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# STRUCTURE AND RESISTIVITY OF FeCoZr-DIELECTRIC GRANULAR NANOCOMPOSITES SINTERED IN MIXED Ar+O<sub>2</sub> ATMOSPHERE

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## ABSTRACT

The response of structure, electrical conductivity, magnetoresistivity and magnetic properties of (FeCoZr)<sub>x</sub>(Al<sub>2</sub>O<sub>3</sub>)<sub>1-x</sub> and (FeCoZr)<sub>x</sub>(PZT)<sub>100-x</sub> (17 at. % <  $x$  < 67 at. %) nanocomposite films on oxygen incorporation into the sputtering ambient was investigated using Mössbauer spectroscopy, resistivity and magnetoresistivity measurements. Change of Fe local states and carrier transport mechanisms due to incorporation of oxygen into FeCoZr-containing granular nanocomposites (*GNCs*) with two different matrixes are discussed with regard to the formation of complex semiconductive FeCo-oxide interlayers and/or shells possessing different types of conductivity as well as the specific properties of Al<sub>2</sub>O<sub>3</sub> and PZT.

**Index Terms** - One, two, three, four, five, six, seven, eight, nine, ten

## 1. INTRODUCTION

Unique magnetic and electroconductivity properties of granular soft ferromagnetic metal-dielectric matrix nanocomposite materials (*GNCs*) make them very attractive in designing of various magnetoelectronic devices [1-3]. It is well established now that for binary metal-dielectric *GNCs* electrical conductivity with and without applied external magnetic field should be considered within the frame of percolation theory [4]. It defines the percolation threshold  $X_C$  as such a concentration of metallic nanoparticles  $X_{FeCoZr}$  in *GNCs* when conductivity mechanism changes from tunnelling or variable-range hopping (VHR) to the metallic carrier transport due to formation of current-conductive cluster. Experimentally observed, that numerous characteristics of *GNCs*, like coercive force, magnetoresistivity effect, etc. revealed maximum values particularly within  $X_C$  [5]. The replacement of tunneling and/or VHR by localized states in dielectric with metallic conductivity through percolative net of nanoparticles has been experimentally observed within  $X_C \approx 50$  % with some variations for different metallic and dielectric compositions. For the case of FeCoZr-Al<sub>2</sub>O<sub>3</sub> sintered in pure Ar ambient experimentally established  $X_C$  was about 45-47 at. % of FeCoZr [6].

It is noteworthy that in some cases formation of current-conductive cluster of metallic nanoparticles within  $X_C$  coincide with nucleation of ferromagnetically-interacting net of metallic nanoparticles, although this is not the general case. In so doing, both magnetic and conductivity properties of *GNCs* are governed with  $X_{FeCoZr}$  value with respect to  $X_C$  for definite combination of metal and dielectric.

Previously obtained results of RT Mössbauer spectroscopy, SQUID-magnetometry and atomic force – magnetic force microscopy [6] confirmed that sputtering of FeCoZr-Al<sub>2</sub>O<sub>3</sub> *GNC* in mixed Ar+O<sub>2</sub> ambient draw to drastic change of correspondent magnetic characteristics and structure. It was related to the progressive oxidation of FeCoZr nanoparticles with  $X_{FeCoZr}$  increase and also formation of complex FeCo-oxides.

Present research was focused on two main purposes: (i) to clarify the correlation between formed local Fe states in oxidized FeCoZr-Al<sub>2</sub>O<sub>3</sub> *GNC* and correspondent electrical properties; (ii) to investigate the role of matrix in oxidation of FeCoZr as well as resistivity and magnetoresistivity on the example of FeCoZr-Pb<sub>0,81</sub>Sr<sub>0,04</sub>(NaBi)<sub>0,15</sub>(ZrTi)O<sub>3</sub> (doped PZT) *GNCs*. Application of segneto- and ferroelectric PZT matrix is supposed to be useful when varying state of *GNCs* by application of external magnetic or electric field.

## 2. EXPERIMENTAL

Nanogranular films of (Fe<sub>45</sub>Co<sub>45</sub>Zr<sub>10</sub>)<sub>x</sub>(*D*)<sub>100-x</sub> (17 at.% <  $x$  < 67 at.%), where dielectric matrix *D* was either Al<sub>2</sub>O<sub>3</sub> or Pb<sub>0,81</sub>Sr<sub>0,04</sub>(Na<sub>0,5</sub>Bi<sub>0,5</sub>)<sub>0,15</sub>(Zr<sub>0,575</sub>Ti<sub>0,425</sub>)O<sub>3</sub> (doped PZT), were deposited on glass-ceramic and Al substrates by ion-beam sputtering in gas mixture Ar+O<sub>2</sub> atmosphere at partial oxygen pressures  $P_O = 2,4 \div 4,2 \cdot 10^{-3}$  Pa. The details of deposition procedure were summarized in [6].

Room temperature 57Fe Mössbauer spectroscopy was applied to reveal of local structure and magnetic state of the nanocomposites. The spectra were recorded at RT in transmission geometry using 57Co/Rh source (40 mCi). The fitting procedure was performed with the use of MOSMOD program assuming the distribution of hyperfine magnetic fields ( $H_{hf}$ ) and quadrupole splittings ( $\epsilon$ ). All isomer shifts

( $\delta$ ) were referenced to  $\alpha$ -Fe. Lorentzian width (FWHM) of the fitted spectral lines was constant and equaled to 0.15 mm/s. Relatively high FWHM value was attributed to the amorphous structure of FeCoZr nanoparticles.

Mechanisms of carrier transport in GNCs were studied using temperature dependences of DC conductivity at the temperature range 4 – 300 K. To make these measurements, the nanocomposite films sputtered onto glass-ceramic substrates were cut into rectangular strips with dimensions of 10 mm  $\times$  2 mm  $\times$   $d$   $\mu$ m ( $d$  is thickness of the films), and 4 indium contacts were deposited using ultrasound soldering. Samples were measured in flow cryostat system or closed-cycle refrigerator with electric field intensities  $E$  up to  $10^5$  V/m and magnetic fields with induction  $B$  up to 600 mT or 8 T using the 4 probe potentiometric method. Ratio error of electric conductance measurements was better than 5 %.

## 2.1. Mössbauer spectroscopy

Mössbauer spectra of the samples deposited at  $P_O = 4.2 \cdot 10^{-3}$  Pa (see Fig. 1) show nonmagnetic behavior in the whole range of  $X_{FeCoZr}$ . It is in contrast with the results reported in [5,6], where spectra of FeCoZr- $Al_2O_3$  sintered in pure Ar ambient revealed the transition from nonmagnetic spectrum to magnetically split spectrum at  $X_{FeCoZr} = X_C \approx 47$  at.%. Such transformation was assigned to the structural change from non-interacting superparamagnetic FeCoZr nanoparticles to the ferromagnetically interacting net of these nanoparticles.

Mössbauer spectra of FeCoZr- $Al_2O_3$  sintered in mixed Ar+O ambient for  $X_{FeCoZr} \leq 55$  at.% were evaluated assuming three subspectra: two doublets ( $D_1^{Al_2O_3}$ :  $\delta=0.08-0.11$  mm/s,  $\epsilon=0.41-0.49$  mm/s;  $D_2^{Al_2O_3}$ :  $\delta=0.40-0.46$  mm/s,  $\epsilon=0.77-0.94$  mm/s) associated with  $Fe^{3+}$  ions and one doublet assigned to  $Fe^{2+}$  state ( $D_3^{Al_2O_3}$ :  $\delta=0.90-0.96$  mm/s,  $\epsilon=1.81-1.91$  mm/s). In accordance with [6]  $D_1^{Al_2O_3}$  doublet should be assigned to superparamagnetic FeCoZr nanoparticles separated in  $Al_2O_3$  matrix.  $D_2^{Al_2O_3}$  subspectrum has parameters similar to nanosized iron oxides and most probably characterizes oxidized FeCoZr nanoparticles. Basing on magnetometry data as well as electrical conductivity results [6]  $D_3^{Al_2O_3}$  doublet may be assigned to complex  $Fe^{2+}$  oxides (like  $FeCo_2O_4$ ) formed either on the interface between FeCoZr and  $Al_2O_3$  matrix or inside  $Al_2O_3$  matrix. Mössbauer spectra of samples containing  $X_{FeCoZr} > 59$  at. % could be fitted with the superposition of only  $D_2^{Al_2O_3}$  and  $D_3^{Al_2O_3}$  doublets, while  $D_1^{Al_2O_3}$  was not detected.

This evidenced that increase of  $X_{FeCoZr}$  caused progressive oxidation of FeCoZr nanoparticles and diminishing of residual pure FeCoZr granulas.

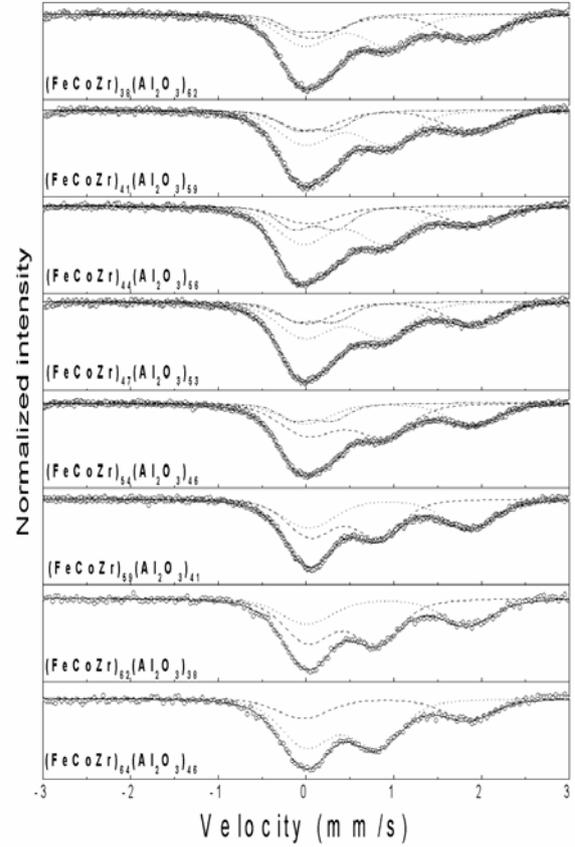


Figure 1 RT Mössbauer spectra recorded for the oxidized FeCoZr- $Al_2O_3$  films sputtered at  $P_O = 4.2 \cdot 10^{-3}$  Pa

## 2.2. Electrical conductivity and magnetoresistivity

Temperature dependences of DC conductivity  $\sigma(T)$  for FeCoZr- $Al_2O_3$  display, as seen in Fig. 2, activation laws  $\ln(\sigma) \sim (1/T)^n$  with the “sliding” activation energies at the whole range of  $X_{FeCoZr}$ .

As was shown in [7,8] and seen from Fig. 2,  $\sigma(T)$  dependences of the most samples studied were close to VRH behaviour

$$\sigma(T) = \sigma_0 \cdot \exp\left(-\left(\frac{T_0}{T}\right)^n\right),$$

where  $\sigma_0$  is an equilibrium conductivity at  $T \rightarrow \infty$ . In so doing, experiments yielded either the Mott law [7],  $\ln(\sigma) \propto (T_0^M/T)^{-0.25}$ , or the Shklovski-Efros law,  $\ln(\sigma) \propto (T_0^{SE}/T)^{-0.5}$ , where parameters  $T_0^M$  and  $T_0^{SE}$  determine average activation energies for carrier hopping by localized states around Fermi level in dielectric matrix for Mott and Shklovski-Efros VRH regimes, respectively.

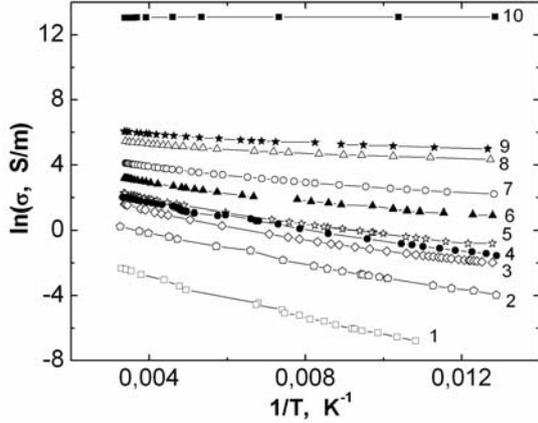


Figure 2 Temperature dependences of conductivity  $\sigma(T)$  for oxidized  $\text{FeCoZr-Al}_2\text{O}_3$  at  $X_{\text{FeCoZr}}$  (in at.%)  $\approx$  31.2 (1), 36.6 (2), 37.4 (3), 38.6 (4), 40.7 (5), 49 (6), 53.6 (7), 62.5 (8), 63.7 (9). Curve 10 -  $\sigma(T)$  for pure  $\text{FeCoZr}$

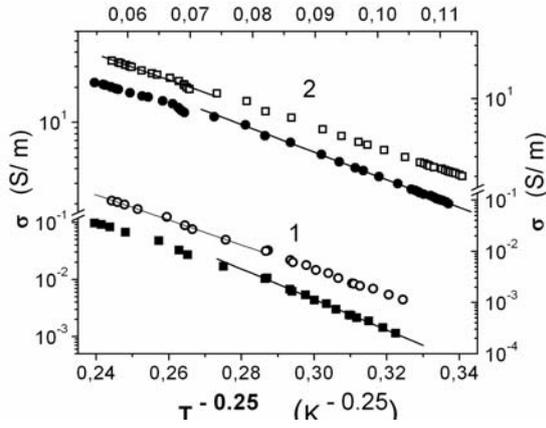


Figure 3 Temperature dependences of conductivity  $\sigma(T)$  for oxidized  $\text{FeCoZr-Al}_2\text{O}_3$  with  $x = 31.2$  at.% (1) and  $x = 45$  at.% (2)

The fitting procedure for the  $\text{FeCoZr-Al}_2\text{O}_3$  showed that the Mott law holds only at low temperatures while at higher temperatures  $T > 160$ - $170$  K we observed the crossover from Mott low to the Shklovski-Efros law (see Fig. 3) describing the hopping transport of the charge carriers at the Coulomb gap in the localized states density near Fermi level [9-11].

The activation energies estimated from the  $\sigma(T)$  showed that the  $T_0^M$  values are essentially higher by the order of magnitude for the oxidized  $\text{FeCoZr-Al}_2\text{O}_3$  as compared to the samples sintered in pure Ar [9]. This result strongly correlate with conclusions based on Mössbauer spectroscopy and magnetometry revealing that in the latter case, the oxide “shells”, that separate  $\text{FeCoZr}$  nanoparticles in the current-conducting routes, created additional barriers for electron transport between nanoparticles. Therefore the Shklovski-Efros behaviour of  $\sigma(T)$  in the oxidized  $\text{FeCoZr}$  can be naturally attributed to the influence of

these oxide “shells” possessing semiconductive electric properties [9-11].

The  $\sigma(T)$  curves allowed us to estimate principally the changes in the localized state density  $N(E_F)$  and localized electron wave function damping length  $a$  in  $\text{Al}_2\text{O}_3$  matrix depending on the  $X_{\text{FeCoZr}}$  using experimentally extracted  $T_0$  and  $\sigma_0$  values (see [12]). When the Mott hopping mechanism of VRH carrier transport dominating the semi-empirical Kirkpatrick formula [10]

$$\sigma = 0,0217 \left( \frac{c}{a} \right) \left( \frac{T_0}{T} \right)^{0,35} \cdot \exp \left[ - \left( \frac{T_0}{T} \right)^{0,25} \right] = \sigma_{AC}^o \cdot \exp \left[ - \left( \frac{T_0}{T} \right)^{0,25} \right]$$

was used to fit  $T_0$  and  $\sigma_0$  values, where

$$\sigma_{AC}^o \approx \sigma_0 \left( \frac{T_0}{T} \right)^{0,35}$$

and

$$C = \left( \frac{E_1^2}{\pi^4 \cdot d \cdot s^5 \cdot \hbar^4} \right) \cdot \left( \frac{2 \cdot e^3}{3 \cdot 4 \cdot \pi \cdot \epsilon \cdot \epsilon_0 \cdot a} \right)^2$$

Here  $a$  is the localized wave function damping length (in cm),  $e$  is the electron charge. The parameters values used in this relations for  $\text{Al}_2\text{O}_3$  were: deformational potential  $E_1 = 10$  eV, density  $d = 3.97$  g/cm<sup>3</sup>, sound velocity  $s = 3 \times 10^3$  cm/s, dielectric constant  $\kappa_0 = 10$  [13]. The  $\sigma_0$  is constant for the sample (with dimensionality  $\Omega^{-1} \text{cm}^{-1}$ ) and  $N(E_F)$  – energy density of localized states (eV<sup>-1</sup>cm<sub>3</sub>) at the Fermi level in dielectric matrix. Using these parameters we can connect the  $T_0$  and  $\sigma_0$  values with  $N(E_F)$  and  $a$ :

$$\sigma_0 = \frac{6.95 \cdot 10^{-15}}{a^3}, \quad T_0 = \frac{8.65 \cdot 10^{-4}}{N(E_F) \cdot a^3}$$

The estimations of  $N(E_F)$  and  $a$  values for the case of  $X_{\text{FeCoZr}} < X_C$  on the base of above presented formulas are shown in Table.

**Table**

$T_0$  parameter, equilibrium conductivity  $\sigma_0$ , energy density  $N(E_F)$  of localized states at the Fermi level in dielectric matrix and the localized wave function damping length  $a$  for oxidized  $\text{FeCoZr-Al}_2\text{O}_3$  at  $T < T_{oc}$  where the Mott law is applied

| $X_{\text{FeCoZr}}$ , at.% | $T_0 \cdot 10^{-3}$ , K | $\sigma_0 \cdot 10^{-3}$ , S/m | $a$ , nm | $N(E_F) \cdot 10^{-18}$ , eV <sup>-1</sup> cm <sup>-3</sup> |
|----------------------------|-------------------------|--------------------------------|----------|---|
| 31.2                       | 1990                    | 12390                          | 4        | 144   |
| 36.6                       | 795                     | 5318                           | 5        | 213   |
| 37.4                       | 323                     | 976                            | 8        | 131   |

As follows from the Table in the range of  $31 < X_{FeCoZr} < 40$  at.% (below the percolation threshold  $X_C$ ) radius of localization of electrons is about  $9 \pm 2$  nm and  $N(E_F)$  is about  $(3 \div 12) \cdot 10^{18} \text{ eV}^{-1} \text{ cm}^{-3}$ . The addition of oxygen to Ar ambient of sputtering chamber results in some lowering of  $a$  down to  $6 \pm 2$  nm and the increase of  $N(E_F)$  up to  $(1 \div 2) \cdot 10^{20} \text{ eV}^{-1} \text{ cm}^{-3}$  (see, Table). This increase of  $N(E_F)$  nearly 2 orders in the FeCoZr-Al<sub>2</sub>O<sub>3</sub> films can be attributed to the influence of oxide “shells” around FeCoZr nanoparticles.

Additional confirmation of tunnelling and/or VRH mechanisms of carrier transport was obtained from the investigation of magnetoresistive effect in the studied nanocomposite films. The DC measurements at room temperature for oxidized FeCoZr-Al<sub>2</sub>O<sub>3</sub> showed (see Fig. 4) that the magnetoresistance  $r(B) = (\Delta\rho(B)/\rho(0))$  of the studied nanocomposites films was negative and obeyed the relationship of  $r(B) \propto -B^k$  as the magnetic induction  $B$  increased.

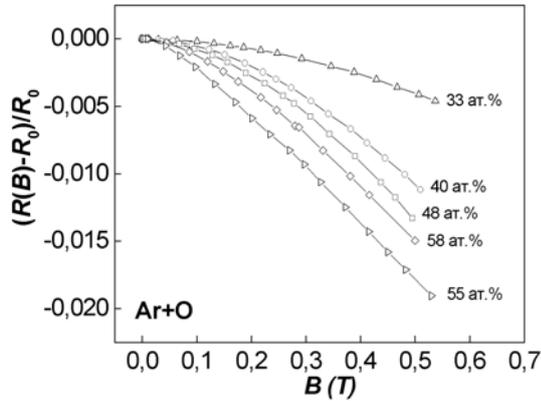


Figure 4 Reduced resistance vs magnetic induction  $B$  at RT for oxidized FeCoZr-Al<sub>2</sub>O<sub>3</sub>

### 3. STRUCTURE AND PROPERTIES OF FeCoZr-PZT SINTERED IN Ar+O AMBIENT

#### 3.1. Mössbauer spectroscopy

Mössbauer spectra of selected granular nanocomposite films  $(FeCoZr)_x(PZT)_{100-x}$  sintered at lower ( $P_O = 2,4 \cdot 10^{-3}$  Pa) and higher ( $P_O = 3,7 \cdot 10^{-3}$  Pa) oxygen pressure are shown in Figs. 5 and 6. It appeared that at lower  $P_O$  Mössbauer spectroscopy revealed the transition from nonmagnetic spectra towards superposition of magnetically split sextets and nonmagnetic subspectra around  $X_{FeCoZr} = 51$  at.%. More particularly, spectra of  $(FeCoZr)_{28}(PZT)_{72}$  and  $(FeCoZr)_{38}(PZT)_{62}$  (not shown here) were fitted with two doublets ( $D_2^{PZT}$  and  $D_3^{PZT}$ ). Parameters of the  $D_2^{PZT}$  doublet ( $\delta=0.32$  mm/s,  $\epsilon=0.96$  mm/s) are quite close to those of  $D_2^{Al_2O_3}$ . This suggests that  $D_2^{PZT}$  is related to Fe<sup>3+</sup> ions located in oxidized

FeCoZr nanoparticles. Assuming similarity of  $D_3^{Al_2O_3}$  and  $D_3^{PZT}$  ( $\delta=0.97$  mm/s,  $\epsilon=1.72$  mm/s) the latter may characterize Fe<sup>2+</sup> ions either on the interface between FeCoZr and PZT matrix or inside PZT matrix.

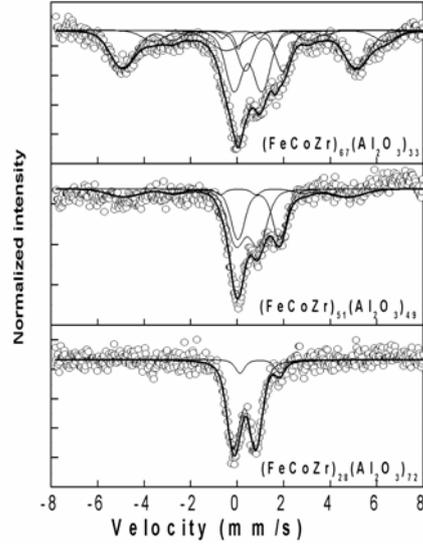


Figure 5 Mössbauer spectra of selected granular oxidized nanocomposite films  $(FeCoZr)_x(PZT)_{100-x}$  sintered at  $P_O = 2,4 \cdot 10^{-3}$  Pa

Drastic change of Mössbauer spectra was observed for FeCoZr-PZT containing  $X_{FeCoZr} = 51$  at.%. At this composition additional magnetically split sextet  $S_1^{PZT}$  ( $H_{hf}=29.9$  T) appeared. In correlation with data reported for FeCoZr-Al<sub>2</sub>O<sub>3</sub> in [8, 14]  $X_{FeCoZr} = 51$  at.% may be considered as a threshold concentration for the formation of ferromagnetically interacting net of FeCoZr nanoparticles. Insufficient increase of  $H_{hf}$  (up to 31.2 T for  $(FeCoZr)_{67}(PZT)_{23}$ ) and evident growth of  $S_1^{PZT}$  contribution at the expense of  $D_2^{PZT}$  and  $D_3^{PZT}$  were observed with further increase of  $X_{FeCoZr}$ .

Mössbauer spectra of FeCoZr-PZT sintered at higher ( $P_O = 3,7 \cdot 10^{-3}$  Pa) pressure (see Fig.6) revealed only Fe<sup>3+</sup> doublet ( $\delta=0.30-0.32$  mm/s,  $\epsilon=0.98-1.03$  mm/s) evidencing the formation of oxidized FeCoZr nanoparticles in the whole range of studied  $X_{FeCoZr}$ .

It should be mentioned that the basic peculiarity of FeCoZr-PZT samples was absence of residual nonoxidized FeCoZr nanoparticles for the cases of both  $P_O$  values ( $2,4 \cdot 10^{-3}$  Pa and  $3,8 \cdot 10^{-3}$  Pa).

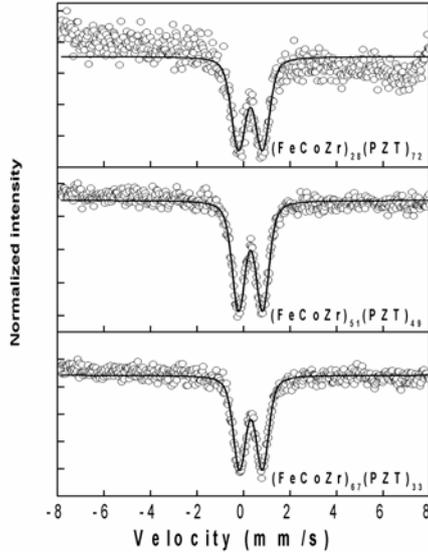


Figure 6 Mössbauer spectra of selected granular oxidized nanocomposite films  $(\text{FeCoZr})_x(\text{PZT})_{100-x}$  sintered at  $P_O = 3,8 \cdot 10^{-3}$  Pa

### 3.2. Electrical conductivity and magnetoresistivity

Electrical resistivity  $\rho$  of  $\text{FeCoZr-Al}_2\text{O}_3$  and  $\text{FeCoZr-PZT}$  films revealed principally different behavior both with  $X_{\text{FeCoZr}}$  (Fig. 7) and temperature  $T$  (see, Figs. 8 and 9). As is seen from comparison of  $\rho(x)$  dependences in Fig. 7,  $\text{FeCoZr-Al}_2\text{O}_3$  films (both oxidized and nonoxidized) showed monotoneous drop of  $\rho(x)$  (curves 1-2 in Fig. 7) while for  $\text{FeCoZr-PZT}$  nanocomposites  $\rho(x)$  possesses “plateau” at  $X_{\text{FeCoZr}} > 42-45$  at.%.

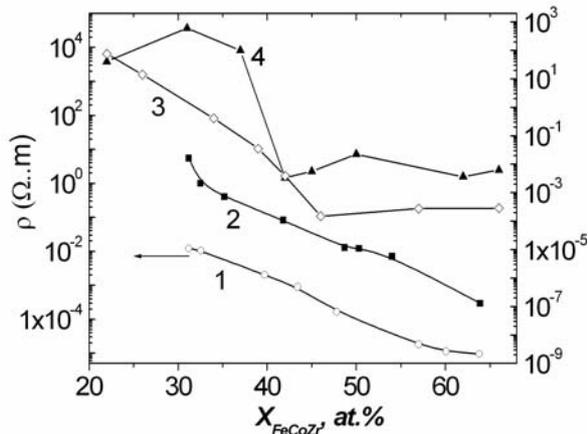


Figure 7 Dependences  $\rho(x)$  for the films  $\text{FeCoZr-Al}_2\text{O}_3$  sintered in pure Ar (curve 1) and in mixture Ar +  $\text{O}_2$  at  $P_O = 4,1 \cdot 10^{-3}$  Pa (curve 2), for  $\text{FeCoZr-PZT}$  sintered in mixture Ar +  $\text{O}_2$  at  $P_O = 2,4 \cdot 10^{-3}$  Pa (curve 3) and at  $P_O = 3,8 \cdot 10^{-3}$  Pa (curve 4)

Remember also that  $\text{FeCoZr-Al}_2\text{O}_3$  (deposited at  $P_O = 4,1 \cdot 10^{-3}$  Pa) and  $\text{FeCoZr-PZT}$  deposited at close pressure ( $P_O = 3,8 \cdot 10^{-3}$  Pa) revealed no formation of current-conducting (percolative) cluster as well as

ferromagnetically interacting net of  $\text{FeCoZr}$  or  $\text{FeCo}$ -based oxide nanoparticles beyond the percolation threshold  $X_C \approx 43-45$  at.% for  $\text{FeCoZr-Al}_2\text{O}_3$  deposited in pure Ar (at  $P_O = 0$ ). This was in strong correlation with tunneling and/or VHR carrier transport mechanism for the samples  $\text{FeCoZr-Al}_2\text{O}_3$  with  $x > 45$  at.% observed in [6]. We believe that this conservation of activational conductivity is due to  $\text{FeCoZr}$  nanoparticles covered with oxide “shells” which also prevent their ferromagnetic coupling.

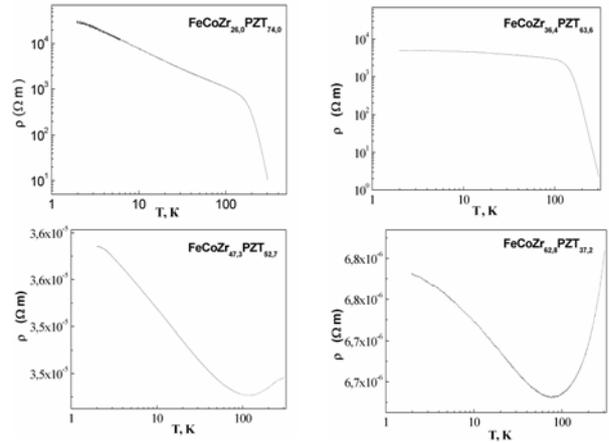


Figure 8 Temperature dependencies of electrical conductivity  $\rho(T)$  in  $\lg \rho - \lg T$  scale for  $\text{FeCoZr-PZT}$  GNC sintered at  $P_O = 2,4 \cdot 10^{-3}$  Pa

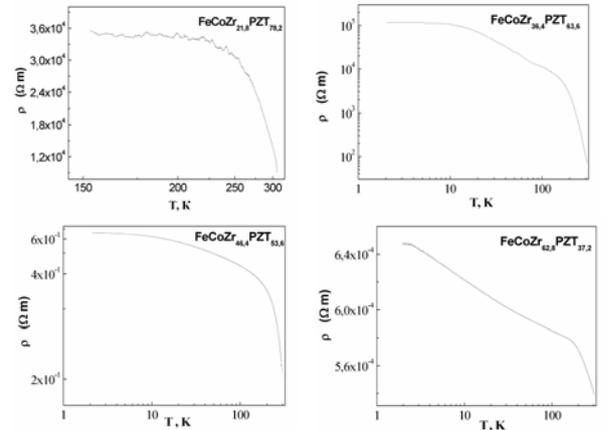


Figure 9 Temperature dependencies of electrical conductivity  $\rho(T)$  in  $\lg \rho - \lg T$  scale for  $\text{FeCoZr-PZT}$  GNC sintered at  $P_O = 3,8 \cdot 10^{-3}$  Pa

Temperature dependencies of electrical conductivity  $\rho(T)$  of  $\text{FeCoZr-PZT}$  sintered in at  $P_O = 2,4 \cdot 10^{-3}$  Pa and  $3,8 \cdot 10^{-3}$  Pa are presented in Figs. 8 and 9, correspondingly. Obtained curves revealed quite different change with  $X_{\text{FeCoZr}}$  increase for both applied  $P_O$  values and also as compared to  $\text{FeCoZr-Al}_2\text{O}_3$  [6]. It was previously suggested (see Fig. 3) that  $\rho(T)$  of  $\text{FeCoZr-Al}_2\text{O}_3$ , sintered at  $P_O = 4,1 \cdot 10^{-3}$  Pa was quite typical for VHR conductivity by Mott and Shklovski-Efros laws. Although  $\rho(T)$  observed for  $\text{FeCoZr-PZT}$  possessed negative temperature

coefficient of resistivity  $d\rho/dT$ , in majority of compositions  $\rho(T)$  cannot be fitted within assumption of VHR Mott conductivity (curves are not linear in Mott scale, see Fig. 3). As evidenced from Figs. 8 and 9  $\rho(T)$  curves for FeCoZr-PZT can be fitted with straight lines only in double logarithmic scale that is typical rather for power-like law.

For  $(\text{FeCoZr})_{47}(\text{PZT})_{53}$  and  $(\text{FeCoZr})_{62,8}(\text{PZT})_{37,2}$  samples sintered at lower  $P_O$  ( $2.4 \cdot 10^{-3}$  Pa) the transition from negative to positive  $d\rho/dT$  value at  $T > 100$  K (see Fig. 8) should be assigned to some contribution of metallic conductivity in strong correlation with Mössbauer data (see Fig. 5) evidencing the formation of ferromagnetically interacting net of FeCoZr nanoparticles.

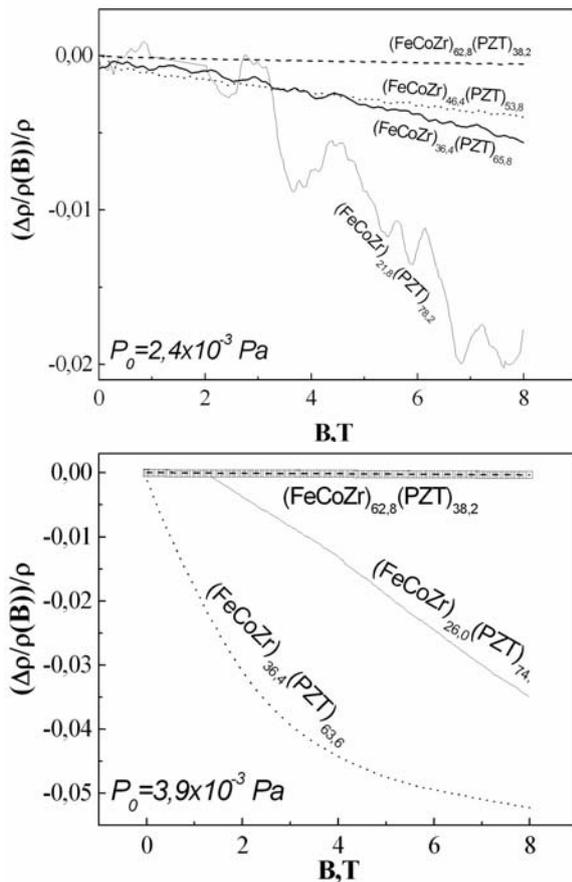


Figure 10 Reduced resistances vs magnetic induction  $B$  at RT for nanocomposite films FeCoZr-PZT sintered at  $P_O = 2.4 \cdot 10^{-3}$  Pa and  $3.8 \cdot 10^{-3}$  Pa.

Similarly to the case of FeCoZr- $\text{Al}_2\text{O}_3$  GNC negative sign of  $\Delta\rho(B)/\rho(0)$  for FeCoZr-PZT (see Fig. 10) supported the superposition of tunneling and VHR conductivity mechanisms. Contributions of both conductivity channels should vary with change of FeCoZr and PZT fractions.

#### 4. SUMMARY

Mössbauer spectroscopy revealed some peculiarities in oxidation of FeCoZr nanogranules in FeCoZr-PZT GNCs as compared to FeCoZr- $\text{Al}_2\text{O}_3$  GNC. Contrary to the case of FeCoZr- $\text{Al}_2\text{O}_3$  GNCs, FeCoZr nanogranules in FeCoZr-PZT sintered both at  $P_O = 2.4 \cdot 10^{-3}$  Pa and  $3.8 \cdot 10^{-3}$  Pa seemed to be totally oxidized in the whole range of  $X_{\text{FeCoZr}}$  values. In the case of FeCoZr- $\text{Al}_2\text{O}_3$  GNCs subspectrum, attributed to nonoxidized FeCoZr granules was detected up to  $X_{\text{FeCoZr}} \approx 55$  at.%. Such a difference evidences the effect of matrix ( $\text{Al}_2\text{O}_3$  and PZT) on the formation of FeCoZr oxides. One may suggest that oxidation of FeCoZr nanoparticles in FeCoZr- $\text{Al}_2\text{O}_3$  at  $X_{\text{FeCoZr}} < 55$  at.% may be slowed down with chemically inert and resistant to oxidation  $\text{Al}_2\text{O}_3$  matrix.

Another specific feature of FeCoZr-PZT (deposited at higher  $P_O = 3.8 \cdot 10^{-3}$  Pa is only one formed  $\text{Fe}^{3+}$  configuration characteristic to oxidized FeCoZr nanoparticles to in the whole  $X_{\text{FeCoZr}}$  range. FeCoZr- $\text{Al}_2\text{O}_3$  sintered at close  $P_O = 4.2 \cdot 10^{-3}$  Pa definitely showed quite large paramagnetic  $\text{Fe}^{2+}$  contribution.

Change of  $\sigma(T)$  observed for FeCoZr-PZT sintered at  $P_O = 2.4 \cdot 10^{-3}$  Pa (see Fig.8) evidenced the metallic conductivity channel at  $X_{\text{FeCoZr}}$  above  $\approx 47$  at. % in strong correlation with Mössbauer data (see Fig. 5) evidencing the coupling of FeCoZr nanoparticles at  $X_{\text{FeCoZr}} \approx 51$  at. %.

Deviations of  $\sigma(T)$  of FeCoZr-PZT sintered at  $P_O = 3.8 \cdot 10^{-3}$  Pa from Mott's law (Fig. 9) as well as untypical shape of  $\sigma(x)$  curve as compared to FeCoZr- $\text{Al}_2\text{O}_3$  (Fig. 8) cannot be confidently explained and assigned to transformation of FeCoZr nanoparticles state. With regard to almost unchanged Mössbauer spectra of FeCoZr-PZT (see Fig. 6) peculiar behavior of  $\sigma(x)$  and  $\sigma(T)$  should be related to the electric properties of PZT matrix rather than to the variation of FeCoZr state with  $X_{\text{FeCoZr}}$ .

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