Numerical Analysis 1982, 19, 409-426, DOI: $10.1137 / 0719026$.
(231) Banerjee, A.; Adams, N.; Simons, J.; Shepard, R. Search for stationary points on surfaces. The Journal of Physical Chemistry 1985, 89, 52-57, DOI: $10.1021 /$ j100247a015.
(232) Bakken, V.; Helgaker, T. The efficient optimization of molecular geometries using redundant internal coordinates. The Journal of Chemical Physics 2002, 117, 9160-9174, DOI: $10.1063 / 1.1515483$.
(233) Asmundis, R. D.; di Serafino, D.; Riccio, F.; Toraldo, G. On spectral properties of steepest descent methods. IMA Journal of Numerical Analysis 2013, 33, 14161435, DOI: $10.1093 /$ imanum/drs056.
(234) Numerical Optimization; Springer New York: 2006, DOI: 10.1007/978-0-387-40065-5.
(235) Broyden, C. G. The Convergence of a Class of Double-rank Minimization Algorithms 1. General Considerations. IMA Journal of Applied Mathematics 1970, 6, 76-90, DOI: $10.1093 /$ imamat/6.1.76.
(236) Fletcher, R. A new approach to variable metric algorithms. The Computer Journal 1970, 13, 317-322, DOI: $10.1093 / \mathrm{comjnl} / 13.3 .317$.
(237) Goldfarb, D. A family of variable-metric methods derived by variational means. Mathematics of Computation 1970, 24, 23-23, DOI: $10.1090 / \mathrm{s} 0025-5718$ -1970-0258249-6.
(238) Shanno, D. F. Conditioning of quasi-Newton methods for function minimization. Mathematics of Computation 1970, 24, 647-647, DOI: $10.1090 / \mathrm{s} 0025-5718$ -1970-0274029-x.
(239) Birkholz, A. B.; Schlegel, H. B. Exploration of some refinements to geometry optimization methods. Theoretical Chemistry Accounts 2016, 135, DOI: 10. 1007/s00214-016-1847-3.
(240) Bofill, J. M. Updated Hessian matrix and the restricted step method for locating transition structures. Journal of Computational Chemistry 1994, 15, 1-11, DOI: $10.1002 / \mathrm{jcc} .540150102$.
(241) Lindh, R.; Bernhardsson, A.; Karlström, G.; Malmqvist, P. A. On the use of a Hessian model function in molecular geometry optimizations. Chemical Physics Letters 1995, 241, 423-428, DOI: 10.1016/0009-2614(95)00646-1.
(242) Fischer, T. H.; Almlof, J. General methods for geometry and wave function optimization. The Journal of Physical Chemistry 1992, 96, 9768-9774, DOI: 10.1021/j100203a036.
(243) Swart, M.; Bickelhaupt, F. M. Optimization of strong and weak coordinates. International Journal of Quantum Chemistry 2006, 106, 2536-2544, DOI: 10 . 1002/qua. 21049 .
(244) Vreven, T.; Morokuma, K.; Farkas, Ö.; Schlegel, H. B.; Frisch, M. J. Geometry optimization with QM/MM, ONIOM, and other combined methods. I. Microiterations and constraints. Journal of Computational Chemistry 2003, 24, 760-769, DOI: $10.1002 / \mathrm{jcc} .10156$.
(245) Farkas, Ö.; Schlegel, H. B. Methods for optimizing large molecules. II. Quadratic search. The Journal of Chemical Physics 1999, 111, 10806-10814, DOI: 10 . 1063/1.480484.
(246) Eckert, F.; Pulay, P.; Werner, H.-J. Ab initio geometry optimization for large molecules. Journal of Computational Chemistry 1997, 18, 1473-1483, DOI: 10. 1002/(sici)1096-987x(199709)18:12<1473::aid-jcc5>3.0.co;2g.
(247) Polak, E.; Ribiere, G. Note sur la convergence de méthodes de directions conjuguées. ESAIM: Mathematical Modelling and Numerical Analysis-Modélisation Mathématique et Analyse Numérique 1969, 3, 35-43.
(248) Polyak, B. The conjugate gradient method in extremal problems. USSR Computational Mathematics and Mathematical Physics 1969, 9, 94-112, DOI: 10. 1016/0041-5553(69)90035-4.
(249) Hager, W. W.; Zhang, H. A New Conjugate Gradient Method with Guaranteed Descent and an Efficient Line Search. SIAM Journal on Optimization 2005, 16, 170-192, DOI: $10.1137 / 030601880$.
(250) Koslover, E. F.; Wales, D. J. Geometry optimization for peptides and proteins: Comparison of Cartesian and internal coordinates. The Journal of Chemical Physics 2007, 127, 234105, DOI: 10.1063/1.2807227.
(251) Nocedal, J. Updating quasi-Newton matrices with limited storage. Mathematics of Computation 1980, 35, 773-773, DOI: 10. 1090/s0025-5718-1980-0572855-7.
(252) Baysal, C.; Meirovitch, H.; Navon, I. M. Performance of efficient minimization algorithms as applied to models of peptides and proteins. Journal of Computational Chemistry 1999, 20, 354-364, DOI: 10.1002/(sici) 1096-987x(199902) 20:3<354: : aid-jcc7>3.0.co;2-8.
(253) Das, B.; Meirovitch, H.; Navon, I. M. Performance of hybrid methods for largescale unconstrained optimization as applied to models of proteins. Journal of Computational Chemistry 2003, 24, 1222-1231, DOI: 10.1002 / jcc. 10275 .
(254) Powell, M. J. D. Algorithms for nonlinear constraints that use lagrangian functions. Mathematical Programming 1978, 14, 224-248, DOI: $10.1007 / \mathrm{bf} 01588967$.
(255) Goldfarb, D.; Ren, Y.; Bahamou, A. Practical Quasi-Newton Methods for Training Deep Neural Networks, 2020.
(256) Besalú, E.; Bofill, J. M. On the automatic restricted-step rational-functionoptimization method. Theoretical Chemistry Accounts: Theory Computation, and Modeling (Theoretica Chimica Acta) 1998, 100, 265-274, DOI: $10.1007 /$ s002140050387.
(257) Moré, J. J.; Thuente, D. J. Line Search Algorithms with Guaranteed Sufficient Decrease. ACM Trans. Math. Softw. 1994, 20, 286-307, DOI: $10.1145 / 192115$. 192132.
(258) Császár, P.; Pulay, P. Geometry optimization by direct inversion in the iterative subspace. Journal of Molecular Structure 1984, 114, 31-34, DOI: $10.1016 /$ s0022-2860(84)87198-7.
(259) Farkas, Ö.; Schlegel, H. B. Methods for optimizing large molecules. Part III. An improved algorithm for geometry optimization using direct inversion in the iterative subspace (GDIIS). Phys. Chem. Chem. Phys. 2002, 4, 11-15, DOI: 10.1039/b108658h.
(260) Li, X.; Frisch, M. J. Energy-Represented Direct Inversion in the Iterative Subspace within a Hybrid Geometry Optimization Method. Journal of Chemical Theory and Computation 2006, 2, 835-839, DOI: 10.1021/ct050275a.
(261) Baker, J.; Hehre, W. J. Geometry optimization in cartesian coordinates: The end of the Z-matrix? Journal of Computational Chemistry 1991, 12, 606-610, DOI: $10.1002 / \mathrm{jcc} .540120510$.
(262) Sellers, H.; Klimkowski, V.; Schäfer, L. Normal coordinate ab initio force relaxation. Chemical Physics Letters 1978, 58, 541-544, DOI: 10.1016/0009-2614(78) 80014-1.
(263) Boŭr, P.; Keiderling, T. A. Partial optimization of molecular geometry in normal coordinates and use as a tool for simulation of vibrational spectra. The Journal of Chemical Physics 2002, 117, 4126-4132, DOI: $10.1063 / 1.1498468$.
(264) Pulay, P.; Fogarasi, G. Geometry optimization in redundant internal coordinates. The Journal of Chemical Physics 1992, 96, 2856-2860, DOI: $10.1063 / 1$. 462844 .
(265) Peng, C.; Ayala, P. Y.; Schlegel, H. B.; Frisch, M. J. Using redundant internal coordinates to optimize equilibrium geometries and transition states. Journal of Computational Chemistry 1996, 17, 49-56, DOI: 10.1002 /(sici) 1096 987x(19960115)17:1<49::aid-jcc5>3.0.co;2-0.
(266) Baiardi, A.; Bloino, J.; Barone, V. General formulation of vibronic spectroscopy in internal coordinates. The Journal of Chemical Physics 2016, 144, 084114, DOI: 10.1063/1.4942165,
(267) Fogarasi, G.; Zhou, X.; Taylor, P. W.; Pulay, P. The calculation of ab initio molecular geometries: efficient optimization by natural internal coordinates and empirical correction by offset forces. Journal of the American Chemical Society 1992, 114, 8191-8201, DOI: $10.1021 / \mathrm{ja00047a032}$.
(268) Baker, J. Techniques for geometry optimization: A comparison of cartesian and natural internal coordinates. Journal of Computational Chemistry 1993, 14, 1085-1100, DOI: $10.1002 / \mathrm{jcc} .540140910$.
(269) Von Arnim, M.; Ahlrichs, R. Geometry optimization in generalized natural internal coordinates. The Journal of Chemical Physics 1999, 111, 9183-9190, DOI: $10.1063 / 1.479510$.
(270) Baker, J.; Kessi, A.; Delley, B. The generation and use of delocalized internal coordinates in geometry optimization. The Journal of Chemical Physics 1996, 105, 192-212, DOI: $10.1063 / 1.471864$.
(271) Billeter, S. R.; Turner, A. J.; Thiel, W. Linear scaling geometry optimisation and transition state search in hybrid delocalised internal coordinates. Physical Chemistry Chemical Physics 2000, 2, 2177-2186, DOI: 10.1039/a909486e.
(272) Wang, L.-P.; Song, C. Geometry optimization made simple with translation and rotation coordinates. The Journal of Chemical Physics 2016, 144, 214108, DOI: 10.1063/1.4952956.
(273) Hoy, A.; Mills, I.; Strey, G. Anharmonic force constant calculations. Molecular Physics 1972, 24, 1265-1290, DOI: $10.1080 / 00268977200102361$.
(274) Jackels, C. F.; Gu, Z.; Truhlar, D. G. Reaction-path potential and vibrational frequencies in terms of curvilinear internal coordinates. The Journal of Chemical Physics 1995, 102, 3188-3201, DOI: $10.1063 / 1.468630$.
(275) Chuang, Y.-Y.; Truhlar, D. G. Reaction-path dynamics with harmonic vibration frequencies in curvilinear internal coordinates: $\mathrm{H}+$ trans- $\mathrm{N} 2 \mathrm{H} 2 \rightarrow \mathrm{~N} 2 \mathrm{H}+\mathrm{H} 2$. The Journal of Chemical Physics 1997, 107, 83-89, DOI: 10.1063/1.474377.
(276) Lee, S.-H.; Palmo, K.; Krimm, S. New out-of-plane angle and bond angle internal coordinates and related potential energy functions for molecular mechanics and dynamics simulations. Journal of Computational Chemistry 1999, 20, 10671084, DOI: $10.1002 /($ sici) 1096-987x(19990730)20:10<1067: : aid-jcc9>3.0.co;2-v.
(277) Németh, K.; Challacombe, M.; Veenendaal, M. V. The choice of internal coordinates in complex chemical systems. Journal of Computational Chemistry 2010, NA-NA, DOI: $10.1002 /$ jcc. 21494 .
(278) Wilson, E. B.; Decius, J. C.; Cross, P. C., Molecular vibrations: the theory of infrared and Raman vibrational spectra; Courier Corporation: 1980.
(279) Baker, J.; Bergeron, D. Constrained optimization in cartesian coordinates. J. Comput. Chem. 1993, 14, 1339-1346, DOI: 10.1002 /jcc. 540141111 .
(280) Baker, J. Constrained optimization in delocalized internal coordinates. J. Comput. Chem. 1997, 18, 1079-1095, DOI: $10.1002 /$ (sici) 1096-987x(199706)18: 8<1079::aid-jcc12>3.0.co;2-8.

## Bibliography

(281) Kruse, H.; Sponer, J. Towards biochemically relevant QM computations on nucleic acids: controlled electronic structure geometry optimization of nucleic acid structural motifs using penalty restraint functions. Phys. Chem. Chem. Phys. 2015, 17, 1399-1410, DOI: $10.1039 / \mathrm{c} 4 \mathrm{cp} 04680 \mathrm{c}$.
(282) Palenik, M. C. Initial estimate for minimum energy pathways and transition states using velocities in internal coordinates. Chemical Physics 2021, 542, 111046, DOI: 10.1016/j.chemphys.2020.111046.
(283) Rybkin, V. V.; Ekström, U.; Helgaker, T. Internal-to-Cartesian back transformation of molecular geometry steps using high-order geometric derivatives. Journal of Computational Chemistry 2013, 34, 1842-1849, DOI: $10.1002 /$ jcc. 23327 .
(284) Golub, G. H., Matrix Computations (Johns Hopkins Studies in the Mathematical Sciences); Johns Hopkins University Press: 2013.
(285) Goodfellow, I.; Bengio, Y.; Courville, A., Deep Learning,http://www. deeplearningbook. org; MIT Press: 2016.
(286) Mones, L.; Ortner, C.; Csányi, G. Preconditioners for the geometry optimisation and saddle point search of molecular systems. Scientific Reports 2018, 8, DOI: 10.1038/s41598-018-32105-x.
(287) Packwood, D.; Kermode, J.; Mones, L.; Bernstein, N.; Woolley, J.; Gould, N.; Ortner, C.; Csányi, G. A universal preconditioner for simulating condensed phase materials. The Journal of Chemical Physics 2016, 144, 164109, DOI: 10.1063/1.4947024
(288) Jiang, L.; Byrd, R. H.; Eskow, E.; Schnabel, R. B. A preconditioned L-BFGS algorithm with application to molecular energy minimization; tech. rep.; COLORADO UNIV AT BOULDER DEPT OF COMPUTER SCIENCE, 2004.
(289) Rappe, A. K.; Casewit, C. J.; Colwell, K. S.; Goddard, W. A.; Skiff, W. M. UFF, a full periodic table force field for molecular mechanics and molecular dynamics simulations. Journal of the American Chemical Society 1992, 114, 10024-10035, DOI: 10.1021/ja00051a040,
(290) Heyden, A.; Bell, A. T.; Keil, F. J. Efficient methods for finding transition states in chemical reactions: Comparison of improved dimer method and partitioned rational function optimization method. The Journal of Chemical Physics 2005, 123, 224101, DOI: $10.1063 / 1.2104507$.
(291) Neese, F.; Wennmohs, F.; Hansen, A.; Becker, U. Efficient approximate and parallel Hartree-Fock and hybrid DFT calculations. A 'chain-of-spheres' algorithm for the Hartree-Fock exchange. Chemical Physics 2009, 356, 98-109, DOI: $10.1016 / j$. chemphys.2008.10.036.
(292) Bykov, D.; Petrenko, T.; Izsák, R.; Kossmann, S.; Becker, U.; Valeev, E.; Neese, F. Efficient implementation of the analytic second derivatives of Hartree-Fock and hybrid DFT energies: a detailed analysis of different approximations. Molecular Physics 2015, 113, 1961-1977, DOI: 10.1080/00268976.2015.1025114.
(293) Ayala, P. Y.; Schlegel, H. B. A combined method for determining reaction paths minima, and transition state geometries. The Journal of Chemical Physics 1997, 107, 375-384, DOI: $10.1063 / 1.474398$.
(294) Henkelman, G.; Jónsson, H. Improved tangent estimate in the nudged elastic band method for finding minimum energy paths and saddle points. The Journal of Chemical Physics 2000, 113, 9978-9985, DOI: $10.1063 / 1.1323224$.
(295) Peters, B.; Heyden, A.; Bell, A. T.; Chakraborty, A. A growing string method for determining transition states: Comparison to the nudged elastic band and string methods. The Journal of Chemical Physics 2004, 120, 7877-7886, DOI: 10.1063/1.1691018.
(296) Smidstrup, S.; Pedersen, A.; Stokbro, K.; Jónsson, H. Improved initial guess for minimum energy path calculations. The Journal of Chemical Physics 2014, 140, 214106, DOI: $10.1063 / 1.4878664$.
(297) Halgren, T. A.; Lipscomb, W. N. The synchronous-transit method for determining reaction pathways and locating molecular transition states. Chemical Physics Letters 1977, 49, 225-232, DOI: 10.1016/0009-2614(77)80574-5.
(298) Govind, N.; Petersen, M.; Fitzgerald, G.; King-Smith, D.; Andzelm, J. A generalized synchronous transit method for transition state location. Computational Materials Science 2003, 28, 250-258, DOI:10.1016/s0927-0256(03)001113.
(299) Behn, A.; Zimmerman, P. M.; Bell, A. T.; Head-Gordon, M. Incorporating Linear Synchronous Transit Interpolation into the Growing String Method: Algorithm and Applications. Journal of Chemical Theory and Computation 2011, 7, 40194025, DOI: $10.1021 /$ ct200654u.
(300) Zimmerman, P. M. Growing string method with interpolation and optimization in internal coordinates: Method and examples. The Journal of Chemical Physics 2013, 138, 184102, DOI: $10.1063 / 1.4804162$.
(301) Zhu, X.; Thompson, K. C.; Martínez, T. J. Geodesic interpolation for reaction pathways. The Journal of Chemical Physics 2019, 150, 164103, DOI: $10.1063 /$ 1.5090303 .
(302) E, W.; Ren, W.; Vanden-Eijnden, E. Simplified and improved string method for computing the minimum energy paths in barrier-crossing events. The Journal of Chemical Physics 2007, 126, 164103, DOI: $10.1063 / 1.2720838$.
(303) Maragakis, P.; Andreev, S. A.; Brumer, Y.; Reichman, D. R.; Kaxiras, E. Adaptive nudged elastic band approach for transition state calculation. The Journal of Chemical Physics 2002, 117, 4651-4658, DOI: $10.1063 / 1.1495401$.
(304) Zhu, T.; Li, J.; Samanta, A.; Kim, H. G.; Suresh, S. Interfacial plasticity governs strain rate sensitivity and ductility in nanostructured metals. Proceedings of the National Academy of Sciences 2007, 104, 3031-3036, DOI: 10.1073/pnas. 0611097104 .
(305) Zhang, J.; Zhang, H.; Ye, H.; Zheng, Y. Free-end adaptive nudged elastic band method for locating transition states in minimum energy path calculation. The Journal of Chemical Physics 2016, 145, 094104, DOI: 10.1063/1.4962019.
(306) Sheppard, D.; Terrell, R.; Henkelman, G. Optimization methods for finding minimum energy paths. The Journal of Chemical Physics 2008, 128, 134106, DOI: $10.1063 / 1.2841941$.
(307) Herbol, H. C.; Stevenson, J.; Clancy, P. Computational Implementation of Nudged Elastic Band Rigid Rotation, and Corresponding Force Optimization. Journal of Chemical Theory and Computation 2017, 13, 3250-3259, DOI: 10.1021/acs. jctc.7b00360.
(308) Melander, M.; Laasonen, K.; Jónsson, H. Removing External Degrees of Freedom from Transition-State Search Methods using Quaternions. Journal of Chemical Theory and Computation 2015, 11, 1055-1062, DOI: $10.1021 / \mathrm{ct} 501155 \mathrm{k}$.
(309) Henkelman, G.; Uberuaga, B. P.; Jónsson, H. A climbing image nudged elastic band method for finding saddle points and minimum energy paths. The Journal of Chemical Physics 2000, 113, 9901-9904, DOI: $10.1063 / 1.1329672$.
(310) Zarkevich, N. A.; Johnson, D. D. Nudged-elastic band method with two climbing images: Finding transition states in complex energy landscapes. The Journal of Chemical Physics 2015, 142, 024106, DOI: $10.1063 / 1.4905209$.
(311) Henkelman, G.; Jónsson, H. A dimer method for finding saddle points on high dimensional potential surfaces using only first derivatives. The Journal of Chemical Physics 1999, 111, 7010-7022, DOI: $10.1063 / 1.480097$.
(312) Kästner, J.; Sherwood, P. Superlinearly converging dimer method for transition state search. The Journal of Chemical Physics 2008, 128, 014106, DOI: 10. 1063/1.2815812,
(313) Shang, C.; Liu, Z.-P. Constrained Broyden Minimization Combined with the Dimer Method for Locating Transition State of Complex Reactions. Journal of Chemical Theory and Computation 2010, 6, 1136-1144, DOI: $10.1021 /$ ct 9005147 .
(314) Schaefer, B.; Mohr, S.; Amsler, M.; Goedecker, S. Minima hopping guided path search: An efficient method for finding complex chemical reaction pathways. The Journal of Chemical Physics 2014, 140, 214102, DOI: 10.1063/1.4878944.
(315) Baker, J.; Chan, F. The location of transition states: A comparison of Cartesian, Z-matrix, and natural internal coordinates. Journal of Computational Chemistry 1996, 17, 888-904, DOI: $10.1002 /($ sici) $1096-987 x(199605) 17: 7<888:$ : aid-jcc12>3.0.co;2-7.
(316) Fukui, K.; Kato, S.; Fujimoto, H. Constituent analysis of the potential gradient along a reaction coordinate. Method and an application to methane + tritium reaction. Journal of the American Chemical Society 1975, 97, 1-7, DOI: 10 . 1021/ja00834a001.
(317) Ishida, K.; Morokuma, K.; Komornicki, A. The intrinsic reaction coordinate. An ab initio calculation for $\mathrm{HNC} \rightarrow \mathrm{HCN}$ and $\mathrm{H}-+\mathrm{CH} 4 \rightarrow \mathrm{CH} 4+\mathrm{H}-$. The Journal of Chemical Physics 1977, 66, 2153-2156, DOI: $10.1063 / 1.434152$.
(318) Maeda, S.; Harabuchi, Y.; Ono, Y.; Taketsugu, T.; Morokuma, K. Intrinsic reaction coordinate: Calculation bifurcation, and automated search. International Journal of Quantum Chemistry 2014, 115, 258-269, DOI: 10.1002/qua. 24757.
(319) Unke, O. T.; Brickel, S.; Meuwly, M. Sampling reactive regions in phase space by following the minimum dynamic path. The Journal of Chemical Physics 2019, 150, 074107, DOI: $10.1063 / 1.5082885$.
(320) Baker, J.; Gill, P. M. W. An algorithm for the location of branching points on reaction paths. Journal of Computational Chemistry 1988, 9, 465-475, DOI: $10.1002 / \mathrm{jcc} .540090505$.
(321) Quapp, W.; Hirsch, M.; Heidrich, D. Bifurcation of reaction pathways: the set of valley ridge inflection points of a simple three-dimensional potential energy surface. Theoretical Chemistry Accounts: Theory, Computation, and Modeling (Theoretica Chimica Acta) 1998, 100, 285-299, DOI: $10.1007 / \mathrm{s} 002140050389$.
(322) Quapp, W.; Hirsch, M.; Heidrich, D. An approach to reaction path branching using valley?ridge inflection points of potential-energy surfaces. Theoretical Chemistry Accounts: Theory, Computation, and Modeling (Theoretica Chimica Acta) 2004, 112, 40-51, DOI: $10.1007 / \mathrm{s} 00214-003-0558-8$.
(323) Curtiss, C. F.; Hirschfelder, J. O. Integration of Stiff Equations. Proceedings of the National Academy of Sciences 1952, 38, 235-243, DOI: 10.1073/pnas. 38 . 3.235 .
(324) Shampine, L. F. In Numerische Behandlung von Differentialgleichungen; Birkhäuser Basel: 1975, pp 287-301, DOI: 10.1007/978-3-0348-5532-7_18.
(325) Hratchian, H. P.; Schlegel, H. B. Accurate reaction paths using a Hessian based predictor-corrector integrator. The Journal of Chemical Physics 2004, 120, 99189924, DOI: $10.1063 / 1.1724823$.
(326) Aguilar-Mogas, A.; Giménez, X.; Bofill, J. M. On the implementation of the Runge-Kutta-Fehlberg algorithm to integrate intrinsic reaction coordinate paths. Chemical Physics Letters 2006, 432, 375-382, DOI: $10.1016 / \mathrm{j}$. cplett. 2006. 10.061 .
(327) Aguilar-Mogas, A.; Giménez, X.; Bofill, J. M. Implementation of an algorithm based on the Runge-Kutta-Fehlberg technique and the potential energy as a reaction coordinate to locate intrinsic reaction paths. Journal of Computational Chemistry 2010, n/a-n/a, DOI: $10.1002 /$ jcc. 21539 .
(328) Melissas, V. S.; Truhlar, D. G.; Garrett, B. C. Optimized calculations of reaction paths and reaction-path functions for chemical reactions. The Journal of Chemical Physics 1992, 96, 5758-5772, DOI: $10.1063 / 1.462674$.
(329) Deng, L.; Ziegler, T. The determination of intrinsic reaction coordinates by density functional theory. International Journal of Quantum Chemistry 1994, 52, 731-765, DOI: 10.1002/qua. 560520406 .
(330) Gonzalez, C.; Schlegel, H. B. Improved algorithms for reaction path following: Higher-order implicit algorithms. The Journal of Chemical Physics 1991, 95, 5853-5860, DOI: $10.1063 / 1.461606$.
(331) Gonzalez, C.; Schlegel, H. B. An improved algorithm for reaction path following. The Journal of Chemical Physics 1989, 90, 2154-2161, DOI: 10.1063/1. 456010.
(332) Hratchian, H. P.; Frisch, M. J.; Schlegel, H. B. Steepest descent reaction path integration using a first-order predictor-corrector method. The Journal of Chemical Physics 2010, 133, 224101, DOI: $10.1063 / 1.3514202$.
(333) Hratchian, H. P.; Frisch, M. J. Integrating steepest-descent reaction pathways for large molecules. The Journal of Chemical Physics 2011, 134, 204103, DOI: 10.1063/1.3593456.
(334) Pechukas, P. On simple saddle points of a potential surface, the conservation of nuclear symmetry along paths of steepest descent, and the symmetry of transition states. The Journal of Chemical Physics 1976, 64, 1516-1521, DOI: 10.1063/1.432370.
(335) Page, M.; McIver, J. W. On evaluating the reaction path Hamiltonian. The Journal of Chemical Physics 1988, 88, 922-935, DOI: 10.1063/1.454172.
(336) Hratchian, H. P.; Schlegel, H. B. Using Hessian Updating To Increase the Efficiency of a Hessian Based Predictor-Corrector Reaction Path Following Method. Journal of Chemical Theory and Computation 2004, 1, 61-69, DOI: 10.1021/ ct0499783.
(337) Bulirsch, R.; Stoer, J. Fehlerabschätzungen und Extrapolation mit rationalen Funktionen bei Verfahren vom Richardson-Typus. Numerische Mathematik 1964, 6, 413-427, DOI: $10.1007 / \mathrm{bf01386092} \mathrm{}$.
(338) Bulirsch, R.; Stoer, J. Numerical treatment of ordinary differential equations by extrapolation methods. Numerische Mathematik 1966, 8, 1-13, DOI: $10.1007 /$ be02165234.
(339) Bulirsch, R.; Stoer, J. Asymptotic upper and lower bounds for results of extrapolation methods. Numerische Mathematik 1966, 8, 93-104, DOI: $10.1007 /$ bf02163179.
(340) Farwig, R. Rate of convergence of Shepard's global interpolation formula. Mathematics of Computation 1986, 46, 577-577, DOI: 10.1090/s0025-5718-1986-0829627-0.
(341) Farwig, R. In Algorithms for Approximation; Clarendon Press: USA, 1987, pp 193211.
(342) Ischtwan, J.; Collins, M. A. Molecular potential energy surfaces by interpolation. The Journal of Chemical Physics 1994, 100, 8080-8088, DOI: $10.1063 / 1$. 466801 .
(343) Thompson, K. C.; Jordan, M. J. T.; Collins, M. A. Molecular potential energy surfaces by interpolation in Cartesian coordinates. The Journal of Chemical Physics 1998, 108, 564-578, DOI: $10.1063 / 1.475419$.
(344) Bettens, R. P. A.; Collins, M. A. Learning to interpolate molecular potential energy surfaces with confidence: A Bayesian approach. The Journal of Chemical Physics 1999, 111, 816-826, DOI: $10.1063 / 1.479368$.
(345) Hratchian, H. P.; Kraka, E. Improved Predictor-Corrector Integrators For Evaluating Reaction Path Curvature. Journal of Chemical Theory and Computation 2013, 9, 1481-1488, DOI: $10.1021 / \mathrm{ct} 301021 \mathrm{y}$.
(346) Meisner, J.; Markmeyer, M. N.; Bohner, M. U.; Kästner, J. Comparison of classical reaction paths and tunneling paths studied with the semiclassical instanton theory. Physical Chemistry Chemical Physics 2017, 19, 23085-23094, DOI: $10.1039 /$ c7cp03722h.
(347) Meisner, J. Theoretical investigations of atom tunneling in the interstellar medium, en, 2018, DOI: $10.18419 /$ OPUS-9841.
(348) IX. The approximate arithmetical solution by finite differences of physical problems involving differential equations, with an application to the stresses in a masonry dam. Philosophical Transactions of the Royal Society of London. Series A, Containing Papers of a Mathematical or Physical Character 1911, 210, 307357, DOI: $10.1098 /$ rsta. 1911.0009 .
(349) VIII. The deferred approach to the limit. Philosophical Transactions of the Royal Society of London. Series A, Containing Papers of a Mathematical or Physical Character 1927, 226, 299-361, DOI: $10.1098 /$ rsta.1927.0008,
(350) Celik, I.; Li, J.; Hu, G.; Shaffer, C. Limitations of Richardson Extrapolation and Some Possible Remedies. Journal of Fluids Engineering 2005, 127, 795-805, DOI: 10.1115/1.1949646.
(351) Monroe, J. L. Extrapolation and the Bulirsch-Stoer algorithm. Physical Review E 2002, 65, DOI: $10.1103 /$ physreve. 65.066116.
(352) Hairer, E.; Wanner, G., Solving Ordinary Differential Equations II; Springer Berlin Heidelberg: 1996, DOI: 10.1007/978-3-642-05221-7.
(353) Hajduk, P. J.; Bures, M.; Praestgaard, J.; Fesik, S. W. Privileged Molecules for Protein Binding Identified from NMR-Based Screening. Journal of Medicinal Chemistry 2000, 43, 3443-3447, DOI: $10.1021 /$ jm000164q.
(354) Michel, M. C.; Foster, C.; Brunner, H. R.; Liu, L. A Systematic Comparison of the Properties of Clinically Used Angiotensin II Type 1 Receptor Antagonists. Pharmacological Reviews 2013, 65, ed. by Perez, D. M., 809-848, DOI: $10.1124 /$ pr. 112.007278 .
(355) Kylmälä, T.; Tois, J.; Xu, Y.; Franzén, R. One step synthesis of Diflunisal using a Pd-diamine complex. Open Chemistry 2009, 7, DOI: 10.2478/s11532-009-0068-1.
(356) Kuuloja, N.; Kylmälä, T.; Xu, Y.; Franzén, R. Synthesis of Xenbucin using Suzuki reaction catalyzed by Pd/C in water. Open Chemistry 2008, 6, DOI: 10.2478/s11532-008-0044-1.
(357) Bringmann, G.; Menche, D. Stereoselective Total Synthesis of Axially Chiral Natural Products via Biaryl Lactones $\dagger$. Accounts of Chemical Research 2001, 34 , 615-624, DOI: 10.1021/ar000106z.
(358) Turner, S. E.; Williams, C. M.; Iversen, L.; Whalley, B. J. In Progress in the Chemistry of Organic Natural Products; Springer International Publishing: 2017, pp 61-101, DOI: 10.1007/978-3-319-45541-9_3.
(359) Fayez, S.; Bruhn, T.; Feineis, D.; Assi, L. A.; Awale, S.; Bringmann, G. Ancistrosecolines A-F Unprecedented seco-Naphthylisoquinoline Alkaloids from the Roots of Ancistrocladus abbreviatus, with Apoptosis-Inducing Potential against HeLa Cancer Cells. Journal of Natural Products 2020, DOI: 10.1021/acs. jnatprod.9b01168.
(360) Xiao, B.-B.; Xia, G.-Y.; Wang, L.-Y.; Qiu, B.-L.; Xia, H.; Zhong, W.-C.; Tian, G.-H.; Lin, S. ( $\pm$ )-Bicoryanhunine A dimeric benzylisoquinoline alkaloid atropoenantiomers from Corydalis yanhusuo. Tetrahedron Letters 2020, 151890, DOI: 10.1016/j.tetlet.2020.151890.
(361) Fayez, S.; Li, J.; Feineis, D.; Assi, L. A.; Kaiser, M.; Brun, R.; Anany, M. A.; Wajant, H.; Bringmann, G. A Near-Complete Series of Four Atropisomeric Jozimine A2-Type Naphthylisoquinoline Dimers with Antiplasmodial and Cytotoxic Activ-
ities and Related Alkaloids from Ancistrocladus abbreviatus. Journal of Natural Products 2019, 82, 3033-3046, DOI: 10.1021/acs.jnatprod.9b00589.
(362) Hirano, K.; Miura, M. Recent Advances in Copper-mediated Direct Biaryl Coupling. Chemistry Letters 2015, 44, 868-873, DOI: $10.1246 / \mathrm{cl} .150354$.
(363) Negishi, E.; King, A. O.; Okukado, N. Selective carbon-carbon bond formation via transition metal catalysis. 3. A highly selective synthesis of unsymmetrical biaryls and diarylmethanes by the nickel- or palladium-catalyzed reaction of aryland benzylzinc derivatives with aryl halides. The Journal of Organic Chemistry 1977, 42, 1821-1823, DOI: $10.1021 /$ jo00430a041.
(364) Guo, L.; Srimontree, W.; Zhu, C.; Maity, B.; Liu, X.; Cavallo, L.; Rueping, M. Nickel-catalyzed Suzuki-Miyaura cross-couplings of aldehydes. Nature Communications 2019, 10, DOI: $10.1038 / \mathrm{s} 41467-019-09766-\mathrm{x}$.
(365) Liao, G.; Zhou, T.; Yao, Q.-J.; Shi, B.-F. Recent advances in the synthesis of axially chiral biarylsviatransition metal-catalysed asymmetric C-H functionalization. Chemical Communications 2019, 55, 8514-8523, DOI: $10.1039 / \mathrm{c} 9 \mathrm{cc} 03967 \mathrm{~h}$.
(366) Shi, S.; Meng, G.; Szostak, M. Synthesis of Biaryls through Nickel-Catalyzed Suzuki-Miyaura Coupling of Amides by Carbon-Nitrogen Bond Cleavage. Angewandte Chemie International Edition 2016, 55, 6959-6963, DOI: $10.1002 /$ anie. 201601914.
(367) Corrie, T. J. A.; Ball, L. T.; Russell, C. A.; Lloyd-Jones, G. C. Au-Catalyzed Biaryl Coupling To Generate 5- to 9 -Membered Rings: Turnover-Limiting Reductive Elimination versus $\pi$-Complexation. Journal of the American Chemical Society 2016, 139, 245-254, DOI: $10.1021 /$ jacs.6b10018.
(368) Simonetti, M.; Cannas, D. M.; Larrosa, I. In Advances in Organometallic Chemistry; Elsevier: 2017, pp 299-399, DOI: $10.1016 / \mathrm{bs}$. adomc.2017.03.002.
(369) Konze, W. V.; Scott, B. L.; Kubas, G. J. C-H Activation and C-C Coupling of Arenes by Cationic Pt(II) Complexes. Journal of the American Chemical Society 2002, 124, 12550-12556, DOI: 10.1021 / ja020798h.
(370) Felpin, F.-X.; Sengupta, S. Biaryl synthesis with arenediazonium salts: crosscoupling CH-arylation and annulation reactions. Chemical Society Reviews 2019, 48, 1150-1193, DOI: $10.1039 / \mathrm{c} 8 \mathrm{cs} 00453 \mathrm{f}$.
(371) Elsler, B.; Schollmeyer, D.; Dyballa, K. M.; Franke, R.; Waldvogel, S. R. Metaland Reagent-Free Highly Selective Anodic Cross-Coupling Reaction of Phenols. Angewandte Chemie International Edition 2014, n/a-n/a, DOI: 10.1002 /anie. 201400627.
(372) Kita, Y.; Morimoto, K.; Ito, M.; Ogawa, C.; Goto, A.; Dohi, T. Metal-Free Oxidative Cross-Coupling of Unfunctionalized Aromatic Compounds. Journal of the American Chemical Society 2009, 131, 1668-1669, DOI: 10 . 1021/ ja808940n.
(373) Chan, T. L.; Wu, Y.; Choy, P. Y.; Kwong, F. Y. A Radical Process towards the Development of Transition-Metal-Free Aromatic Carbon-Carbon Bond-Forming Reactions. Chemistry - A European Journal 2013, 19, 15802-15814, DOI: 10 . 1002/chem. 201301583.
(374) Yanagi, T.; Otsuka, S.; Kasuga, Y.; Fujimoto, K.; Murakami, K.; Nogi, K.; Yorimitsu, H.; Osuka, A. Metal-Free Approach to Biaryls from Phenols and Aryl Sulfoxides by Temporarily Sulfur-Tethered Regioselective C-H/C-H Coupling. Journal of the American Chemical Society 2016, 138, 14582-14585, DOI: 10 . 1021/jacs.6b10278.
(375) Dohi, T.; Ito, M.; Itani, I.; Yamaoka, N.; Morimoto, K.; Fujioka, H.; Kita, Y. Metal-Free C-H Cross-Coupling toward Oxygenated Naphthalene-Benzene Linked Biaryls. Organic Letters 2011, 13, 6208-6211, DOI: $10.1021 / 01202632$ h.
(376) Morimoto, K.; Yamaoka, N.; Ogawa, C.; Nakae, T.; Fujioka, H.; Dohi, T.; Kita, Y. Metal-Free Regioselective Oxidative Biaryl Coupling Leading to Head-toTail Bithiophenes: Reactivity Switching a Concept Based on the Iodonium(III) Intermediate. Organic Letters 2010, 12, 3804-3807, DOI: $10.1021 /$ ol101498r.
(377) Morimoto, K.; Sakamoto, K.; Ohnishi, Y.; Miyamoto, T.; Ito, M.; Dohi, T.; Kita, Y. Metal-Free OxidativeparaCross-Coupling of Phenols. Chemistry - A European Journal 2013, 19, 8726-8731, DOI: $10.1002 /$ chem. 201301028.
(378) Leroux, F. R.; Berthelot, A.; Bonnafoux, L.; Panossian, A.; Colobert, F. Transition-Metal-Free Atropo-Selective Synthesis of Biaryl Compounds Based on Arynes. Chemistry - A European Journal 2012, 18, 14232-14236, DOI: $10.1002 /$ chem. 201202739.
(379) Morimoto, K.; Dohi, T.; Kita, Y. Metal-free Oxidative Cross-Coupling Reaction of Aromatic Compounds Containing Heteroatoms. Synlett 2017, 28, 1680-1694, DOI: $10.1055 / \mathrm{s}-0036-1588455$.
(380) Ali, M. A. Computational studies on the gas phase reaction of methylenimine (CH2NH) with water molecules. Scientific Reports 2020, 10, DOI: $10.1038 /$ s41598-020-67515-3.
(381) Weigend, F.; Ahlrichs, R. Balanced basis sets of split valence triple zeta valence and quadruple zeta valence quality for H to Rn: Design and assessment of accuracy. Physical Chemistry Chemical Physics 2005, 7, 3297, DOI: 10.1039/b508541a.
(382) Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. A consistent and accurate ab initio parametrization of density functional dispersion correction (DFT-D) for the 94 elements H-Pu. The Journal of Chemical Physics 2010, 132, 154104, DOI: 10.1063/1.3382344.
(383) Grimme, S.; Ehrlich, S.; Goerigk, L. Effect of the damping function in dispersion corrected density functional theory. Journal of Computational Chemistry 2011, 32, 1456-1465, DOI: 10.1002 / jcc. 21759 .
(384) Frisch, M. J. et al. Gaussian 16 Revision B.01, Gaussian Inc. Wallingford CT, 2016.
(385) Paulechka, E.; Kazakov, A. Efficient DLPNO-CCSD(T)-Based Estimation of Formation Enthalpies for C-, H-, O-, and N-Containing Closed-Shell Compounds Validated Against Critically Evaluated Experimental Data. The Journal of Physical Chemistry A 2017, 121, 4379-4387, DOI: 10.1021/acs.jpca.7.603195,
(386) Riplinger, C.; Neese, F. An efficient and near linear scaling pair natural orbital based local coupled cluster method. The Journal of Chemical Physics 2013, 138, 034106, DOI: $10.1063 / 1.4773581$.
(387) Riplinger, C.; Sandhoefer, B.; Hansen, A.; Neese, F. Natural triple excitations in local coupled cluster calculations with pair natural orbitals. The Journal of Chemical Physics 2013, 139, 134101, DOI: $10.1063 / 1.4821834$.
(388) Neese, F. Software update: the ORCA program system, version 4.0. WIREs Computational Molecular Science 2017, 8, DOI: $10.1002 / \mathrm{wcms} .1327$.
(389) Weigend, F. Accurate Coulomb-fitting basis sets for H to Rn. Physical Chemistry Chemical Physics 2006, 8, 1057, DOI: $10.1039 / \mathrm{b} 515623 \mathrm{~h}$.
(390) Hellweg, A.; Hättig, C.; Höfener, S.; Klopper, W. Optimized accurate auxiliary basis sets for RI-MP2 and RI-CC2 calculations for the atoms Rb to Rn. Theoretical Chemistry Accounts 2007, 117, 587-597, DOI: 10.1007/s00214-007-02505.
(391) Barone, V.; Cossi, M. Quantum Calculation of Molecular Energies and Energy Gradients in Solution by a Conductor Solvent Model. The Journal of Physical Chemistry A 1998, 102, 1995-2001, DOI: $10.1021 / \mathrm{jp9716997}$.
(392) Cossi, M.; Rega, N.; Scalmani, G.; Barone, V. Energies, structures, and electronic properties of molecules in solution with the C-PCM solvation model. Journal of Computational Chemistry 2003, 24, 669-681, DOI: $10.1002 /$ jcc. 10189 .
(393) Liakos, D. G.; Sparta, M.; Kesharwani, M. K.; Martin, J. M. L.; Neese, F. Exploring the Accuracy Limits of Local Pair Natural Orbital Coupled-Cluster Theory. Journal of Chemical Theory and Computation 2015, 11, 1525-1539, DOI: $10.1021 / \mathrm{ct} 501129 \mathrm{~s}$.
(394) Pedregosa, F.; Varoquaux, G.; Gramfort, A.; Michel, V.; Thirion, B.; Grisel, O.; Blondel, M.; Prettenhofer, P.; Weiss, R.; Dubourg, V.; Vanderplas, J.; Passos, A.; Cournapeau, D.; Brucher, M.; Perrot, M.; Duchesnay, E. Scikit-learn: Machine Learning in Python. Journal of Machine Learning Research 2011, 12, 2825-2830.
(395) Buitinck, L.; Louppe, G.; Blondel, M.; Pedregosa, F.; Mueller, A.; Grisel, O.; Niculae, V.; Prettenhofer, P.; Gramfort, A.; Grobler, J.; Layton, R.; VanderPlas, J.; Joly, A.; Holt, B.; Varoquaux, G. In ECML PKDD Workshop: Languages for Data Mining and Machine Learning, 2013, pp 108-122.
(396) Mu, Y.; Nguyen, P. H.; Stock, G. Energy landscape of a small peptide revealed by dihedral angle principal component analysis. Proteins: Structure, Function, and Bioinformatics 2004, 58, 45-52, DOI: $10.1002 /$ prot. 20310.
(397) Peach, M. J. G.; Benfield, P.; Helgaker, T.; Tozer, D. J. Excitation energies in density functional theory: An evaluation and a diagnostic test. The Journal of Chemical Physics 2008, 128, 044118, DOI: $10.1063 / 1.2831900$.
(398) Lu, T.; Chen, F. Multiwfn: A multifunctional wavefunction analyzer. Journal of Computational Chemistry 2011, 33, 580-592, DOI: $10.1002 /$ jcc. 22885 .
(399) Hawkins, P. C. D.; Skillman, A. G.; Warren, G. L.; Ellingson, B. A.; Stahl, M. T. Conformer Generation with OMEGA: Algorithm and Validation Using High Quality Structures from the Protein Databank and Cambridge Structural Database. Journal of Chemical Information and Modeling 2010, 50, 572-584, DOI: $10.1021 /$ ci100031x.
(400) O’Boyle, N. M.; Banck, M.; James, C. A.; Morley, C.; Vandermeersch, T.; Hutchison, G. R. Open Babel: An open chemical toolbox. Journal of Cheminformatics 2011, 3, DOI: $10.1186 / 1758-2946-3-33$.
(401) Hawkins, P. C. D.; Nicholls, A. Conformer Generation with OMEGA: Learning from the Data Set and the Analysis of Failures. Journal of Chemical Information and Modeling 2012, 52, 2919-2936, DOI: $10.1021 / \mathrm{ci} 300314 \mathrm{k}$.
(402) Kanal, I. Y.; Keith, J. A.; Hutchison, G. R. A sobering assessment of smallmolecule force field methods for low energy conformer predictions. International Journal of Quantum Chemistry 2017, 118, e25512, DOI: 10.1002/qua. 25512.
(403) Chan, L.; Hutchison, G. R.; Morris, G. M. Bayesian optimization for conformer generation. Journal of Cheminformatics 2019, 11, DOI: $10.1186 /$ s13321 019-0354-7.
(404) Grimme, S. Exploration of Chemical Compound, Conformer, and Reaction Space with Meta-Dynamics Simulations Based on Tight-Binding Quantum Chemical Calculations. Journal of Chemical Theory and Computation 2019, 15, 2847-2862, DOI: $10.1021 /$ acs.jctc.9b00143.
(405) De Vijver, R. V.; Zádor, J. KinBot: Automated stationary point search on potential energy surfaces. Computer Physics Communications 2020, 248, 106947, DOI: $10.1016 / \mathrm{j} . \mathrm{cpc} .2019 .106947$.
(406) Chandramouli, B.; Galdo, S. D.; Fusè, M.; Barone, V.; Mancini, G. Two-level stochastic search of low-energy conformers for molecular spectroscopy: implementation and validation of MM and QM models. Physical Chemistry Chemical Physics 2019, 21, 19921-19934, DOI: $10.1039 / \mathrm{c} 9 \mathrm{cp} 03557 \mathrm{e}$.
(407) Pracht, P.; Bohle, F.; Grimme, S. Automated exploration of the low-energy chemical space with fast quantum chemical methods. Physical Chemistry Chemical Physics 2020, 22, 7169-7192, DOI: $10.1039 / \mathrm{c} 9 \mathrm{cp} 06869 \mathrm{~d}$.
(408) Hanson, K.; Roskop, L.; Djurovich, P. I.; Zahariev, F.; Gordon, M. S.; Thompson, M. E. A Paradigm for Blue- or Red-Shifted Absorption of Small Molecules Depending on the Site of $\pi$-Extension. Journal of the American Chemical Society 2010, 132, 16247-16255, DOI: $10.1021 / \mathrm{ja1075162}$.
(409) Chaitanya, K.; Ju, X.-H.; Heron, B. M. Can elongation of the $\pi$-system in triarylamine derived sensitizers with either benzothiadiazole and/or ortho-fluorophenyl moieties enrich their light harvesting efficiency? - a theoretical study. RSC Advances 2015, 5, 3978-3998, DOI: $10.1039 / \mathrm{c} 4 \mathrm{ra09914a}$.
(410) Tsuneda, T.; Singh, R. K.; Nakata, A. Relationship between orbital energy gaps and excitation energies for long-chain systems. Journal of Computational Chemistry 2016, 37, 1451-1462, DOI: $10.1002 /$ jcc. 24357 .
(411) Giereth, R.; Reim, I.; Frey, W.; Junge, H.; Tschierlei, S.; Karnahl, M. Remarkably long-lived excited states of copper photosensitizers containing an extended $\pi$ system based on an anthracene moiety. Sustainable Energy \& Fuels 2019, 3, 692-700, DOI: $10.1039 / \mathrm{c} 8 \mathrm{se} 00521 \mathrm{~d}$.
(412) Kimber, P.; Plasser, F. Toward an understanding of electronic excitation energies beyond the molecular orbital picture. Physical Chemistry Chemical Physics 2020, 22, 6058-6080, DOI: $10.1039 / \mathrm{d0cp} 00369 \mathrm{~g}$.
(413) Van Rossum, G.; Drake Jr, F. L., Python tutorial; Centrum voor Wiskunde en Informatica Amsterdam: 1995; Vol. 620.
(414) Van Rossum, G. et al. In USENIX annual technical conference, 2007; Vol. 41, p 36 .
(415) Kästner, J.; Carr, J. M.; Keal, T. W.; Thiel, W.; Wander, A.; Sherwood, P. DL-FIND: An Open-Source Geometry Optimizer for Atomistic Simulations $\dagger$. The Journal of Physical Chemistry A 2009, 113, 11856-11865, DOI: 10.1021/ jp9028968.
(416) Denzel, A.; Kästner, J. Gaussian process regression for geometry optimization. The Journal of Chemical Physics 2018, 148, 094114, DOI: 10.1063/1.5017103.
(417) Raggi, G.; Galvan, I. F.; Ritterhoff, C. L.; Vacher, M.; Lindh, R. RestrictedVariance Molecular Geometry Optimization Based on Gradient-Enhanced Kriging. Journal of Chemical Theory and Computation 2020, 16, 3989-4001, DOI: 10 . 1021/acs.jctc.0c00257.
(418) Bannwarth, C.; Ehlert, S.; Grimme, S. GFN2-xTB-An Accurate and Broadly Parametrized Self-Consistent Tight-Binding Quantum Chemical Method with Multipole Electrostatics and Density-Dependent Dispersion Contributions. Journal of Chemical Theory and Computation 2019, 15, 1652-1671, DOI: $10.1021 /$ acs.jctc.8b01176.
(419) Murtagh, B. A. Computational experience with quadratically convergent minimisation methods. The Computer Journal 1970, 13, 185-194, DOI: $10.1093 /$ comjnl/13.2.185.
(420) POWELL, M. J. D. On the Convergence of the Variable Metric Algorithm. IMA Journal of Applied Mathematics 1971, 7, 21-36, DOI: $10.1093 /$ imamat/7.1. 21.
(421) Galvan, I. F. et al. OpenMolcas: From Source Code to Insight. Journal of Chemical Theory and Computation 2019, 15, 5925-5964, DOI: $10.1021 /$ acs. jctc. 9b00532.
(422) Aquilante, F. et al. Modern quantum chemistry with [Open]Molcas. The Journal of Chemical Physics 2020, 152, 214117, DOI: $10.1063 / 5.0004835$.
(423) Jurečka, P.; Šponer, J.; Černý, J.; Hobza, P. Benchmark database of accurate (MP2 and $\operatorname{CCSD}(\mathrm{T})$ complete basis set limit) interaction energies of small model complexes, DNA base pairs, and amino acid pairs. Phys. Chem. Chem. Phys. 2006, 8, 1985-1993, DOI: $10.1039 / \mathrm{b} 600027 \mathrm{~d}$.
(424) Takatani, T.; Hohenstein, E. G.; Malagoli, M.; Marshall, M. S.; Sherrill, C. D. Basis set consistent revision of the S 22 test set of noncovalent interaction energies. The Journal of Chemical Physics 2010, 132, 144104, DOI: $10.1063 / 1.3378024$.
(425) Cremer, D. Møller-Plesset perturbation theory: from small molecule methods to methods for thousands of atoms. WIREs Computational Molecular Science 2011, 1, 509-530, DOI: https://doi.org/10.1002/wcms.58.
(426) Ditchfield, R.; Hehre, W. J.; Pople, J. A. Self-Consistent Molecular-Orbital Methods. IX. An Extended Gaussian-Type Basis for Molecular-Orbital Studies of Organic Molecules. J. Chem. Phys. 1971, 54, 724-728, DOI: $10.1063 / 1$. 1674902 .
(427) Hariharan, P. C.; Pople, J. A. The influence of polarization functions on molecular orbital hydrogenation energies. Theor. Chim. Acta 1973, 28, 213-222, DOI: 10.1007/b£00533485.
(428) Hehre, W. J.; Ditchfield, R.; Pople, J. A. Self-Consistent Molecular Orbital Methods. XII. Further Extensions of Gaussian-Type Basis Sets for Use in Molecular Orbital Studies of Organic Molecules. J. Chem. Phys. 1972, 56, 2257-2261, DOI: 10.1063/1.1677527.
(429) Weigend, F.; Häser, M. RI-MP2: first derivatives and global consistency. Theor. Chem. Acta 1997, 97, 331-340, DOI: $10.1007 / \mathrm{s} 002140050269$.
(430) Hättig, C. Optimization of auxiliary basis sets for RI-MP2 and RI-CC2 calculations. Phys. Chem. Chem. Phys. 2005, 7, 59-66, DOI: 10.1039/B415208E.
(431) Munkres, J. Algorithms for the Assignment and Transportation Problems. Journal of the Society for Industrial and Applied Mathematics 1957, 5, 32-38, DOI: 10.1137/0105003.
(432) Kuhn, H. W. The Hungarian method for the assignment problem. Naval Research Logistics Quarterly 1955, 2, 83-97, DOI: $10.1002 /$ nav. 3800020109 .
(433) Allen, W. J.; Rizzo, R. C. Implementation of the Hungarian Algorithm to Account for Ligand Symmetry and Similarity in Structure-Based Design. Journal of Chemical Information and Modeling 2014, 54, 518-529, DOI: $10.1021 /$ ci400534h.
(434) Wagner, A.; Himmel, H.-J. aRMSD: A Comprehensive Tool for Structural Analysis. Journal of Chemical Information and Modeling 2017, 57, 428-438, DOI: 10. 1021/acs.jcim.6b00516.
(435) Pettersen, E. F.; Goddard, T. D.; Huang, C. C.; Couch, G. S.; Greenblatt, D. M.; Meng, E. C.; Ferrin, T. E. UCSF Chimera?A visualization system for exploratory research and analysis. Journal of Computational Chemistry 2004, 25, 1605-1612, DOI: $10.1002 /$ jcc. 20084 .
(436) Coutsias, E. A.; Seok, C.; Dill, K. A. Using quaternions to calculate RMSD. Journal of Computational Chemistry 2004, 25, 1849-1857, DOI: $10.1002 /$ jcc. 20110.
(437) Ahuja, K.; Green, W. H.; Li, Y.-P. Learning to Optimize Molecular Geometries Using Reinforcement Learning. Journal of Chemical Theory and Computation 2021, DOI: 10.1021 /acs.jctc.0c00971.
(438) Helgaker, T. Transition-state optimizations by trust-region image minimization. Chemical Physics Letters 1991, 182, 503-510, DOI: 10.1016/0009-2614(91) 90115-p.
(439) Spicher, S.; Grimme, S. Single-Point Hessian Calculations for Improved Vibrational Frequencies and Rigid-Rotor-Harmonic-Oscillator Thermodynamics. Journal of Chemical Theory and Computation 2021, DOI:10.1021/acs.jctc.0c01306.
(440) Hermes, E. D.; Sargsyan, K.; Najm, H. N.; Zador, J. Accelerated Saddle Point Refinement through Full Exploitation of Partial Hessian Diagonalization. J. Chem. Theory Comput. 2019, 15, 6536-6549, DOI: $10.1021 /$ acs.jctc.9b00869.
(441) Virtanen, P. et al. SciPy 1.0: fundamental algorithms for scientific computing in Python. Nature Methods 2020, 17, 261-272, DOI: 10.1038/s41592-019-0686-2.
(442) Birkholz, A. B.; Schlegel, H. B. Using bonding to guide transition state optimization. J. Comput. Chem. 2015, 36, 1157-1166, DOI: $10.1002 /$ jcc. 23910.
(443) Peng, C.; Schlegel, H. B. Combining Synchronous Transit and Quasi-Newton Methods to Find Transition States. Isr. J. Chem. 1993, 33, 449-454, DOI: 10.1002/ijch.199300051.
(444) Ess, D. H. Transition-Structure Catalog of Organic Reactions. J. Chem. Educ. 2012, 89, 817-818, DOI: $10.1021 / \mathrm{ed} 2005856$.
(445) Stewart, J. J. P. Optimization of parameters for semiempirical methods V: Modification of NDDO approximations and application to 70 elements. J Mol Model 2007, 13, 1173-1213, DOI: 10.1007/s00894-007-0233-4.
(446) Vosko, S. H.; Wilk, L.; Nusair, M. Accurate spin-dependent electron liquid correlation energies for local spin density calculations: a critical analysis. Can. J. Phys. 1980, 58, 1200-1211, DOI: $10.1139 / \mathrm{p} 80-159$.
(447) Lee, C.; Yang, W.; Parr, R. G. Development of the Colle-Salvetti correlationenergy formula into a functional of the electron density. Physical Review $B$ 1988, 37, 785-789, DOI: $10.1103 /$ physrevb.37.785.
(448) Becke, A. D. Density-functional thermochemistry. III. The role of exact exchange. The Journal of Chemical Physics 1993, 98, 5648-5652, DOI: $10.1063 / 1$. 464913 .
(449) Stephens, P. J.; Devlin, F. J.; Chabalowski, C. F.; Frisch, M. J. Ab Initio Calculation of Vibrational Absorption and Circular Dichroism Spectra Using Density Functional Force Fields. J. Phys. Chem. 1994, 98, 11623-11627, DOI: 10.1021/j100096a001.
(450) Iron, M. A.; Janes, T. Evaluating Transition Metal Barrier Heights with the Latest Density Functional Theory Exchange-Correlation Functionals: The MOBH35 Benchmark Database. J. Phys. Chem. A 2019, 123, 3761-3781, DOI: 10.1021/ acs.jpca.9b01546.
(451) Dohm, S.; Bursch, M.; Hansen, A.; Grimme, S. Semiautomated Transition State Localization for Organometallic Complexes with Semiempirical Quantum Chemical Methods. J. Chem. Theory Comput. 2020, 16, 2002-2012, DOI: $10.1021 /$ acs.jctc.9b01266.
(452) Shukla, M.; Mishra, P. A gas phase ab initio excited state geometry optimization study of thymine cytosine and uracil. Chemical Physics 1999, 240, 319-329, DOI: $10.1016 / \mathrm{s} 0301-0104(98) 00374-7$.
(453) Peon, J.; Zewail, A. H. DNA/RNA nucleotides and nucleosides: direct measurement of excited-state lifetimes by femtosecond fluorescence up-conversion. Chemical Physics Letters 2001, 348, 255-262, DOI: 10.1016/s0009-2614(01) 01128-9.
(454) Mennucci, B.; Toniolo, A.; Tomasi, J. Theoretical Study of the Photophysics of Adenine in Solution: Tautomerism Deactivation Mechanisms, and Comparison with the 2-Aminopurine Fluorescent Isomer. The Journal of Physical Chemistry A 2001, 105, 4749-4757, DOI: $10.1021 /$ jp0 045843 .
(455) Shukla, M. K.; Leszczynski, J. Interaction of Water Molecules with Cytosine Tautomers: An Excited-State Quantum Chemical Investigation. The Journal of Physical Chemistry A 2002, 106, 11338-11346, DOI: 10.1021/jp021317j.
(456) Ismail, N.; Blancafort, L.; Olivucci, M.; Kohler, B.; Robb, M. A. Ultrafast Decay of Electronically Excited Singlet Cytosine via a $\pi, \pi^{*}$ to $\mathrm{n}_{O}, \pi^{*}$ State Switch. Journal of the American Chemical Society 2002, 124, 6818-6819, DOI: 10. 1021/ja0258273.
(457) Malone, R. J.; Miller, A. M.; Kohler, B. Singlet Excited-state Lifetimes of Cytosine Derivatives Measured by Femtosecond Transient Absorption. Photochemistry and Photobiology 2007, 77, 158-164, DOI: 10.1562/0031-8655(2003) 0770158 sesloc2.0.co2.
(458) Improta, R.; Santoro, F.; Blancafort, L. Quantum Mechanical Studies on the Photophysics and the Photochemistry of Nucleic Acids and Nucleobases. Chemical Reviews 2016, 116, 3540-3593, DOI: 10.1021/acs.chemrev.5b00444.
(459) Taylor, J. S. Unraveling the Molecular Pathway from Sunlight to Skin Cancer. Accounts of Chemical Research 1994, 27, 76-82, DOI: 10.1021/ar00039a003.
(460) Crespo-Hernández, C. E.; Cohen, B.; Hare, P. M.; Kohler, B. Ultrafast ExcitedState Dynamics in Nucleic Acids. Chemical Reviews 2004, 104, 1977-2020, DOI: $10.1021 /$ cr0206770.
(461) Kleinermanns, K.; Nachtigallová, D.; de Vries, M. S. Excited state dynamics of DNA bases. International Reviews in Physical Chemistry 2013, 32, 308-342, DOI: $10.1080 / 0144235 \times .2012 .760884$.
(462) Richter, M.; Marquetand, P.; González-Vázquez, J.; Sola, I.; González, L. Femtosecond Intersystem Crossing in the DNA Nucleobase Cytosine. The Journal of Physical Chemistry Letters 2012, 3, 3090-3095, DOI: 10.1021/jz301312h.
(463) Adamo, C.; Barone, V. Toward reliable density functional methods without adjustable parameters: The PBE0 model. The Journal of Chemical Physics 1999, 110, 6158-6170, DOI: $10.1063 / 1.478522$.
(464) Eichkorn, K.; Treutler, O.; Öhm, H.; Häser, M.; Ahlrichs, R. Auxiliary basis sets to approximate Coulomb potentials. Chemical Physics Letters 1995, 240, 283-290, DOI: 10.1016/0009-2614(95)00621-a.
(465) Mustafa, A. K.; Gadalla, M. M.; Snyder, S. H. Signaling by Gasotransmitters. Science Signaling 2009, 2, re2-re2, DOI: $10.1126 /$ scisignal. 268 re 2 ,
(466) Frostell, C.; Fratacci, M. D.; Wain, J. C.; Jones, R.; Zapol, W. M. Inhaled nitric oxide. A selective pulmonary vasodilator reversing hypoxic pulmonary vasoconstriction. Circulation 1991, 83, 2038-2047, DOI: $10.1161 / 01$. cir. 83 , 6.2038
(467) Williams, S. B.; Cusco, J. A.; Roddy, M.-A.; Johnstone, M. T.; Creager, M. A. Impaired nitric oxide-mediated vasodilation in patients with non-insulin-dependent diabetes mellitus. Journal of the American College of Cardiology 1996, 27, 567574, DOI: $10.1016 / 0735-1097$ (95) 00522-6.
(468) Murad, F. Nitric Oxide and Cyclic GMP in Cell Signaling and Drug Development. New England Journal of Medicine 2006, 355, 2003-2011, DOI: 10 . $1056 /$ nejmsa063904.
(469) Bogdan, C. Nitric oxide and the immune response. Nature Immunology 2001, 2, 907-916, DOI: $10.1038 / \mathrm{ni1001-907}$.
(470) Stuehr, D. J.; Gross, S. S.; Sakuma, I.; Levi, R.; Nathan, C. F. Activated murine macrophages secrete a metabolite of arginine with the bioactivity of endotheliumderived relaxing factor and the chemical reactivity of nitric oxide. The Journal of Experimental Medicine 1989, 169, 1011-1020, DOI: $10.1084 / \mathrm{jem} .169 .3$. 1011.
(471) Divakaran, S.; Loscalzo, J. The Role of Nitroglycerin and Other Nitrogen Oxides in Cardiovascular Therapeutics. Journal of the American College of Cardiology 2017, 70, 2393-2410, DOI: $10.1016 / \mathrm{j} \cdot \mathrm{jacc} .2017 .09 .1064$.
(472) Steinhorn, B. S.; Loscalzo, J.; Michel, T. Nitroglycerin and Nitric Oxide A Rondo of Themes in Cardiovascular Therapeutics. New England Journal of Medicine 2015, 373, 277-280, DOI: $10.1056 / \mathrm{ne}$ jmsr1503311.
(473) Carneiro, Z. A.; Biazzotto, J. C.; Alexiou, A. D.; Nikolaou, S. Nitric oxide photorelease from a trinuclear ruthenium nitrosyl complex and its in vitro cytotoxicity against melanoma cells. Journal of Inorganic Biochemistry 2014, 134, 36-38, DOI: $10.1016 / j . j i n o r g b i o .2014 .01 .012$.
(474) Roose, M.; Sasaki, I.; Bukhanko, V.; Mallet-Ladeira, S.; Barba-Barba, R. M.; Ramos-Ortiz, G.; Enriquez-Cabrera, A.; Farfán, N.; Lacroix, P. G.; Malfant, I. Nitric oxide photo-release from a ruthenium nitrosyl complex with a 4,4/-bisfluorenyl-2,2/-bipyridine ligand. Polyhedron 2018, 151, 100-111, DOI: 10 . 1016/j.poly.2018.05.028.
(475) Bukhanko, V.; Lacroix, P. G.; Sasaki, I.; Tassé, M.; Mallet-Ladeira, S.; Voitenko, Z.; Malfant, I. Mechanism and oxidation state involved in the nitric oxide (NO) photorelease in a terpyridine-bipyridine-based ruthenium nitrosyl complex. Inorganica Chimica Acta 2018, 482, 195-205, DOI: $10.1016 /$ j.ica.2018.05.038.
(476) De Lima, R. G.; Sauaia, M. G.; Bonaventura, D.; Tedesco, A. C.; Bendhack, L. M.; da Silva, R. S. Influence of ancillary ligand $L$ in the nitric oxide photorelease by the $[\mathrm{Ru}(\mathrm{L})($ tpy $) \mathrm{NO}] 3+$ complex and its vasodilator activity based on visible light irradiation. Inorganica Chimica Acta 2006, 359, 2543-2549, DOI: $10.1016 / j$. ica.2006.02.020.
(477) De Lima, R. G.; Sauaia, M. G.; Bonaventura, D.; Tedesco, A. C.; Lopez, R. F. V.; Bendhack, L. M.; da Silva, R. S. Controlled nitric oxide photo-release from nitro ruthenium complexes: The vasodilator response produced by UV light irradiation. Inorganica Chimica Acta 2005, 358, 2643-2650, DOI: 10.1016/j.ica. 2005 . 03.019 .
(478) Rose, M. J.; Mascharak, P. K. Fiat Lux: selective delivery of high flux of nitric oxide (NO) to biological targets using photoactive metal nitrosyls. Current Opinion in Chemical Biology 2008, 12, 238-244, DOI: $10.1016 / \mathrm{j} . \mathrm{cbpa} .2008 .02 .009$.
(479) García, J. S.; Alary, F.; Boggio-Pasqua, M.; Dixon, I. M.; Heully, J.-L. Is photoisomerization required for NO photorelease in ruthenium nitrosyl complexes? Journal of Molecular Modeling 2016, 22, DOI: 10.1007/s00894-016-3138-2.
(480) De Lima Batista, A. P.; de Oliveira-Filho, A. G. S.; Galembeck, S. E. Photophysical properties and the NO photorelease mechanism of a ruthenium nitrosyl model complex investigated using the CASSCF-in-DFT embedding approach. Physical Chemistry Chemical Physics 2017, 19, 13860-13867, DOI: 10.1039/c7cp01642e.
(481) Roose, M.; Tassé, M.; Lacroix, P. G.; Malfant, I. Nitric oxide (NO) photo-release in a series of ruthenium-nitrosyl complexes: new experimental insights in the search for a comprehensive mechanism. New Journal of Chemistry 2019, 43, 755-767, DOI: $10.1039 / \mathrm{c} 8 \mathrm{nj} 03907 \mathrm{k}$.
(482) Giri, B.; Kumbhakar, S.; Selvan, K. K.; Muley, A.; Maji, S. Formation reactivity, photorelease, and scavenging of NO in ruthenium nitrosyl complexes. Inorganica Chimica Acta 2020, 502, 119360, DOI: 10.1016/j.ica.2019.119360.
(483) González, L.; Boggio-Pasqua, M.; Talotta, F. Early relaxation dynamics in the photoswitchable trans- $[\mathrm{RuCl}(\mathrm{NO})(\mathrm{py}) 4] 2+$. Chemistry - A European Journal 2020, DOI: $10.1002 /$ chem. 202000507.
(484) Sasaki, I.; Amabilino, S.; Mallet-Ladeira, S.; Tassé, M.; Sournia-Saquet, A.; Lacroix, P. G.; Malfant, I. Further studies on the photoreactivities of ruthe-nium-nitrosyl complexes with terpyridyl ligands. New Journal of Chemistry 2019, 43, 11241-11250, DOI: $10.1039 / \mathrm{c} 9 \mathrm{nj} 02398 \mathrm{~d}$.
(485) Becke, A. D. A new mixing of Hartree-Fock and local density-functional theories. The Journal of Chemical Physics 1993, 98, 1372-1377, DOI: $10.1063 / 1$. 464304.
(486) Hehre, W. J.; Ditchfield, R.; Pople, J. A. Self-Consistent Molecular Orbital Methods. XII. Further Extensions of Gaussian-Type Basis Sets for Use in Molecular Orbital Studies of Organic Molecules. The Journal of Chemical Physics 1972, 56, 2257-2261, DOI: $10.1063 / 1.1677527$.
(487) Spitznagel, G. W.; Clark, T.; von Ragué Schleyer, P.; Hehre, W. J. An evaluation of the performance of diffuse function-augmented basis sets for second row elements Na-Cl. Journal of Computational Chemistry 1987, 8, 1109-1116, DOI: $10.1002 /$ jcc. 540080807.
(488) Clark, T.; Chandrasekhar, J.; Spitznagel, G. W.; Schleyer, P. V. R. Efficient diffuse function-augmented basis sets for anion calculations. III. The $3-21+\mathrm{G}$ basis set for first-row elements Li-F. Journal of Computational Chemistry 1983, 4, 294-301, DOI: $10.1002 /$ jcc. 540040303.
(489) T. H. Dunning, P. J. H., In Modern Theoretical Chemistry; Plenum: 1977.
(490) Hay, P. J.; Wadt, W. R. Ab initio effective core potentials for molecular calculations. Potentials for the transition metal atoms Sc to Hg. The Journal of Chemical Physics 1985, 82, 270-283, DOI: $10.1063 / 1.448799$.
(491) Lu, T.; Chen, F. Multiwfn: A multifunctional wavefunction analyzer. Journal of Computational Chemistry 2011, 33, 580-592, DOI: $10.1002 /$ jcc. 22885 .
(492) Hanson, R. M. Jmol- a paradigm shift in crystallographic visualization. Journal of Applied Crystallography 2010, 43, 1250-1260, DOI:10.1107/s0021889810030256.
(493) Larsen, C. B.; van der Salm, H.; Shillito, G. E.; Lucas, N. T.; Gordon, K. C. Tuning the Rainbow: Systematic Modulation of Donor-Acceptor Systems through Donor Substituents and Solvent. Inorganic Chemistry 2016, 55, 8446-8458, DOI: 10.1021/acs.inorgchem.6b01039.
(494) Shillito, G. E.; Hall, T. B. J.; Preston, D.; Traber, P.; Wu, L.; Reynolds, K. E. A.; Horvath, R.; Sun, X. Z.; Lucas, N. T.; Crowley, J. D.; George, M. W.; Kupfer, S.; Gordon, K. C. Dramatic Alteration of 3ILCT Lifetimes Using Ancillary Ligands in $[\operatorname{Re}(\mathrm{L})(\mathrm{CO}) 3($ phen-TPA $)] \mathrm{n}+$ Complexes: An Integrated Spectroscopic and Theoretical Study. Journal of the American Chemical Society 2018, 140, 45344542, DOI: $10.1021 /$ jacs. 7b12868.
(495) Kupfer, S.; Zedler, L.; Guthmuller, J.; Bode, S.; Hager, M. D.; Schubert, U. S.; Popp, J.; Gräfe, S.; Dietzek, B. Self-healing mechanism of metallopolymers investigated by QM/MM simulations and Raman spectroscopy. Physical Chemistry Chemical Physics 2014, 16, 12422, DOI: $10.1039 / \mathrm{c} 4 \mathrm{cp} 00562 \mathrm{~g}$.
(496) Schindler, J.; Kupfer, S.; Zedler, L.; Wächtler, M.; Gräfe, S.; Ryan, A. A.; Senge, M. O.; Dietzek, B. Spectroelectrochemical Investigation of the One-Electron Reduction of Nonplanar Nickel(II) Porphyrins. ChemPhysChem 2016, 17, 34803493, DOI: $10.1002 /$ cphc. 201600698 .
(497) Reiher, M.; Salomon, O.; Hess, B. A. Reparameterization of hybrid functionals based on energy differences of states of different multiplicity. Theoretical Chemistry Accounts: Theory Computation, and Modeling (Theoretica Chimica Acta) 2001, 107, 48-55, DOI: 10.1007/s00214-001-0300-3.
(498) Salomon, O.; Reiher, M.; Hess, B. A. Assertion and validation of the performance of the B3LYP $\star$ functional for the first transition metal row and the G2 test set. The Journal of Chemical Physics 2002, 117, 4729-4737, DOI:10.1063/1.1493179.
(499) Dolg, M.; Wedig, U.; Stoll, H.; Preuss, H. Ab initio pseudopotential study of the first row transition metal monoxides and iron monohydride. The Journal of Chemical Physics 1987, 86, 2123-2131, DOI: $10.1063 / 1.452110$.
(500) Mennucci, B.; Cappelli, C.; Guido, C. A.; Cammi, R.; Tomasi, J. Structures and Properties of Electronically Excited Chromophores in Solution from the Polarizable Continuum Model Coupled to the Time-Dependent Density Functional Theory. The Journal of Physical Chemistry A 2009, 113, 3009-3020, DOI: 10 . 1021/jp8094853.
(501) Yanai, T.; Tew, D. P.; Handy, N. C. A new hybrid exchange-correlation functional using the Coulomb-attenuating method (CAM-B3LYP). Chemical Physics Letters 2004, 393, 51-57, DOI: $10.1016 / \mathrm{j}$. cplett. 2004.06 .011 .
(502) GNU General Public License, version 3, Free Software Foundation, 2007.
(503) Smith, D. et al. Psi4 1.4: Open-Source Software for High-Throughput Quantum Chemistry. 2020, DOI: $10.26434 /$ chemrxiv.11930031.v1,
(504) Sun, Q.; Berkelbach, T. C.; Blunt, N. S.; Booth, G. H.; Guo, S.; Li, Z.; Liu, J.; McClain, J. D.; Sayfutyarova, E. R.; Sharma, S.; Wouters, S.; Chan, G. K.-L. PySCF: the Python-based simulations of chemistry framework. WIREs Computational Molecular Science 2017, 8, DOI: $10.1002 / \mathrm{wcms} .1340$.
(505) Eastman, P.; Swails, J.; Chodera, J. D.; McGibbon, R. T.; Zhao, Y.; Beauchamp, K. A.; Wang, L.-P.; Simmonett, A. C.; Harrigan, M. P.; Stern, C. D.; Wiewiora, R. P.; Brooks, B. R.; Pande, V. S. OpenMM 7: Rapid development of high performance algorithms for molecular dynamics. PLoS Comput Biol Computational Biology 2017, 13, ed. by Gentleman, R., e1005659, DOI: 10.1371 / journal. pcbi. 1005659 .
(506) Olsen, J. M. H.; Reine, S.; Vahtras, O.; Kjellgren, E.; Reinholdt, P.; Dundas, K. O. H.; Li, X.; Cukras, J.; Ringholm, M.; Hedegard, E. D.; Remigio, R. D.; List, N. H.; Faber, R.; Tenorio, B. N. C.; Bast, R.; Pedersen, T. B.; Rinkevicius, Z.; Sauer, S. P. A.; Mikkelsen, K. V.; Kongsted, J.; Coriani, S.; Ruud, K.; Helgaker, T.; Jensen, H. J. A.; Norman, P. Dalton Project: A Python platform for molecularand electronic-structure simulations of complex systems. J. Chem. Phys. 2020, 152, 214115, DOI: $10.1063 / 1.5144298$.
(507) Herbst, M. F.; Scheurer, M.; Fransson, T.; Rehn, D. R.; Dreuw, A. adcc: A versatile toolkit for rapid development of algebraic-diagrammatic construction methods. WIREs Comput Mol Sci Computational Molecular Science 2020, 10, DOI: $10.1002 / \mathrm{wcms} .1462$,
(508) Harris, C. R. et al. Array programming with NumPy. Nature 2020, 585, 357-362, DOI: $10.1038 / \mathrm{s} 41586-020-2649-2$.
(509) Stewart, J. J. P. MOPAC2016, 2016.
(510) Smith, D.; Altarawy, D.; Burns, L.; Welborn, M.; Naden, L. N.; Ward, L.; Ellis, S.; Crawford, T. The MolSSI QCArchive Project: An Open-Source Platform to Compute Organize, and Share Quantum Chemistry Data. 2020, DOI: 10 . 26434/chemrxiv.11908356.v1.
(511) Valiev, M.; Bylaska, E.; Govind, N.; Kowalski, K.; Straatsma, T.; Dam, H. V.; Wang, D.; Nieplocha, J.; Apra, E.; Windus, T.; de Jong, W. NWChem: A comprehensive and scalable open-source solution for large scale molecular simulations. Computer Physics Communications 2010, 181, 1477-1489, DOI: 10.1016/j. срс.2010.04.018.
(512) Gordon, M. S.; Schmidt, M. W. In Theory and Applications of Computational Chemistry; Elsevier: 2005, pp 1167-1189, DOI: 10.1016/b978-044451719-7/50084-6.
(513) Shao, Y. et al. Advances in molecular quantum chemistry contained in the QChem 4 program package. Molecular Physics 2014, 113, 184-215, DOI: 10 . 1080/00268976.2014.952696.
(514) For dynamic task scheduling, D. L. Dask Development Team, 2016.
(515) Meurer, A. et al. SymPy: symbolic computing in Python. PeerJ Computer Science 2017, 3, e103, DOI: $10.7717 /$ peerj-cs. 103.
(516) Ceriotti, M.; More, J.; Manolopoulos, D. E. i-PI: A Python interface for ab initio path integral molecular dynamics simulations. Computer Physics Communications 2014, 185, 1019-1026, DOI: $10.1016 /$ j.cpc.2013.10.027.
(517) Kapil, V. et al. i-PI 2.0: A universal force engine for advanced molecular simulations. Computer Physics Communications 2019, 236, 214-223, DOI: $10.1016 /$ j.cpc.2018.09.020.
(518) Giannozzi, P. et al. QUANTUM ESPRESSO: a modular and open-source software project for quantum simulations of materials. J. Phys.: Condens. Matter 2009, 21, 395502, DOI: 10.1088/0953-8984/21/39/395502.
(519) Giannozzi, P.; Baseggio, O.; Bonfa, P.; Brunato, D.; Car, R.; Carnimeo, I.; Cavazzoni, C.; de Gironcoli, S.; Delugas, P.; Ruffino, F. F.; Ferretti, A.; Marzari, N.; Timrov, I.; Urru, A.; Baroni, S. Quantum ESPRESSO toward the exascale. J. Chem. Phys. 2020, 152, 154105, DOI: $10.1063 / 5.0005082$.
(520) Artacho, E.; Anglada, E.; Dieguez, O.; Gale, J. D.; Garcia, A.; Junquera, J.; Martin, R. M.; Ordejon, P.; Pruneda, J. M.; Sanchez-Portal, D.; Soler, J. M. The SIESTA method - developments and applicability. J. Phys.: Condens. Matter 2008, 20, 064208, DOI: $10.1088 / 0953-8984 / 20 / 6 / 064208$.
(521) Apra, E. et al. NWChem: Past, present, and future. J. Chem. Phys. 2020, 152, 184102, DOI: $10.1063 / 5.0004997$.
(522) Meyer, M. Continuous Integration and Its Tools. IEEE Software 2014, 31, 14-16, DOI: $10.1109 / \mathrm{ms} .2014 .58$.
(523) Jones, G. Make Dimer Method N-Vector Translationally Invariant, https: //github. com/eljost/pysisyphus/pull/116, [Online; accessed at 2021-02-09], 2021.
(524) Mloston, G.; Buday, P.; Seeber, P.; Neumann, C.; Abul-Futouh, H.; Görls, H.; Gräfe, S.; Matczak, P.; Kupfer, S.; Weigand, W. $\operatorname{Iron}(0)$ mediated stereoselective (3+2)-cycloaddition of thiochalcones via a diradical intermediate. Chemistry - A European Journal 2020, DOI: $10.1002 / \mathrm{chem} .202001412$.
(525) eljost/pywfo, https://github.com/eljost/pywfo, Accessed on Wed, April 29, 2020.
(526) Krekel, H.; Oliveira, B.; Pfannschmidt, R.; Bruynooghe, F.; Laugher, B.; Bruhin, F. pytest 6.2, 2004.
(527) Hubertz, J. In Softwaretests mit Python; Springer Berlin Heidelberg: Berlin, Heidelberg, 2016, pp 117-169, DOI: 10.1007/978-3-662-48603-0_5.
(528) Pajankar, A. In Python Unit Test Automation : Practical Techniques for Python Developers and Testers; Apress: Berkeley, CA, 2017, pp 87-100, DOI: $10.1007 /$ 978-1-4842-2677-3_5.
(529) Hunt, J. In Advanced Guide to Python 3 Programming; Springer International Publishing: Cham, 2019, pp 175-186, DOI: 10.1007/978-3-030-259433_15.


[^0]:    ${ }^{1}$ Henry VIII, (1491-1547), King of England, presumably died of sepsis. 10,11 Napoleon Bonaparte (1769-1821) and Ivan the Terrible (1530-1584) presumably died of syphilis. 12

[^1]:    ${ }^{1}$ Matrix $\boldsymbol{A}$ is normal if $\boldsymbol{A} \boldsymbol{A}^{*}=\boldsymbol{A}^{*} \boldsymbol{A}$

[^2]:    ${ }^{1}$ If the Hessian is calculated every 10th cycle, the DM becomes more efficient than PRFO for molecules with $\geq 15$ atoms. 290

[^3]:    Parts of this chapter are based on:
    Haensch, V. G.; Neuwirth, T.; Steinmetzer, J.; Kloss, F.; Beckert, R.; Gräfe, S.; Kupfer, S.; Hertweck, C. Metal-Free Aryl Cross-Coupling Directed by Traceless Linkers. Chemistry - A European Journal 2019, 25, 16068-16073, DOI: $10.1002 /$ chem. 201903582
    Open Access publication, licensed under Creative Commons - Attribution 4.0 International (CC BY 4.0).

[^4]:    ${ }^{1}$ Two dihedrals of $179^{\circ}$ and $-179^{\circ}$ are nearly identical, even though their naively calculated (Euclidean) distance is big.

[^5]:    ${ }^{2} \Delta E\left(S_{1}\right)+$ ES barrier height $-\Delta E\left(S_{3}\right)=5.12 \mathrm{eV}+0.69 \mathrm{eV}-5.40 \mathrm{eV}=0.41 \mathrm{eV}$

[^6]:    ${ }^{1}$ Using RIC usually results in different units for individual coordinates, e.g., lengths for stretches, and radians for angles (bends, dihedrals). Certain definitions for out-of-plane angles and linear bends even result in unitless coordinates. 276 Mixed units are denoted as $a_{0}$ (rad) in the present work.

[^7]:    ${ }^{2}$ The author of this thesis did not contribute in any way to the work by Ahuja et al.

[^8]:    ${ }^{3}$ Numerical double- $\zeta$ basis and GGA•PBE XCffunctional

[^9]:    
    
    
    

[^10]:    $\sqrt[4]{ }$ https://pysisyphus.readthedocs.io

